



# Controlled Atmosphere Storage of Fruit & Vegetables

3rd Edition

**A. Keith Thompson,  
Robert K. Prange, Roger D. Bancroft  
and Tongchai Puttongsiri**



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# Preface

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The name 'controlled atmosphere storage' replaced the archaic term 'gas storage' and is synonymous with the term 'modified atmosphere storage', though the latter is little used and usually only specifically in modified atmosphere packaging. Modified atmosphere packaging refers to using packaging material (usually plastic film) that is partially permeable to gases and that uses the metabolism of the product it contains to change the atmosphere to one that is more beneficial for storage. Another way of changing the atmosphere is to coat the individual fruit or vegetable in such a way that gas exchange is restricted, resulting in a more beneficial atmosphere in the cells and intercellular spaces.

For almost a century an enormous volume of literature has been published on the subject of controlled atmosphere storage of fresh fruits and vegetables. It would be the work of a lifetime to begin to do those results justice in presenting a comprehensive and focused view, interpretation and digest for its application in commercial practice. Such a review is in demand to enable those engaged in the commerce of fruits and vegetables to be able to utilize this technology and reap its benefits in terms of the reduction of postharvest losses and maintenance of their nutritive value and organoleptic characteristics. The potential use of controlled atmosphere storage as an alternative to the application of preservation and pesticide chemicals is of continuing interest.

In order to facilitate the task of reviewing the literature we have had to rely on a combination of reviewing original publications as well as consulting reviews and technical books. The latter are not always entirely satisfactory, since they may not give their source of information and we may have inadvertently quoted the same work more than once. Much reliance has been made on conference proceedings, especially the International Controlled Atmosphere Research Conference held every few years, the European Co-operation in the Field of Scientific and Technical Research (COST 94) which held postharvest meetings throughout Europe between 1992 and 1995, the International Society for Horticultural Science's regular international conferences and the CAB Abstracts.

Different views exist on the usefulness of controlled atmosphere storage. Blythman (1996) described controlled atmosphere storage as a system that 'amounts to deception' from the consumer's point of view. The reason behind this assertion seems to be that the consumer thinks that the fruits and vegetables that they purchase are fresh and that controlled atmosphere storage technology 'bestows a counterfeit freshness'. Also she claimed that storage may change produce in a detrimental way and cited changes in texture of apples, 'potatoes that seem watery and fall apart when cooked and bananas that have no

flavour'. Some of these contentions are true and need addressing, but others are oversimplifications of the facts. Another view was expressed by David Sainsbury in 1995 and reported in the press as: 'These techniques [controlled atmosphere storage], could halve the cost of fruit to the customer. It also extends the season of availability, making good eating-quality fruit available for extended periods at reasonable costs.'

The purpose of this book is primarily to help the fresh produce industry in storage and transport of fruit and vegetables, but it also provides an easily accessible reference source for those studying agriculture, horticulture, food science and technology and food marketing. In addition it will be useful to researchers in this area, giving an overview of our present knowledge of controlled atmosphere storage that will indicate areas where there is a need for further research.

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# Glossary

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**Active packaging** usually involves the inclusion of a desiccant or O<sub>2</sub> absorber within the package or as part of the packaging material.

**Controlled atmosphere (CA) storage** has been defined (Bishop, 1996) as ‘A low O<sub>2</sub> and/or high CO<sub>2</sub> atmosphere created by natural respiration or artificial means ... controlled by a sequence of measurements and corrections throughout the storage period’.

**Dynamic controlled atmosphere (DCA) storage** has been defined as where the gas mixture will constantly change due to metabolic activity of the respiring fruit or vegetables in the store and leakage of gases through doors and walls.

**Equilibrium modified atmosphere (EMA) packaging** uses films that are highly permeable for O<sub>2</sub> and CO<sub>2</sub> to achieve an atmosphere inside the packages that remains constant during the marketing chain (Jacxsens *et al.*, 2002).

**Humidity** is referred to as percentage relative humidity (RH) (humidity relative to that which is saturated) but can also be referred to as **vapour pressure deficit**, which refers to the gaseous water in the atmosphere in relation to its maximum capacity at a given temperature.

**Modified atmosphere (MA) packaging** is where the fruit or vegetable is enclosed within sealed plastic film that is slowly permeable to the respiratory gases. The gases will change within the package, thus producing lower concentrations of O<sub>2</sub> and higher concentrations of CO<sub>2</sub> than exist in fresh air.

**Permselectivity**, used in modified atmosphere packaging, is the ratio of CO<sub>2</sub>:O<sub>2</sub> permeation coefficients.

**Scrubbing** is the selective removal of gases from the atmosphere by adsorption or absorption. In some cases this is referred to as product generated controlled atmosphere or injected controlled atmosphere.

**Shrink-films** are plastic films usually made from polyolefin (PO), but polyethylene (PE) and polyvinyl chloride (PVC) are also used. The film should have a low permeability to water vapour and O<sub>2</sub> but high permeability to CO<sub>2</sub>. It should be thin (about 17–19 µm thick) but with a high tensile strength and should be transparent. The film is placed over the product, which is then passed through a tunnel where heated air is blown over it, causing the film to shrink (i.e. the product is shrink-wrapped). The temperature and residence time in the tunnel are developed by experience, but should not be more than 180–210°C for individual wraps and 140–145°C for tray wraps. The residence time is usually 15–30 seconds.

**Static CA** is where levels of O<sub>2</sub> and CO<sub>2</sub> are set, constantly monitored and adjusted. It has been described as 'static CA storage' and 'flushed CA storage', to define the two most commonly used systems (D.S. Johnson, personal communication). 'Static' is where the product generates the atmosphere; 'flushed' is where the atmosphere is supplied from a flowing gas stream that purges the store continuously. Systems may be designed that utilize flushing initially to reduce the O<sub>2</sub> content and then either inject CO<sub>2</sub> or allow it to build up through respiration, after which the maintenance of this atmosphere is by ventilation and scrubbing. The gases are measured periodically and adjusted to the predetermined level by the introduction of fresh air or N<sub>2</sub> or by passing the store atmosphere through a chemical to remove CO<sub>2</sub>.

**Ultra-low oxygen (ULO) storage** was first defined as O<sub>2</sub> levels of 2 kPa or less. It is now more commonly referred to as O<sub>2</sub> levels of less than 1 kPa.

**Vacuum packing** uses a range of low-permeable or non-permeable films (barrier films) or containers into which the fresh fruit or vegetable is placed and the air is sucked out.

# Units of Measurement

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## Pressure

In **hypobaric and hyperbaric storage** of fruit and vegetables, different publications refer to the pressure data using different units. It would be cumbersome to give SI equivalents to all the examples given in this book, as would be normal practice. For guidance, these units might include millimetres of mercury (**mm Hg**) where atmospheric pressure at sea level is 760 mm Hg, but this is not an SI unit. Pressure can also be referred to in terms of a standard atmosphere (**atm**) and reduced pressures can be given as a fraction of atmosphere or of normal pressure. The most commonly used SI unit in hypobaric storage and in CA storage is the **pascal** (Pa), usually as kilopascals (**kPa**, i.e. 1000 Pa) and sometimes in hyperbaric storage as megapascals (**MPa**, i.e. 1000 kPa). Another SI unit is the **Newton** (**N**), a unit of force which, acting on 1 kg of mass, increases its velocity by 1 metre per second squared ( $1 \text{ m/s}^2$ ). By definition,  $1 \text{ Pa} = 1 \text{ N/m}^2$ . Very occasionally measurements are given in pounds per square inch (**psi**), where  $1 \text{ psi} = 6.895 \text{ kPa}$  or  $6895 \text{ N/m}^2$ . Note:  $1 \text{ mm Hg} = 0.133 \text{ kPa}$ ;  $1 \text{ atm} = 101.325 \text{ kPa} = 760 \text{ mm HG}$  or  $14.7 \text{ psi}$ .

In **controlled atmosphere storage**, the level of a gas in the store is measured as partial pressure in kPa or as a **percentage** (%), the latter being the proportion of the atmosphere that contains the gas. In this edition, the gas measurements are usually in kPa; 1% is approximately the equivalent of 1 kPa.

## Volume

Volumes in this edition are measured on the basis of litres (**l**), including millilitres (**ml**) and microlitres (**µl**). There are also a few examples of parts per million (**ppm**), based on mass for mass, or volume for volume.

Some American data are given in **bushels** (e.g. in Chapter 1). By volume, 1 US bushel = 35.24 litres. By weight, the bushel varies according to the commodity being measured; for example, for oats 1 US bushel = 14.52 kg. For fruit and vegetables the equivalents also vary from State to State, though for apples it is typically 40 lb (18.15 kg) per bushel. In general, weights are given in metric tonnes (**t**), i.e. 1000 kg.



# Abbreviations

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<b>1-MCP</b>	1-methylcyclopropene
<b>AA</b>	ascorbic acid
<b>ACC</b>	1-aminocyclopropane 1-carboxylic acid ACC oxidase = an enzyme involved in biosynthesis of ethylene ACC synthase = an enzyme involved in biosynthesis of ethylene ACC synthesis = the reaction catalysed by ACC synthase
<b>ACP</b>	anaerobic compensation point
<b>ACPd</b>	granular activated carbon impregnated with palladium
<b>ACR</b>	advanced control of respiration
<b>atm</b>	atmosphere
<b>ATP</b>	adenosine triphosphate
<b>AVG</b>	aminoethoxyvinylglycine
<b>BBD</b>	Braeburn browning disorder
<b>benomyl</b>	methyl-1-[butylcarbamoyl] benzimidazol-1-yl carbamate = Benlate
<b>BOP</b>	bi-axially oriented polyester
<b>BOPP</b>	bi-oriented polypropylene
<b>CA</b>	controlled atmosphere
<b>CA</b>	(plastic film) = cellulose acetate
<b>CF</b>	chlorophyll fluorescence
<b>CFC</b>	chlorofluorocarbon
<b>CFU</b>	colony-forming units
<b>CMC</b>	carboxymethyl cellulose
<b>CO<sub>2</sub>e</b>	carbon dioxide equivalent
<b>COPP</b>	coextruded oriented polypropylene
<b>CPP</b>	cast polypropylene
<b>DAT</b>	delivery air temperature
<b>DCA</b>	dynamic controlled atmosphere
<b>DCR</b>	dynamic control of respiration
<b>DCS</b>	dynamic control system
<b>DLOCA</b>	dynamic low oxygen controlled atmosphere
<b>DPA</b>	diphenylamine
<b>DPPH</b>	1,1-diphenyl-2-picrylhydrazyl
<b>EMA</b>	equilibrium modified atmosphere

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<b>EMR</b>	East Malling Research
<b>EMRS</b>	East Malling Research Station
<b>EP</b>	ethanol production
<b>EU</b>	European Union
<b>EVA</b>	ethylene-vinyl acetate
<b>EVAL</b>	ethylene-vinyl alcohol copolymer; also trade name for EVOH resin
<b>EVOH</b>	ethylene-vinyl alcohol copolymer
<b>FCA</b>	fluorescence controlled atmosphere
<b>GA</b>	gibberellins
<b>GA<sub>3</sub></b>	gibberellic acid
<b>HDPE</b>	high-density polyethylene
<b>HOA</b>	high-oxygen atmosphere
<b>ILOS</b>	initial low oxygen stress
<b>Imazalil</b>	1-[allyloxy-2,4-dichlorophenethyl] imidazole
<b>ISHS</b>	International Society for Horticultural Science
<b>kPa</b>	kilopascal
<b>l</b>	litre
<b>LDPE</b>	low-density polyethylene
<b>LLDPE</b>	linear low-density polyethylene
<b>LOL</b>	lower oxygen limit
<b>LPE</b>	lysophosphatidylethanolamine
<b>MA</b>	modified atmosphere
<b>MAP</b>	modified atmosphere packaging
<b>MB</b>	methyl bromide
<b>MDA</b>	malondialdehyde
<b>MDPE</b>	medium-density polyethylene
<b>MENA</b>	methyl ester of naphthalene acetic acid
<b>MH</b>	maleic hydrazide
<b>MHO</b>	6-methyl-5-hepten-2-one
<b>MIP</b>	modified interactive packaging
<b>MJ</b>	methyl jasmonate
<b>mRNA</b>	messenger ribonucleic acid
<b>NAA</b>	naphthaleneacetic acid
<b>NO</b>	nitric oxide
<b>NRI</b>	Natural Resources Institute
<b>OPP</b>	oriented polypropylene
<b>PA</b>	polyamide, also called nylon
<b>PAL</b>	phenylalanine ammonia-lyase
<b>PB</b>	polybutylene
<b>PBS</b>	polybutylene succinate
<b>PCR-PET</b>	post-consumer recycled PET
<b>PE</b>	polyethylene
<b>PET</b>	polyethylene terephthalate
<b>PFM</b>	pulse frequency modulated
<b>PG</b>	polygalacturonase
<b>PLA</b>	polylactic acid
<b>PME</b>	pectin methylesterase
<b>PO</b>	polyolefin
<b>POD</b>	pyrogallol peroxidase
<b>PP</b>	polypropylene
<b>PPO</b>	polyphenol oxidase
<b>PQC</b>	Produce Quality Centre

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<b>PR</b>	pathogenesis-related
<b>PRSV</b>	papaya ring spot virus
<b>PS</b>	polystyrene
<b>PSA</b>	pressure swing adsorption
<b>PU</b>	polyurethane
<b>PVB</b>	polyvinyl butyral
<b>PVC</b>	polyvinyl chloride
<b>PVDC</b>	polyvinylidene chloride
<b>QR</b>	quinone reductase
<b>RA</b>	refrigerated air
<b>RAT</b>	return air temperature
<b>RFID</b>	radio frequency identification
<b>RH</b>	relative humidity
<b>RNA</b>	ribonucleic acid
<b>RQ</b>	respiratory quotient
<b>RQB</b>	respiratory quotient breakpoint
<b>SAM</b>	S-adenosyl-methionine
<b>SAT</b>	supply air temperature
<b>SCA</b>	standard controlled atmosphere
<b>SOD</b>	superoxide dismutase
<b>TA</b>	total acidity
<b>TDCA</b>	two-dimensional dynamic controlled atmosphere storage
<b>TBZ</b>	thiabendazole = 2-thiazol-4-yl benzimidazole
<b>TPI</b>	Tropical Products Institute
<b>TSS</b>	total soluble solids
<b>ULO</b>	ultra-low oxygen
<b>USDA</b>	US Department of Agriculture
<b>UV</b>	ultraviolet
<b>ULDPE</b>	very low-density PE
<b>WTO</b>	World Trade Organization
<b>XLO</b>	extreme low oxygen





# 1

## Introduction

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The maintenance or improvement of the postharvest life and quality of fresh fruit and vegetables is becoming increasingly important. This has been partly as a response to a free market situation where the supply of good quality fruit and vegetables constantly exceed demand. Therefore to maintain or increase market share there is increasing emphasis on quality. Also consumer expectation in the supply of all types of fresh fruit and vegetables throughout the year is often taken for granted. This latter expectation is partly supplied by long-distance transport but also long-term storage of many crops. With growing awareness and concern for climate change, long-distance transport of fruit and vegetables is being questioned.

Controlled atmosphere (CA) storage has been shown to be a technology that can contribute to these consumer requirements in that in certain circumstances, with certain cultivars of crop and appropriate treatments, the marketable life can be greatly extended. An enormous amount of interest and research has been reported on CA storage and modified atmosphere (MA) packaging of fruit and vegetables to prolong their availability and retain their quality for longer. This book seeks to evaluate the history and current technology reported and used in CA storage and MA packaging and its applicability and restrictions for the use in a variety of crops in

different situations. While it is not exhaustive in reviewing the enormous quantity of science and technology that has been developed and published on the subject, it will provide an access into CA and MA for those applying the technology in commercial situations. The book can also be used as a basis for determination of researchable issues in the whole area of CA storage and MA packaging.

The scientific basis for the application of CA technology to the storage of fresh fruit and vegetables has been the subject of considerable research, which seems to be progressively increasing. Some of the science on which it is based has been known for over 200 years but was refined and applied commercially for the first time in the first half of the 20th century.

### History of the Effects of Gases on Crops

The effects of gases on harvested crops have been known for centuries. In eastern countries some types of fruit were taken to temples, where incense was burned, to improve ripening. Bishop (1996) indicated that there was evidence that Egyptians and Samaritans used sealed limestone crypts for crop storage in the second century BCE. He also quoted from the Bible suggesting that the technology might have been used in Old Testament

Egypt when Joseph prevented famine by storing grain for 7 years. Dilley (1990) mentioned the storage of fresh fruit and vegetables in tombs and crypts. This was combined with the gas-tight construction of the inner vault so that the fruit and vegetables would consume the oxygen ( $O_2$ ) and thus help to preserve the crops. An interpretation of this practice would indicate that knowledge of the respiration of fruit pre-dates the work described in the 19th century (Dalrymple, 1967). Wang (1990) quotes a Tang dynasty 8th century poem which described how litchis were shown to keep better during long-distance transport when they were sealed in the hollow centres of bamboo stems with some fresh leaves. Burying fruit and vegetables in the ground to preserve them is a centuries-old practice (Dilley, 1990). In Britain crops were stored in pits, which would have restricted ventilation and may have improved their storage life. Currently CA research and commercial CA storage is used in many countries, some of which are described in this chapter.

### France

The earliest documented scientific study of CA storage was by Jacques Etienne Berard at the University of Montpellier in 1819 (Berard, 1821), who found that harvested fruit absorbed  $O_2$  and gave out carbon dioxide ( $CO_2$ ). He also showed that certain types of fruit stored in atmospheres containing no  $O_2$  did not ripen, but if they were held for only a short period and then placed in air they continued to ripen. These experiments showed that storage in zero  $O_2$  gave a shelf-life of about 1 month for peaches, prunes and apricots and about 3 months for apples and pears. Zero  $O_2$  was achieved by placing a paste composed of water, lime and iron sulfate in a sealed jar which, as Dalrymple (1967) pointed out, would also have absorbed  $CO_2$ . Considerable CA research has been carried out over the intervening period in France. Berard's method of achieving an atmosphere that had no  $O_2$  and likely had  $CO_2$  absorbed as well is still in use today. There are commercial sources of 'oxygen

absorbers' available on e-commerce sites that will absorb virtually all  $O_2$  in a sealed container. The product consists of food-safe sachets containing iron powder which consumes  $O_2$  to form iron oxides. At least one of the products contains activated carbon, which can serve to absorb  $CO_2$  (though the supplier does not make this claim) (Robert Prange, personal communication).

### USA

In 1856 Benjamin Nyce built a commercial cold store in Cleveland, Ohio, using ice to keep it below 34°F. In the 1860s he experimented with modifying the  $CO_2$  and  $O_2$  in the store by making it airtight. This was achieved by lining the store with casings made from iron sheets, thickly painting the edges of the metal and having tightly fitted doors. It was claimed that 4000 bushels (about 72.6 t; see Glossary) of apples were kept in good condition in the store for 11 months. However, he mentioned that some fruits were injured in a way that Dalrymple (1967) interpreted as possibly being  $CO_2$  injury. The carbonic acid level was so high in the store (or the  $O_2$  level was so low) that a flame would not burn. He also used calcium chloride to control the moisture level in the mistaken belief that low humidity was necessary.

Dalrymple (1967) stated that R.W. Thatcher and N.O. Booth working in Washington State University around 1903 studied fruit storage in jars containing different gases. They found that 'the apples which had been in  $CO_2$  were firm of flesh, possessed the characteristic apple colour, although the gas in the jar had a slight odour of fermented apple juice, and were not noticeably injured in flavour'. The apples stored in hydrogen, nitrogen ( $N_2$ ),  $O_2$  and sulfur dioxide ( $SO_2$ ) did not fare so well. They subsequently studied the effects of  $CO_2$  on raspberries, blackberries and loganberries and 'found that berries which softened in three days in air would remain firm for from 7 to 10 days in  $CO_2$ '.

Fulton (1907) observed that fruit could be damaged where large amounts of  $CO_2$

were present in the store, but strawberries were 'damaged little, if any ... by the presence of a small amount of CO<sub>2</sub> in the air of the storage room'. Thatcher (1915) published a paper describing work in which he experimented with apples sealed in boxes containing different levels of gases and concluded that CO<sub>2</sub> greatly inhibited ripening.

G.R. Hill Jr reported work carried out at Cornell University in 1913 in which the firmness of peaches had been retained by storage in inert gases or CO<sub>2</sub>. He also observed that the respiration rate of the fruit was reduced and did not return to normal for a few days when removed from storage in a CO<sub>2</sub> atmosphere to air. C. Brooks and J.S. Cooley, working for the US Department of Agriculture, stored apples in sealed containers in which the air was replaced three times each week with air plus 5 kPa CO<sub>2</sub>. After 5 weeks storage they noted that the apples were green, firm and crisp, but were also slightly alcoholic and had 'a rigor or an inactive condition from which they do not entirely recover' (Brooks and Cooley, 1917). J.R. Magness and H.C. Diehl in 1924 described a relationship between apple softening and CO<sub>2</sub> concentration in that an atmosphere containing 5 kPa CO<sub>2</sub> slowed the rate of softening with a greater effect at higher concentrations, but at 20 kPa CO<sub>2</sub> the flavour was impaired. Work on CA storage that had been carried out at the University of California at Davis was reported by Overholser (1928). This work included a general review and some preliminary results on Fuerte avocados. In 1930 Overholser left the University and was replaced by F.W. Allen, who had been working on storage and transport of fresh fruit in artificial atmospheres. Allen began work on CA storage of 'Yellow Newtown' apples. 'Yellow Newtown', like 'Cox's Orange Pippin' and 'Bramley's Seedling' grown in England, was subject to low-temperature injury at temperatures higher than 0°C. These experiments (Allen and McKinnon, 1935) led to a successful commercial trial on Yellow Newtown apples in 1933 at the National Ice and Cold Storage Company in Watsonville. Thornton (1930) carried out trials where the concentration of CO<sub>2</sub> tolerated by selected fruit, vegetables and flowers was examined at six

temperatures over the range of 0–25°C. To illustrate the commercial importance of this type of experiment the project was financed by the Dry Ice Corporation of America.

From 1935 Robert M. Smock worked in the University of California at Davis on apples, pears, plums and peaches (Allen and Smock, 1938). In 1936 and 1937 F.W. Allen spent some time with Franklin Kidd and Cyril West at the Ditton laboratory in England and then continued his work at Davis, while in 1937 Smock moved to Cornell University. Smock and his PhD student, Archie Van Doren, conducted CA storage research on apples, pears and stone fruit (Smock, 1938; Smock and Van-Doren, 1938, 1939). New England farmers in the USA were growing a number of apple cultivars, particularly 'McIntosh'. 'McIntosh' is subject to chilling injury and cannot be stored at or below 0°C. It was thought that if the respiration rate could be slowed, storage life could be extended, lengthening the marketing period, and CA storage was investigated to address this problem (Smock and Van-Doren, 1938). Sharples (1989a) credited Smock with the 'birth of CA storage technology to North America'. In fact it was apparently Smock who coined the term 'controlled atmosphere storage' as he felt it better described the technology than the term 'gas storage' which was used previously by Kidd and West. Smock and Van-Doren (1941) stated: 'There are a number of objections to the use of the term *gas storage* as the procedure is called by the English. The term *controlled atmosphere* has been substituted since control of the various constituents on the atmosphere is the predominant feature of this technique. A substitute term or synonym is *modified atmosphere*.' The term 'controlled atmosphere storage' was not adopted in Britain until 1960 (Fidler *et al.*, 1973). Smock also spent time with Kidd and West at the Ditton laboratory. The CA storage work at Cornell included strawberries and cherries (Van Doren *et al.*, 1941). A detailed report of the findings of the Cornell group was presented in a comprehensive bulletin (Smock and Van Doren, 1941) which gave the results of research on atmospheres, temperatures and varietal response of fruit as well as store

construction and operation. In addition to the research efforts in Davis and Cornell in the 1930s, several other groups in the USA were carrying out research into CA storage. CA storage research for a variety of fruit and vegetables was described by Miller and Brooks (1932) and Miller and Dowd (1936). Work on apples was described by Fisher (1939) and Massachusetts Agricultural Experiment Station (1941), work on citrus fruit by Stahl and Cain (1937) and Samisch (1937) and work on cranberries by Massachusetts Agricultural Experiment Station (1941). Smock's research in New York State University was facilitated in 1953 with the completion of large new storage facilities designed specifically to accommodate studies on CA storage. During the intervening years the work at Davis has been prolific, with an enormous number of publications (many are referred to in subsequent chapters), and the campus has become a widely recognized world leader in CA science and technology.

Commercial CA storage of apples in the USA began in New York State with 'McIntosh'. The first three CA rooms with a total capacity of 24,000 bushels were put into operation in 1940 with Smock and Van Doren acting as consultants. This had been increased to 100,000 bushels by 1949, but the real expansion in the USA began in the early 1950s. In addition to a pronounced growth in commercial operation in New York State, CA stores were constructed in New England in 1951, in Michigan and New Jersey in 1956, in Washington, California and Oregon in 1958, and in Virginia in 1959 (Dalrymple, 1967). According to Love (1988) the delayed adoption in Washington State in the 1940s and early 1950s may have been due to the high cost of CA installation and management coupled with a certain scepticism among USDA postharvest researchers. But after 1957, the Washington apple industry increased CA technology at a rapid rate. Washington's initial reluctance to adopt CA technology changed with new economic developments and new USDA scientific leadership in postharvest horticulture. A CA store for 'Red Delicious' was set up in Washington State in the late 1950s in a Mylar tent where some 1000 bushels

were stored with good results. By the 1955–1956 season the total CA storage holdings had grown to about 814,000 bushels, some 684,000 bushels in New York State and the rest in New England. In the spring of that year Dalrymple did a study of the industry in New York State and found that typically CA stores were owned by large and successful fruit farmers (Dalrymple, 1967). The total CA storage holdings per farm were large, averaging 31,200 bushels and ranging from 7500 to 65,000 bushels. An average storage room held some 10,800 bushels. A little over three-quarters of the capacity represented new construction while the other quarter was remodelled from refrigerated stores. About 68% of the capacity was rented out to other farmers or speculators. In 2004 it was reported that some 75% of cold stores in USA had CA facilities (DGCL, 2004). On 1 December 2016, the amount of the apple crop remaining in storage in the USA was about 155 million bushels, representing 62% of the total crop. About 79% of these holdings were in CA storage (AAFC, 2016). Washington (WA) is the largest apple-producing state in the USA with a crop of c. 7,300,000 bins (one WA bin holds about 400–450 kg) in 2017 (Robert Blakey, personal communication). In a survey of packinghouses that represents some 50% of the crop, about 87.5% of the crop was stored in CA. Many of the respondents indicated that they planned to increase their CA capacity in the next 2 years and, based on their planned construction, CA capacity will increase to 88.9%

## UK

Sharpley (1989a) in his review in *Classical Papers in Horticultural Science* stated that '[Franklin] Kidd and [Cyril] West can be described as the founders of modern CA storage'. Sharpley described the background to their work and how it came about. Dalrymple (1967) in reviewing early work on the effects of gases on postharvest of fruit and vegetables stated: 'The real start of CA storage had to await the later work of two British scientists [Kidd and West], who started from quite a different vantage point.'

During the First World War the British Government was concerned about food shortages. It was decided that one of the methods of addressing the problem should be through research, and the Food Investigation Organisation was formed at Cambridge in 1917 under the direction of W.B. Hardy, who was later to be knighted and awarded the fellowship of the Royal Society (Sharples, 1989a). In 1918 the work being carried out at Cambridge was described as 'a study of the normal physiology, at low temperatures, of those parts of plants which are used as food. The influence of the surrounding atmosphere, of its content of O<sub>2</sub>, CO<sub>2</sub> and water vapour was the obvious point to begin at, and such work has been taken up by Dr. F. Kidd. The composition of the air in fruit stores has been suspected of being of importance and this calls for thorough elucidation. Interesting results in stopping sprouting of potatoes have been obtained, and a number of data with various fruits proving the importance of the composition of the air' (Food Investigation Board, 1919).

One problem, which was identified by the Food Investigation Organisation, was the high levels of wastage that occurred during the storage of apples. Kidd and West were working at that time at the Botany School in the University of Cambridge on the effects of CO<sub>2</sub> levels in the atmosphere on seeds (Kidd and West, 1917a, b) and Kidd was also working on the effects of CO<sub>2</sub> and O<sub>2</sub> on sprouting of potatoes (Kidd, 1919). Kidd and West transferred to the Low Temperature Laboratory for Research in Biochemistry and Biophysics (later called the Low Temperature Research Station) at Cambridge in 1918 and conducted experiments on what they termed 'gas storage' of apples (Sharples, 1986). The Low Temperature Research Station had a wide remit, including research problems basic to the preservation, storage and transport of perishable food-stuffs. It went on to focus on meat, eggs and poultry. With the assistance of the Empire Marketing Board the station was later enlarged and supplemented by the Ditton Laboratory at East Malling, which was set up in 1928 and focused on fruit and vegetables. By 1920 Kidd and West were able to set up

semi-commercial trials at a farm at Histon in Cambridgeshire to test their laboratory findings in small-scale commercial practice. In 1929 a commercial gas store for apples was built by a grower near Canterbury in Kent. From this work they published a series of papers on various aspects of storage of apples in mixtures of CO<sub>2</sub>, O<sub>2</sub> and N<sub>2</sub> (Kidd and West, 1925, 1934, 1935a, b, 1938, 1939, 1949). They also worked on pears, plums and soft fruit (Kidd and West, 1930). In 1927 Kidd toured Australia, Canada, the USA, South Africa and New Zealand discussing gas storage. In 1934 Kidd became Superintendent of the Low Temperature Research Station. By 1938 there were over 200 commercial gas stores for apples in England.

The Food Investigation Organisation was subsequently renamed, since the title did not describe fully the many agents used in food preservation. It was renamed the Food Investigation Board with the following term of reference: 'To organise and control research into the preparation and preservation of food'. The first step towards the formation of the Food Investigation Board was 'taken by the Council of the Cold Storage and Ice Association' (Food Investigation Board, 1919). The committee given the task of setting up the Food Investigation Board consisted of Mr W.B. Hardy, Professor F.G. Hopkins, Professor J.B. Farmer FRS and Professor W.M. Bayliss to prepare a memorandum surveying the field of research in connection with cold storage. The establishment of a 'Cold Storage Research Board' was approved.

Work carried out by the Food Investigation Board at Cambridge under Dr F.F. Blackman FRS consisted of experiments on gas storage of strawberries at various temperatures by Kidd and West at the Botany School (Food Investigation Board, 1920). Results were summarized as follows.

Strawberries picked ripe may be held in cold store (temperature 1°C to 2°C) in a good marketable condition for six to seven days. Unripe strawberries do not ripen normally in cold storage; neither do they ripen when transferred to normal temperatures after a period of cold storage. The employment of certain artificial atmospheres in the storage chambers has

been found to extend the storage life of strawberries. For example, strawberries when picked ripe can be kept in excellent condition for the market for three to four weeks at 1°C to 2°C if maintained:

1. in atmospheres of O<sub>2</sub>, soda lime being used to absorb the CO<sub>2</sub> given off in respiration;
2. in atmospheres containing reduced amounts of O<sub>2</sub> and moderate amounts of CO<sub>2</sub> obtained by keeping the berries in a closed vessel fitted with an adjustable diffusion leak.

Under both these conditions of storage the growth of parasitic and saprophytic fungi is markedly inhibited, but in each case the calyces of the berries lose their green after two weeks.

The Food Investigation Board (1920) said that CA storage of plums, apples and pears at low temperature 'has been continuing' with large-scale gas storage tests on apples and pears. It was reported that storage of plums in total N<sub>2</sub> almost completely inhibited ripening. Plums can tolerate, for a considerable period, an almost complete absence of O<sub>2</sub> without being killed or developing an alcoholic or unpleasant flavour.

The Food Investigation Board (1920) described work by Kidd and West at the John Street store of the Port of London Authority on 'Worcester Pearmain' and 'Bramley's Seedling' apples at 1°C and 85% relative humidity (RH), 3°C and 85% RH and 5°C and approximately 60% RH. 'Sterling Castle' apples were stored in about 14 kPa CO<sub>2</sub> and 8 kPa O<sub>2</sub> from 17 September 1919 to 12 May 1920. Ten per cent of the fruit were considered unmarketable at the end of November for the controls, whereas the gas-stored fruit had the same level of wastage 3 months later (by the end of February). The Covent Garden Laboratory was set up as part of the Empire Marketing Board in 1925. It was situated in London close to the wholesale fruit market with R.G. Tomkins as Superintendent. The Food Investigation Board (1958) described some of their work on pineapples and bananas as well as pre-packaging work on tomatoes, grapes, carrots and rhubarb.

Besides defining the appropriate gas mixture required to extend the storage life

of selected apple cultivars, Kidd and West were able to demonstrate an interaction. They showed that the effects of the gases in extending storage life varied with temperature in that at 10°C gas storage increased the storage life of fruit by 1.5–1.9 times longer than when stored in air, while at 15°C the storage life was about the same in both gas storage and in air. They also showed that apples were more susceptible to low temperature breakdown when stored in controlled atmospheres than in air (Kidd and West, 1927b).

In 1928 the Ditton Laboratory (Fig. 1.1) was established close to the East Malling Research Station in Kent by the Empire Marketing Board with J.K. Hardy as Superintendent. At that time it was an out-station of the Low Temperature Research Station at Cambridge. The research facilities were comprehensive and novel with part of the Station designed to simulate the refrigerated holds of ships in order to carry out experiments on sea freight transport of fruit. Cyril West was appointed Superintendent of the Ditton Laboratory in 1931. West retired in 1948 and R.G. Tomkins was appointed Superintendent (later the title was changed to Director) until his retirement in 1969. At that time the Ditton Laboratory was incorporated into the East Malling Research Station as the Fruit Storage Section (later the Storage Department) with J.C. Fidler as head. When the Low Temperature Research Station had to move out from its Downing Street laboratories in Cambridge in the mid 1960s, part of it was used to form the Food Research Institute in Norwich. Subsequently it was reorganized, in November 1986, as the Institute of Food Research. Most of the staff of the Ditton Laboratory were transferred elsewhere, mainly to the new Food Research Institute. The UK Government's Agricultural Research Council had decided in the mid to late 1960s that it should reorganize its research institutes on a crop basis, rather than by discipline. For example, Dr W.G. Burton, who worked on postharvest and CA storage of potatoes at the Low Temperature Research Station at Cambridge and subsequently at the Ditton Laboratory, was appointed



**Fig. 1.1.** The Ditton Laboratory at East Malling in Kent. The photograph was taken in June 1996 after it had been closed and the controlled atmosphere storage work had been transferred to the adjacent Horticulture Research International.

Deputy Director at Food Research Institute. Some of the Ditton Laboratory staff thought this government action was just a ploy to dismember the Laboratory as it had got out of control. Apparently at one time visitors from Agricultural Research Council's headquarters were neither met off the train nor offered refreshment (John Stow, personal communication).

B.G. Wilkinson, R.O. Sharples and D.S. Johnson were subsequent successors to the post of Head of the Storage Department at East Malling Research Station. The laboratory continued to function as a centre for CA storage research until 1992, when new facilities were constructed in the adjacent East Malling Research Station and the research activities were transferred to the Jim Mount Building. In 1990 East Malling Research Station (EMRS) became part of Horticulture Research International and subsequently it became 'privatized' to East Malling Research (EMR). In an interview on 24 July 2009 David Johnson indicated that the team of scientists and engineers he joined at East

Malling in 1972 'has come down to me', and he was about to retire (Abbott, 2009). Since then the Produce Quality Centre has been set up under the direction of Dr Deborah Rees. This is a collaborative initiative between the Natural Resources Institute (NRI) and East Malling Research. The NRI, which was a successor to the Tropical Products Institute (TPI), became part of the University of Greenwich in 1996. The work of TPI in London 'developed from that of the Plant and Animal Products Department of the Imperial Institute, which from the 1890s aimed to furnish scientific and technical information necessary for the better production and marketing of plant and animal products, which British territories overseas produced or might become able to produce'. The name Imperial Institute was changed to Colonial Products Laboratory in 1953 then to TPI in 1957 and changed again to Tropical Development and Research Institute (with the amalgamation of two other institutes) and then to NRI. The NRI Produce Quality Centre focused on tropical and subtropical



perishable commodities and EMR on UK fruit. As UK government funding for agricultural and horticultural research was limited, the two organizations decided to benefit from the increased efficiency of pooling expertise and sharing facilities. In 2017 EMR was taken over by the National Institute of Agricultural Botany (NIAB) to become NIAB EMR and the NRI took over the management of the Jim Mount Building (Fig. 1.2). Previously, in 2015, as interest increased in the use of dynamic controlled atmosphere (DCA) storage, David Bishop and Jim Shaeffer of Storage Control Systems Ltd worked with the Product Quality Centre (PQC) (primarily D. Rees, R. Colgan and K. Thurston) to set out the key areas of research

to determine how to use respiratory characteristics to optimize CA protocols, thereby creating DCA standards for the UK's important apple and pear cultivars (Deborah Rees, personal communication).

Although Kidd and West collaborated on 46 papers during their lifetimes, they rarely met outside the laboratory. Kidd was an avid walker, a naturalist, gardener and beekeeper. He also wrote poetry and painted. West was interested in systematic botany and was honoured for his contribution to that field (Kupferman, 1989). West retained an office in the Ditton Laboratory until the 1970s from which he continued to pursue his interests in systematic botany (John Stow, personal communication).



**Fig. 1.2.** The Jim Mount Building at East Malling containing the CA laboratory of the Natural Resources Institute of the University of Greenwich.

## Australia

Little *et al.* (2000) provided an excellent review of the history of CA in Australia. In 1926 G.B. Tindale was appointed by the state of Victoria to carry out research into post-harvest science and technology of fruit. He collaborated with Kidd during his visit there in 1927. The Council for Scientific and Industrial Research Organization (CSIRO) was formed about the same time and F.E. Huelin and S. Trout from CSIRO worked with Tindale on gas storage of apples and pears. For example, in one study in 1940 they used 5 kPa CO<sub>2</sub> + 16 kPa O<sub>2</sub> for storage of 'Jonathan' apples by controlled ventilation with air and no CO<sub>2</sub> scrubbing. Huelin and Tindale (1947) reported on gas storage research of apples and CA work was subsequently started on bananas (McGlasson and Wills, 1972) but the work was not applied commercially until the early 1990s. In Australia most commercial fruit storage until 1968 was in air and in 1972 CA generators were introduced. So the reality was that commercial CA storage was probably not used before 1968 and presumably not to any significant extent until after 1972. It seems the real problem of introducing CA storage technology was that the old cold stores were leaky (Little *et al.*, 2000; and John Faragher, personal communication) and therefore not easily adapted to CA. In 2015 a successful commercial trial was conducted at Valley Pack Storage Pty Ltd in the Goulbourn Valley on 'Granny Smith' apples and 'Packham's Triumph' pears which maintained quality and controlled superficial scald without chemicals using DCA-CF (chlorophyll fluorescence). Combined with support from Australian supermarkets that are promoting less postharvest chemical usage, there is now an increase in modern CA construction in Australia (Mark Novotny, personal communication).

## Brazil

Commercial apple production started in Brazil in the mid 1970s and the first commercial CA storage of apples occurred in the

late 1970s. Apple production has increased and Brazil now produces 1–1.2 million tonnes of apples each year (62% 'Gala', 33% 'Fuji' and < 5% other cultivars). About 278,000 t are stored in air and about 645,060 t are stored in CA. In Brazil, 750–1000 t of kiwifruit are stored in CA and approximately 2000 t in air each year (R. Prange personal communication, 2017).

## Canada

The commercial development of CA storage of apples and pears in Canada can be largely attributed to the research of Charles Eaves of the Canada Department of Agriculture (the current name is Agriculture and Agri-Food Canada (AAFC)). Eaves culminated a distinguished research career at the AAFC Kentville Research and Development Centre, Kentville, Nova Scotia, where he made numerous significant and innovative contributions (Dilley, 2006; Hoehn *et al.*, 2009). Eaves worked on fruit and vegetable storage in Nova Scotia and initiated the construction of the first commercial CA store in 1939 in Canada at Port Williams. Eaves was born in England but studied and worked in Nova Scotia and then spent a year in England with Kidd and West at the Low Temperature Research Station in Cambridge in 1932–1933. On returning to Canada he was active in the introduction of CA technology (Eaves, 1934). The use of high calcium hydrated lime (calcium hydroxide, Ca (OH)<sub>2</sub>) (also known as just 'lime' in the CA literature) for removing CO<sub>2</sub> in CA stores was also first developed by him in the 1950s (Eaves, 1959). Hydrated lime is still used today as a simple means to scrub CO<sub>2</sub> in CA stores, particularly for some cultivars of apples and pears that are very sensitive to CO<sub>2</sub>, i.e. stored at 0.5 kPa CO<sub>2</sub> or less. He was involved, with others, in the development of a propane burner for rapidly reducing the O<sub>2</sub> concentration in CA stores (Eaves, 1963). He retired in 1972. CA is important in Canada especially for apples and in December 2016 there were 172,020 t of apples in CA storage and 37,811 t in refrigerated storage (AAFC, 2016).

## China

It was reported that in 1986 the first CA store was established in Yingchengzi in China to contain some 1000 t of fruit (DGCL, 2004). Apple is the most produced fruit in China (Zhang Xuejie, personal communication). In 2016, China produced about 43 million tonnes of apples (the total output of fruits was about 0.27 billion tonnes), and over 50% of the apples were stored. About 14.7% of the apples were stored in CA (which reached over 35% in some advanced areas like Shandong province), 55.2% of apples were stored in cold storage, and 30.1% were stored in short-term non-cold storage. The CA recommendations for apple and some other fruits in China are shown below (Table 1.1).

## Holland

A considerable amount of CA research was carried out by the Sprenger Institute, which was part of the Ministry of Agriculture and Fisheries and eventually was incorporated within Wageningen University. Although the effects of CA on flowers dates back to work in the USA in the late 1920s (Thornton, 1930), much of the research that has been applied

commercially was done by the Sprenger Institute, which started in the mid 1950s (Staden, 1986). From 1955 to 1960, research was carried out at the Sprenger Institute on 'normal' CA storage of apples, which meant that only the CO<sub>2</sub> concentration was measured and regulated. Also at that time the first commercial CA stores were developed and used with good results. From 1960, both O<sub>2</sub> and CO<sub>2</sub> were measured and controlled and several types of CO<sub>2</sub> scrubbers were tested, including those using lime, molecular sieves, sodium hydroxide and potassium hydroxide. For practical situations lime scrubbers were mainly used and gave good results. From 1967 to 1975, active carbon scrubbers were used, which gave the opportunity to develop central scrubber systems for multiple rooms. From 1975 to 1980, the first pull-down equipment was used in practice, mainly using the system of ammonia cracking called 'Oxydrain', which produced N<sub>2</sub> and hydrogen (H<sub>2</sub>) gases. The N<sub>2</sub> gas was used to displace storage air to lower O<sub>2</sub> levels. At this time most apple cultivars were stored in 3kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. From 1980 to 1990, an enormous development in CA storage occurred with much attention on low O<sub>2</sub> storage (1.2 kPa) sometimes in combination with low CO<sub>2</sub>. In this system the quality of the stored apples improved significantly (Schouten *et al.*, 1997). There

**Table 1.1.** CA recommendations for selected fruits in China.

Fruit	°C	% RH	O <sub>2</sub> kPa	CO <sub>2</sub> kPa	Storage time
Apple	0–5	85–95	2–4	3–5	4–6 months
Waxberry ( <i>Myrica rubra</i> )	0.5–1.5	95	3–5	5–7	40 days
Strawberry	0	95–100	10	5–10	4 weeks
Hami melon	3~4	80	3~5	1~1.5	6 months
Date	0~1	90	3	5	2–3 months
Hawthorn	–1	90	5	3	7 months
Grape	3	90–95	5	5~20	7 months
Persimmon	0~–1	90	2	4–7	6 months
Peach	0~1	85	1–3	5	10 weeks
Plum	0~1	90	1~3	5	10–12 weeks
Cherry	0~1	90	3–5	20~25	6–7 weeks
Banana	11–13	90–95	0.5–10	0.5–7	100 days
Pineapple	7	85	2	0	40 days
Mango	10~12	85	3–5	2.5~10	40 days
Lychee	1~5	90	5	5	2 months
Kiwifruit	0~1	90–95	2~4	5	3~6 months



**Fig. 1.3.** An experimental CA storage room in Holland. Reproduced with permission from Wageningen Food & Biobased Research.

was also improvement in active CO<sub>2</sub> scrubbers, gas-tight rooms, much better pull-down systems for O<sub>2</sub> including membrane systems, pressure swing adsorption and centralized measurement and controlling systems for CO<sub>2</sub> and O<sub>2</sub>. From 1990 to 2000, there was further improvement of the different systems and the improvement of defrosting water. During 2000–2009, there was further development of the Dynamic Control System using the measurement of ethanol as the control method for the O<sub>2</sub> concentration. This enables the level to be lowered sometimes down to 0.4 kPa in commercial storage. A lot of sophisticated ultra-low oxygen (ULO) storage rooms now use this technique (Fig. 1.3). At the same time the use of 1-methylcyclopropene (1-MCP) was introduced in combination with ULO storage. A next step in the 1-MCP application, especially for ‘Elstar’ apples, was the application during and after ULO storage, to get a better shelf-life. From 2015, a new dynamic CA (DCA) technique was introduced which is based on the respiration coefficient of the apples. This advanced control of respiration (ACR) system regularly measures the CO<sub>2</sub> production and oxygen uptake of the entire CA cell (Fig. 1.3). The oxygen content in the cell is adjusted on the basis of the calculated RQ (Alex van Schaik, personal communication).

### Italy

By the late 1950s, CA was being used in the apple and pear growing regions of Northern

Italy. Since then, CA has been extensively adopted. As an example, in the Alto Adige (South Tyrol) region, in the 2016–2017 storage season 88.43% of a total 864,000 t of apples was stored in CA (Angelo Zanella, personal communication).

### India

Originally established in 1905, the Indian Agricultural Research Institute initiated a coordinated project on postharvest technology of fruit and vegetables in 1970. Under the direction of Dr R.K. Pal the IARI conducted a considerable amount of research on several fruit species, including work on CA. Jog (2004) reported that there are a large number of cold storages in India and some of the old ones have been revamped, generators added and the availability of CA and MA facilities was increasing, but CA stores remain rare.

### Mexico

In countries like Mexico crops are not widely stored, partly because of long harvesting seasons. However, CA storage is increasingly used, for example, in Chihuahua and Coahuila for apples. It was estimated by Yahia (1995) that there were about 50 CA storage rooms in Mexico storing about 33,000 t of apples each year.

## New Zealand

Research work for the Department of Industrial and Scientific Research on gas storage of apples between 1937 and 1949 was summarized in a series of papers (for example, Mandeno and Padfield, 1953). The first experimental CA shipments of 'Cox's Orange Pippin' apples were carried out to Holland and UK in the early 1980s supervised by Stella McCloud. These experiments were in reefer containers at 3.5–4°C and a small bag of lime was placed in each box to absorb CO<sub>2</sub>, though this presented a problem with the Dutch customs officers who needed to be convinced that the white powder was in fact lime. Oxygen was probably controlled by N<sub>2</sub> injection. Before that time, apples were shipped only in reefer containers or reefer ships and the use of CA was in response to the apples developing 'bitter pit', which was controlled by the increased CO<sub>2</sub> levels (John Stow, personal communication). Subsequently the Department of Scientific and Industrial Research developed predictive levels of fruit calcium needed for 'Cox's Orange Pippin' apples free of bitter pit.

## Poland

In Poland in 2017, there was about 1,450,000 t of fruit storage capacity of which 950,000 t was CA. There was 1,720,000 t of vegetable storage capacity but only about 20 t in CA in 2017 (Franciczek Adamicki, personal communication).

## South Africa

Kidd and West (1923) investigated the levels of CO<sub>2</sub> and O<sub>2</sub> in the holds of ships carrying stone fruit and citrus fruit to UK from South Africa. However, the first commercial CA storage facilities in South Africa were installed in Elgin in 1935 and 1936 (Dilley, 2006). Others were commissioned in 1978 and by 1989 the CA storage volume had increased to a total of 230,000 bulk bins catering for > 40% of the annual apple and

pear crop (Eksteen *et al.*, 1989). The adoption of CA has increased in South Africa since 1989. CA capacity was 660,000 bins in 1997 and > 1.3million bins in 2017, with 75% of CA capacity being used for apples (341,000 t assuming an apple bin contains 350 kg) and 25% used for pears (136,000 t assuming a pear bin contains 420 kg) (Richard Hurndall, personal communication.). There are no CA stores for avocados in South Africa but avocado export shipment takes place in CA containers. Export volumes are 50,000–60,000 t per annum.

## South-east Asia

Considerable work has been carried out in many South-east Asian countries on the postharvest effects of gases on fruit, but since long-term storage of fruit is rarely carried out because of constant availability most of the work has been related to MA packaging and coatings. In Thailand, there is no commercial CA storage (Ratiporn Haruenkit, personal communication), but some CA research is currently carried out at Kasetsart University in Bangkok, mostly on rambutan. In the Philippines, CA transport trials by the Central Luzon State University were reported for mango exports by Angelito T. Carpio, FreshPlaza on 12 September 2005, but results were not published.

## Turkey

In Turkey, CA research started in Yalova Central Horticultural Research Institute in 1979, mainly by Dr Umit Ertan, Dr Sozar Ozelkok and Dr Kenan Kaynaş, initially on apples and subsequently at the Scientific and Technical Research Council of Turkey (TÜBİTAK) and various Turkish universities. Commercial CA storage also started near Yalova and one private company was reported to have some 5000 t CA capacity (Kenan Kaynaş, personal communication, 2008). The most significant and accelerated expansion of CA storage in Turkey began between 2000 and 2010 and the CA storage

capacity in 2017 for fruit and vegetables (especially apple, pear and pomegranate) has reached 200,000 t, which is about 10% of the total cold-storage capacity in Turkey, and the amount of CA-stored pomegranate was about 4000–5000 t (Kenan Kaynaş and Mustafa Erkan, personal communication, 2017).

### West Indies

In 1928 the Low Temperature Research Station was established in St Augustine in Trinidad at the Imperial College of Tropical Agriculture at a cost of £5800. The initial work was confined to:

improving storage technique as applied to Gros Michel [bananas] ... for investigating the storage behaviour of other varieties and hybrids which might be used as substitutes for Gros Michel in the event of that variety being eliminated by the epidemic spread on Panama Disease.

Due to demand the work was extended to include 'tomatoes, limes, grapefruit, oranges, avocados, mangoes, pawpaws, eggplant fruit, cucurbits of several kinds and to the assortment of vegetables that can be grown in the tropics' (Wardlaw and Leonard, 1938). An extension to the building was completed in 1937 (Fig. 1.4) at a cost of £4625. 'Dr F. Kidd and Members of the Low Temperature Research Station at Cambridge gave assistance and advice in technical and scientific matters' and CA storage work was carried out using imported cylinders of mixtures of  $N_2$ ,  $O_2$  and  $CO_2$ . Professor C.W. Wardlaw was Officer-in-Charge and a considerable amount of work was carried out by him and his staff on fresh fruit and vegetables,



**Fig. 1.4.** Laboratories at the Low Temperature Research Station in St Augustine, Trinidad, in 1937. Photo Tucker Picture Production Ltd (Wardlaw and Leonard, 1938).

including work on CA storage (Wardlaw, 1938). Wardlaw returned to Manchester University as Professor of Cryptogamic Botany under Professor S.C. Harland FRS and in 1961, after Harland's retirement, he was appointed Head of Department.

With the formation of the University of the West Indies in the 1960s, postharvest work in Trinidad continued mainly under the supervision of Professor L.A. Wilson and subsequently by Keith Thompson and Lynda Wickham on postharvest science and technology of fruit and vegetables. There has been considerable work on MA packaging but limited work on CA storage. Errol Reid and Keith Thompson carried out experiments in 1997 on CA transport of bananas in reefer containers from the Windward Islands to Britain and subsequently Reid supervised the use of CA reefer ships in 1998. CA reefer ships were being used for transport of bananas from the Windward Islands to the UK with some 50,000–60,000 t/year in 2010, often combined with bananas from the Dominican Republic to reduce the risk of dead-freight.

## 2

# Harvest and Pre-harvest Factors

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The conditions and environment in which a crop is grown can affect its postharvest life and therefore can result in variability in storage recommendations and variable responses to recommendations. Factors that might affect optimum storage conditions include soil, climate and weather as well as the cultural conditions. When and how they are harvested can also have a substantial influence. For example, Johnson (1994b) found that the ester content of 'Cox's Orange Pippin' apples was influenced by growing season and source of fruit. Delaying harvesting maximized flavour potential, but harvesting too late reduced storage life and had adverse effects on texture. Johnson (2008) also reported that the significant variability between orchards in internal ethylene concentration and quality of 'Bramley's Seedling' apples demonstrated the need to maximize storage potential through attention to pre-harvest factors. In Moldova, Tsiprush *et al.* (1974) found that the place where 'Jonathan' apples were grown affected the optimum storage conditions. They recommended 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for fruit from the Kaushanskii and Kriulyanskii regions and 5 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub> for fruit from the Kotovskii region, all at 4°C and 83–86% RH. Variability in response of fruit to 1-MCP treatment has been shown, not only to the concentration used and exposure time, but also to the

location of the orchard where the apples were grown (Johnson, 2008). At one site, 'McIntosh' apples treated with 1-MCP had more internal browning but there was no effect at another site (Robinson *et al.*, 2006).

Being able to identify a characteristic of the crop at harvest that could be used to predict its postharvest behaviour would have advantages. For example, Hatoum *et al.* (2016) studied the postharvest development of internal browning in 'Braeburn' apples in order to identify the main metabolic changes associated with their pre-harvest and post-harvest factors. They stored the apples in CA conditions and observed internal browning as early as 14 days after harvest, but they were unable to identify any metabolite at harvest that could be used to reliably predict the early incidence of internal browning. Hyperspectral imaging and magnetic resonance imaging of crops have been tested for use for this purpose before storage (Suchanek *et al.*, 2017). Resonant frequency was tested (Darko, 1984) and is used commercially to predict the ripening speed of avocados (Fig. 2.1).

The way that the crop is handled can clearly have an impact on its postharvest life, particularly if it is damaged. For example, the effect of impact stress on the expression of genes related to respiratory by-products and ethylene biosynthesis of damaged cabbages



**Fig. 2.1.** Avocados, on sale in a British supermarket, that had been graded by resonant frequency to predict ripening time.

was studied by Thammawong *et al.* (2014). The cabbages were damaged by dropping them and within 1 h this damage increased the expression of the genes *BoAPX2*, *BoPAL*, *BoSAMS* and *BoACS2* in the wounded and surrounding areas. They proposed a mathematical model based on a modified Weibull distribution to describe the stress response characteristics of the cabbage, which could be used to predict their stress-responsive cellular metabolisms and quality changes and therefore have an impact on their post-harvest life.

## Organic Production

There is conflicting information on the effects of organic production of fruit and vegetables on their postharvest characteristics. Organic production has been shown to result in crops having higher levels of post-harvest diseases. Massignan *et al.* (1999) grew 'Italia' grapes both conventionally and organically and after storage at 0°C and 90–95% RH for 30 days they found that organic grapes were more prone to storage decay than those grown conventionally. Gasser and von Arx (2015) compared the storage of apples grown under an organic system with those produced under 'integrated production' and found that organic fruit had higher levels of rot during DCA storage than those from integrated production. The anaerobic compensation point (ACP) during DCA storage

was not influenced by the production method. In another case there was evidence that organic production reduced disease level. In samples from organically cultivated 'Bintje' and 'Ukama' potato tubers the gangrene disease (*Phoma foveata*) levels were lower compared with conventionally cultivated ones. However, there was no such difference in the cultivars 'King Edward' and 'Ulama' tested 4 months later. The dry rot (*Fusarium solani* var. *coeruleum*) levels were generally lower in organically cultivated potatoes compared with tubers grown conventionally (Povolny, 1995). Louarn *et al.* (2013) compared storage of organically and conventionally grown carrots and found no effect of the system on the quality of the carrots during 6 months of storage. Vermathen *et al.* (2017) compared organic, low-input and integrated production of apples on their metabolic changes during storage. Metabolic changes over time were similar for all three production systems: mainly decreasing lipid and sucrose and increasing fructose, glucose and acetaldehyde levels. However, organic apples had consistently higher levels of fructose and monomeric phenolic compounds but lower levels of condensed polyphenols than integrated and low-input produced apples.

## Fertilizers

The chemical composition of fruit and vegetables is affected by the nutritional status of the soil in which they are grown, which in turn can affect their storage life. An adequate supply of potassium (K) fertilizer in tomato production was shown to prolong their postharvest life as well as improving fruit colour and reducing the incidence of yellow shoulder (Hartz *et al.*, 2005). In the UK, Sharples (1980) found that 'Cox's Orange Pippin' apples were commonly stored at 3.5–4°C in 2 kPa O<sub>2</sub> + < 1 kPa CO<sub>2</sub>. In order to achieve this storage period he found that the composition of the fruit at harvest should be 50–70% N, a minimum of 11% phosphorus (P), 130–160% K, 5% magnesium (Mg) and 4.5 kPa Ca (on a dry matter



basis). However, an increase in N fertilizer to greenhouse-grown tomatoes, beyond a threshold level, may reduce the sugar content of the fruits and shorten their postharvest life (Passam *et al.*, 2007). N fertilizers over about 250 kg/ha reduced the TSS of tomatoes (Senevirathna and Daundasekera, 2010). In Brazil the effects of soil K application on storage of 'Fuji' apples from a long-term trial was evaluated. Fruits were stored at 0°C with 1 kPa O<sub>2</sub> + < 0.3 kPa CO<sub>2</sub>, 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> or air for 8 months and then 7 days at 20°C in air. Fruit weight losses during storage, ground colour development and rot incidence were not affected by soil K application. There was a significant interaction between K application and storage atmospheres for internal breakdown. When fruits were stored in 1 kPa O<sub>2</sub> + < 0.3 kPa CO<sub>2</sub> no differences were detected between treatments, but storing apples in 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> resulted in higher breakdown in fruits with lower K concentrations (Hunsche *et al.*, 2003). Streif *et al.* (2001) found no relationship between CA storage disorders and K, Ca, Mg and P levels in pears, but boron (B) was related to the occurrence of browning development during storage. In citrus there is some indication that higher N may reduce the development of stem-end rot (*Diplodia natalensis*) and green mould (*Penicillium digitatum*) during storage (Ritenour *et al.*, 2009). Ortiz *et al.* (2011) reported that apple softening was mediated by Ca loss from the middle lamella, and accordingly pre-harvest Ca sprays (7 weekly applications at 1.6%, w/v, 81–123 days after full bloom) were applied to 'Fuji Kiku-8' apples, which improved cell-to-cell adhesion and reduced softening during CA storage. Holb *et al.* (2012) also evaluated the effects of pre-harvest Ca applications on brown rot (*Monilinia fructigena*) on 'Idared' apples during storage at 1°C for 4 months in air, CA or ULO. Both CA and particularly ULO storage reduced rot incidence compared with air and the Ca treatment significantly reduced the number of infected fruit from 75% in non-treated to 45.5% in treated fruit. Passam *et al.* (2007) showed that pre-harvest Ca application to tomatoes had a positive effect on the

prevention of some diseases and delayed softening during storage. Grape clusters were sprayed with calcium chloride either from fruit set to veraison or from veraison to harvest and compared with non-treated controls over a 2-year period (veraison is the onset of ripening). In both years, Ca applications to bunches was effective in maintaining flesh firmness and berry breaking force as well as reducing *Botrytis cinerea* rots during storage. The applications were particularly effective if carried out between fruit set and veraison (Ciccarese *et al.*, 2013).

## Climate and Weather

Tudela *et al.* (2017) investigated, over 26 months, whether different weather conditions influenced the quality of iceberg and romaine lettuce. After being grown at different times of the year the lettuce was fresh-cut and stored for 11 days in active MA packaging and then transferred to air for 24 h at 7°C. They determined that the mean temperature during cultivation was the climatic variable that contributed most to quality loss of the lettuce. Lau (1997) and Watkins *et al.* (1997) found that 'Braeburn' apples were most susceptible to BBD and internal CO<sub>2</sub> injury during CA storage after cool growing seasons. 'Cambridge Favourite' strawberries, harvested early, were less susceptible to fungal spoilage during storage at 2°C or 15°C than fruit harvested from the same crops later during each of these seasons (Browne *et al.*, 1984). Lachapelle *et al.* (2013) reported an association of soft scald in 'Honeycrisp' apples with weather conditions, particularly cool and relatively wet weather. They developed a model to test the effects and found it could be used, prior to storage, to predict soft scald development.

## Fungicides

Diseases, particularly those caused by fungal infection, are a major limiting factor in post-harvest of fruit and vegetables. Kiwifruit

harvested from orchards infected with bacterial canker (*Pseudomonas syringae* pv. *actinidiae*) showed significantly lower total soluble solids (TSS) during CA storage and after 1-MCP treatment, compared with those not treated with 1-MCP or stored in air. The TSS of kiwifruits from healthy orchards was not influenced by the storage conditions. They also showed that in kiwifruit the presence of bacterial canker affected postharvest quality, shelf-life and susceptibility to post-harvest rots (Prencipe *et al.*, 2016). Youssef *et al.* (2012) sprayed aqueous solutions of various salts at 2% w/v at 2000 l/ha on citrus trees. After harvest, fruits were stored at  $4 \pm 1^\circ\text{C}$  and 95–98% RH, followed by 7 days of shelf-life at  $20 \pm 2^\circ\text{C}$ . They reported that the pre-harvest sprays of sodium carbonate, potassium carbonate and sodium silicate completely inhibited the incidence of decay during storage as compared with the water control. When the salts were applied after harvest, the activity was in general less pronounced, with sodium carbonate and potassium carbonate being the most effective. Kobilier *et al.* (2011) reported that *Alternaria alternata* infects persimmon fruit in the orchard and remains quiescent until harvest. During storage at  $0^\circ\text{C}$  the pathogen slowly colonizes the fruit, causing black

spot symptoms. Pre-harvest treatment with a single spray of the fungicide polyoxin B 14 days before harvest in combination with a postharvest chlorine dip reduced the black spot infected area by 60%, compared with the chlorine dip alone. Thompson (1987) reviewed pre-harvest sprays of mangoes to control anthracnose (*Colletotrichum gloeosporioides*). Recommendations were for field application of chemical fungicides often followed by postharvest hot-water treatment usually combined with a fungicide (Fig. 2.2). In Florida (McMillan, 1972) cupric hydroxide at 2.4 g/l or tribasic copper sulfate at 3.6 g/l plus the organic sticker Nu-Film 17 at 0.125% applied at monthly intervals at 57 l per tree from flowering to harvest gave good anthracnose control. Benomyl at 0.3 g/l plus Triton B1956 at 0.15 ml/l at 57 l per tree at monthly intervals from flowering to 30 days before harvesting was also shown to be very effective against anthracnose on mangoes (McMillan, 1973). Mancozeb, chlorothalonil and ferbam were shown to be equally effective as field sprays against mango anthracnose in Florida (Spalding, 1982). Trials carried out on 'Keitt' mangoes in South Africa showed that two pre-flowering applications of copper oxychloride then two applications of triadimenol



**Fig. 2.2.** Mango hot-water treatment tank being tested in Jamaica on a commercial shipment of fruit to be exported to UK.

during flowering followed by monthly applications of copper oxychloride from fruit set to just before harvest gave effective control of anthracnose (Lonsdale, 1992). Sprays were applied to run off at about 20 l per tree. In Australia mancozeb applied at 1.6 g/l (active ingredient) as a weekly spray during flowering then as a monthly spray until just before harvesting gave good control of mango anthracnose (Grattidge, 1980). In studies in the Philippines, field sprays with mancozeb or copper were effective in controlling mango anthracnose and superior to either captan or zineb (Quimio and Quimio, 1974). Postharvest water and/or sodium hydroxide washes in the orchard and a hot water spray applied over rollers, without brushes, in the packhouse were effective in reducing the incidence of *Alternaria* and stem-end rots (*Phomopsis* spp.) in mangoes (Feygenberg *et al.*, 2014). Some fungicides are not permitted in many countries and checks should be made before they are used.

Clearly microorganisms exist on plants that can affect pathogens. For example, Korsten *et al.* (1993) showed that *Bacillus* spp. isolated from leaves and fruit of avocados were more effective in controlling anthracnose (*C. gloeosporioides*) and stem-end rot (*Botryodiplodia theobromae*) of avocados when applied as a postharvest dip than the fungicide prochloraz applied in the same way. Therefore they recommended that consideration should be given to field control of diseases that do not reduce the effects of these 'beneficial' organisms.

## Water Relation

Irrigation and rainfall can affect the storage life of fruit and vegetables, but no information could be found on water relations having a differential effect specifically in CA storage. Citrus fruit harvested early in the morning, during rainy periods, or from trees with poor canopy ventilation, had a higher risk of postharvest decay (sour rot, brown rot and green mould) than fruit from trees on well drained soils with good canopy ventilation. Conversely, fruit grown in climates

that are more arid tend to develop less rot during postharvest handling, transportation and marketing. If fruit must be harvested from trees during rainy periods, it may be best to avoid fruit from lower branches that may be exposed to more pathogens.

Increased irrigation has been reported to decrease postharvest incidence of stem-end rot but increased the incidence of green mould (Ritenour *et al.*, 2009). In New York State, McKay and Van Eck (2006) found that redcurrant plants that lacked water in the first growth phase (May–June) and had excess water in the phase just before harvest may burst during storage. They also reported that any stress in the first growth phase can cause the fruit to turn yellow. High rainfall or heavy irrigation increased skin cracking in fruit including cherries, apples and tomatoes. Generally crops that have higher moisture content have poorer storage characteristics. For example, hybrid onion cultivars that tend to give high yield of bulbs with low dry matter content had a comparatively shorter storage life (Thompson *et al.*, 1972b; Thompson, 1985). Srikul and Turner (1995) reported that low irrigation reduced the green life of bananas, but if bananas were allowed to mature fully before harvest and harvesting was shortly after rainfall or irrigation the fruit could easily split during handling operations, allowing microorganism infection and postharvest rotting (Thompson and Burden, 1995). If oranges are too turgid at harvest the oil glands in the skin can be ruptured, releasing phenolic compounds and causing oleocellosis (Fig. 2.3).

Too much rain or irrigation can result in the leaves of leaf vegetables becoming brittle (Wardlaw, 1938). Irrigating crops can have other effects on their postharvest life. In carrots, heavy irrigation during the first 90 days after drilling resulted in up to 20% growth splitting, while minimal irrigation for the first 120 days followed by heavy irrigation resulted in virtually split-free carrots with a better skin colour and finish and only a small reduction in yield (McGarry, 1993). Shibairo *et al.* (1998) grew carrots with different irrigation levels and found that pre-harvest water stress lowered membrane



**Fig. 2.3.** Oleocellosis is a browning and pitting of the skin. It is due to the oil vesicles being ruptured, releasing phenolic compounds that damage the cells of the skin.

integrity of carrots, which may enhance moisture loss during storage.

The effects of water stress, applied for 30 or 45 days before flowering on 'Haden' mangoes, on their storage life was studied by Pina *et al.* (2000). They found that the 45-day fruits exhibited a higher incidence and severity of internal darkening, were firmer, more acid and had redder skins than 30-day fruits during storage at 13°C for 21 days.

In a study of the storage of onions grown in Tajikistan under various irrigation regimes, Pirov (2001) found that if the onions were to be used fairly quickly, maximum yields could be achieved by keeping the soil at 80–90% of field capacity. However, when they were to be stored for 7 months at 0–1°C and 75–80% RH, the best irrigation regime was 70% of field capacity throughout the growing season. Broccoli was cultivated under low soil water content (0.40 MPa) or field capacity (0.04 MPa). Low soil water content gave the best preservation of colour, antioxidant activity, L-ascorbic acid and 5-methyl-tetrahydrofolate contents (Cogo *et al.*, 2012).

Splitting during growth can affect post-harvest losses. The incidence of damage in carrots was shown to be affected by the total amount of irrigation and the time when it was applied. Heavy irrigation during the first 90 days after sowing resulted in up to 20% growth splitting, while minimal irrigation for the first 120 days followed by heavy

irrigation resulted in virtually no splitting, with a better skin colour and finish and only a small reduction in yield (McGarry, 1993). Carrots contain an antifreeze protein that improves storage performance and those grown in temperatures of less than 6°C accumulated higher levels of this protein and subsequently showed less electrolyte leakage from cells, slightly higher dry matter and less fungal infestation than carrots grown in warmer temperatures. However, levels also varied between the environments in which the carrots had been grown (Galindo *et al.*, 2004; Kidmose *et al.*, 2004).

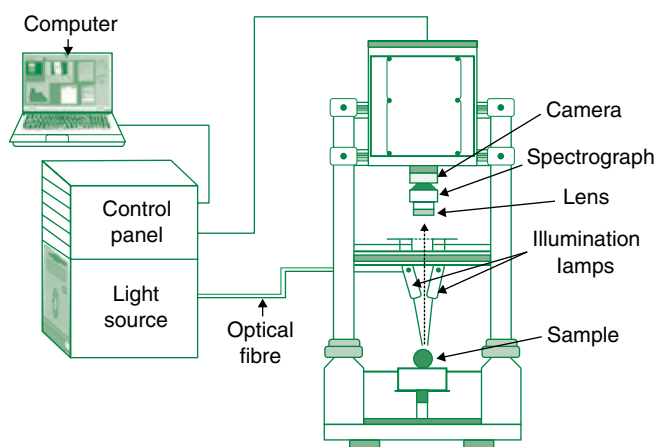
Luna *et al.* (2012) studied the effects of five drip irrigation systems (excess 50%, excess 25%, control, deficit 25% and deficit 50%) on the postharvest quality of minimally processed iceberg lettuce. They found that visual quality was lower on those from the highest irrigation regime and they developed increased off-odours. Also the mid-rib tissue had a more than 17-fold increase in phenylalanine ammonia-lyase (PAL) activity from the highly irrigated lettuce. They concluded that the quality and shelf-life of the minimally processed lettuce was better preserved by reduced irrigation.

## Harvest Maturity

The correct harvest maturity in terms of quality and postharvest changes of fruit and vegetables is crucial, but particularly so for long-term storage including CA. Many ways have been developed and used commercially to determine the stage of maturity or ripeness before or at harvest, including pressure testers, TSS, total acidity (TA) and starch content. These are all methods that rely on taking a representative sample, testing it and assuming the rest of the crop is similar within defined limits. Non-destructive methods of measuring fruit and vegetable maturity have been successfully tested, including surface colour and near-infrared hyperspectral imaging to measure TSS and juice content (Teerachaichayut and Ho, 2017) (Fig. 2.4).



(b)



**Fig. 2.4.** Infra-red hyperspectral imaging being developed for non-destructive testing of fruit maturity at KMITL in Thailand. (a) Equipment being used by a student. (b) Schematic of the equipment. Reproduced with permission from Miss Huang HoThanh.

There can be interactions between optimum storage conditions and harvest maturity. For example, Trierweiler *et al.* (2004) indicated that 'Braeburn' apples harvested at advanced maturity had increased Braeburn browning disorder (BBD) during storage in 3.0 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> compared with fruit harvested at a less mature stage. This confirmed the findings of Lau (1997) and Streif *et al.* (2001), who also found that late harvesting of pears increased the appearance and intensity of browning disorders. In 'Gala Must' apples, stored in air or CA at 0°C, flesh browning symptoms increased with delayed harvesting (Ben, 2001). Awad *et al.* (1993) found that for 'Jonagold' apples stored at 3 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> the level of scald was 39% where the harvest was delayed compared with apples from a normal harvest date, which was 13%. However, in 'Fuji' apples Jobling *et al.* (1993) found that harvest maturity was not

critical and all the maturity stages tested had a good level of consumer acceptability for up to 9 months in CA storage followed by 25 days at 20°C in air. Johnson (2001) recommended that in 'Gala' apples later harvests should be consigned to shorter-term storage and quoted previous work which suggested that 'Gala' stored well when harvested at above 7 kg firmness (measured with a Magness and Taylor penetrometer) and over a broad range of starch levels of 50–90% black. He also pointed out the difficulties in being sufficiently precise using firmness and starch levels in assessing the optimum maturity for CA storage of 'Gala'. Increased ethylene production was found to be the best physiological marker of ripening. Internal ethylene concentration of more than 100 ppb normally indicates that a fruit has commenced its ripening and fruits for storage are usually picked just prior to this. The internal ethylene concentration data for

the six sites in Johnson's study indicated that a high proportion of fruit in five of the six orchards had already begun ripening when harvested. Whilst it is economically important that fruit achieves sufficient size and red coloration and can be harvested with stalks intact, these criteria should not be used as the primary indicators of maturity for storage. Johnson recommended that 'picking over' to ensure that fruit maturity matches the storage expectation appears to be the solution in orchards that are slow to develop the desired visual characteristics.

Harvest maturity has also been shown to affect postharvest life in other crops. Heltoft *et al.* (2016) stored 'Saturna' and 'Asterix' potatoes and found that immature potatoes had a lower dry matter content and higher weight losses and respiration rates during storage than more mature potatoes. In tomatoes, Batu (1995) showed that MA packaging was more effective in delaying ripening of fruit harvested at the mature green stage than for those harvested at a more advanced stage of maturity.

### Chemical Sprays

Crops are sprayed during production for pest, disease and weed control. These sprays may affect their postharvest behaviour. Also sprays are applied, or have been investigated, specifically to improve maintenance of quality and reduce postharvest losses. Some of the sprays have also been tested as postharvest dips (see Chapter 7). Pre-storage sprays include abscisic acid, methyl jasmonate (MJ), daminozide, gibberellins (GA), aminoethoxyvinylglycine (AVG), lysophosphatidylethanolamine (LPE), Ethrel, maleic hydrazide (MH) and naphthaleneacetic acid (NAA).

### Abscisic acid

Abscisic acid is common in plants and is involved in leaf retention and abscission and therefore is particularly important in the storage of leafy vegetables. Also Foukaraki

*et al.* (2016) reported that abscisic acid and sucrose may play a role in sprout suppression when ethylene is exogenously applied to stored potatoes. Singh *et al.* (2014) sprayed litchis with abscisic acid at 150 or 300 mg/l at the colour-break stage. They found that this increased anthocyanins levels in the pericarp without adversely affecting the quality or stability of the fruit during 14 days of subsequent storage at 5°C.

### Methyl jasmonate

MJ has been tested as a field spray as well as a postharvest treatment (see Chapter 7). Zapata *et al.* (2014) sprayed plum trees with MJ at 0, 0.5, 1.0 or 2.0 millimolar (mM) at 63, 77 and 98 days after full blossom. The fruit was harvested at the 'commercial ripening stage' and stored for 9 days at 20°C or for 50 days at 2°C then 1 day at 20°C. Treatment with 2.0 mM significantly accelerated ripening, whereas 0.5 mM delayed ripening. Ethylene production, respiration rate and softening were reduced significantly in both storage conditions for fruit treated with 0.5 mM. In these fruit, total phenolics, total antioxidant activity, pyrogallol peroxidase (POD), catalase and ascorbate peroxidase were higher in treated than non-treated plums during storage, which could account for the delay in ripening. Similarly Martínez-Esplá *et al.* (2014) found that the total phenolics, total antioxidant activity, firmness, colour, size and retention of weight of two plum cultivars were increased after MJ application to trees but there were interactions between cultivar and concentration. Application of 0.5 mM was the most effective for the cultivar 'Black Splendor' and 2.0 mM for 'Royal Rosa'.

### Daminozide

*N*-dimethylamino succinamic acid, also called Alar, B9 and B995, was first marketed in the early 1960s mainly as a dwarfing agent (Brooks, 1964). It was also claimed to delay ripening of apples on the tree so that

all the fruit on the same tree could be harvested at the same time. When applied to 'Cox's Orange Pippin' apples at 2500 µl/l in late June and mid-August, they developed a redder colour and were firmer than fruit from non-sprayed trees (Sharples, 1967). Sprayed apples were less susceptible to *Gloeosporium* rots but had more core flush during storage. There was some indication that sprayed fruits were slower to mature, since daminozide tended to retard the climacteric rise in respiration rate. In a comparison between pre-harvest and postharvest application of daminozide to 'Cox's Orange Pippin' apples, immersion of fruit in a solution containing 4.25 g/l for 5 min delayed the rise in ethylene production at 15°C by about 2 days, whereas orchard application of 0.85 g/l caused delays of about 3 days (Knee and Looney, 1990). Both modes of application depressed the maximum rate of ethylene production in ripe apples by about 30%. They also found that daminozide-treated fruit were less sensitive to exogenous ethylene during storage than non-treated fruit, but this response varied between cultivars. Daminozide has been used on other fruit; for example, Nagar (1994) found that treated litchis had increased shelf-life by up to 4 days. Daminozide has been withdrawn from the market in several countries.

### Gibberellins

Pre-harvest sprays of gibberellic acid ( $GA_3$ ) have been shown to have effects on the postharvest science and technology of fruit and vegetables. These include reduced yellowing and extended storage life of parsley (Lers *et al.*, 1998), reduced chilling injury in persimmon (Besada *et al.*, 2008), decreased fruit cracking in cherry (Yildirim and Koyuncu, 2010) and delayed ripening and reduced decay in prickly pear (Schirra *et al.*, 1999). Postharvest treatment with gibberellins has been shown to reduce degreening in lemons (Mizobutsi *et al.*, 2000) and limes (Sposito *et al.*, 2000) and improved retention of vitamin C also in limes (Keleg *et al.*, 2003). A negative effect of  $GA_3$  was reported by Zoffoli *et al.* (2009) who

found that it could induce shatter and pre-disposed grapes to grey mould caused by *B. cinerea*. Ding *et al.* (2015) found that storage of tomatoes in low temperatures inhibited increase of endogenous levels of  $GA_3$ . This effect was associated with lower expression of key GA metabolic genes (*GA3ox1*, *GA20ox1* and *GA2ox1*).  $GA_3$  treatment reduced chilling injury and decreased electrolyte leakage and malondialdehyde (MDA) content, increased proline content, and improved antioxidant enzyme activities.

### Aminoethoxyvinylglycine

AVG ([S]-trans-2-amino-4-(2-aminoethoxy)-3-butenic acid hydrochloride) is an ethylene biosynthesis inhibitor, which is marketed commercially as ReTain® and is used in apple orchard sprays. Its mode of action is to inhibit the activity of 1-aminocyclopropane 1-carboxylic acid (ACC) synthase. Effects of AVG at both 100 and 200 mg/l applied to plums 2 weeks before harvesting decreased their firmness during subsequent storage compared with those not treated. Also with AVG treatment at 100 mg/l there was a decrease in total phenolics and total antioxidant activity (Ozturk *et al.*, 2012). They concluded that AVG generally had negative impacts on individual phenolic compounds. Mission muskmelon harvested from plots sprayed with AVG had lower rates of ethylene production, at harvest and after cold storage, than melons harvested from control plots. Melon ethylene production after storage was consistently lower when AVG was applied 7 days after full net formation, especially at the highest rate they tested (260 mg/l). AVG delayed initial development of an abscission zone and increased leaf chlorosis, but had no effects on flesh firmness, TSS or incidence of decay at harvest and after storage (Shellie, 1999). AVG at 1 g/l reduced ethylene production, respiration rate and softening of apricots (Muñoz-Robredo *et al.*, 2012). Lulai and Suttle (2004) found that treatment of potato tubers with AVG inhibited wound-induced ethylene production by about 90%, but did not affect wound-induced suberization. Cheema *et al.* (2013)

found that stored sweet potatoes could be exposed to ethylene to suppress sprouting, but it also resulted in increased respiration rate. However, AVG inhibited this increase in respiration rate and counteracted the decrease in monosaccharide concentrations. The effect of AVG was attributed to its possible inhibitory effect on protein synthesis.

AVG treatment delayed the onset of the climacteric in 'McIntosh' apples during storage but did not reduce internal ethylene concentration (Robinson *et al.*, 2006). Johnson and Colgan (2003) showed that 'Queen Cox' apples, which had been sprayed with AVG at 123.5g/ha before harvest, were firmer than those that had not been sprayed after 6 months of storage at 3.5°C under 1.2 kPa O<sub>2</sub> + 98.8 kPa N<sub>2</sub> combined with ethylene removal. The additive effect on firmness of AVG treatment and ethylene removal was negated by the development of core flush and after a simulated marketing period 57% of the fruit was affected after this combination of treatments. Drake *et al.* (2006) found that AVG reduced starch loss and ethylene production, retained firmness and reduced cracking in 'Gala' apples, but reduced the sensory acceptance of apples and apple juice. AVG followed by ethylene treatment (Ethephon) reduced starch loss, ethylene production and cracking and maintained firmness. This combination also improved the sensory acceptance of apples but reduced sensory preference of apple juice. Harb *et al.* (2008b) showed that the biosynthesis of volatiles was highly reduced in apples after treatments with AVG, especially after extended storage periods in ULO. They found that after storage at 3.5°C in 1.2 kPa O<sub>2</sub> + < 1 kPa CO<sub>2</sub>, with no removal of ethylene, until late March or early April, the AVG-treated fruit had a marked increase in the production of the corresponding volatiles. However, this effect was transitory in both AVG-treated fruits and those stored in ULO.

Application of 1-MCP has been investigated on fruits that had been treated with AVG. After 8 months in CA storage at 2°C, 'McIntosh' apples that had been treated with AVG had more internal browning disorders than non-treated controls but fruits with the combination of AVG + 1-MCP had less

internal browning and were similar to the non-treated controls, but with better retention of firmness (Robinson *et al.*, 2006). Moran (2006) also reported that AVG + 1-MCP maintained firmness in 'McIntosh' apples more than 1-MCP alone after 120 or 200 days of CA storage. It was reported that AVG + 1-MCP could be used to maintain fruit firmness even when internal ethylene concentration at harvest was as high as 240 µl/l, but CA storage life was limited to 4 months. AVG was not effective in increasing the efficacy of 1-MCP on Cortland apples when internal ethylene concentration at harvest was not significantly different between AVG treated and non-treated fruit and internal ethylene concentration was less than 2 µl/l. AVG increased the efficacy of 1-MCP in the apples when internal ethylene concentration was 36 µl/l in non-treated fruits compared with undetectable in AVG treated fruits. Compared with non-treated control, AVG at 60 mg/l applied to 'Bartlett' pears 1 week before harvest suppressed ethylene production, respiration rate and softening and reduced senescence (Wang *et al.*, 2016). However, they found that the effectiveness of AVG was influenced by fruit maturity at the time of application.

### Lysophosphatidylethanolamine

LPE is a naturally occurring lipid that has regulatory effects on senescence and ripening of fruit. In their review, Amaro and Almeida (2013) reported that the benefits of LPE include delayed leaf senescence, stimulation of maturation in grapes, acceleration of colour development and extension of shelf-life in cranberry and tomato. Özgen *et al.* (2015) also indicated that there is evidence that LPE can 'accelerate ripening of fruits and prolong shelf-life at the same time'. Hong *et al.* (2007) sprayed LPE (10 mg/l) on grape vines at either 4 or 6 weeks after fruit set and found that it increased berry size and TSS content. They also showed increases in anthocyanin levels, improved fruit quality and enhanced phytochemical characteristics of sweet cherries after pre-harvest application of LPE at 10 mg/l. Wan Zaliha and



Singh (2013) showed that spraying 'Cripps Pink' apples with LPE (125 mg/l or 250 mg/l) 2 and 4 weeks before 'commercial harvest' enhanced accumulation of total anthocyanins and polyphenolic compounds. LPE has also been shown to be effective as a post-harvest treatment. For example, Ahmed and Palta (2016) dipped bananas at ripening stage 2 in a combination of LPE (200 mg/l) plus lecithin for 30 min and found delayed ripening during 10 days of storage at room temperature. LPE plus lecithin was better than LPE alone.

### Ethylene

Ethylene is a common postharvest treatment to initiate ripening in climacteric fruit, de-greening of citrus and inhibition of sprouting in potatoes. Pre-harvest use of ethylene uses 2-chloroethyl-phosphonic acid, which is marketed as Ethrel or Ethephon. Ethrel C will release 74.4 l of ethylene gas from 1 l of Ethrel C. Ethrel has been used as a source of ethylene for decades. In order to reduce the difficulty of dislodging the fruit from the tree, Ethrel, abscissic acid and cycloheximide have all been shown to be effective but are not permitted for use in all countries. Ethrel is used to initiate flowering in pineapples. It has also been applied just before harvesting to accelerate de-greening and therefore the development of the orange colour in the skin. In Queensland, 'Smooth Cayenne' pineapples treated prior to harvest with Ethrel, at a concentration of 2.5 l in 1000 l of water, had superior eating quality, de-greened more evenly, but had a shorter shelf-life due to accelerated skin senescence than non-treated fruit 10 days after harvest. Treated fruit left on the plant for 23 days had inferior eating quality compared with the non-treated fruit (Smith, 1991). These effects are probably because of the ethylene speeding up the maturation of the fruit. Ferrer *et al.* (1996) showed that after the application of Ethrel (in the range of 0.1–1.0 mM) to lettuces, brown spots appeared in the leaves, but both soluble and cell wall-bound peroxidase activity in the leaves was reduced.

### Maleic hydrazide

Chemicals may also be applied to certain crops in the field to prevent them sprouting during storage and thus to extend their storage period. An example of this is the application of MH (a plant growth regulator) to onions. MH is permitted in most countries, including the USA, Australia and most European countries. It has an acute mammalian oral LD<sub>50</sub> of > 2000 mg/kg. Because it is necessary for the chemical to be translocated to the apex of the growing point towards the centre of the bulb for it to be effective, it must be applied to the leaves of the growing crop some 2 weeks before harvesting so that the chemical has time to be translocated into the middle of the bulb into the meristematic tissue. MH should be applied according to the manufacturer's instructions, but 2500 µl/l sprayed on the crop shortly before the leaves die down was recommended by Tindall (1983). Eshel *et al.* (2014) reported that onions grown in southern Israel were treated with MH before storage, and then stored for up to 8 months at 0°C with minimal losses. However, they recommended curing at 30°C and 98% RH for up to 9 days directly after harvest to reduced skin cracks.

### Naphthalene acetic acids

In Trinidad pre-harvest spraying of sweet potatoes with MH or treating the tubers postharvest with methyl ester of naphthalene acetic acid (MENA) has been reported to inhibit sprouting. MENA was applied by interleaving it with layers of paper soaked in MENA in acetone at a ratio of 40 ml/100 kg of tubers (Kay, 1987). Paton and Scriven (1989) applied NAA either under reduced pressure as a 1 g/l solution containing a wetting agent or as a dust of 1, 10 or 100 mg/g of talc. During storage of at least 40 days, NAA applied as a solution reduced sprouting by more than 50% and when it was applied in talc the reduction in sprouting was 29%. However, in both cases there was an increase in water loss.

# 3

## Effects and Interactions

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CA storage is now used worldwide on a variety of fresh fruit and vegetables. The stimulation for the development of CA storage was arguably the requirement for extended availability of fruit and vegetables, especially certain cultivars of apple that were subject to chilling injury and therefore had reduced maximum storage periods. CA storage has been the subject of an enormous number of biochemical, physiological and technological studies and has been demonstrated to reduce the metabolism of fruit and vegetables in certain circumstances. However, for some crops in certain conditions high CO<sub>2</sub> or low O<sub>2</sub> or a combination of both can have either no effect or can even increase metabolism and therefore shorten storage life. The reasons for this variability are many. Interactions with temperature would mean that the metabolism of the crop could be changed so that it would be anaerobic and thus higher, especially where the O<sub>2</sub> content was low. High levels of CO<sub>2</sub> can actually injure the crop, which again could affect its rate of respiration. These effects on metabolism could also affect the eating quality of fruit and vegetables. Generally crops stored in CA have a longer storage life because the rates of the metabolic processes are slower. Particularly with climacteric fruit, this would slow ripening and deterioration so that when fruits have been stored for protracted periods

they may well be less ripe than fruits stored in air. The actual effects that varying the levels of O<sub>2</sub> and CO<sub>2</sub> in the atmosphere have on crops varies with factors such as:

- species;
- cultivar;
- concentration of gases in store;
- temperature;
- stage of maturity of crop at harvest;
- degree of ripeness of climacteric fruit;
- growing conditions;
- ethylene in store; and
- pre-storage treatments.

There are also interactive effects of the two gases, so that the effect of CO<sub>2</sub> and O<sub>2</sub> in extending the storage life of a crop may be increased when they are combined. The effects of O<sub>2</sub> on postharvest responses of fruit, vegetables and flowers were reviewed and summarized by Thompson (1996) as follows:

- reduced respiration rate;
- reduced substrate oxidation;
- delayed ripening of climacteric fruit;
- prolonged storage life;
- delayed breakdown of chlorophyll;
- reduced rate of production of ethylene;
- changed fatty acid synthesis;
- reduced degradation rate of soluble pectin;
- formation of undesirable flavour and odours;

- altered texture; and
- development of physiological disorders.

Thompson (2003) also reviewed some of the effects of increased CO<sub>2</sub> levels on stored fruits and vegetables as follows:

- decreased synthetic reactions in climacteric fruit;
- delaying the initiation of ripening;
- inhibition of some enzymatic reactions;
- decreased production of some organic volatiles;
- modified metabolism of some organic acids;
- reducing the rate of breakdown of pectic substances;
- inhibition of chlorophyll breakdown;
- production of off-flavour;
- induction of physiological disorders;
- retarded fungal growth on the crop;
- inhibition of the effect of ethylene;
- changes in sugar content (potatoes);
- effects on sprouting (potatoes);
- inhibition of postharvest development;
- retention of tenderness; and
- decreased discoloration levels.

The recommendations for the optimum storage conditions have varied over time due mainly to improvements in the control technology over the levels of gases within the stores. Bishop (1994) showed the evolution of storage recommendations by illustration of the recommendation for the storage of the apple cultivar ‘Cox’s Orange Pippin’ since 1920 (Table 3.1).

CA storage is still mainly applied to apples but studies of other fruit and vegetables have shown it has wide application and an

**Table 3.1.** Recommended storage conditions for Cox’s Orange Pippin apples all at 3.5°C (modified from Bishop, 1994).

O <sub>2</sub> kPa	CO <sub>2</sub> kPa	Storage time in weeks	Approximate date of implementation
21	0	13	–
16	5	16	1920
3	5	21	1935
2	< 1	27	1965
1¼	< 1	31	1980
1	< 1	33	1986

increasing number of crops are being stored and transported under CA conditions. The technical benefits of CA storage have been amply demonstrated for a wide range of flowers, fruit and vegetables but the economic implications of using this comparatively expensive technology have often limited its commercial application. However, with technological developments, more precise control equipment and the reducing cost, CA storage is being used commercially for an increasing range of crops.

The question of changes in quality of fruit after long-term storage is important. Johnson (1994a) found that the storage practices that retard changes in senescence in apples generally reduced the production of volatile aroma compounds. Reduced turnover of cell lipids under CA conditions is thought to result in lack of precursors (long-chain fatty acids) for ester synthesis. The lower O<sub>2</sub> levels required to increase storage duration and to maximize retention of the desired textural characteristics can further reduce aroma development. Using hydrated lime or activated carbon to reduce the level of CO<sub>2</sub> did not affect aromatic flavour development, though there may be concern over continuous flushing of N<sub>2</sub>/air mixtures. The prospect of improving flavour in CA stored apples by raising O<sub>2</sub> levels prior to the opening of the store is limited by the need to retain textural quality.

### Carbon Dioxide and Oxygen Damage

CO<sub>2</sub> and O<sub>2</sub> injury symptoms on some apple cultivars (Table 3.2) and selected fruit and vegetables (Table 3.3) have been reviewed. It should be noted that the exposure time to different gases will affect susceptibility to injury and there may be interactions with ripeness, harvest maturity or the storage temperature.

Fidler *et al.* (1973) gave a detailed description of injury caused to different cultivars of apples stored in atmospheres containing low O<sub>2</sub> or high CO<sub>2</sub> levels. Internal injury (Fig. 3.1) was described as often beginning

**Table 3.2.** Threshold level of O<sub>2</sub> or CO<sub>2</sub> required causing injury to apples and typical injury symptoms (Kader, 1989; Saltveit, 1989; Meheriuk, 1989b; Kader, 1993).

Cultivar	CO <sub>2</sub> injury level	CO <sub>2</sub> injury symptoms	O <sub>2</sub> injury level	O <sub>2</sub> injury symptoms
Boskoop	> 2 kPa	CO <sub>2</sub> cavitation	< 1.5 kPa	
Cox's Orange Pippin	> 1 kPa	Core browning	< 1 kPa	Alcoholic taste
Golden Delicious	> 6 kPa continuous > 15 kPa for 10 days or more	CO <sub>2</sub> injury	< 1 kPa	Alcoholic taint
Red Delicious	> 3 kPa	Internal browning	< 1 kPa	Alcoholic taste with late-picked fruit
Starking Delicious	> 3 kPa	Internal disorders	< 1 kPa	Alcoholic taste
Elstar	> 2 kPa	CO <sub>2</sub> injury	< 2 kPa	Coreflush
Empire	> 5 kPa	CO <sub>2</sub> injury	< 1.5 kPa	Flesh browning
Fuji	> 5 kPa	CO <sub>2</sub> injury	< 2 kPa	Alcoholic taint
Gala	> 1.5 kPa	CO <sub>2</sub> injury	< 1.5 kPa	Ribbon scald
Gloster	> 1 kPa	Core browning	< 1 kPa	
Granny Smith	> 1 kPa with O <sub>2</sub> at < 1.5 kPa > 3 kPa with O <sub>2</sub> < 2 kPa	Severe coreflush	< 1 kPa	Alcoholic taint, ribbon scald and core browning
Idared	> 3 kPa	CO <sub>2</sub> injury	< 1 kPa	Alcoholic taint
Jonagold	> 5 kPa	Unknown	< 1.5 kPa	Unknown
Jonathan	> 5 kPa	CO <sub>2</sub> injury, flesh browning	< 1 kPa	Alcoholic taint, core browning
Karmijn	> 3 kPa	Core browning and low temperature breakdown	< 1 kPa	Alcoholic taint
McIntosh	> 5 kPa continuous > 15 kPa for short periods	CO <sub>2</sub> injury, core flush	< 1.5 kPa	Corky browning, skin discoloration, flesh browning, alcoholic taint
Melrose	> 5 kPa	Unknown	< 2 kPa	Alcoholic taint
Mutsu	> 5 kPa with O <sub>2</sub> > 2.5 kPa	Unknown	< 1.5 kPa	Alcoholic taint
Rome	> 5 kPa	Unknown	< 1.5 kPa	Alcoholic taint
Spartan	> 3 kPa	Coreflush, CO <sub>2</sub> injury	< 1.5 kPa	Alcoholic taint
Stayman		Unknown	< 2 kPa	Alcoholic taint

in the vascular tissue and then increases to involve large areas of the cortex. At first the injury zones are firm, and have a 'rubbery' texture when a finger is drawn over the surface of the cut section of the fruit. Later, the damaged tissue loses water and typical cork-like cavities appear.

They also showed that the appearance of CO<sub>2</sub> injury symptoms is a function of concentration, exposure time and temperature. They described external CO<sub>2</sub> injury where 'initially the damaged area is markedly sunken, deep green in colour and with sharply defined edges. Later in storage the damaged tissue turns brown and finally almost black'.

Injury caused as a result of low O<sub>2</sub> levels is due to fermentation, also called anaerobic respiration, resulting in the accumulation of the toxic by-products alcohols and aldehydes. These can result in necrotic tissue, which tends to begin at the centre of the fruit (Fig. 3.2). It may be that the injury that is referred to as spongy tissue in 'Alphonso' mangoes and some other mango varieties may be related to CO<sub>2</sub> injury, since they display morphological symptom similar to those described for apples (Fig. 3.3).

The lower O<sub>2</sub> limit for apples was found to be cultivar dependent, ranging from a low of about 0.8 kPa for 'Northern Spy' and 'Law Rome' to a high of about 1.0 kPa for

**Table 3.3.** Threshold level of O<sub>2</sub> or CO<sub>2</sub> required causing injury to some fruits and vegetables and typical injury symptoms (Kader, 1989; Saltveit, 1989; Meheriuk, 1989a and b; Kader, 1993).

Crop	CO <sub>2</sub> injury level	CO <sub>2</sub> injury symptoms	O <sub>2</sub> injury level	O <sub>2</sub> injury symptoms
Apricot	> 5 kPa	Loss of flavour, flesh browning	< 1 kPa	Off-flavour development
Artichoke, Globe	> 3 kPa	Stimulates papus development	< 2 kPa	Blackening of inner bracts and receptacle
Asparagus	> 10 kPa at 3 to 6°C > 15 kPa at 0 to 3°C	Increased elongation, weight gain and sensitivity to chilling and pitting	< 10 kPa	Discoloration
Avocado	> 15 kPa	Skin browning, off-flavour	< 1 kPa	Internal flesh breakdown, off-flavour
Banana °	> 7 kPa	Green fruit softening, undesirable texture and flavour	< 1 kPa	Dull yellow or brown skin discoloration, failure to ripen, off-flavour
Beans, Green, Snap	> 7 kPa for more than 24 h	Off-flavour	< 5 kPa for more than 24 h	Off-flavour
Blackberry	> 25 kPa	Off-flavour	< 2 kPa	Off-flavour
Blueberry	> 25 kPa	Skin browning, off-flavour	< 2 kPa	Off-flavour
Broccoli	> 15 kPa	Persistent off-odours	< 0.5 kPa	Off-odours, can be lost upon aeration if slight
Brussels sprouts	> 10 kPa	–	< 1 kPa	Off-odours, internal discoloration
Cabbage	> 10 kPa	Discoloration of inner leaves	< 2 kPa	Off-flavour, increased sensitivity to freezing
Cauliflower	> 5 kPa	Off-flavour, aeration removes slight damage, curd must be cooked to show symptoms	< 2 kPa	Persistent off-flavour and odour after cooking
Celery	> 10 kPa	Off-flavour and odour, internal discoloration	< 2 kPa	Off-flavour and odour
Cherimoya	Not determined	Not known	< 1 kPa	off-flavour
Cherry, sweet	> 30 kPa	Brown, discoloration of skin, off-flavour	< 1 kPa	Skin pitting, off-flavour
Cranberry	Not determined	Not determined	< 1 kPa	Off-flavour
Cucumber	> 5 kPa at 8°C > 10 kPa at 5°C	Increased softening, increased chilling injury, surface discoloration and pitting	< 1 kPa	Off-odours, breakdown and increased chilling injury
Custard apple	15 kPa +	Flat taste, uneven ripening	< 1 kPa	Failure to ripen
Durian	> 20 kPa	Not known	< 2 kPa	Failure to ripen, grey discoloration of pulp
Fig	> 25 kPa (?)	Loss of flavour (?)	< 2 kPa (?)	Off-flavour (?)
Grape	> 5 kPa	Browning of berries and stems	< 1 kPa	Off-flavour
Grapefruit	> 10 kPa	Scald-like areas on the rind, off-flavour	< 3 kPa	Off-flavour due to increased ethanol and acetaldehyde contents
Kiwifruit	> 7 kPa	Internal breakdown of the flesh	< 1 kPa	Off-flavour
Lemon	> 10 kPa	Increased susceptibility to decay, decreased acidity	< 5 kPa	Off-flavour
Lettuce, Crisphead	> 2 kPa	Brown stain	< 1 kPa	Breakdown at centre

Lime	> 10 kPa	Increased susceptibility to decay	< 5 kPa	Scald-like injury, decreased juice content
Mango	> 10 kPa	Softening, off-flavour	< 2 kPa < 5 kPa	Skin discoloration, greyish flesh colour, off-flavour
Melon, Cantaloupe	> 20 kPa	Off flavour and odours, impaired ripening	< 1 kPa	Off-flavour and odours, impaired ripening
Mushroom	> 20 kPa	Surface pitting	Near 0 kPa	Off-flavour and odours, stimulation of cap opening and stipe elongation
Nectarine	> 10 kPa	Flesh browning, loss of flavour	< 1 kPa	Failure to ripen, skin browning, off-flavour
Olive	> 5 kPa	Increased severity of chilling injury at 7°C	< 2 kPa	Off-flavour
Onion, bulb	> 10 kPa for short term, > 1 kPa for long term	Accelerated softening, rots and putrid odour	< 1 kPa	Off-odours and breakdown
Orange	> 5 kPa	Off-flavour	< 5 kPa	Off-flavour
Papaya	> 8 kPa	May aggravate chilling injury at < 12°C, off-flavour	< 2 kPa	Failure to ripen, off-flavour
Peach, clingstone	> 5 kPa	Internal flesh browning severity increases with CO <sub>2</sub> kPa	< 1 kPa	Off-flavour in the canned product
Peach, freestone	> 10 kPa	Flesh browning, off-flavour	< 1 kPa	Failure to ripen, skin browning, off-flavour
Pepper, Bell	> 5 kPa	Calyx discoloration, internal browning, and increased softening	< 2 kPa	Off-odours and breakdown
Pepper, Chilli	> 20 kPa at 5°C, > 5 kPa at 10°C	Calyx discoloration, internal browning, and increased softening	< 2 kPa	Off odours and breakdown
Persimmon	> 10 kPa	Off-flavour	< 3 kPa	Failure to ripen, off-flavour
Pineapple	> 10 kPa	Off-flavour	< 2 kPa	Off-flavour
Plum	> 1 kPa	Flesh browning	< 1 kPa	Failure to ripen, off-flavour
Rambutan	> 20 kPa	Not known	< 1 kPa	Increased decay incidence
Raspberry	> 25 kPa	Off-flavour, brown discoloration	< 2 kPa	Off-colours
Strawberry	> 25 kPa	Off-flavour, brown discoloration of berries	< 2 kPa	Off-flavour
Sweetcorn	> 10 kPa	Off-flavour and odours,	< 2 kPa	Off-flavour and off-odours,
Tomato	> 2 kPa for mature-green, > 5 kPa for turning, also depends on length of exposure and temperature	Discoloration, softening and uneven ripening	< 2 kPa depending on length of exposure	Off flavour, softening and uneven ripening



**Fig. 3.1.** CO<sub>2</sub> injury on apples from a CA store in UK. Photograph courtesy of Dr R.O. Sharples.



**Fig. 3.2.** O<sub>2</sub> injury on an apple taken from a CA store in UK. Photograph courtesy of Dr R.O. Sharples.



**Fig. 3.3.** Spongy tissue in mangoes.

‘McIntosh’ in cold storage. For blueberries, the lower O<sub>2</sub> limit increased with temperature and CO<sub>2</sub> level. Raising the temperature from 0°C to 25°C caused the lower O<sub>2</sub> limit to increase from about 1.8 kPa to approximately 4 kPa. Raising CO<sub>2</sub> levels from 5 kPa to 60 kPa increased the lower O<sub>2</sub> limit for blueberry fruits from approximately 4.5 kPa

to > 16 kPa (Beaudry and Gran, 1993). ‘Marshall McIntosh’ apples held at 3°C in 2.5–3 kPa O<sub>2</sub> + 11–12 kPa CO<sub>2</sub> developed small desiccated cavities in the cortex associated with CO<sub>2</sub> injury (DeEll *et al.*, 1995). Mencarelli *et al.* (1989) showed that high CO<sub>2</sub> concentrations (5, 8 or 12 kPa) during storage resulted in CO<sub>2</sub> injury of aubergines, characterized by external browning without tissue softening. Wardlaw (1938), also working with aubergines, showed that high CO<sub>2</sub> can cause surface scald, browning, pitting and excessive decay, and these symptoms are similar to those caused by chilling injury. Mencarelli *et al.* (1989) showed that the spherical-shaped aubergine cultivars ‘Black Beauty’ and ‘Sfumata Rosa’ were more tolerant of high CO<sub>2</sub> concentrations than the long-fruited cultivar ‘Violetta Lunga’. Burton (1974a) also found that the level of CO<sub>2</sub> that can cause damage varied between cultivars of the same crop and he put forward a hypothesis that these differences could be due to anatomical rather than biochemical differences. He stated that:

Variability in plant material prevents precise control of intercellular atmosphere; recommendations can be designed only to avoid complete anaerobic conditions and a harmful level of CO<sub>2</sub> in the centre of the least permeable individual fruit or vegetable.

High CO<sub>2</sub> levels (10 kPa) have been shown to cause damage to stored onions, particularly resulting in internal browning (Gadalla, 1997). Adamicki *et al.* (1977) suggested that the decomposition of the cell walls was the result of the influence of hydrolytic enzymes of the pectinase group. Their preliminary comparative studies on cells of sound and physiologically disordered onions due to CO<sub>2</sub> injury indicated destructive changes in the ultra-structure of the mitochondria. The mitochondria displayed fragmentation, reduction in size and changes in shape from elliptical to spherical. These changes have also been observed in studies on the influence of CO<sub>2</sub> on the ultra-structure of pears (Frenkel and Patterson, 1974). Elevated CO<sub>2</sub> concentrations were shown to inhibit the activity of succinic dehydrogenase, resulting in accumulation of

succinic acid, a toxicant to plant tissues (Hulme, 1956; Williams and Patterson, 1964; Frenkel and Patterson, 1974). Adamicki *et al.* (1977) discovered that the highest amino acid content was found in the physiologically disordered onion bulbs stored at 1°C, with lower values found at 5°C. They added that this may be due to greater enzyme activity, especially in physiologically disordered bulbs. Low O<sub>2</sub> levels (0.21 kPa and 3 kPa) on stored avocados from early-season harvests could lead to severe external browning (Allwood and Cutting, 1994).

There may be interactions between low O<sub>2</sub> and high CO<sub>2</sub> levels. Riad and Brecht (2003) reported that at 5°C sweetcorn tolerated 2 kPa O<sub>2</sub> or 25 kPa CO<sub>2</sub> alone for 2 weeks, but may be damaged by these two gas levels in combination.

### High Oxygen Storage

Increased O<sub>2</sub> is also achieved by hyperbaric storage (see also Chapter 10). Kidd and West (1934) showed that storage of apples in pure O<sub>2</sub> can be detrimental and Yahia (1989, 1991) showed that exposure of 'McIntosh' and 'Cortland' apples to 100 kPa O<sub>2</sub> at 3.3°C for 4 weeks did not enhance the production of aroma volatiles. Pears kept in 100 kPa O<sub>2</sub> showed an increase in the rate of softening, chlorophyll degradation and ethylene evolution (Frenkel, 1975). Various publications have shown that high O<sub>2</sub> levels in citrus fruit stores can affect the fruit colour. Navel and 'Valencia' oranges were stored for 4 weeks at 15°C in a continuous flow of gases and after 2 weeks of storage in 40 or 80 kPa O<sub>2</sub>, with the balance N<sub>2</sub>, the endocarp of navel oranges, but not of 'Valencia', had turned a perceptibly darker orange than fruit stored in air or in air with ethylene. The orange colour intensified with increased storage time and persisted at least 4 weeks after the fruit was removed from high O<sub>2</sub>. Juice from navel oranges, but not from 'Valencia', stored in 80 kPa O<sub>2</sub> for 4 weeks was slightly darker than juice from those stored in air (Houck *et al.*, 1978). Storage at 15°C for 4 weeks in 40 or 80 kPa O<sub>2</sub> deepened the red of flesh and juice of 'Ruby', 'Tarocco' and 'Sanguinello' blood oranges,

but TSS and TA of the juice were not affected. 'Hamlin', 'Parson Brown' and 'Pineapple' oranges showed similar results with the respiration rate of 'Pineapple' being highest in 80 kPa O<sub>2</sub> and lowest in 40 kPa O<sub>2</sub> or air (Aharoni and Houck, 1980).

Ripe-green tomatoes were stored at either 19°C or 13°C followed by 19°C in air or in 100 kPa O<sub>2</sub> at normal or reduced atmospheric pressure until they were fully red in colour. Fruit softening was exacerbated in 100 kPa O<sub>2</sub> and reduced pressure (0.25 atmospheres) compared with a normal gaseous atmosphere and reduced pressure. Removal of ethylene in the atmosphere made little difference to the fruit but in 100 kPa O<sub>2</sub> ethylene accumulation was detrimental (Stenvers, 1977). Li *et al.* (1973) reported that ripening of tomatoes was accelerated at 12–13°C in 40–50 kPa O<sub>2</sub> compared with those ripened in air.

At 4°C the respiration rate of potato tubers was lower in 1 or 3 kPa O<sub>2</sub> compared with those stored in air or in 35 kPa O<sub>2</sub>. Sprouting was inhibited and sugar content increased in 1 and 35 kPa O<sub>2</sub> compared with those in air, but those in 3 kPa O<sub>2</sub> showed no effects (Hartmans *et al.*, 1990). Storage atmospheres containing 40 kPa O<sub>2</sub> reduced mould infection compared with low O<sub>2</sub> levels, but increased sprouting and rooting of stored carrots (Abdel-Rahman and Isenberg, 1974). Blueberries in jars ventilated continuously with 60, 80 or 100 kPa O<sub>2</sub> had improved antioxidant capacity, total phenolic and anthocyanin contents and less decay compared with those in 0 or 40 kPa O<sub>2</sub> during 35 days of storage at 5°C (Zheng *et al.*, 2003). Dick and Marcellin (1985) showed that bananas held in 50 kPa O<sub>2</sub> during cooling periods of 12 h at 20°C had reduced high temperature damage during subsequent storage at 30–40°C. Belay *et al.* (2017) compared the storage of pomegranate arils at 5°C for 12 days in air, 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, 10 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 70 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, all with balance N<sub>2</sub>. They found that the highest relative composition of volatile organic compounds was in arils stored under 70 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>. Day (1996) indicated highly positive effects of storing minimally processed fruit and vegetables in 70 and 80 kPa O<sub>2</sub> in MA film packs.



He indicated that high O<sub>2</sub> inhibited undesirable fermentation reactions and delayed browning that was caused by damage during processing and that the O<sub>2</sub> levels of over 65 kPa inhibited both aerobic and anaerobic microbial growth. The technique used involved flushing the packs with the required gas mixture before they were sealed. The O<sub>2</sub> level within the package would then fall progressively as storage proceeded, due to the respiration of the fruit or vegetable contained in the pack. High O<sub>2</sub> levels have also been used in the marketing of other commodities. For example, Champion (1986) mentioned that O<sub>2</sub> levels above ambient were used to help preserve 'redness and eye appeal' of red meat.

### Carbon Dioxide Shock Treatment

Treating fruit and vegetables with high levels of CO<sub>2</sub> prior to storage can have beneficial effects on their subsequent storage life. Hribar *et al.* (1994) described experiments where 'Golden Delicious' apples were held in 15 kPa CO<sub>2</sub> for 20 days prior to storage in 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> for 5 months. The CO<sub>2</sub> treated fruits retained a better green skin colour during storage, which was shown to be due to inhibition of carotenoid production. Other effects of the CO<sub>2</sub> shock treatment were that it increased fruit firmness during storage, ethanol production was greater due to fermentation, but TA was not affected. Pesis *et al.* (1993a) described experiments where apples were treated with 95 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> + 4 kPa N<sub>2</sub> at 20°C for 24–48 h for 'Golden Delicious' and 24–96 h for 'Braeburn', then transferred to either 20°C or 5°C for 6 weeks in air followed by a shelf-life of 10 days at 20°C. High CO<sub>2</sub> pre-treatment increased respiration rate and induced ethylene, ethanol and ethyl acetate production in 'Golden Delicious' apples during storage at 20°C. The fruits became softer and more yellow but 'tastier' than non-treated fruits after 2 weeks at 20°C. For 'Braeburn', respiration rate and ethylene production were reduced and volatiles were increased by the high CO<sub>2</sub> pre-treatment during shelf-life at 20°C following

6 weeks of cold storage, and treated fruits remained firmer but more yellow than control fruits.

Two cultivars of table grapes were stored by Laszlo (1985) in CA: 'Waltham Cross' in 1–21 kPa O<sub>2</sub> + 0–5 kPa CO<sub>2</sub> and 'Barlinka' in 1–21 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, both at –0.5°C for 4 weeks. Some of the grapes were exposed to 10 kPa CO<sub>2</sub> for 3 days before storage in air at 10°C for 1 week. With both cultivars the incidence of berry decay (mainly due to infections by *Botrytis cinerea*) was highest with grapes that had been subjected to the CO<sub>2</sub> treatment. The incidence of berry cracking was less than 1% in 'Waltham Cross' but in 'Barlinka' it was higher with fruits that had been subjected to the CO<sub>2</sub> treatment, high O<sub>2</sub> concentrations and controls without SO<sub>2</sub>. In a trial with 'Passe Crassane' pears stored in air at 0°C for 187 days, internal browning was prevented by treatment with 30 kPa CO<sub>2</sub> for 3 days at intervals of 14–18 days. Also the fruits that had been subjected to CO<sub>2</sub> shock treatment had excellent quality after being ripened at 20°C (Marcellin *et al.*, 1979). They also reported that, with 'Comice' pears stored at 0°C for 169 days followed by ripening for 7 days, 15 kPa CO<sub>2</sub> shock treatment every 14 days greatly reduced the incidence of scald and internal browning. Park *et al.* (1970) showed that PE film packaging and/or CO<sub>2</sub> shock treatment markedly delayed ripening, preserved freshness and reduced spoilage and core browning of pears during storage.

'Fuerte' avocados were treated with 25 kPa CO<sub>2</sub> for 3 days, commencing 1 day after harvest, followed by storage for 28 days at 5.5°C in either 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> or in air. Fruits from the CA and CO<sub>2</sub> shock treatments showed a lower incidence of physiological disorders (mesocarp discoloration, pulp spot and vascular browning) than fruits that had not had the CO<sub>2</sub> treatment. Total phenols tended to be lower in CO<sub>2</sub> shock fruits than in fruits from other treatments (Bower *et al.*, 1990). Wade (1979) showed that intermittent exposure of unripe avocados to 20 kPa CO<sub>2</sub> reduced chilling injury when they were stored at 4°C.

Pesis and Sass (1994) showed that exposure of feijoa fruits (*Acca sellowiana*

cultivar 'Slor') to total  $N_2$  or  $CO_2$  for 24 h prior to storage induced the production of aroma volatiles including acetaldehyde, ethanol, ethyl acetate and ethyl butyrate. The enhancement of flavour was mainly due to the increase in volatiles and not to changes in TSS or the TSS:TA ratio. Eaks (1956) showed that high  $CO_2$  in the storage atmosphere could have detrimental effects on cucumbers in that it appeared to increase their susceptibility when stored at low temperatures. Broccoli was treated with  $CO_2$  at 5°C with 20, 30 and 40 kPa  $CO_2$  for 3 and 6 days before storage.  $CO_2$  treatment delayed yellowing and loss of both chlorophyll and ascorbic acid and retarded ethylene production, but 30 or 40 kPa  $CO_2$  for 6 days resulted in the development of an offensive odour and flavour dissipated when the broccoli was transferred to air (Wang, 1979).

$CO_2$  behaves as a supercritical fluid above its critical temperature of 31.1°C and above its critical pressure of 7.39 MPa, and expands to fill the container like a gas but with a density like a liquid. Gui *et al.* (2006) exposed horseradish to supercritical  $CO_2$  and found some reduction in enzyme activity which was reversed during subsequent storage in air at 4°C.

### Total Nitrogen or High Nitrogen Storage

Since fresh fruit and vegetables are living organisms, they require  $O_2$  for aerobic respiration. Where this is available to individual cells below a threshold level, the fruit or vegetable or any part of the organism can go into anaerobic respiration, usually called fermentation, the end products of which are organic compounds that can affect their flavour. It was reported by Food Investigation Board (1920) that storage of plums in total  $N_2$  almost completely inhibited ripening. Plums were reported to be able to tolerate, for a considerable period, an almost complete absence of  $O_2$  without being killed or developing an alcoholic or unpleasant flavour. Parsons *et al.* (1964) successfully stored several fruit and vegetables at 1.1°C

in either total  $N_2$  or 1 kPa  $O_2$  + 99 kPa  $N_2$ . During 10 days storage of lettuce, both treatments reduced the physiological disorder russet spotting and butt discoloration compared with those stored in air without affecting their flavour. At 15.5°C, ripening of both tomatoes and bananas was retarded when they were stored in total  $N_2$ , but their flavour was poorer only if they were held in these conditions for longer than 4 days or for over 10 days in 99 kPa  $N_2$  + 1 kPa  $O_2$ . In strawberries, 100 kPa and 99 kPa  $N_2$  were shown to reduce mould growth during 10 days of storage at 1.1°C with little or no effect on flavour. Decay reduction was also observed on peaches stored in either 100 kPa or 99 kPa  $N_2$  at 1.1°C; off-flavours were detected after 4 days in 100 kPa, but none in those stored in 99 kPa  $N_2$ . Klieber *et al.* (2002) found that storage of bananas in total  $N_2$  at 22°C did not extend their storage life compared with those stored in air but resulted in brown discoloration. Storing potatoes in total  $N_2$  prevented accumulation of sugars at low temperature, but it had undesirable side effects (Harkett, 1971). It was reported that the growth of fungi was inhibited by atmospheres of total  $N_2$  but not in atmospheres containing 99 kPa  $N_2$  + 1 kPa  $O_2$  (Ryall, 1963). Initial low oxygen stress (ILOS) is a name that has been given to a pre-storage treatment where apples and pears are exposed to anaerobic conditions, commonly 7 days in 100%  $N_2$ , to inhibit superficial scald and increase their storage life (Eaves *et al.*, 1969a; Robatscher *et al.*, 2012) (see Chapter 9). Treatment of fruit in atmospheres of total  $N_2$  prior to storage was shown to retard the ripening of tomatoes (Kelly and Saltveit, 1988) and avocados (Pesis *et al.*, 1993b). 'Fuerte' avocados were exposed to 97 kPa  $N_2$  for 24 h or 40 h and then successfully stored at 17°C in air for 7 days by Dori *et al.* (1995) and Pesis *et al.* (1993b). They exposed 'Fuerte' avocados to 97 kPa  $N_2$  for 24 h at 17°C then stored them at 2°C for 3 weeks followed by shelf-life at 17°C in air. Pre-treatment with  $N_2$  reduced chilling injury symptoms and softening was also delayed and they had lower respiration rates and ethylene production during cold storage and shelf-life.

## Ethylene

The physiological effects of ethylene on plant tissue have been known for several decades. Gane (1934), at the Low Temperature Research Station at Cambridge, showed that the volatile agent given off by ripening apples and pears is ethylene and at that time ethylene began to be considered as a ripening hormone and shipping companies stopped the transport of apples and bananas in adjacent holds. The concentration of ethylene causing half maximum inhibition of growth of etiolated pea seedlings was shown to be related to the CO<sub>2</sub> and O<sub>2</sub> levels in the surrounding atmosphere (Table 3.4) by Burg and Burg (1967). They suggested that high levels of CO<sub>2</sub> in stores can compete with ethylene for binding sites in fruit and the biological activity of 1 kPa ethylene was negated in the presence of 10 kPa CO<sub>2</sub>. CO<sub>2</sub> levels in store affected ethylene biosynthesis and it was shown that increased levels of CO<sub>2</sub> in CA stores containing apples reduced ethylene levels (Tomkins and Meigh, 1968). Yang (1985) showed that CO<sub>2</sub> accumulation in the intercellular spaces of fruit acts as an ethylene antagonist. Woltering *et al.* (1994) therefore suggested that most of the beneficial effects of CA storage in climacteric fruit are due to suppression of ethylene action. However, Knee (1990) had pointed out earlier that ethylene was not normally removed in commercial CA apple stores, but CA technology was successful and so there must be an effect of reduced O<sub>2</sub> and or increased CO<sub>2</sub> apart from those on ethylene synthesis or ethylene action. Knee also mentioned that laboratory experiments on CA storage commonly used

small containers which were constantly purged with the appropriate gas mixture. This meant that ethylene produced by the crop under such conditions was constantly being removed. In contrast, the concentration of gases in commercial CA stores was usually adjusted by scrubbing and limited ventilation, which could allow ethylene to accumulate to high levels in the stores. Knee also pointed out that ethylene concentrations of up to 1000 µl/l had been reported in CA stores containing apples and 100 µl/l in CA stores containing cabbage.

Deng *et al.* (1997) found that as the O<sub>2</sub> level in the store reduced, the rate of ethylene production was also reduced. Wang (1990) also showed that ethylene production was suppressed in CA storage. He described experiments with sweet peppers stored at 13°C where exposure to combinations of 10–30 kPa CO<sub>2</sub> with either 3 or 21 kPa O<sub>2</sub> suppressed ethylene production down to less than 10 nl/100 g/h compared with fruits stored in air at 13°C which had ethylene production in the range of 40–75 nl/100 g/h. He also showed that the ethylene levels rapidly increased to a similar level to those stored in air when removed from exposure to CA storage for just 3 days. The concentrations of CO<sub>2</sub> and O<sub>2</sub> in such conditions should therefore inhibit ethylene biosynthesis. Plich (1987) found that the presence of a high concentration of CO<sub>2</sub> may strongly inhibit ACC synthesis in ‘Spartan’ apples and, consequently, the rate of ethylene evolution. The lowest ethylene evolution was found in fruit stored in 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> or 1 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>, but in 3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> there was considerably more ethylene production whereas in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> ethylene production was markedly decreased.

Before the availability of gas chromatography it was generally thought that only climacteric fruit produced ethylene. Now it has been shown that all plant material can produce ethylene and Shinjiro *et al.* (2002) found that ethylene production in pineapples (a non-climacteric fruit) increased the longer the fruits were stored, but the maximum production rate was less than 1nl/g/h.

CA storage has also been shown to stimulate ethylene production. After 70 days in

**Table 3.4.** Sensitivity at 20°C of pea seedlings to ethylene in different levels of O<sub>2</sub>, CO<sub>2</sub> and N<sub>2</sub> (Burg and Burg, 1967).

O <sub>2</sub> kPa	CO <sub>2</sub> kPa	N <sub>2</sub> kPa	Sensitivity (µl ethylene/l)
0.7	0	99.3	0.6
2.2	0	97.8	0.3
18.0	0	82.0	0.14
18.0	1.8	80.2	0.3
18.0	7.1	74.9	0.6

cold storage of celeriac in air the ethylene content did not exceed 2 ppm, but the ethylene level in CA stores of celeriac was nearly 25 ppm (Golias, 1987). Plich (1987) found that the lowest ethylene production was in apples stored in 1 kPa O<sub>2</sub> + 0 or 2 kPa CO<sub>2</sub>, but there was considerably more ethylene production in 3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>. Bangerth (1984) in studies on apples and bananas suggested that they became less sensitive to ethylene during prolonged storage.

The level of CO<sub>2</sub> and O<sub>2</sub> in the surrounding environment of climacteric fruit can affect the ripening rate by suppressing the synthesis of ethylene. The biosynthesis of ethylene in ripening fruit was shown in early work by Gane (1934) to cease in the absence of O<sub>2</sub>. Wang (1990) reviewed the literature on the effects of CO<sub>2</sub> and O<sub>2</sub> on the activity of enzymes associated with fruit ripening and cited many examples of the activity of these enzymes being reduced in CA storage. This is presumably due, at least partly, to many of these enzymes requiring O<sub>2</sub> for their activity. Ethylene biosynthesis was studied in 'Jonathan' apples stored at 0°C in 0–20 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 0–20 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub> for up to 7 months. Internal ethylene concentration, ACC levels and ACC-oxidase activity were determined in fruits immediately after removal from storage and after holding at 20°C for one week. Ethylene production by fruits was inhibited by increasing CO<sub>2</sub> concentration over the range of 0–20 kPa at both 3 and 15 kPa O<sub>2</sub>. ACC levels were similarly reduced by increasing CO<sub>2</sub> concentrations even in 3 kPa O<sub>2</sub>, but at 3 kPa O<sub>2</sub> ACC accumulation was enhanced but only in the absence of CO<sub>2</sub>. ACC oxidase activity was stimulated by CO<sub>2</sub> up to 10 kPa but was inhibited by 20 kPa CO<sub>2</sub> at both O<sub>2</sub> concentrations. The inhibition of ethylene production by CO<sub>2</sub> may, therefore, be attributed to its inhibitory effect on ACC-synthase activity (Levin *et al.*, 1992). In other work, storing pre-climacteric apples of the cultivars 'Barnack Beauty' and 'Wagner' at 20°C for 5 days in 0 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> + 99 kPa N<sub>2</sub> or in air containing 15 kPa CO<sub>2</sub> inhibited ethylene production and reduced ACC concentration and ACC-oxidase activity compared with storage in air (Lange

*et al.*, 1993). Storage of kiwifruit in 2–5 kPa O<sub>2</sub> + 0–4 kPa CO<sub>2</sub> reduced ethylene production and ACC oxidase activity (Wang *et al.*, 1994). Ethylene production was lower in 'Mission' figs stored at 15–20 kPa CO<sub>2</sub> concentrations compared with those kept in air (Colelli *et al.*, 1991). Low O<sub>2</sub> and increased CO<sub>2</sub> concentrations were shown to decrease ethylene sensitivity of 'Elstar' apples and 'Scania' carnation flowers during CA storage (Woltering *et al.*, 1994).

Quazi and Freebairn (1970) showed that high CO<sub>2</sub> and low O<sub>2</sub> delayed the production of ethylene in pre-climacteric bananas, but the application of exogenous ethylene was shown to reverse this effect. Wade (1974) showed that bananas could be ripened in atmospheres of reduced O<sub>2</sub>, even as low as 1 kPa, but the peel failed to de-green, which resulted in ripe bananas that were still green. Similar effects were shown at O<sub>2</sub> levels as high as 15 kPa. Since the de-greening process in 'Cavendish' bananas is entirely due to chlorophyll degradation (Seymour *et al.*, 1987a and b) the CA storage treatment was presumably due to suppression of this process. Hesselman and Freebairn (1969) showed that ripening of bananas, which had already been initiated to ripen by ethylene, was slowed in a low O<sub>2</sub> atmosphere. Goodenough and Thomas (1980, 1981) also showed suppression of de-greening of fruits during ripening; in this case it was with tomatoes ripened in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. Their work, however, showed that this was due to a combination of suppression of chlorophyll degradation and the suppression of the synthesis of carotenoids and lycopene. Jeffery *et al.* (1984) also showed that lycopene synthesis was suppressed in tomatoes stored in 6 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub>.

Ethylene is used as a commercial sprouting inhibitor in potatoes, but ethylene is required to break dormancy, while continuous exposure inhibits sprout growth. In sweet potatoes Cheema *et al.* (2013) showed that continuous exposure to ethylene at 10 µl/l for 24 h, exposure to 1-MCP at 625 nl/l or dipping in AVG at 100 µl/l all inhibited sprout growth over 4 weeks of storage at 25°C. Both ethylene on its own and 1-MCP inhibited sprout growth, which they concluded indicated that while continuous exposure to exogenous

ethylene leads to sprout growth inhibition, ethylene is also required for sprouting.

### Carbon Monoxide

CO is a colourless, odourless gas which is flammable and explosive in air at concentrations between 12.5 and 74.2 kPa. It is extremely toxic and human exposure to 0.1 kPa for 1 h can cause unconsciousness and exposure for 4 h can cause death. Goffings and Herregods (1989) reported that CA storage of leeks could be improved by the inclusion of 5 kPa CO. If added to CA stores, with levels of 2–5 kPa O<sub>2</sub>, CO can inhibit discoloration of lettuce on the cut butts or from mechanical damage on the leaves, but the effect was lost when the lettuce was removed to air at 10°C (Kader, 1992). Storage in 4 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> + 5 kPa CO was shown to be optimum in delaying ripening and maintaining good quality of mature-green tomatoes stored at 12.8°C (Morris *et al.*, 1981). Reyes and Smith (1987) reported positive effects of CO during the storage of celery. CO has fungistatic properties, especially when combined with low O<sub>2</sub> (Kader, 1993). *Botrytis cinerea* on strawberries was reduced in CO levels of 5 kPa or higher in the presence of 5 kPa O<sub>2</sub> or lower (El-Goorani and Sommer, 1979). Decay in mature-green tomatoes stored at 12.8°C was reduced when 5 kPa CO was included in the storage atmosphere (Morris *et al.*, 1981). They also reported that in capsicums and tomatoes the level of chilling injury symptoms could be reduced, but not eliminated, when CO was added to the store.

### Temperature

Kidd and West (1927a, b) were the first to show that the effects of the gases in extending storage life could vary with temperature. At 10°C, CA storage could increase the postharvest life of apples compared with those in air, while at 15°C their storage life was the same in both CA storage and in air. The opposite effect was shown by Ogata *et al.* (1975). They

found that CA storage of okra at 1°C did not increase storage life compared with storage in air, but at 12°C CA storage increased postharvest life. Izumi *et al.* (1996a) also showed that the best CA conditions varied with temperature. They found that the best storage conditions for the broccoli cultivar 'Marathon' were 0.5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> at 0°C and 5°C, and 1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> at 10°C.

The relationship between CA storage and temperature has been shown to be complex. Respiration rates of apples were progressively reduced by lowering O<sub>2</sub> levels from 21 kPa to 1 kPa. Although lowering the temperature from 4°C to 2°C also reduced the respiration rate, fruits stored in 1 or 2 kPa O<sub>2</sub> were shown to be respiring faster after 100 days at 0°C than at 2°C or 4°C. After 192 days the fruit stored in air also showed an increase in respiration rate at 0°C. These higher respiration rates preceded the development of low temperature breakdown in fruit stored in air, 1 or 2 kPa O<sub>2</sub> at 0°C and in 1 kPa O<sub>2</sub> at 2°C. Progressively lowering the O<sub>2</sub> concentration reduced ethylene production but increased the retention of acidity, TSS, chlorophyll and firmness. In the absence of low temperature breakdown the effects of reduced temperature on fruit ripening were similar to those of lowered O<sub>2</sub> concentrations. The quality of apples stored at 4°C in 1 kPa O<sub>2</sub> was markedly better than in 2 kPa; the fruits were also free of core flush (brown core) and other physiological disorders (Johnson and Ertan, 1983).

### Humidity

Kajiura (1973) stored *Citrus natsudaidai* at 4°C for 50 days in either 98–100 or 85–95% RH in air mixed with 0, 5, 10 or 20 kPa CO<sub>2</sub>. It was found that, at the higher humidity, high CO<sub>2</sub> increased the water content of the peel and the ethanol content of the juice and produced abnormal flavour and reduced the internal O<sub>2</sub> content, causing watery breakdown. At the lower humidity, no injury occurred and CO<sub>2</sub> was beneficial, its optimum level being much higher. In another trial, *C. natsudaidai* fruits stored at

4°C in 5 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub> had abnormal flavour development at 98–100% RH but those in 85–95% RH did not. Bramlage *et al.* (1977) showed that treatment of ‘McIntosh’ apples with high CO<sub>2</sub> in a non-humidified room reduced CO<sub>2</sub> injury without reducing other benefits compared with those treated in a room with humidified air. In a comparison by Polderdijk *et al.* (1993) of 85, 90, 95 or 100% RH during CA storage of capsicums, they found that after removal from storage the incidence of decay increased as humidity increased but weight loss and softening increased as humidity decreased. The physiological disorder vascular streaking in cassava that develops rapidly during storage was almost completely inhibited at high humidity irrespective of the O<sub>2</sub> level (Aracena *et al.*, 1993).

### Delayed CA Storage

Generally, placing fruit or vegetables in store as quickly as possible after harvest gives the longest storage period. Drake and Eisele (1994) found that immediate establishment of CA conditions of apples after harvest resulted in good-quality fruits after 9 months of storage. Reduced quality was evident when CA establishment was delayed by as little as 5 days, even though the interim period was in air at 1°C. Colgan *et al.* (1999) also showed that apple scald was controlled less effectively when establishment of CA conditions was delayed. In contrast, Streif and Saquet (2003) found that flesh browning in apples was reduced when the establishment of the CA conditions was delayed, with 30 days’ delay being optimum. However, softening was quicker than in fruits where CA conditions were established directly after harvest. Landfald (1988) recommended prompt CA storage to reduce the incidence of lenticel rot in ‘Aroma’ apples. BBD appeared to develop during the first 2 weeks of storage and storage in air at 0°C prior to CA storage decreased incidence and severity of the disorder (Elgar *et al.*, 1998). Watkins *et al.* (1997) also showed that the sensitivity of ‘Braeburn’ to CO<sub>2</sub> induced injury was greatest soon after harvest, but declined if fruits were held in

air storage before CO<sub>2</sub> application. The external skin disorders in ‘Empire’ apples could be reduced when fruits were held in air before CA was established (Watkins *et al.*, 1997). Streif and Saquet (2003) found that flesh browning in ‘Elstar’ apples was reduced when the establishment of the CA conditions was delayed for up to 40 days. A delay of 30 days was optimum, but firmness loss was much higher than in fruits where CA conditions were established directly after harvest. The storage conditions they compared at 1°C were 0.6 kPa CO<sub>2</sub> + 1.2 kPa O<sub>2</sub>, 2.5 kPa CO<sub>2</sub> + 1.2 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 1.2 kPa O<sub>2</sub>, 0.6 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub>, 2.5 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> and 5 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub>. Wang *et al.* (2000c) also found that CO<sub>2</sub> linked disorders were reduced in ‘Elstar’ apples by storage at 3°C in air before CA storage, but excessive ripening and associated loss of flesh firmness occurred. Argenta *et al.* (2000) found that delaying establishment of CA conditions for 2–12 weeks significantly reduced the severity of brown heart in ‘Fuji’ apples, but resulted in lower firmness and TA compared with establishment of 1.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. DeEll *et al.* (2016) showed that slow cooling of apples after harvest may exacerbate the development of external CO<sub>2</sub> injury during CA storage.

Storage of ‘Jupiter’ capsicums for 5 days at 20°C in 1.5 kPa O<sub>2</sub> resulted in post-storage respiratory rate suppression for about 55 h after transfer to air (Rahman *et al.*, 1995). Saquet *et al.* (2017) stored ‘Rocha’ pears for 257 days at –0.5°C under 0.6 kPa CO<sub>2</sub> combined with either 0.5 or 3.0 kPa O<sub>2</sub>. The atmospheres were imposed either at the beginning of storage or after 46 days. They found that delaying CA for 46 days did not benefit the pears but development of internal disorders depended on O<sub>2</sub> level, in that when CA was imposed immediately under 0.5 kPa O<sub>2</sub> there were no internal disorders, but under 3.0 kPa O<sub>2</sub>, 63.3% of fruit developed internal disorders.

In work on kiwifruit, Tonini and Tura (1997) showed that storage in 4.8 kPa CO<sub>2</sub> + 1.8 kPa O<sub>2</sub> reduced rots caused by infections with *Botrytis cinerea*. If the fruit was cooled to –0.5°C immediately after harvest then the effect was greatest, the quicker the CA storage

conditions were established. With a delay of 30 days the CA storage conditions were ineffective in controlling the rots.

In pears, delaying CA conditions for 46 days did not benefit 'Rocha' pears during storage for 257 days. Pears tolerated immediate storage in 0.6 kPa CO<sub>2</sub> + 0.5 kPa O<sub>2</sub> without developing internal disorders during subsequent storage. However, 63.3% of fruit stored immediately in 0.6 kPa CO<sub>2</sub> + 3.0 kPa O<sub>2</sub> developed internal disorders, while delayed pull-down of O<sub>2</sub> for 46 days reduced their internal disorders to 35.5%, but increased the disorder incidence in fruit in 0.6 kPa CO<sub>2</sub> + 0.5 kPa O<sub>2</sub> to 27.3% (Saquet *et al.*, 2017). Both delayed storage of 'Flavourtop' nectarines held for 48 h at 20°C before storage, and CA (10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>), alleviated or prevented chilling injury manifested as woolliness in nectarines stored for up to 6 weeks at 0°C (Zhou *et al.*, 2000).

### Interrupted CA Storage

Where CA storage has been shown to have detrimental side effects on fruit and vegetables, the possibility of alternating CA storage with air storage has been studied. Results have been mixed with positive, negative and in some cases no effect. Neuwirth (1988) described storage of 'Golden Delicious' apples from January to May at 2°C in 1 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. These regimes were interrupted by a 3-week period of ventilation with air, beginning on 15 January, 26 January, 17 February or 11 March, after which the CA treatment was reinstated. Ventilation at the time of the climacteric rise in respiration in late February to early March produced large increases in respiration rate and volatile flavour substances, but ventilation at other times had little effect. After CA conditions were restored, respiration rate and the production of flavour substances declined again, sometimes to below the level of fruit stored in continuous controlled atmospheres.

Storage of bananas at high temperatures, as may happen in producing countries, can

cause physiological disorders and unsatisfactory ripening. In trials with the variety 'Poyo' from Cameroon, storage at 30–40°C was interrupted by 1–3 periods at 20°C for 12 h either in air or in atmospheres containing 50 kPa O<sub>2</sub> or 5 kPa O<sub>2</sub>. Damage was reduced, especially when fruits were stored at 30°C and received three cooling periods in 50 kPa O<sub>2</sub> (Dick and Marcellin, 1985). Parsons *et al.* (1974) interrupted CA storage of tomatoes at 3 kPa O<sub>2</sub> + 0, 3 or 5 kPa CO<sub>2</sub> each week by exposing them to air for 16 h at 13°C. This interrupted storage had no measurable effect on the storage life of the fruit, but increased the level of decay that developed on fruit when removed from storage to higher temperatures to simulate shelf-life. Intermittent exposure of 'Haas' avocados to 20 kPa CO<sub>2</sub> increased their storage life at 12°C and reduced chilling injury during storage at 4°C compared with those stored in air at the same temperatures (Marcellin and Chaves, 1983). Anderson (1982) described experiments where peaches and nectarines were stored at 0°C in 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, which was interrupted every 2 days by removing the fruit to 18–20°C in air. When subsequently ripened, fruits in this treatment had little of the internal breakdown found in fruits stored continuously in air at 0°C.

### Residual Effects of CA Storage

There is considerable evidence in the literature that storing fruit and vegetables in CA storage can affect their subsequent shelf-life or marketable life. Day (1996) indicated that minimally processed fruit and vegetables stored in 70 kPa and higher O<sub>2</sub> levels deteriorated more slowly on removal than those freshly prepared. Hill (1913) described experiments on peaches stored in increased levels of CO<sub>2</sub> and showed that their respiration rate was reduced, not only during exposure, but he also showed that respiration rate only returned to the normal level after a few days in air. Bell peppers exposed to 1.5 kPa O<sub>2</sub> for 1 day exhibited suppressed respiration rate for at least 24 h after transfer to air (Rahman *et al.*, 1993a). Berard (1985) showed that cabbages stored at 1°C and

92% RH in 2.5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> had reduced losses during long-term storage compared with those stored in air, but also the beneficial effects persisted after removal from CA. Burdon *et al.* (2008) showed that avocados that had been stored in CA had a longer shelf-life than those that had been stored in air for a similar period.

Goulart *et al.* (1990) showed that when 'Bristol' raspberries were stored at 5°C in 2.6, 5.4 or 8.3 kPa O<sub>2</sub> + 10.5, 15.0 or 19.6 kPa CO<sub>2</sub> or in air the weight loss was greatest after 3 days for berries stored in air. When berries were removed from CA after 3 days and held for 4 days in air at 1°C, those that had been stored in 15 kPa CO<sub>2</sub> had less deterioration than any other treatment except for those stored in air. Deterioration was greatest in the berries that had previously been stored in the 2.6, 5.4 or 8.3 kPa O<sub>2</sub> + 10.5 kPa CO<sub>2</sub> treatments. Berries removed after 7 days and held for up to 12 days at 1°C showed least deterioration after the 15.0 kPa CO<sub>2</sub> storage.

The climacteric rise in respiration rate of cherimoyas was delayed by storage in 15 or 10 kPa O<sub>2</sub> and fruit kept in 5 kPa O<sub>2</sub> did not show a detectable climacteric rise and did not produce ethylene. All fruit ripened normally after being transferred to air storage at 20°C, but the time needed to reach an edible condition differed with O<sub>2</sub> level and was inversely proportional to O<sub>2</sub> concentration during storage. The actual data showed that following 30 days of storage in 5, 10 and 20 kPa O<sub>2</sub> fruit took 11, 6 and 3 days respectively to ripen at 20°C (Palma *et al.*, 1993).

Fruit firmness can be measured by inserting a metal probe into a fruit and measuring its resistance to the insertion. This is called a pressure test and generally the greater the resistance, the firmer and more immature is the fruit. The plum cultivars 'Santa Rosa' and 'Songold' were partially ripened to a firmness of approximately 4.5 kg pressure then kept at 0.5°C in 4 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 7–14 days. This treatment kept the fruits in an excellent condition for an additional 4 weeks when they were removed to 7.5°C in air (Truter and Combrink, 1992). Mencarelli *et al.* (1983) showed that storage of courgettes in low O<sub>2</sub> protected them against chilling injury, but on removal to air

storage at the same chilling temperature the protection disappeared within 2 days.

Khanbari and Thompson (1996) stored the potato cultivars 'Record', 'Saturna' and 'Hermes' in different CA conditions and found that there was almost complete sprout inhibition, low weight loss and maintenance of a healthy skin for all cultivars stored in 9.4 kPa CO<sub>2</sub> + 3.6 kPa O<sub>2</sub> at 5°C for 25 weeks. When tubers from this treatment were stored for a further 20 weeks in air at 5°C the skin remained healthy and they did not sprout while the tubers that had been previously stored in air or other CA combinations sprouted quickly. The fry colour of the crisps made from these potatoes was darker than the industry standard, but when they were reconditioned, tubers of 'Saturna' produced crisps of an acceptable fry colour while crisps from the other two cultivars remained too dark. This residual effect of CA storage could have major implications in that it presents an opportunity to replace chemical treatments in controlling sprouting in stored potatoes. The reverse was the case with bananas that had been initiated to ripen by exposure to exogenous ethylene and then immediately stored in 1 kPa O<sub>2</sub> at 14°C. They remained firm and green for 28 days, but then ripened almost immediately when transferred to air at 21°C (Liu, 1976a). Conversely, Wills *et al.* (1982) showed that pre-climacteric bananas exposed to low O<sub>2</sub> took longer to ripen when subsequently exposed to air than fruit kept in air for the whole period.

Hardenburg *et al.* (1977) showed that apples in CA storage for 6 months and then for 2 weeks at 21°C in air were firmer and more acid, and had a lower respiration rate, than those that had previously been stored in air. Storage of 'Jupiter' capsicums for 5 days at 20°C in 1.5 kPa O<sub>2</sub> resulted in post-storage suppression of the respiration rate for about 55 h after transfer to air and a marked reduction in the oxidative capacity of isolated mitochondria. Mitochondrial activity was suppressed for 10 h after transfer to air but within 24 h had recovered to values comparable to those of mitochondria from fruit stored continuously in air (Rahman *et al.*, 1993b, 1995).



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## Quality

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Storage of crops *per se* can detrimentally affect eating quality through senescent breakdown or, in climacteric fruits, being overripe. There is also evidence in the literature that fruit ripened after storage does not taste as good as fruit ripened directly after harvest. This is not always the case and would depend on the length of storage and the type of crop. Generally the level of flavour volatile compounds is lower after storage than in freshly harvested fruit, but there is little evidence that CA storage affects flavour volatiles in any way that is different to storage in air. However, taking phenolics as an example, Tomás-Barberán and Gil (2008) showed that the method of preparation can affect their nutritional status. For example, steam-cooked spinach had 80 mg phenolic antioxidants per 100 g, and boiled, with the water removed, had only 33 mg/100 g. They also showed that the cultivar and parts that are consumed can have a very big effect on their nutritional value (Table 4.1).

Sugar levels clearly affect flavour. Since they are a primary source of energy for plant metabolism, generally the longer the storage period the lower are the sugar levels. However, in many fruits starch is the main storage carbohydrate, which is broken down to sugars during storage. In those cases there is a balance between sugars being produced and sugars being used in metabolism. The acidity of fruit and vegetables also has an effect on

flavour. The data from storage trials show widely different effects on acidity, including levels of ascorbic acid (AA). Generally the acid levels of fruit and vegetables should be greater after CA storage in increased CO<sub>2</sub> atmospheres compared with air, because CO<sub>2</sub> in solution is acidic and there is lower acid metabolism associated with CA storage.

### Flavour

CA storage can affect the flavour of fruit and vegetables and both positive and negative effects have been cited in the literature. CA-stored apples were shown to retain a good flavour longer than those stored in air (Reichel, 1974) and in many other crops CA storage has generally been shown to maintain better flavour than storage in air (e.g. Zhao and Murata, 1988; Wang, 1990). However, the stage of ripeness of fruit when they are put into storage has the major effect on their flavour, sweetness, acidity and texture. It is therefore often difficult to specify exactly what effect CA storage has, since the effect on flavour may well be confounded with the effect of maturity and ripening. An example of this is that the storage of tomatoes in low concentrations of O<sub>2</sub> had less effect on fruits, which were subsequently ripened,

**Table 4.1.** Amount of phenolic antioxidants contained in 150 g of selected fruit and vegetables (adapted from Tomás-Barberán and Gil, 2008).

Crop	Cultivar	Part	Phenolic antioxidants intake mg/150g
Peach	Snow King	With peel	110
	Flavour Crest	Without peel	15
Grape	Napoleon	With seeds and peel	150
	Napoleon	Without seeds and peel	5
Orange	Navel	With albedo	400
	Navel	Without albedo	100
Lettuce	Lollo Rosso	With external leaves	300
	Iceberg	White midribs	10

than the stage of maturity at which they were harvested (Kader *et al.*, 1978). Stoll (1976) showed that 'Louise Bonne' pears were still of good eating quality after 3-4 months storage in air but were of similar good eating quality after 5-6 months in CA storage. In contrast it was reported that the flavour of satsumas stored at low O<sub>2</sub> with high CO<sub>2</sub> was inferior to those that had been stored in air (Ito *et al.*, 1974). Truter *et al.* (1994) showed that after 6 weeks storage at -0.5°C under 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, apricots had an inferior flavour compared with those stored for the same period in air (Table 4.2).

Spalding and Reeder (1974b) showed that limes (*Citrus latifolia*) stored for 6 weeks at 10°C in 5 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> or 21 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> lost more acid and sugars than those stored in air, but the limes from all the treatments had acceptable flavour and juice content. 'Golden Delicious', 'Idared' and 'Gloster' apples were stored for 100 or 200 days at 2°C and 95% RH under 15-16 kPa O<sub>2</sub> + 5-6 kPa CO<sub>2</sub>, then moved to 5°C, 10°C or 15°C (all at approximately 60% RH) for 16 days to determine their shelf-life. Flavour improvement occurred only in fruits that were removed from CA storage after 100 days and the apples removed from CA storage after 200 days showed a decline in flavour during subsequent storage at 5°C, 10°C or 15°C (Urban, 1995).

Blednykh *et al.* (1989) showed that Russian cultivars of cherry stored under 8-16 kPa CO<sub>2</sub> + 5-8 kPa O<sub>2</sub> still had good flavour after 3-3½ months. A 6-year experiment on the apple cultivars 'Golden Delicious',

**Table 4.2.** Effects of CA storage and time on the flavour of apricots which were assessed by a taste panel 3 months after being canned, where a score of 5 or more was acceptable (Truter *et al.*, 1994).

Storage period (weeks)	kPa CO <sub>2</sub>	kPa O <sub>2</sub>	Flavour score	
			Cultivar Bulida	Cultivar Peeka
0	-	-	7.5	4.6
4	1.5	1.5	6.3	6.3
5	1.5	1.5	5.0	5.0
6	1.5	1.5	1.3	1.3
4	5.0	2.0	6.3	5.0
5	5.0	2.0	5.5	5.4
6	5.0	2.0	2.9	5.4
4	0	21	6.7	5.8
5	0	21	4.2	5.4
6	0	21	2.1	4.6

'Auralia', 'Spartan' and 'Starkrimson' harvested at four different dates was carried out by Kluge and Meier (1979). These were stored at 3°C and 85-90% RH in air, 13 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> or 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. For prolonged storage, early harvesting and CA storage proved best in terms of flavour maintenance. There were interactions reported between storage conditions and cultivar on flavour. In a second experiment they stored 11 cultivars, including 'Golden Delicious', 'Jonathan', 'Gold Spur', 'Idared' and 'Boskoop'. For those stored in air only 'Boskoop' was considered to have had a good flavour after 180 days but both 'Gloster 69' and 'Jonathan' fruit also had a good flavour after CA storage for 180 days. Storage of 'Abbe Fetel' pears at -0.5°C under 0.5 kPa O<sub>2</sub> resulted in

losses in aroma and flavour and it was suggested that this may be overcome by raising the O<sub>2</sub> level to 1 kPa (Bertolini *et al.*, 1991). At 2°C sensory acceptance of peaches stored in air was lower than those stored in 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> for up to 15 days when they were ripened after storage. Higher acceptance scores were associated mainly with the perception of juiciness, emission of volatiles and TSS (Ortiz *et al.*, 2009). Meheriuk (1989a) found that storage of pears in CA did not have a deleterious effect on fruit flavour compared with those stored in air and were generally considered better by a taste panel, although those in CA generally had significantly higher acid levels (Table 4.3). Litchis were packed in punnets and stored in air or over-wrapped with plastic film with 10 kPa CO<sub>2</sub> or vacuum packed. They were then stored for 28 days at 1°C followed by 3 days of shelf-life at 20°C. It was found that the taste and flavour of the fruit from both the plastic film packs were unacceptable, while they remained acceptable for the non-wrapped fruits (Ahrens and Milne, 1993).

Some of the effects on flavour of fruit and vegetables during storage are the result of fermentation. Off-flavour development is associated with increased production of ethanol and acetaldehyde. For example, Mateos *et al.* (1993) reported that off-flavour developed in intact lettuce exposed to 20 kPa CO<sub>2</sub>

was associated with increased concentrations of ethanol and acetaldehyde. It was found by Magness and Diehl (1924) that the flavour of apples was impaired when CO<sub>2</sub> exceeded 20 kPa. Karaoulanis (1968) showed that oranges stored in 10–15 kPa CO<sub>2</sub> had increased alcohol content while those stored in 5 kPa CO<sub>2</sub> did not. He also showed that grapes stored in 12 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> + 67 kPa N<sub>2</sub> for 30 days had only 17 mg alcohol/100 g fresh weight while those stored in 25 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> + 54 kPa N<sub>2</sub> had 170 mg/100 g at the same time. Corrales-Garcia (1997) stored avocados at 2°C or 5°C for 30 days in air, 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 15 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. The fruits stored in air had a higher ethanol and acetaldehyde content than the fruits in CA. The alcohol content of strawberries increased with the length of storage and with higher concentrations of CO<sub>2</sub>. Storage in 20 kPa CO<sub>2</sub> resulted in high levels after 30 days (Woodward and Topping, 1972). Colelli and Martelli (1995) stored 'Pajaro' strawberries in air, or air with 10 kPa, 20 kPa or 30 kPa CO<sub>2</sub> for 5 days at 5°C, followed by an additional 4 days in air at the same temperature. Ethanol and acetaldehyde accumulation was very slight, though sensory evaluation of the fruits showed that off-flavours were present at transfer from CA, but not after the subsequent storage in air. With storage at 5°C for 28 days, CO<sub>2</sub> concentrations of melons

**Table 4.3.** Effect of controlled atmosphere storage on the acidity and firmness of Spartlett pears stored at 0°C and their taste after 7 days of subsequent ripening at 20°C (adapted from Meheriuk, 1989a).

Storage atmosphere			Malic acid mg/100 ml	Taste (score 1 = like, 9 = dislike)	Firmness (Newtons)
CO <sub>2</sub> kPa	O <sub>2</sub> kPa	Storage time (days)			
0	21	60	450b	6.1	62a
5	3	60	513a	7.6	61ab
2	2	60	478ab	7.6	60b
0	21	120	382b	4.3	55b
5	3	120	488a	6.7	58a
2	2	120	489a	6.9	58a
0	21	150	342b	2.9	51b
5	3	150	451a	7.7	58a
2	2	150	455a	7.1	58a
0	21	180	318b	–	40b
5	3	180	475a	–	62a
2	2	180	468a	–	61a

Figures followed by the same letter were not significantly different ( $p = 0.05$ ) by the Duncan's Multiple Range Test.

in sealed polyethylene (PE) and perforated PE packs were 1-4% and no off-flavours were found (Zhao and Murata, 1988). Storage of snow pea pods in either 2.5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> resulted in the development of slight off-flavours, but this effect was reversible since it was partially alleviated after ventilation (Pariasca *et al.*, 2001). Delate and Brecht (1989) showed that exposure of sweet potatoes to 2 kPa O<sub>2</sub> + 60 kPa CO<sub>2</sub> resulted in less sweet potato flavour and more off-flavour.

Low O<sub>2</sub> levels have also been shown to result in fermentation and the production of off-flavours. Ke *et al.* (1991a) described experiments where 'Granny Smith' and 'Yellow Newtown' apples, '20th Century' pear and 'Angeleno' plums were kept in either 0.25 or 0.02 kPa O<sub>2</sub> with the consequent development of an alcoholic off-flavour. Ke *et al.* (1991b) also stored the peach cultivar 'Fairtime' in air or in 0.25 or 0.02 kPa O<sub>2</sub> at 0°C or 5°C for up to 40 days. They found that flavour was affected by ethanol and that acetaldehyde accumulated in 0.02 kPa O<sub>2</sub> at 0°C or 5°C or in 0.25 kPa O<sub>2</sub> at 5°C. The fruits kept in air or 0.25 kPa O<sub>2</sub> at 0°C for up to 40 days and those stored in 0.02 kPa O<sub>2</sub> at 0°C or in air, 0.25 or 0.02 kPa O<sub>2</sub> at 5°C for up to 14 days had good to excellent taste, but the flavour of the fruits stored at 5°C for 29 days was unacceptable. Mattheis *et al.* (1991) stored 'Delicious' apples in 0.05 kPa O<sub>2</sub> + 0.2 kPa CO<sub>2</sub> at 1°C for 30 days and found that they developed high concentrations of ethanol and acetaldehyde; these included ethyl propanoate, ethyl butyrate, ethyl 2-methylbutyrate, ethyl hexanoate, ethyl heptanoate and ethyl octanoate. The increase in the emission of these compounds was accompanied by a decrease in the amounts of other esters requiring the same carboxylic acid group for synthesis. In contrast, plums were shown to tolerate an almost complete absence of O<sub>2</sub> for a considerable period without developing an alcoholic or unpleasant flavour (Food Investigation Board, 1920). Ke *et al.* (1991a) reported that strawberries could be stored in 1.0, 0.5 or 0.25 kPa O<sub>2</sub> or air plus 20 kPa CO<sub>2</sub> at 0°C or 5°C for 10 days without detrimental effects on their eating quality. Fruit could also be stored in 0 kPa O<sub>2</sub> or 50 kPa or

80 kPa CO<sub>2</sub> for up to 6 days without affecting their appearance. The taste panel found slight off-flavour in the cultivar G3 kept in 0 or 0.25 kPa O<sub>2</sub> which correlated with ethanol, ethyl acetate and acetaldehyde in juice. Transfer of fruit to air at 0°C for several days after treatment reduced ethanol and acetaldehyde levels, leading to an improvement in final sensory quality.

## Volatile Compounds

CA storage has also been shown to affect volatile compounds that are produced by fruit and give the characteristic flavour and aroma. There is some indication that CA storage can improve flavour development in grapes. Dourtoglou *et al.* (1994) found grapes exposed to 100 kPa CO<sub>2</sub> for 20 h had 114 volatile compounds after subsequent storage in air, compared with only 60 for those stored in air throughout. Forney *et al.* (2015) showed that the majority of fruit volatiles in raspberries were composed of C13 norisoprenoids (13-carbon butene cyclohexene degradation products formed by the cleavage of carotenoids), which totalled 65–93% of their total volatile content, and monoterpenes comprised 2.6–20.2% of total volatiles. They reported no effects of CA storage on these volatiles when compared with storage in air.

The onion cultivar 'Hysam' was stored at 0.5 ± 0.5°C and 80 ± 3% RH for 9 weeks in air, 2 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub>. Both the CA levels resulted in reduced pungency and flavour, compared with those stored in air, which was assumed to be due to the reduction both in flavour precursors and in enzyme activity (Uddin and MacTavish, 2003).

Willaert *et al.* (1983) isolated 24 aroma compounds from 'Golden Delicious' apples and showed that the relative amounts of 18 of these components declined considerably during CA storage. CA storage of apples in either 2 kPa O<sub>2</sub> + 98 kPa N<sub>2</sub> or 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> + 93 kPa N<sub>2</sub> resulted in few organic volatile compounds being produced during storage (Hatfield and Patterson, 1974). Even

when the fruits were removed from storage they did not synthesize normal amounts of esters during ripening. Yahia (1989) showed that 'McIntosh' and 'Cortland' apples stored in 100 kPa O<sub>2</sub> at 3.3°C for 4 weeks did not have enhanced production of aroma volatiles compared with those stored in air at the same temperature. He showed that most organic volatile compounds were produced at lower rates during ripening after CA storage than those produced from fruits ripened immediately after harvest. Hatfield (1975) also showed that 'Cox's Orange Pippin' apples, after storage in 2 kPa O<sub>2</sub> at 3°C for 3½ months and subsequent ripening at 20°C, produced lower amounts of volatile esters than they did when ripened directly after harvest or after storage in air. This was correlated with a marked loss of flavour. They showed that the inhibition of volatile production could be relieved considerably if the apples were kept in air at 5–15°C after storage and before being transferred to 20°C. Hansen *et al.* (1992) found that volatile ester production by 'Jonagold' apples was reduced after prolonged CA storage in low O<sub>2</sub>. After removal from storage, large differences were seen in the production of esters. Esters with the alcohol 2-methylbut-2-enol were produced in negative correlation to increased O<sub>2</sub> concentration. Aroma volatiles in 'Golden Delicious' apples were suppressed during storage for up to 10 months at 1°C in CA compared with storage in air (Brackmann, 1989). 'Jonagold' apples were stored at 0°C in air or in 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> for 6 months by Girard and Lau (1995). They found that CA storage decreased production of esters, alcohols and hydrocarbons by about half. The measurement of the organic volatile production of the apples was over a 10-day period at 20°C after removal from the cold store. In CA storage of 'Gala Must' apples, there was some slight reduction in volatile production compared with fruit stored in air. Similar results were obtained from sensory evaluation of aroma intensity after a further 1 week in air (Miszczak and Szymczak, 2000). Miszczak and Szymczak (2000) found that storage of 'Jonagold' at 0°C under 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> for 6 months decreased production of volatile compounds by half. Girard and Lau

(1995) also found that at 1°C in 1 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> there was some slight reduction in volatile production compared with fruit stored in air. Harb *et al.* (1994) found that storage of 'Golden Delicious' apples in 3 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, 1 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> or 1 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> suppressed volatile production but fruit stored in 1 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> had higher volatile production. Belay *et al.* (2017) compared the storage of pomegranate arils at 5°C for 12 days in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> + 85 kPa N<sub>2</sub>, 10 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> + 85 kPa N<sub>2</sub>, 70 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> + 85 kPa N<sub>2</sub> or air. They found that the highest level of volatile organic compounds was in arils stored under 70 kPa O<sub>2</sub>.

Harb *et al.* (2008b) found that the biosynthesis of volatiles in apples was greatly reduced following ULO storage, especially after an extended storage period. Treating fruit with volatile precursors (alcohols and aldehydes) stimulated the biosynthesis of the corresponding volatiles, mainly esters, but this effect was transitory with fruit that had been stored under ULO. Brackmann *et al.* (1993) found that the largest reduction in aroma production was in ULO at 1 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. However, there was a partial recovery when they were subsequently stored for 14 days in air at 1°C.

## Acidity

The acid levels in fruits and vegetables can obviously affect their flavour and acceptability. Knee and Sharples (1979) reported that acidity could fall by as much as 50% during storage of apples and that there was a good correlation between fruit acidity and sensory evaluation. Meheriuk (1989a) found that storage of pears in CA generally resulted in significantly higher acid levels than those stored in air. Girard and Lau (1995) also found that CA storage significantly reduced the loss of acidity in apples. Similarly Kollas (1964) found that the titratable acid of 'McIntosh' apples was much greater for fruits at 3.3°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> than those stored in air at 0°C and

concluded that it was likely to be due to lower oxidation but that significant rates of 'CO<sub>2</sub> fixation' may have occurred. Kays (1997) described an experiment where 'Valencia' oranges lost less acid during storage at 3.5°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> than in air at 0°C.

Batu (1995) also confirmed changes in acidity in relation to CA storage and MA packaging in tomatoes. Those stored in sealed plastic film bags generally had a lower rate of loss in acidity than those stored without film bags and generally the fruit stored in the less permeable films had a lower rate of acid loss. The acidity levels of fruit sealed in polypropylene (PP) film, where the equilibrium atmosphere was 5–8 kPa O<sub>2</sub> + 11–13 kPa CO<sub>2</sub>, were similar to fruit stored in air and both were lower than the fruit sealed in 30 or 50 µm PE film. He also showed that TA of tomatoes in CA storage generally increased during the first 20 days at both 13°C and 15°C (Figs 4.1 and 4.2). After 20 days, acidity levels tended to decrease until about 70 days. Although there was no correlation

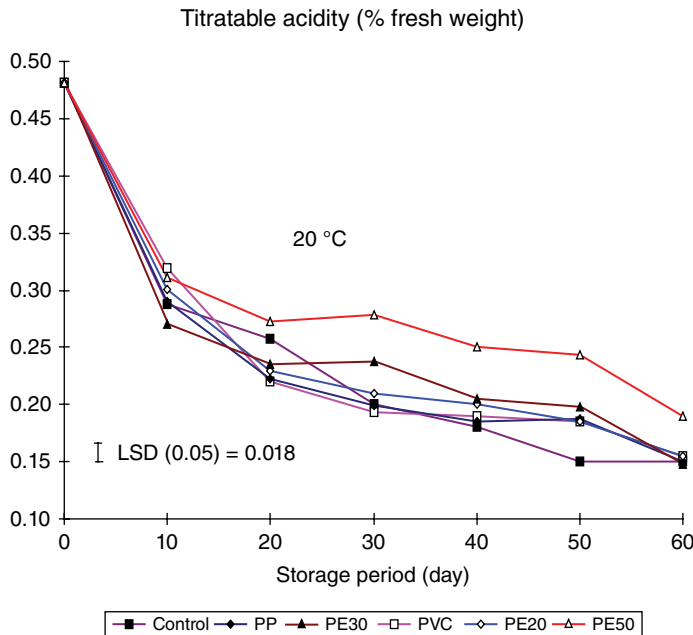
between the O<sub>2</sub> or CO<sub>2</sub> concentrations and acidity levels of fruits during CA storage, the acidity of tomatoes stored in 6.4 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> were the highest among the treatments and the lowest was in 9.1 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub>.

## Nutrition

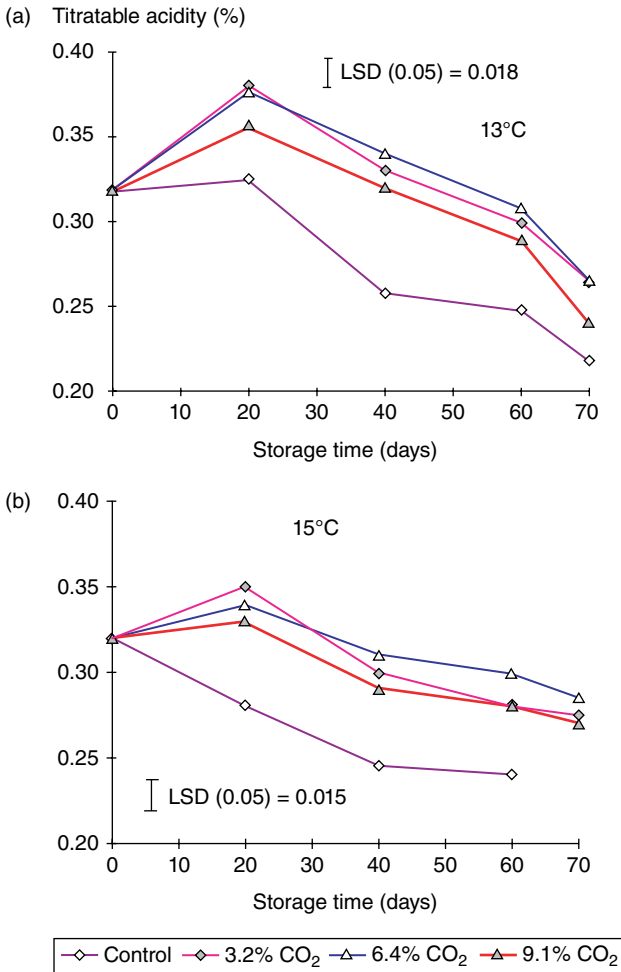
CA storage has been shown to have both positive and negative effects on the synthesis and retention of the phytonutrients required for human nutrition. The following are some examples reported of the different effects.

## Ascorbic acid

Storage of 'Conference' pears in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> resulted in 60% loss in AA content (Veltman *et al.*, 1999). The AA content



**Fig. 4.1.** Changes of titratable acidity values of tomatoes harvested at pink stage of maturity and sealed in various thickness of various packaging films and stored at 20°C (modified from Batu, 1995). *Control*: unwrapped; *PE20*: sealed with 20 µm polyethylene; *PE30*: sealed with 30 µm polyethylene; *PE50*: sealed with 50 µm polyethylene; *PVC*: sealed with 10 µm polyvinyl chloride; *PP*: sealed with 25 µm polypropylene.



**Fig. 4.2.** Changes of titratable acidity values of tomatoes harvested at mature-green stage of maturity and kept in controlled atmosphere storage (all with 5.5% O<sub>2</sub>) for 60 days storage time at (a) 13°C and (b) 15°C, plus 10 days in air at 20°C (modified from Batu, 1995).

of the tomato cultivars 'Punjab Chuhara' and 'Punjab Kesri' decreased as the CO<sub>2</sub> concentration in the storage atmosphere increased and increased as the storage period was lengthened (Singh *et al.*, 1993). 'Elvira' strawberries and 'Thornfree' blackberries were stored at 0–1°C in up to 20 kPa CO<sub>2</sub> for strawberries and 30 kPa CO<sub>2</sub> for blackberries combined with either 1–3 kPa or > 14 kPa O<sub>2</sub>. Loss of AA was highest in the higher CO<sub>2</sub> atmosphere and degradation after 20 days of storage was more rapid with the low O<sub>2</sub> concentrations. This degradation was even more pronounced during simulated shelf-life in ambient conditions following storage (Agar *et al.*, 1994a). Wang (1983) showed that AA levels of Chinese cabbage were not

affected in storage in 10 or 20 kPa CO<sub>2</sub>, but in 30 or 40 kPa CO<sub>2</sub> the rate of loss of AA was much higher than for cabbage stored in air. Ogata *et al.* (1975) stored okra at 1°C in air or 3 kPa O<sub>2</sub> combined with 3, 10 or 20 kPa CO<sub>2</sub> or at 12°C in air or 3 kPa O<sub>2</sub> combined with 3 kPa CO<sub>2</sub>. At 1°C there was no effect of any of the CA storage combinations on AA, but at 12°C CA resulted in lower AA retention. Ito *et al.* (1974) found that storage of satsumas at 1–4°C in low O<sub>2</sub> and high CO<sub>2</sub> resulted in the AA content of the flesh and peel gradually declining but the dehydroascorbic acid content increasing. However, they found that such changes were smaller at high O<sub>2</sub> levels. Fruit held in CA storage lost more acid and sugars than those held in

air. Haruenkit and Thompson (1996) showed that storage of pineapples in O<sub>2</sub> levels below 5.4% helped to retain AA levels but generally had little effect on TSS (Table 4.4). Kurki (1979) showed some greater loss in AA in CA storage compared with storage in air, but CA storage gave much better retention of AA than storage in air (Table 4.5). Trierweiler *et al.* (2004) found that the AA content was reduced during storage for 7 months in air or 3 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> in 'Bohnapfel' apples but total antioxidant capacity remained constant.

In contrast, Vidigal *et al.* (1979) in studies of tomatoes found that the AA levels increased during CA storage at 10°C. Storage of peas in plastic film bags at 5°C had an equilibrium atmosphere of 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, which resulted in better maintenance of AA, compared with those not stored in bags (Pariasca *et al.*, 2001). Serrano *et al.* (2006) found that broccoli in MA packaging film had half the loss of AA compared with those stored non-wrapped during 21 days. The AA content of fresh-cut kiwifruit at 0°C under 0.5, 2 or 4 kPa O<sub>2</sub> decreased by 7%, 12% or 18%, respectively, after 12 days while

those in air, 5, 10, or 20 kPa CO<sub>2</sub> decreased by 14%, 22% or 34%, respectively. Generally, high CO<sub>2</sub> concentration in the storage atmosphere caused degradation of vitamin C in fresh-cut kiwifruit slices (Lee and Kader, 2000).

## Chlorophyll

Ubiquitous in plants for photosynthesis, chlorophylls are also an important phytonutrient. For example, Ferruzzi *et al.* (2001) demonstrated the uptake of chlorophyll derivatives by human intestinal cells and their potential importance as health-promoting phytochemicals. Knee and Sharples (1979) found that chlorophyll content of 'Cox's Orange Pippin' apples stored in CA remained the same or only reduced slightly over a 6-month storage period. Storage of peas in plastic film bags at 5°C gave an equilibrium atmosphere of 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> that resulted in better retention of chlorophyll compared with those not stored in bags (Pariasca *et al.*, 2001). Bangerth (1974) showed that hypobaric storage at 2–3°C improved retention of chlorophyll in stored parsley.

## Lycopene

Lycopene consumption is important in human health and its antioxidant function has been shown to offer protection against lung, colorectal, breast, uterine and prostate cancers. Lycopene is a carotenoid in the same family as β-carotene and gives a red colour to fruit. It occurs in many fruit, including tomatoes, pink grapefruit, apricots, blood oranges, watermelon, rose hips and

**Table 4.4.** Effects of CA storage on ascorbic acid, °brix and acidity (w/v) of Smooth Cayenne pineapples stored at 8°C for 3 weeks and then 5 days at 20°C (Haruenkit and Thompson, 1996).

Composition (kPa)			Ascorbic acid mg/100 ml
O <sub>2</sub>	CO <sub>2</sub>	N <sub>2</sub>	
1.3	0	98.7	7.14
2.2	0	97.8	8.44
5.4	0	94.6	0.76
1.4	11.2	87.4	9.16
2.3	11.2	86.5	7.94
20.8	0	79.2	0.63
LSD ( $p = 0.05$ )			5.28

**Table 4.5.** Effects of CA storage and storage in air on the retention of vitamins in leeks at 0°C and 95% RH (Kurki, 1979).

	Initial level before storage	After 4 months storage	
		Air	10 kPa CO <sub>2</sub> + 1 kPa O <sub>2</sub>
Ascorbic acid (mg/100 g)	37.2	24.1	20.2
Vitamin A (IU/100g)	2525	62	1350



guavas. Plant breeders have developed high-pigment hybrid processing tomatoes that yield improved lycopene, which can constitute 80–90% of the total pigments present. Lycopene in fresh tomato fruits occurs essentially in the *all-trans* configuration (Shi and Le Maguer, 2000). Isomerization converts *all-trans* isomers to *cis*-isomers due to additional energy input and results in an unstable, energy-rich station. Lycopene in tomatoes was shown to increase during storage at 8°C or 16°C while at lower temperatures there was no change or a slight decrease. The optimum temperature for lycopene synthesis in tomatoes is 24°C. Lycopene is not formed at 30°C and above, and temperatures greater than 32.2°C during the growing season resulted in lower lycopene concentrations in the fruit (Garcia and Barrett, 2006). CA storage can also affect lycopene synthesis. Jeffery *et al.* (1984) showed that lycopene synthesis in tomatoes was suppressed during storage in 6 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub>, while tomatoes stored in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> had suppressed synthesis of lycopene and xanthophylls, as well as suppressed chlorophyll degradation (Goodenough and Thomas, 1981). In an extensive study of Californian processing tomatoes over 3 years, Garcia and Barrett (2006) found that the range of lycopene content was 55–181 mg/kg, depending on year, cultivar and growing conditions. The mean lycopene concentrations were 106 mg/kg in 2000, 101 mg/kg in 1999 and 88 mg/kg in 2001. During maturation of red cultivars there was a 10–14-fold increase in the concentration of carotenoids, mainly lycopene (Fraser *et al.*, 1994). In general, dehydrated and powdered tomatoes have poor lycopene stability unless carefully processed and promptly placed in hermetically sealed containers in an inert atmosphere for storage. Frozen foods and heat-sterilized foods exhibit excellent lycopene stability throughout normal temperature storage shelf-life. Farneti *et al.* (2012) reported a survey among consumers that indicated that tomatoes were most often stored in refrigerators at well below 10°C. They tested two cultivars ('Cocktail' and 'Round Type') at the red-ripe stage during 20 days storage at 4°C, 8°C, 12°C and 16°C

and found that temperatures below 12°C reduced their lycopene content. Application of ethylene to stored ripening tomatoes was shown to increase their lycopene content (Haard and Salunkhe, 1975). This appears to be additional to the increase that is normally associated with ripening.

## Phenolics

Rogiers and Knowles (2000) stored four cultivars of amelanchier (*Amelanchier alnifolia*) at 0.5°C for 56 days in various CA and found that 5 kPa CO<sub>2</sub> + 21 O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> were most effective at minimizing losses in anthocyanins. In blueberries, CA storage had little or no effect on phenolic content (Schotsmans *et al.*, 2007). Matityahu *et al.* (2016) showed that storage of pomegranates at 7°C in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> was better than air in retaining their quality, but the anthocyanin levels and off-flavours in the CA-stored arils was higher than those in air at the same temperature. However, Zheng *et al.* (2003) found that total phenolics increased in blueberries during storage at 5°C in 60–100 kPa O<sub>2</sub> for 35 days to a greater extent than those stored in air or 40 kPa O<sub>2</sub>. In grapes, anthocyanin levels were lower after storage in air at 0°C for those that had been pre-treated for 3 days in 20 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> compared with those that had not been pre-treated (Romero *et al.*, 2008). Artés-Hernández *et al.* (2006) stored 'Superior Seedless' grapes in MA packaging and found no changes in the phenolics during storage at 0°C and 8°C, but a slight decrease during 2 days of subsequent shelf-life at 20°C. However, Sanz *et al.* (1999) found that 'Camarosa' strawberries in MA packaging had lower anthocyanin levels compared with those stored non-wrapped. Shin *et al.* (2008) found that the total flavonoid and phenolic concentrations and total antioxidant activity of strawberries harvested at the white-tip stage were greater than those harvested at the red-ripe stage and these differences were maintained during storage at 3°C or 10°C for 12 days. The latex that spurts from the broken pedicel of immature mango

fruit is structurally similar to phenolic allergens found in other genera of Anacardiaceae. The main component of mango latex was identified as 5-[2(Z)-heptadecenyl] resorcinol (Bandyopadhyay *et al.*, 1985). Phenolic phytoalexins in carrots and parsnips accumulate in response to ethylene and give them a bitter flavour. In carrots, isocoumarin accumulation increased as respiration rate increased in response to ethylene and there was more accumulation of isocoumarin in immature carrots in response to ethylene than in more mature ones (Seljasen *et al.*, 2001).

Banana and plantain can contain high levels of phenolic compounds, especially in the peel (Von Loesecke, 1950). As fruit ripens the astringency becomes lower, which seems to be associated with a change in structure

of the tannins, rather than a reduction in their levels, in that they form polymers. The enzyme polyphenol oxidase (PPO) is responsible for the oxidative browning reaction when the pulp of fruit is cut. In 'Carabao' mangoes, there was a progressive loss in total phenolic content during ripening which was associated with loss in astringency (Tirtosoekotjo, 1984). In guava, levels of polyphenols, particularly the high-molecular-weight polyphenols, were shown to decrease as the fruit matured (Ito *et al.*, 1987). Young guava fruit contained 620 mg/100 g fresh weight, of which 65% were condensed tannins. In persimmon, the soluble tannin content in fruit varied between cultivars and in astringent cultivars it was particularly high in immature fruit (Ito *et al.*, 1987).

## 5

# Physiology, Ripening and Genetics

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### Respiration Rate

Within the physiological temperature range of the crop, the respiration rate increases exponentially with temperature, so that for every 10°C rise the increase in metabolism is in the order of two to three times. This is called the  $Q_{10}$  value and is predicated by the Van't Hoff rule. Respiration rate is also affected by the gaseous environment, especially  $O_2$ ,  $CO_2$  and ethylene. Respiratory quotient (RQ) is the ratio of moles (mol)  $CO_2$  evolved to moles  $O_2$  absorbed in plant cells. The value is 1 when the substrate is carbohydrate, about 0.7 for lipids and about 0.8 for proteins.

Generally fruit and vegetables have a lower respiration rate during CA storage compared with storage in air, but there are some exceptions. Most work on respiration rate has been reported on apples and Kidd and West (1927a, b) showed that the apple respiration rate in storage at 8°C in 12 kPa  $CO_2$  + 9 kPa  $O_2$  was 54–55% of that of fruit stored in air at the same temperature. Robinson *et al.* (1975) showed this lowering effect for a range of fresh fruit and vegetables during storage under 3%  $O_2$  at various temperatures (Table 5.1). They studied the respiration rate of a variety of crops at different temperatures stored in air or in 3 kPa  $O_2$  (Tables 5.2 and 5.3). Respiration rate was generally lower when crops were stored in

3 kPa  $O_2$  than when they were stored in air, but there were some interactions between  $O_2$  and temperature in that the effects of the reduced  $O_2$  level were more marked at higher storage temperatures. Exceptions are also common; for example, Fonseca *et al.* (2004) found that 'Royal Sweet' watermelons had a higher respiration rate at 8 kPa  $O_2$  compared with 14 kPa  $O_2$ . Respiratory heat at 'normal storage temperatures' in watts/tonne (W/t) was reported by Cambridge Refrigeration Technology (2001) as: avocado 183–465, asparagus 81–237, mango 133, banana 59–130, cauliflower 44, apple 9–18, grape 3.9–6.8, orange 18.9 and cherry 17–39.

Knee (1973) showed that  $CO_2$  could inhibit succinate dehydrogenase in the tricarboxylic acid cycle which is part of the respiratory pathway. McGlasson and Wills (1972) also suggested that low  $O_2$  limited the operation of the Krebs cycle between pyruvate and citrate and 2-oxoglutarate and succinate, but they apparently found no similar effect of high  $CO_2$  in bananas. The tricarboxylic acid cycle is inhibited at very low levels of  $O_2$  but the glycolytic pathway may not be affected. This results in less efficient energy production during respiration, since there is insufficient  $O_2$  to metabolize stored carbohydrates to water and  $CO_2$ . Instead the blocking of the glycolytic pathway results in a build-up of acetaldehyde and

**Table 5.1.** Effects of temperature and reduced O<sub>2</sub> level on the respiration rate (CO<sub>2</sub> production<sup>a</sup>, mg/kg/h<sup>b</sup>) and storage life of selected fruit and vegetables (Robinson *et al.*, 1975).

Temperature (°C)	In air					In 3 kPa O <sub>2</sub>			Water loss <sup>c</sup>
	0	5	10	15	20	0	10	20	
Asparagus	28	44	63	105	127	25	45	75	3.6 <sup>d</sup>
Beans, broad	35	52	87	120	145	40	55	80	(2.1)
Beans, runner	21	28	36	54	90	15	25	46	(1.8)
Beetroot, storing	4	7	11	17	19	6	7	10	1.6
Beetroot, bunching with leaves	11	14	22	25	40	7	14	32	(1.6)
Blackberries, Bedford Giant	22	33	62	75	155	15	50	125	0.5
Blackcurrants, Baldwin	16	27	39	90	130	12	30	74	–
Brussels sprouts	17	30	50	75	90	14	35	70	(2.8)
Cabbage, Primo	11	26	30	37	40	8	15	30	1.0
Cabbage, January King	6	13	26	33	57	6	18	28	–
Cabbage, Decema	3	7	8	13	20	2	6	12	0.1
Carrots, storing	13	17	19	24	33	7	11	25	1.9
Carrots, bunching with leaves	35	51	74	106	121	28	54	85	(2.8)
Calabrese	42	58	105	200	240	–	70	120	(2.4)
Cauliflower, April Glory	20	34	45	67	126	14	45	60	(1.9)
Celery, white	7	9	12	23	33	5	9	22	(1.8)
Cucumber	6	8	13	14	15	5	8	10	(0.4)
Gooseberries, Leveller	10	13	23	40	58	7	16	26	–
Leeks, Musselburgh	20	28	50	75	110	10	30	57	(0.9)
Lettuce, Unrivalled	18	22	26	50	85	15	20	55	(7.5)
Lettuce, Kordaat	9	11	17	26	37	7	12	25	–
Lettuce, Kloek	16	24	31	50	80	15	25	45	–
Onion, Bedfordshire Champion	3	5	7	7	8	2	4	4	0.02
Parsnip, Hollow Crown	7	11	26	33	49	6	12	30	(2.4)
Potato, maincrop (King Edward)	6 <sup>e</sup>	3	4	5	6	5 <sup>s</sup>	3	4	(0.05)
Potato, 'new' (immature) <sup>f</sup>	10	15	20	30	40	10	18	30	(0.5)
Pea (in pod) early, (Kelvedon Wonder)	40	61	130	180	255	29	84	160	(1.3) (cv. Onward)
Peas, main crop (Dark Green Perfection)	47	55	120	170	250	45	60	160	
Peppers, green	8	11	20	22	35	9	14	17	0.6
Raspberry (Malling Jewel)	24	55	92	135	200	22	56	130	2.5

*Continued*

Table 5.1. Continued.

Temperature (°C)	In air					In 3 kPa O <sub>2</sub>			Water loss <sup>c</sup>
	0	5	10	15	20	0	10	20	
Rhubarb (forced)	14	21	35	44	54	11	20	42	2.3
Spinach (Prickly True)	50	70	80	120	150	51	87	137	(11.0) (glasshouse grown)
Sprouting broccoli	77	120	170	275	425	65	115	215	(7.5)
Strawberries, Cambridge Favourite	15	28	52	83	127	12	45	86	(0.7)
Sweetcorn	31	55	90	142	210	27	60	120	(1.4)
Tomato, Eurocross BB	6	9	15	23	30	4	6	12	0.1
Turnip, bunching with leaves	15	17	30	43	52	10	19	39	1.1 (without leaves)
Watercress	18	36	80	136	207	19	72	168	(35.0)

<sup>a</sup>These figures, which give the average rates of respiration of the samples, are a guide only. Other samples could differ, but the rates could be expected to be of the same order of magnitude, ± about 20 %.

<sup>b</sup>Heat production in Btu/t/h is given by multiplying CO<sub>2</sub> output in mg/kg/h by 10 (1 Btu approximates to 0.25 kcal and to 1.05 kJ).

<sup>c</sup>Values in parentheses were determined at 15°C and 6–9 mb water vapour pressure deficit (wvpd). Remainder at 10°C and 3–5 mb wvpd.

<sup>d</sup>Water loss as a % of the initial weight per day per millibar wvpd.

<sup>e</sup>After storage for a few weeks at 0°C, sufficient time having elapsed for low-temperature sweetening. The figures for potato are mid-season (December) values and could be 50 % greater in October and March.

<sup>f</sup>Typical rounded-off values, for tubers with an average weight of 60 g immediately after harvest. The rates are too labile for individual values to be meaningful. After a few weeks' storage they approximate to the values for maincrop potatoes.

<sup>g</sup>After storage for a few weeks at 0°C, sufficient time having elapsed for low-temperature sweetening. The figures for potatoes are mid-season (December values) and could be 50 % greater in October and March.

**Table 5.2.** Effects of temperature and reduced O<sub>2</sub> level on the respiration rate in CO<sub>2</sub> production in mg/kg/h of various fruit and vegetables (adapted from Robinson *et al.*, 1975).

Crop	Gaseous atmosphere	Storage temperature		
		0°C	10°C	20°C
Asparagus	Air	28	63	127
	3 kPa O <sub>2</sub>	25	45	75
Beetroot without leaves	Air	4	11	19
	3 kPa O <sub>2</sub>	6	7	10
Beetroot with leaves	Air	11	22	40
	3 kPa O <sub>2</sub>	7	14	32
Blackberries	Air	22	62	155
	3 kPa O <sub>2</sub>	15	50	125
Blackcurrants	Air	16	39	130
	3 kPa O <sub>2</sub>	12	30	74
Broad beans	Air	35	87	145
	3 kPa O <sub>2</sub>	40	55	80
Sprouting broccoli	Air	77	170	425
	3 kPa O <sub>2</sub>	65	115	215
Brussels sprouts	Air	17	50	90
	3 kPa O <sub>2</sub>	14	35	70
Green peppers	Air	8	20	35
	3 kPa O <sub>2</sub>	9	14	17
Carrots	Air	13	19	33
	3 kPa O <sub>2</sub>	7	11	25
Cauliflowers	Air	20	45	126
	3 kPa O <sub>2</sub>	14	45	60
Cucumbers	Air	6	13	15
	3 kPa O <sub>2</sub>	5	8	10
Gooseberry	Air	10	23	58
	3 kPa O <sub>2</sub>	7	16	26
Onions	Air	3	7	8
	3 kPa O <sub>2</sub>	2	4	4
Strawberry	Air	15	52	127
	3 kPa O <sub>2</sub>	12	45	86
Sweetcorn	Air	31	90	210
	3 kPa O <sub>2</sub>	27	60	120
Tomato	Air	6	15	30
	3 kPa O <sub>2</sub>	4	6	12

ethanol, which can be toxic to the cells if allowed to accumulate. Compared with storage in air, storing broccoli in 50 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> led to a decreased respiration rate, inhibition of the reduction of ATP levels, energy charge levels, succinic dehydrogenase and cytochrome oxidase and increases in glucose-6-phosphate dehydrogenase and 6-phosphogluconate dehydrogenase activity. Storage in 50 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> also suppressed reduction of the tricarboxylic acid cycle and cytochrome oxidase pathway rate and increased the phosphopentose pathway, but had little impact on the

Embden–Meyerhof–Parnas rate and phosphohexose isomerase activity (Li *et al.*, 2016c).

The effect of reduced O<sub>2</sub> and increased CO<sub>2</sub> on respiration rate can be affected by temperature. Kubo *et al.* (1989b) showed that the effects of high CO<sub>2</sub> on respiration rate varied according to the crop and its stage of development. The relationship between O<sub>2</sub> and CO<sub>2</sub> levels and respiration rate is not a simple one. It also varies between cultivars (Table 5.4) as in the comparison between ‘Golden Delicious’ and ‘Cox’s Orange Pippin’ apples where the respiration rate

**Table 5.3.** Effects of temperature and reduced O<sub>2</sub> level on the respiration rate (CO<sub>2</sub> production in mg/kg/h) of potatoes (adapted from Robinson *et al.*, 1975).

Temperature (°C)	Air					3 kPa O <sub>2</sub>		
	0	5	10	15	20	0	10	20
Main crop (King Edward)	6	3	4	5	6	5	3	4

**Table 5.4.** Rates of respiration rate of some cultivars of apples stored in different CA conditions; modified from Fidler *et al.* (1973).

Cultivar	Storage conditions			Respiration rate (l/t/day)	
	°C	CO <sub>2</sub> kPa	O <sub>2</sub> kPa	CO <sub>2</sub>	O <sub>2</sub>
Bramley	3.5	8–10	11–13	40–45	40
Cox's Orange Pippin	3.5	5	16	62	57
Cox's Orange Pippin	3.5	5	3	42	40
Cox's Orange Pippin	3.5	< 1	2.5	80	55
Golden Delicious	3.5	5	3	20	20
Delicious	0	5	3	18	–
Jonathan	3.5	7	13	33	38
McIntosh	3.5	5	3	35	–

was suppressed more in the former than the latter in CA storage (Fidler *et al.*, 1973). Olsen (1986) also showed an interaction between O<sub>2</sub> and CO<sub>2</sub> on the respiration rate of apples, with the strongest effect of O<sub>2</sub> at 0 kPa CO<sub>2</sub> and only a small effect of O<sub>2</sub> at 1 kPa CO<sub>2</sub>. Andrich *et al.* (1994) showed some evidence that the respiration rate of 'Golden Delicious' apples was affected differentially by CO<sub>2</sub> concentration in storage at 21°C and 85% RH, depending on the O<sub>2</sub> concentration in the store. The effect was that in anaerobic conditions, or near-anaerobic conditions, respiration rates were more affected in increasingly high CO<sub>2</sub> levels than for the same fruit stored in aerobic conditions. When apples and melons were stored in 60 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> + 20 kPa N<sub>2</sub> their respiration rate fell to about half the initial level. Ripening tomatoes and bananas also showed a reduction in respiration in response to high CO<sub>2</sub>, but showed little response when tested before the climacteric (Kubo *et al.*, 1989a).

Where the O<sub>2</sub> level in plant cells is too low, the respiration rate becomes anaerobic. In some CA stores in England where they store apples in 1 kPa O<sub>2</sub>, an alcohol detector is fitted which sounds an alarm if ethanol fumes are detected because of

fermentation. This enables the store operator to increase the O<sub>2</sub> level and no damage should have been done to the fruit. The detector technology is based on that used by the police to detect alcohol fumes on the breath of motorists. The level of O<sub>2</sub> at which fermentation may occur in plant cells may be as low as 0.2 kPa, but the gradient of O<sub>2</sub> concentration between the store atmosphere and the cells of the crop requires the maintenance of a higher level around the crop (Kader, 1986). The gradient is affected by the cuticle or periderm of the crop, the ease of gas penetration through the cells and the utilization of O<sub>2</sub> by the crop.

Production of ethylene by guavas and tomatoes was substantially reduced by all levels of CO<sub>2</sub>, but 30 kPa CO<sub>2</sub> accelerated ethylene production in bananas, carrots, cucumbers, onions and potatoes, possibly due to an injury response (Pal and Buescher, 1993). Green bananas were held in humidified gas streams comprising air, 5 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> + 75 kPa N<sub>2</sub>, 0 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> + 97 kPa N<sub>2</sub> or 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> + 92 kPa N<sub>2</sub>. Ripening in all three CA combinations and the respiration rates were reduced over the period before the beginning of the climacteric compared with storage in air (McGlasson

and Wills, 1972). Wade (1974) showed that the respiratory climacteric was induced in bananas by Ethephon (a source of ethylene) at O<sub>2</sub> concentrations of 3–21 kPa, but their respiration rate was not affected by Ethephon at O<sub>2</sub> concentrations of 1 kPa or less. Awad *et al.* (1975) showed that green bananas immersed for 2 min in Ethephon at 500 ppm had their climacteric advanced by 5 days, whereas fruits treated with GA at 100 ppm had their climacteric delayed by 2 days, compared with the control.

CA can also affect the respiration rate of vegetables and non-climacteric fruit. Kubo *et al.* (1989a) reported that exposure to high CO<sub>2</sub> had little or no effect in *Citrus natsudaidai*, lemons, potatoes, sweet potatoes or cabbage, but reduced the respiration rate in broccoli. High CO<sub>2</sub> stimulated the respiration rate of lettuce, aubergine and cucumber. Pal and Buescher (1993) found that short-term exposure to 20–30 kPa CO<sub>2</sub> reduced the respiration rate of pickling cucumbers but could increase the respiration rate of potatoes and carrots, and had no effect on respiration rate of oranges and onions. They found that changes in respiration rate seldom coincided with changes in ethylene production. The respiration rate of the strawberry cultivar ‘Cambridge Favourite’ held at 4.5°C in air or 1, 2 or 5 kPa O<sub>2</sub> fell to a minimum after 5 days. Thereafter the rate increased, more rapidly in air (in which rotting was more prevalent) than in 1 or 2 kPa O<sub>2</sub> (Woodward and Topping, 1972). Storage of strawberries in 10–30 kPa CO<sub>2</sub> or 0.5–2 kPa O<sub>2</sub> was shown to slow their respiration rate (Hardenburg *et al.*, 1990). Weichmann (1973) showed that the respiration rate of five cultivars of carrots stored at 1.5°C in atmospheres containing 0.03, 2.5, 5.0 or 7.5 kPa CO<sub>2</sub>, but with no regulation of the O<sub>2</sub> level, increased with increasing CO<sub>2</sub> level. Weichmann (1981) found that horseradish stored in 7.5 kPa CO<sub>2</sub> had a higher respiration rate than those stored in air. Izumi *et al.* (1996b) showed that storage of freshly cut carrots in 10 kPa CO<sub>2</sub> + 0.5 kPa O<sub>2</sub> reduced their respiration rate by about 55% at 0°C, about 65% at 5°C and 75% at 10°C.

In summary, it has been shown that low O<sub>2</sub> levels in store can affect the crop in

several ways (Richardson and Meheriuk, 1982), including:

- reduced respiration rate;
- reduced substrate oxidation;
- delayed ripening of climacteric fruit;
- prolonged storage life;
- delayed breakdown of chlorophyll;
- reduced rate of production of ethylene;
- changed fatty acid synthesis;
- reduced degradation rate of soluble pectins;
- formation of undesirable flavours and odours;
- altered texture; and
- development of physiological disorders.

The effects of CO<sub>2</sub> on postharvest responses of crops include:

- decreased synthetic reactions in climacteric fruit;
- delaying the initiation of ripening;
- inhibition of some enzyme reactions;
- decreased production of some organic volatiles;
- modified metabolism of some organic acids;
- reducing the rate of breakdown of pectic substances;
- inhibition of chlorophyll breakdown;
- production of ‘off-flavours’;
- induction of physiological disorders;
- retarded fungal growth on the crop;
- inhibition of the effect of ethylene;
- changes in sugar content (potatoes);
- effects on sprouting (potatoes);
- inhibition of postharvest development;
- retention of tenderness; and
- decreased discoloration levels.

## Fruit Ripening

Fruit can be classified into two groups, climacteric and non-climacteric, though some fruits appear to be intermediate between the two. Climacteric can be defined as fruit that can be ripened after harvest and non-climacteric as fruit that do not ripen after harvest. The term climacteric was first applied to fruit ripening in the mid 1920s by Kidd and West



(1927b). They observed an increase in the respiration rate as measured by the production of CO<sub>2</sub> of Bramley's Seedling apples around the time of the normal commercial harvest. Biale and Barcus (1970) published measurements of the respiration rate of some selected fruits and classified them into climacteric, non-climacteric and indeterminate on the basis of their respiration rate. For example, they classified banana, biriba, breadfruit, mango, papaya and soursop as climacteric; cacao, cashew and guava as non-climacteric; and jackfruit, jambo, passionfruit and sapote as indeterminate. Other fruits that have been shown to display typical climacteric respiration patterns include annonas, apricot, avocado, blueberry, cantaloupe melon, durian, feijoa, kiwifruit, peach, persimmon, plum, pome fruits and watermelon. Other fruits that can be classified as non-climacteric include aubergine, blackberry, capsicum, cherry, citrus fruits, cucumber, grape, litchi, olive, pineapple, pomegranate, raspberry, strawberry and tamarillo. In some cases there is conflict of opinion on classification. For example, fig and melon were classified as climacteric by Wills *et al.* (1989), but Biale (1960) classified them both as non-climacteric and Pratt (1971) showed that both honeydew muskmelon and 'PMR-45' cantaloupes were climacteric. Biale and Barcus (1970) classified passionfruit as indeterminate while Akamine *et al.* (1957) classified them as climacteric. Grierson (1993), commenting on the classification of fruit into climacteric and non-climacteric, reported that parts of citrus fruits are climacteric and other parts of the same fruit are non-climacteric.

### Initiation of ripening

In climacteric fruit, ripening is initiated when a threshold level of ethylene is reached in the cells. Ethylene biosynthesis involves a series of steps culminating in the synthesis of *S*-adenosyl-methionine (SAM) from the amino acid methionine. SAM is converted to ACC (1-aminocyclopropane-1-carboxylic acid) by the action of the enzyme

ACC synthase and ACC is converted to ethylene by the action of ACC oxidase, which is sensitive to O<sub>2</sub>. The synthesis of ACC was reported to be the rate-limiting step in the biosynthesis of ethylene (Kende, 1993). ACC synthase is controlled by a number of ACS genes (Li *et al.*, 2013). The biosynthesis of ethylene can be inhibited at temperatures above 35°C. High levels of CO<sub>2</sub> can compete with ethylene for binding sites in fruit and the biological activity of ethylene at 1 µl/l ceased in the presence of 10 kPa CO<sub>2</sub> (Burg and Burg, 1967). Yang (1985) showed that CO<sub>2</sub> accumulation in the intercellular spaces of fruit acts as an ethylene antagonist. The inhibition of ethylene production by CO<sub>2</sub> may be attributed to its inhibitory effect on ACC synthase activity, but ACC oxidase activity was stimulated by CO<sub>2</sub> up to 10 kPa but inhibited by 20 kPa CO<sub>2</sub> (Levin *et al.*, 1992). Koyakumar *et al.* (1995) also found that ethylene production rates were suppressed at high CO<sub>2</sub> concentrations or at O<sub>2</sub> levels of less than 5 kPa in the fruit of *Prunus mume* (commonly known as Japanese apricot, or Chinese plum). In 3 kPa O<sub>2</sub>, ethylene production rates peaked half or one day after the respiratory climacteric peak. Ethylene production in ripening fruit was shown to stop in the absence of O<sub>2</sub> (Gane, 1934). Treating fruit with chemicals, including 1-MCP and AVG, can inhibit ACC synthase activity. Han *et al.* (2016) identified three different ripening characteristics caused by ethylene that correlated well with ethylene production. Gene expression analysis showed a significant induction of *MaC2H2-1/2* transcripts during the ripening of bananas and they suggested that 'these genes are transcriptional repressors and may mediate a finely tuned regulation of ethylene production during ripening, possibly via transcriptional repression of ethylene biosynthetic genes'. Clendennen and May (1997) found that differential screening of complementary DNA libraries in banana pulp during ripening led to the isolation of 11 non-redundant groups of differentially expressed mRNA. Identification of these transcripts indicated that two of the mRNA encoded proteins were involved in carbohydrate metabolism, whereas the other nine were thought to be

associated with pathogenesis, senescence or stress responses. Binding of ethylene requires a copper cofactor and proper receptor function relies on a copper transporter.

### Changes in fruit during ripening

Ripening is associated with climacteric fruit, but it is sometimes used for non-climacteric fruit, especially as they develop on the plant. Perhaps a better term for non-climacteric would be maturation. The most obvious change in many fruits during ripening is their external colour, which is due to chlorophyll destruction and biosynthesis of pigments such as anthocyanins, carotenoids and lycopene. Textural changes are normally progressive softening during ripening. Softening is due to several factors; for example, Cordeiro *et al.* (2013) showed that softening of cherimoya was mainly attributed to depolymerization of pectin and lipid deterioration rather than hemicellulose degradation. During the developmental stage of climacteric fruit there is a general increase in starch content, which is hydrolysed to simple sugars during ripening. Acidity of fruit generally decreases during ripening. Acids help to form the desirable sugar-to-acid balance considered necessary for a pleasant taste. Organic acids present in fruit vary with different types of fruit, with malate and citrate being the most common. Some types of fruit can contain high levels of phenolic compounds – for example, in bananas and plantain. During ripening phenolics, such as tannins, are polymerized to insoluble compounds, resulting in a reduction of astringency in the ripe fruit. Phenolics are common in many fruit and are responsible for the oxidative browning reaction when the pulp is cut, especially if the fruit is immature. The enzyme PPO is responsible for this reaction. Besides the changes described above that affect the flavour of the fruit, the characteristic flavour and aroma of ripe fruit is due to the production of a complex mixture of individual volatile components that is characteristic to that particular fruit or cultivar. Toxins may exist in unripe fruit that reduce as the fruit ripens. Tomatoes at

the green stage of maturity contain a toxic alkaloid called solanine that decreases during ripening (Laval Martin *et al.*, 1975) and ackee fruit contains the toxin hypoglycin in the arils that gradually reduces as the fruit matures (Larson *et al.*, 1994). Profuse ethylene biosynthesis occurs in climacteric fruit, which initiates the ripening process. This takes place on the plant or tree when the fruit reaches maturity. Avocados are an exception, since they will not ripen on the tree and are harvested and either placed in ripening rooms or allowed to ripen postharvest (Gazit and Blumenfeld, 1970). Immature fruit may also be harvested and exposed to exogenous ethylene, which initiates ripening. Also most pear cultivars, if allowed to become too mature or to ripen on the tree, develop a coarse, mealy texture and often have core breakdown.

### Controlled atmosphere storage on ripening

In order to initiate climacteric fruit to ripen and control the ripening processes, the level of CO<sub>2</sub> and O<sub>2</sub> in the surrounding environment as well as ethylene and even moisture content have to be controlled. As indicated above, ethylene production and ripening can be suppressed at high CO<sub>2</sub> or low O<sub>2</sub> levels. Exposing fruit to sufficient stress can also initiate ripening. Stress during growing, caused by diseases or low water levels in the soil, can cause bananas to initiate to ripen before the fruit is fully grown or mature. Low humidity in the storage atmosphere can also result in stress and can initiate ripening. Maintaining high humidity around the fruit can help to keep fruit in the pre-climacteric stage (Thompson *et al.*, 1974d). Temperature is also critical; for example, Hatton *et al.* (1965) reported that ripening and softening rates of Florida mango cultivars increased as temperature increased over the range of 16–27°C and the optimum temperature range was 21–24°C. If mangoes were ripened at 27–30°C their skin became mottled and the fruit acquire a ‘strong flavour’, while ripening was retarded when they are held above 30°C (Paull and Chen,

2004). Exposure of Florida mango cultivars, picked at the mature-green stage, to ethylene at 20–100  $\mu\text{l/l}$  for 24 h resulted in faster and more uniform ripening at 21°C and 92–95% RH (Barmore and Mitchell, 1977).

CA storage has been shown to suppress the biosynthesis of ethylene in ripening fruit and in early work by Gane (1934) biosynthesis of ethylene was shown to cease in the absence of  $\text{O}_2$ . Ethylene concentrations in the core of apples were generally progressively lower with reduced  $\text{O}_2$  concentrations in the store over the range of 0.5–2 kPa  $\text{O}_2$  (Stow, 1989). Wang (1990) reviewed the literature on the effects of  $\text{CO}_2$  and  $\text{O}_2$  on the activity of enzymes associated with fruit ripening and cited many examples of the activity of these enzymes being reduced in CA storage. This is presumably due, at least partly, to many of these enzymes requiring  $\text{O}_2$  for their activity. Quazi and Freebairn (1970) showed that high  $\text{CO}_2$  and low  $\text{O}_2$  delayed the high production of ethylene in bananas, but the application of exogenous ethylene was shown to reverse this effect. Wade (1974) showed that bananas could be ripened in atmospheres of reduced  $\text{O}_2$ , even as low as 1 kPa, but the peel failed to de-green, which resulted in ripe fruits that were still green. Similar effects were shown at  $\text{O}_2$  levels as high as 15 kPa. Since the de-greening process in ‘Cavendish’ bananas is entirely due to chlorophyll degradation (Seymour *et al.*, 1987a and b), the CA storage effect was presumably due to suppression of this process. Hesselman and Freebairn (1969) showed that ripening of bananas, which had already been initiated to ripen by ethylene, was slowed in low  $\text{O}_2$  atmospheres. Goodenough and Thomas (1980, 1981) also showed suppression of de-greening of fruits during ripening; in this case it was with tomatoes ripened in 5 kPa  $\text{CO}_2$  combined with 5 kPa  $\text{O}_2$ . Their work, however, showed that this was due to a combination of suppression of chlorophyll degradation and the suppression of the synthesis of carotenoids, lycopene and xanthophyll. Jeffery *et al.* (1984) also showed that lycopene synthesis was suppressed in tomatoes stored in 6 kPa  $\text{CO}_2$  + 6 kPa  $\text{O}_2$ .

Ethylene biosynthesis was studied in ‘Jonathan’ apples stored at 0°C under CA

storage conditions of 0–20 kPa  $\text{CO}_2$  + 3 kPa or 15 kPa  $\text{O}_2$ . Fruits were removed from storage after 3, 5 and 7 months. Internal ethylene concentration, ACC levels and ethylene-forming enzyme activity were determined immediately after removal from storage and after holding at 20°C for 1 week. Increasing  $\text{CO}_2$  concentration from 0 kPa to 20 kPa at both low and high  $\text{O}_2$  concentrations inhibited ethylene production by the fruits. ACC levels were similarly reduced by increasing  $\text{CO}_2$  concentrations even in low  $\text{O}_2$ ; low  $\text{O}_2$  enhanced ACC accumulation but only in the absence of  $\text{CO}_2$ . One of the metabolic pathways used for ethylene synthesis by some plants and microbes is via an ethylene-forming enzyme, which uses  $\alpha$ -ketoglutarate and arginine as substrates. Ethylene-forming enzyme activity was stimulated by  $\text{CO}_2$  up to 10 kPa, but was inhibited by 20 kPa  $\text{CO}_2$  at both  $\text{O}_2$  concentrations. The inhibition of ethylene production by  $\text{CO}_2$  may, therefore, be attributed to its inhibitory effect on ACC synthase activity (Levin *et al.*, 1992). In other work, storing pre-climacteric fruit of apple cultivars ‘Barnack Beauty’ and ‘Wagner’ at 20°C for 5 days in 0 kPa  $\text{O}_2$  with 1 kPa  $\text{CO}_2$  (99 kPa  $\text{N}_2$ ) or in air containing 15 kPa  $\text{CO}_2$  inhibited ethylene production and reduced ACC concentration and ACC synthase activity compared with storage in normal atmospheres (Lange *et al.*, 1993).

Low  $\text{O}_2$  and increased  $\text{CO}_2$  concentrations were shown to have an effect on the decrease in ethylene sensitivity of ‘Elstar’ apples and ‘Scania’ carnation flowers during controlled atmosphere storage (Woltering *et al.*, 1994). Storage of kiwifruit in 2–5 kPa  $\text{O}_2$  with 0–4 kPa  $\text{CO}_2$  reduced ethylene production and ACC oxidase activity (Wang *et al.*, 1994). Ethylene production was lower in ‘Mission’ figs stored at 15–20 kPa  $\text{CO}_2$  concentrations compared with those kept in air (Colelli *et al.*, 1991).

## Genetics

The optimum storage conditions for fresh fruit and vegetables vary, not only between

species (from very short in fruit such as strawberries to very long as in natural storage organs such as potato tubers), but also between varieties and cultivars of the same species. For example, in apples some cultivars will be damaged when exposed to low levels of CO<sub>2</sub> in storage while others can tolerate comparatively high levels. In pears, optimum CA conditions and maximum storage time have been shown to vary considerably between cultivars (Table 5.5), but these recommendations interact and can also vary between countries for the same cultivar (Table 5.6).

The effects of the various pre-harvest and postharvest conditions on the postharvest life of fresh fruit and vegetables are clear and are dealt with in detail in other parts of this book. However, these effects and variations in their postharvest life depend on not only the genetic make-up of the plants but also the expression of their genes. Gene expression can be influenced by various factors, including the environment in which the crop is grown, as well as the postharvest environment to which it is exposed. However, changes in gene expression of crops

during storage, as well as their response to their environment, can affect their postharvest responses.

The genetic make-up of plants gives them their characteristics and their responses to postharvest conditions. Brizzolaro *et al.* (2017) reported that the response of different cultivars of apple to low O<sub>2</sub> storage depended on their genetic make-up. They showed that, in ULO or DCA-CF storage, ‘Granny Smith’ apples had low levels of fermentation while fermentation was high under the same conditions in ‘Red Delicious’. The expression of those genes may be different and can affect their responses to storage conditions. For example, Koch (1996) commented that ‘Plant gene responses to changing carbohydrate status can vary markedly. Some genes are induced, some are repressed, and others are minimally affected.’ Thammawong *et al.* (2015) found that changes in sugar concentration in cabbages during 2 days of storage seemed not to be influenced by CA and gene expression. They therefore interpreted it that the responses of crops to postharvest conditions can be a complex of genes, gene expression, the chemical changes occurring in the crop as well as the storage environment. They stored cabbages under different CA conditions and found that the changing pattern of each gene appeared to depend on the different percentages of CO<sub>2</sub> and O<sub>2</sub>. Obvious elevated gene expression was detected in the *BoGAPDH* (glyceraldehyde 3-phosphate dehydrogenase [GAPDH]-encoding gene), *BoADH* (alcohol dehydrogenase [ADH]-encoding gene) and *BoCTS* (citrate synthase [CTS]-encoding gene) after 24 hours storage. At the same time the expression of *BoHXK1*

**Table 5.5.** Recommended CA storage conditions for pears stored at -0.5°C; adapted from Van der Merwe (1996).

Cultivar	O <sub>2</sub> kPa	CO <sub>2</sub> kPa	Storage duration (months)
Bon Chretien	1.0	0	4
Buerré Bosc	1.5	1.5	4
Forelle	1.5	0–1.5	7
Packham’s Triumph	1.5	1.5	9

**Table 5.6.** Optimum CA conditions for selected pear cultivars; adapted from Kupferman (2001b).

Cultivar	Country	°C	O <sub>2</sub> kPa	CO <sub>2</sub> kPa	Duration (months)
Doyenne du Comice	Netherlands	-0.5	2.5	0.7	5
Doyenne du Comice	South Africa	-0.5	1.5	1.5	6
Doyenne du Comice	New Zealand	-0.5	2	< 1	3
Packham’s Triumph	New Zealand	-0.5	2	< 1	5
Packham’s Triumph	South Africa	-0.5	1.5	2.5	9
Williams Bon Chretien	South Africa	-0.5 to 0	1	0	4
Williams Bon Chretien	USA	-0.5 to 1	1.5	0.5	4

and *BoHXX2* (hexokinase [HXK]-encoding gene) and *BoGDH* (glutamate dehydrogenase [GDH]-encoding gene) was at a low level. Genes are expressed only under particular physiological conditions. Many genes might be involved, and interactions at different host–pathogen levels are continuously occurring during pathogen penetration and host reaction. Wild *et al.* (1997) found that failure to inhibit blue mould *Penicillium digitatum* in apples during the early stages of infection resulted in the subsequent development of the pathogen in tissue that is normally resistant. Li *et al.* (2018) reported that waxing pineapples reinforced their ‘antioxidant system and enhanced the expression levels of genes related to defence, such as *PGIP*’.

The elements of the storage environment that have been shown to interact with genetic factors include temperature, ethylene, O<sub>2</sub> and CO<sub>2</sub>. The postharvest changes that have been shown to be influenced by gene expression include ripening, textural and colour changes and susceptibility to diseases. Pre-storage treatments can affect the genetic response of the crop to storage conditions. For example, Sheng *et al.* (2018) found that several key genes involved in phenylpropanoid, flavonoid and stilbenoid pathways, including *PAL*, *CHS*, *F3H*, *LAR*, *ANS* and *STS*, were more expressed in response to pre-storage UV treatment, particularly with UV-C treatment.

### Ethylene

A number of genes involved in ethylene signalling have been identified, mainly from the isolation of ethylene response mutants in *Arabidopsis*. Many have been cloned, defining a pathway from ethylene perception to changes in gene expression. Johnson and Ecker (1998) found a negative regulator of ethylene responses, *CTR1*, acting downstream of the ethylene receptors. This pathway suggests that the ethylene signal is propagated through a mitogen-activated protein kinase cascade. *EIN2*, *EIN5*, *EIN6* and *EIN7* function downstream of *CTR1* and appear to lead ultimately to the regulation of gene

expression. During storage of Bartlett pears at 10°C, expression of ethylene biosynthesis genes (*ACS1a* and *ACO*) and ethylene receptor genes (*ETR2*, *ERS1a* and *ETR1a*) increased, while *CTR1* expression decreased (Nham *et al.*, 2017). Han *et al.* (2016) showed a significant induction of *MaC2H2-1/2* transcripts during the ripening of bananas with three different ripening characteristics caused by natural, ethylene-induced and 1-MCP delayed treatments, which correlated well with ethylene production. Also *MaC2H2-1/2* bound to the promoters of the key ethylene biosynthetic genes *MaACS1* and *MaACO1* and repressed their activities. They concluded that *MaC2H2-1/2* are transcriptional repressors and may mediate a finely tuned regulation of ethylene production during banana ripening, possibly via transcriptional repression of ethylene biosynthetic genes. Jian-fei Kuang *et al.* (2013) reported that EIN3 binding F-box protein (EBF) is an essential signalling component necessary for ethylene response. They isolated two EBF genes designated *MaEBF1* and *MaEBF2* from bananas and found that *MaEBF2* was enhanced by ethylene during fruit ripening, while *MaEBF1* changed only slightly. They concluded that *MaEBF* may be involved in banana ripening, at least partly via interaction with *MaEIL5*. Downstream, ethylene insensitive proteins (EIN2) are structurally novel containing an integral membrane domain. In the nucleus, the EIN3 family of DNA-binding proteins regulates transcription in response to ethylene, and an immediate target of EIN3 is a DNA-binding protein of the AP2/EREBP family (Alonso *et al.*, 1999).

### Carbon dioxide

Exposure of fruit and vegetables to increased levels of CO<sub>2</sub> can have negative or positive effects. Increased levels are commonly used in CA storage (Table 5.7) and in other cases exposure to high levels has been used to control pests. Short exposures to high levels of CO<sub>2</sub> have been shown to be effective in controlling quarantine insect pests while maintaining fruit quality. It also enhances the storage life of fruit.

**Table 5.7.** Recommended CO<sub>2</sub> and O<sub>2</sub> levels for the apple cultivar Bramley's Seedling at 4–4.5°C; Johnson (1994b and personal communication 1997).

CO <sub>2</sub> kPa	O <sub>2</sub> kPa	Storage time (weeks)
8–10	11–13	39
6	2	39
5	1	44

Romero *et al.* (2008) stored grapes under 20 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> + 60 kPa N<sub>2</sub> or air for 3 days, then air for 15 days. The grapes that had been exposed to CA for 3 days maintained their quality and activated the induction of transcription factors belonging to different families, such as ethylene response factors. Their results suggested that the beneficial effect of high CO<sub>2</sub> in maintaining grape quality appeared to be mediated by the regulation of ERFs and in particular *VviERF2-c* might play an important role by modulating the expression of PR genes. Ponce-Valadeza *et al.* (2009) reported that the effects of CO<sub>2</sub> on the metabolism of different fresh fruit and vegetables include changes in the activity of enzymes involved in glycolysis, fermentative metabolism, the tricarboxylic acid cycle and malate metabolism. Rothan *et al.* (1997) found that CO<sub>2</sub> induced the expression of certain stress-related genes and blocked the expression of both ethylene-dependent and ethylene-independent ripening associated genes. Responses of strawberries to high CO<sub>2</sub> concentrations have been shown to vary between cultivars (Pelayo *et al.*, 2003). For example, the cultivar 'Jewel' was shown to accumulate acetaldehyde and ethanol in storage in 20 kPa CO<sub>2</sub> while the cultivar 'Cavendish' did not. In 'Jewel', 168 gene sequences suggested differential expression, but in 'Cavendish' only 51 were differentially expressed (Ponce-Valadeza *et al.*, 2009).

Mature-green tomatoes were exposed to 20 kPa CO<sub>2</sub> for 3 days and then transferred to air. The high CO<sub>2</sub> effectively inhibited colour changes and ethylene production and reduced their protein content (Rothan *et al.*, 1997). They found that exposure of fruit to high CO<sub>2</sub> also resulted in the strong

induction of two genes encoding stress-related proteins: a ripening-regulated heat-shock protein and glutamate decarboxylase. In addition, high CO<sub>2</sub> blocked the accumulation of mRNA for genes involved in the main ripening-related changes: ethylene synthesis (ACC and ACC oxidase), colour (phytoene synthase), firmness (polygalacturonase) and sugar accumulation (acid invertase). They postulated that the inhibition of tomato fruit ripening by high CO<sub>2</sub> was due, in part, to the suppression of the expression of ripening-associated genes, which was probably related to the stress effect exerted by high CO<sub>2</sub>. Mathooko *et al.* (2004) found that exposure of peaches to 20 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> + 60 kPa N<sub>2</sub> for 1 h, sealed in a jar, and then to injected ethylene, regulated ethylene biosynthesis during ripening, at least in part, by antagonizing ethylene action. They also found that exogenous application of ethylene stimulated ethylene production at concentrations of 100, 500 and 1000 ppm. ACC oxidase activity was stimulated in a concentration-dependent manner. *PP-ACO1* was slightly expressed and exogenous ethylene stimulated accumulation of its mRNA transcript in a concentration-dependent manner at 0.1, 1, 10 or 100 ppm, then the level remained constant at 100, 500 and 1000 ppm. CO<sub>2</sub> and 1-MCP inhibited the ethylene-stimulated ethylene production, ACC oxidase activity and accumulation of *PP-ACO1* transcripts by about 50%.

## Oxygen

The molecular basis for the effects of low O<sub>2</sub> storage on fresh fruit and vegetables remains largely unknown. In climacteric fruit the final step in the biosynthesis of ethylene (ACC to ethylene) requires O<sub>2</sub> (Adams and Yang, 1979). Therefore a lower O<sub>2</sub> concentration might result in an inhibition of ethylene biosynthesis and slow the ripening process. Also all fresh fruit and vegetables synthesize O<sub>2</sub>, many in very low levels, so the effects on ethylene biosynthesis may also be a factor. Loulakakis *et al.* (2006) concluded from their work on low O<sub>2</sub> storage of avocados that the:

low oxygen, in addition to its inhibitory effect on ethylene biosynthesis and action, exerts its effect on prolonging the storage life of fruit by inducing a number of genes and proteins which possibly participate in the adaptation of fruit to low oxygen and by suppressing or by un-affecting others without the involvement of ethylene.

Passaris *et al.* (2007) showed that low O<sub>2</sub> regulated genes from *Citrus flavedo* responded differently in terms of their earliness, band intensity and their specificity to stress. They reported that this showed that some genes can be termed hypoxia-induced or anoxia-induced. Tonutti (2015) reported that in apples, in addition to the expression of genes involved in primary metabolism, ULO (about 1 kPa O<sub>2</sub>) affects specific secondary metabolic pathways that appear to be selectively modulated by different low O<sub>2</sub> levels. Members of the ethylene responsive factor VII (transcription factors gene family) displayed differential expression, suggesting their involvement in the modulation or controlling mechanisms where there is O<sub>2</sub> deficiency in cells. Tomato fruit stored under low O<sub>2</sub> conditions showed that this induced *HSP17.7* and *HSP21* genes, indicating that their primary role is in regulating cellular activity so as to remain very nearly constant after this stress (Pegoraro *et al.*, 2012). Hypoxic conditions inhibited the accumulation of RNA, protein and DNA synthesis associated with wounding in potato tubers (Butler *et al.*, 1990). In a number of plant tissues, anoxia and hypoxia caused marked alterations in the profiles of proteins, stability of mRNA species and gene expression (Fukao and Bailey-Serres, 2004). Kanellis *et al.* (1989a, 1998b) found that storing ripening-initiated avocados to 2.5 kPa O<sub>2</sub> for 6 days suppressed cellulase activity immune-reactive protein and abundance of its mRNA and also produced an alteration in the profile of total proteins, which involved suppression, enhancement and induction of new polypeptides (Kanellis *et al.*, 1989a, 1993). In addition, it has been shown that O<sub>2</sub> levels of 2.5–5.5 kPa, which suppressed the appearance of ripening enzymes, at the protein and mRNA levels, was similar to those O<sub>2</sub> levels that induced the synthesis of

new isoenzymes of alcohol dehydrogenase (Kanellis *et al.*, 1993).

Genetic engineering has been used to produce tomatoes that do not soften normally. These can be harvested green and ripened in an atmosphere containing ethylene. One genetically engineered cultivar of tomato called 'Flavr Savr' was marketed in the USA in 1993. It was also canned and marketed in the UK, but the market was restricted due to restrictions in marketing products of biotechnology in the European Union (EU) and other countries and consumer resistance. El-Sharkawy *et al.* (2016) showed that there is a direct involvement of growth substances in advancing fruit ripening independent of ethylene action through stimulating the transcription of several genes that encode cell wall metabolism-related proteins critical for determining fruit softening rate. Their results supported the hypothesis that the autonomous role played by growth substances is as important as that of ethylene in determining fruit ripening and in mediating other fruit quality traits, including their shelf-life.

## Diseases

Controlling postharvest diseases and pests of fruit and vegetables has been shown to be effective using CA conditions as well as applying fungicidal chemicals. For example, storage of figs in high CO<sub>2</sub> atmospheres reduced mould growth without detrimentally affecting the flavour of the fruit (Wardlaw, 1938). Mitcham *et al.* (1997) showed that high CO<sub>2</sub> levels in the storage atmosphere could be used for controlling insect pests of grapes, using 45 kPa CO<sub>2</sub> + 11.5 kPa O<sub>2</sub> at 0°C, which gave complete control of *Platynota stultana*, *Tetranychus pacificus* and *Frankliniella occidentalis* without injury to the grapes. In papaya, low O<sub>2</sub> (1–5 kPa), with or without increased CO<sub>2</sub> (2–10 kPa), reduced decay and delayed ripening (Chen and Paull, 1986). However, new technologies have enabled the evaluation of gene expression for host defence enhancement for disease control; for example, increasing levels of phenolic compounds in the plant

tissues. This approach to plant development meets the requirements of integrated disease management on sustainable use of pesticides that is included in the EU implementation through Directive 128/2009.

Fruit and vegetables have physical and chemical defence mechanisms, some of which are produced in response to an attack by a microorganism, to prevent or limit damage. The reaction of plant cells to a pathogenic attack leads to the induction of numerous genes encoding defence proteins. These include structural proteins that are incorporated into the cell walls and contribute to the confinement of the pathogen, enzymes involved in the biosynthesis of phytoalexins and pathogenesis-related (PR) proteins. Phytoalexins are antimicrobial compounds that are both synthesized by and accumulated in plants after an attack by microorganisms. Production of these chemical defences is associated with the activation of a large number of genes encoding various types of stress proteins, including PR proteins. Ding *et al.* (2002) suggested that the pre-treatment of tomato fruit with MJ or methyl salicylate induced the synthesis of some stress proteins, including PR proteins, which led to increased chilling tolerance and resistance to pathogens. The gene *MaPR1a* has a high level of identity with PR proteins and Kesari *et al.* (2007) also suggested that jasmonic acid and salicylic acid as well as ethylene could induce expression of *MaPR1a* in fruit tissue. Peng *et al.* (2014) reported that all calmodulin genes were upregulated by salicylic acid and MJ. Calmodulin is a multifunctional intermediate messenger protein expressed in plant cells. They performed expression studies on a family of six calmodulin genes (*SICaMs*) in mature green tomatoes in response to mechanical injury and *Botrytis cinerea* infection. Both wounding and pathogen inoculation triggered expression of all those genes, with *SICaM2*

being the most responsive to both treatments. Nitric oxide acts as an important signal molecule in plants. Hu *et al.* (2014b) inoculated mangoes with a spore suspension of *Colletotrichum gloeosporioides*, which causes anthracnose disease. Nitric oxide treatment enhanced the activities of defence-related enzymes including PAL, cinnamate-hydroxylase, 4 coumarate: CoA ligase, peroxidase,  $\beta$ -1,3-glucanase and chitinase and also promoted the accumulation of total phenolics, flavonoids and lignin that might contribute to inhibition of the pathogen.

Several viruses attack papaya, including mosaic and bunchy top. Papaya ring spot virus (PRSV) causes chlorosis and if young plants are infected they remain stunted, while infection of adult plants can result in low yield of poor quality fruit. The virus is primarily spread by aphids and some control can be achieved by controlling the aphids. In 1985, a group from the US Pacific Basin Agricultural Research Center began work on a genetic modification of papaya (González, 1998). They isolated and sequenced the gene that coded for the protein coat of PRSV and introduced it into callus cells from the papaya cultivar 'Sunset', which eventually resulted in a resistant plant line that was successfully field tested in 1991 (Tenant *et al.*, 2001). The result was the cultivar 'SunUp'. 'SunUp' was later crossed with the non-genetically modified 'Kapoho Solo' to create the 'Rainbow' cultivar. Both 'SunUp' and 'Rainbow' passed regulatory testing by the US government and were officially released to farmers in 1998; and in 2000 more than half papayas grown in Hawai'i were transgenic. 'Rainbow' subsequently passed Japanese regulatory testing in 2011. The team in Hawai'i has collaborated in producing PRSV-resistant transgenic papaya for other countries, including Brazil, Jamaica, Venezuela, Thailand, Bangladesh, Uganda and Tanzania (Ferreira *et al.*, 2002).



# 6

## Pests, Diseases and Disorders

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Pests and particularly diseases are among the main factors limiting the postharvest life of fresh fruit and vegetables. Clearly when fruit are infected in the field it can affect their postharvest characteristics. For example, bacterial canker (*Pseudomonas syringae* pv. *actinidiae*) in kiwifruit orchards affected the quality and storage life of the fruit (Prencipe *et al.*, 2016). Additionally the storage conditions, especially temperature and to a lesser extent humidity, can affect pests and diseases. CA storage has been shown to affect diseases and pest infestation but particularly physiological disorders. Changing the O<sub>2</sub> and CO<sub>2</sub> can also affect the metabolism of microorganisms and insect pests in storage and can therefore be a factor in controlling them.

Postharvest control of pests and diseases by CA storage can be used as a method of reducing the amount of chemicals used and thus chemical residues in the food we eat. Low O<sub>2</sub> but especially high CO<sub>2</sub> levels in storage have been generally shown to have a negative effect on the growth and development of disease-causing microorganisms. There is also some evidence that fruit develops less disease on removal from high CO<sub>2</sub> storage than after previously being stored in air. However, in certain cases the levels of CO<sub>2</sub> necessary to give effective disease control have detrimental effects on the quality of the fruit and vegetables. The mechanism

for reduction of diseases appears to be a reaction of the fruit rather than the low O<sub>2</sub> or high CO<sub>2</sub> directly affecting the microorganism, though there is some evidence for the latter. The mode of action of changing the O<sub>2</sub> and CO<sub>2</sub> contents around the fruit or vegetable is related to modifying their physiology. Therefore the levels of the two gases or the balance between them can change the metabolism of fruit and vegetables in a way that is not beneficial resulting in physiological disorders. Physiological disorders of fruit can result from CA storage, but in other cases levels of disorders can be reduced. The mechanisms for reduction of disorders vary and are not always well understood. Exposing fruit and vegetables to either high levels of CO<sub>2</sub> or a combination of high CO<sub>2</sub> and low O<sub>2</sub> can control insects infecting fruit and vegetables. However, extended exposure to insecticidal levels of CO<sub>2</sub> may be phytotoxic. The CA treatment may be applied for just a few days at the beginning of storage, which may be sufficient to kill the insects without damaging the crop.

### Physiological Disorders

There is a whole range of disorders that can occur in fresh produce during storage that

are not primarily associated with infection with microorganisms. These are collectively referred to as physiological disorders, physiological diseases or physiological injury. As would be expected, CA storage has been shown to have positive, negative and no effect on physiological disorders of stored fresh fruit and vegetables. The incidence and extent of storage disorders are influenced not only by the concentration of CO<sub>2</sub> and O<sub>2</sub> and duration of exposure but also by the conditions in which they were grown, cultivar, harvest maturity, storage temperature and humidity. Cultivars and individual fruits vary in their susceptibility to injury because of biochemical and anatomical differences, including the size of intercellular spaces and the rate of gas diffusion through the cuticle, the epidermis and other cells. External injuries to the epidermis or internal disorders and cavities in the tissue usually become visible as brown spots as a result of oxidation of phenolic compounds. This is the last step in a reaction chain beginning with the impairment of the viability of cell membrane by fermentation metabolites, shortage of energy or possibly excess of free radicals (Streif *et al.*, 2003). Physiological disorders associated with high CO<sub>2</sub> are shown in [Table 6.1](#).

A more comprehensive discussion of the subject can be found in Fidler *et al.* (1973) and Snowden (1990,1992) but to indicate

the symptoms and some of the causes of disorders mentioned in this book a few disorders are described in the following sections.

### Bitter pit

The incidence and severity of bitter pit in apples is influenced by the dynamic balance of minerals in different parts of the fruit as well as the storage temperature and levels of O<sub>2</sub> and CO<sub>2</sub> in the store (Sharples and Johnson, 1987) and can be controlled by chemical treatment, especially with calcium. Jankovic and Drobnjak (1994) described experiments where ‘Idared’, ‘Cacanska Pozna’, ‘Jonagold’ and ‘Melrose’ apples were stored at 1°C and 85–90% RH, either in < 7 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub> or in air. Fruits stored in CA exhibited no physiological disorder, whereas bitter pit was observed in ‘Melrose’ stored in air. Bitter pit was observed in ‘Red Delicious’ apples stored in PE bags when CO<sub>2</sub> exceeded 5 kPa and O<sub>2</sub> fell to 15–16 kPa (Hewett and Thompson, 1988). In contrast, the incidence of bitter pit was reduced in ‘Cox’s Orange Pippin’ apples in storage at 1°C in PE bags of 25 µm thickness in which there were 50–70 holes per bag to fit inside an 18.5 kg carton. However, it was not reduced to commercially acceptable

**Table 6.1.** Physiological disorders associated with high CO<sub>2</sub>.

	Disorder	Source
Apples	Brown heart	Kidd and West, 1923
Apples	Core flush	Lau, 1983
Apples	Low temperature breakdown	Fidler, 1968
Broad bean	Pitting	Tomkins, 1965
Broccoli	Accelerated softening	Lipton and Harris, 1974
Broccoli	Off-flavours	Lipton and Harris, 1974
Cabbage	Internal browning	Isenberg and Sayles, 1969
Capsicums	Internal browning	Morris and Kader, 1977
Kiwifruit	Off-flavours	Harman and McDonald, 1983
Lettuce	Brown stain	Stewart and Uota, 1971
Mushrooms	Cap discoloration	Smith, 1965
Potato	Curing inhibition	Butchbaker <i>et al.</i> , 1967
Spinach	Off-flavours	McGill <i>et al.</i> , 1966
Strawberry	Off-flavours	Couey and Wells, 1970
Tomato	Uneven ripening	Morris and Kadar, 1977
Tomato	Surface blemishes	Tomkins, 1965

levels when the non-treated fruit had levels of bitter pit exceeding 20–30%. Stella McCloud (unpublished results), working in New Zealand, reported that CA, with increased CO<sub>2</sub> levels, reduced the development of bitter pit in ‘Cox’s Orange Pippin’ apples (John Stow, personal communication). De Freitas *et al.* (2010) reported that the development of bitter pit was a complex process that involved the total input of calcium into the fruit, but also a proper calcium homeostasis at the cellular level. They found that calcium accumulation into storage organelles and calcium binding to the cell wall represented an important contribution to bitter pit development.

### Cell wall destruction

Onions stored for 162 or 224 days by Adamicki *et al.* (1977) at 1°C in 10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> developed a physiological disorder where the cell walls of the fleshy scales of the epidermis and parenchyma were destroyed and the ultra-structure of mitochondria were altered.

### Chilling injury

There is some evidence that CA storage can affect the development of chilling injury in stored fruit. Flesh browning and flesh breakdown of plums was shown to occur when they were stored in air at 0°C for 3–4 weeks (Sive and Resnizky, 1979), but when the same fruits were stored at 0°C in 2–8 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> they could be stored for 2–3 months followed by 7 days in air without showing the symptoms. Wade (1981) showed that storage of the peach cultivar ‘J.H. Hale’ at 1°C resulted in flesh discoloration and the development of a soft texture after 37 days, but in atmospheres containing 20 kPa CO<sub>2</sub> fruit had only moderate symptoms even after 42 days. Conversely, Visai *et al.* (1994) showed a higher incidence of chilling injury in the form of internal browning in ‘Passe Crassane’ pears stored at 2°C in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> compared with fruit stored in air at

0°C. They accounted for this effect as being due to the stimulation of the production of free radicals in the fruit stored in CA.

### Discoloration

Core flush has been described on several cultivars of apple where it develops during storage as a brown or pink discoloration of the core, while the flesh remains firm. It has been associated with CO<sub>2</sub> injury but may also be related to chilling injury and senescent breakdown. Wang (1990) reviewed the effects of CO<sub>2</sub> on brown core and concluded that it was due to exposure to high levels of CO<sub>2</sub> at low storage temperatures. Johnson and Ertan (1983) found that in storage at 4°C apples were free of core flush and other physiological disorders in 1 kPa O<sub>2</sub>, which was markedly better than in 2 kPa O<sub>2</sub>. Resnizky and Sive (1991) found that keeping ‘Jonathan’ apples in 0 kPa O<sub>2</sub> for the first 10 days of storage prevented core flush in early-picked apples from highly affected orchards. No damage due to fermentation was observed in any of the treatments.

Skin browning was reduced in litchis that were stored for 28 days at 1°C in plastic film bags with 10 kPa CO<sub>2</sub> or vacuum packed compared with non-wrapped fruit. However, their taste and flavour were unacceptable, while fruit stored non-wrapped remained acceptable (Ahrens and Milne, 1993). *Phaseolus vulgaris* beans, broken during harvesting and handling, developed a brown discoloration on the exposed surfaces; exposure to O<sub>2</sub> levels of 5 kPa or less controlled this browning, but resulted in off-flavours in the canned products (Henderson and Buescher, 1977). High CO<sub>2</sub> concentrations have also been reported to inhibit browning of beans at the sites of mechanical injury (Costa *et al.*, 1994) and were not injurious to quality as long as O<sub>2</sub> was maintained at 10 kPa or higher.

Wang *et al.* (2000c) reported a CO<sub>2</sub> linked disorder whose symptoms resembled superficial scald. The ‘Empire’ CO<sub>2</sub>-linked disorder was effectively controlled by conditioning fruit at 3°C (but not at 0°C) for 3–4 weeks at 1.5–3% O<sub>2</sub> without CO<sub>2</sub> prior to CA storage at

1.5% O<sub>2</sub> + 3% CO<sub>2</sub>. Reduction of the disorder was also achieved by storage at 3°C in air, but excessive ripening and associated loss of flesh firmness occurred during subsequent CA storage (Wang *et al.*, 2000c). Castro *et al.* (2007) reported flesh browning in 'Pink Lady' apples 2 months after harvest for those stored at 1°C in CA, but the browning did not increase after a longer storage time and was not observed during storage in air. Lipton and Mackey (1987) showed that Brussels sprouts stored in 0.5 kPa O<sub>2</sub> occasionally had a reddish-tan discoloration of the heart leaves and frequently an extremely bitter flavour in the non-green portion of the sprouts.

The apple cultivar 'Braeburn' is highly susceptible to internal browning (Braeburn browning disorder) (BBD), which has been associated with increased levels of CO<sub>2</sub> in the storage atmosphere (see Chapters 7, 8 and 12). Scott and Wills (1974) showed that all 'Williams' Bon Chretien' ('Bartlett') pears, which had been stored at -1°C for 18 weeks in about 5 kPa CO<sub>2</sub>, were externally in excellent condition but were affected by brown heart. When the apples were placed in micro-perforated PE film bags after vacuum infiltration with up to 20% calcium chloride, they developed a brown heart-like disorder (Hewett and Thompson, 1989). Deuchande *et al.* (2017) stored 'Rocha' pears at -0.5°C in air and various atmospheres and found that only those in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> developed internal browning. They found faster depletion of ascorbate under high CO<sub>2</sub> and concluded that it was associated with the down-regulation of glutathione reductase, ascorbate peroxidase and monodehydroascorbate reductase. The genes involved were *PcGR*, *PcAPX* and *PcMDHAR*, respectively. They also reported a synergistic effect between high CO<sub>2</sub> and low O<sub>2</sub>. When, after 60 days, the pears were transferred from 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> to 1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> for 80 days, internal browning disorder incidence increased, which was associated with increases in fermentation.

Stewart and Uota (1971) showed that lettuce during cold storage had increasing levels of a brown stain on the leaves with increasing levels of CO<sub>2</sub> in the storage atmosphere.

### Internal breakdown

Symptoms of internal breakdown are browning or darkening of the flesh, which eventually becomes soft and the fruit breaks down. It is associated with storage for too long, especially for fruit harvested too mature, and eventually all fruits that do not breakdown due to disease infection will succumb to internal breakdown. There was no internal breakdown of Cox's Orange Pippin apples after 4 months of storage at 2.5°C in 2.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, but there was a perceptible but low incidence in air (Schulz, 1974).

Tonini *et al.* (1993) showed that storage of nectarine and plums for 40 days at 0°C in 2 kPa O<sub>2</sub> with either 5 or 10 kPa CO<sub>2</sub> reduced internal breakdown compared with those stored in air. When the nectarine cultivar 'Flamekist' was stored by Lurie *et al.* (1992) in CA for 6 or 8 weeks, the 10 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> atmosphere prevented internal breakdown and reddening that occurred in fruits stored in air. Cooper *et al.* (1992) showed that CA storage with up to 20 kPa CO<sub>2</sub> reduced the incidence of internal browning in nectarine without any other adverse effect on fruit quality and good control was also shown in storage in 4 kPa O<sub>2</sub>. Streif *et al.* (1994) found that exposure of nectarine to 25 kPa CO<sub>2</sub> prior to CA storage had little effect, but storage at high CO<sub>2</sub> levels, especially in combination with low O<sub>2</sub>, significantly delayed ripening, retained fruit firmness and prevented both storage disorders. There was no deleterious effect on flavour with storage at high CO<sub>2</sub> levels.

### Necrotic spots

Bohling and Hansen (1977) described the development of necrotic spots on the outer leaves of stored cabbage, which was largely prevented by low O<sub>2</sub> atmospheres in the store, but increased CO<sub>2</sub> had no effects.

### Superficial scald

Scald is a physiological disorder that can develop in apples and pears during storage.



**Fig. 6.1.** Superficial scald on apples in New Zealand (photograph courtesy of Dr R.O. Sharples).

Kupferman (undated) described the typical symptoms of scald as a diffuse irregular browning of the skin with no crisp margins between affected and unaffected tissue and it is usually only skin deep (Fig. 6.1). It can occur on any part of the skin and can vary in colour from light to dark brown. In some cases the lenticels are not affected, leaving green spots.

Scald has been associated with ethylene levels in the store atmosphere and can be controlled by a pre-storage treatment with a suitable antioxidant. Ethoxyquin (1,2-dihydro-2,2,4-trimethylquinoline-6-yl ether) marketed as ‘Stop-Scald’ or DPA (diphenylamine) marketed as ‘No Scald’ or ‘Coraza’ can be effective when applied directly to the fruit within a week of harvesting (Hardenburg and Anderson, 1962; Knee and Bubb, 1975). Postharvest treatment with Ethoxyquin gave virtually complete control of scald and stem-end browning on fruit stored at 3.9°C or 5°C in 8–10 kPa CO<sub>2</sub> (Knee and Bubb, 1975). Ethoxyquin was initially registered as a pesticide in 1965 as an antioxidant used to prevent scald in pears and apples through a pre-harvest spray and post-harvest dip or spray. The Codex Alimentarius Commission established maximum residue limits for Ethoxyquin residues in or on pears at 3 ppm under US Environment Protection Agency 738-R-04-011 November 2004. DPA was first registered as a pesticide in the USA in 1947. However, the Codex Alimentarius maximum residue limit for DPA on apples was set at 5 mg/kg, compared with

the 10 ppm US tolerance for apples (*Guide to Codex Maximum Limits For Pesticides Residues* under EPA-738-F-97-010 April 1998). Residue levels of DPA in apples were found to vary depending on the application method and the position of the fruit in the pallet box. Villatoro *et al.* (2009) reported that there was an interaction between DPA and CA storage, in that ‘Pink Lady’ apples stored at 1°C in 2.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> retained higher concentrations of DPA residues compared with those stored in 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> or in air.

Very low levels of O<sub>2</sub> in CA storage have been used commercially to control scald in apples with 0.7 kPa O<sub>2</sub> in British Columbia and 1 kPa O<sub>2</sub> in Washington State, and 1 kPa for pears in Oregon State (Lau and Yastremski, 1993). Coquinot and Richard (1991) stored apples in 1.2 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub>, with or without removal of ethylene, and found that in this atmosphere scald was controlled and ethylene removal was not necessary. Johnson *et al.* (1993) found that fruits respired normally for 150 days in storage in 0.4 O<sub>2</sub> + nominally 0 kPa CO<sub>2</sub> or 0.6 kPa O<sub>2</sub> + nominally 0 kPa CO<sub>2</sub> but ethanol accumulated thereafter. Retardation of scald development by 0.4 and 0.6 kPa O<sub>2</sub> was as effective as 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> and ethylene removal from 9 kPa CO<sub>2</sub> + 12 kPa O<sub>2</sub> storage provided scald-free fruits for 216 days. However, rapid loss of firmness occurred in fruits stored in all low O<sub>2</sub> levels + nominally 0 kPa CO<sub>2</sub> after 100 days storage and was the major limitation to storage life.

It was recommended that scrubbed low O<sub>2</sub> storage, e.g. 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, and ethylene removal from scrubbed or non-scrubbed CA stores should be considered as alternatives to chemical antioxidants for the control of scald. In 'Jonagold' apples, Awad *et al.* (1993) found that both O<sub>2</sub> and CO<sub>2</sub> affected the occurrence of scald. At 3 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> the level of scald was 39%, which was significantly higher in early-harvested apples compared with 13% in apples from a normal harvest date. Scald in 'Granny Smith' apples was reported by Gallerani *et al.* (1994) to be successfully controlled by storing fruits in targeted low O<sub>2</sub> concentrations of 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub>, and their findings enabled low O<sub>2</sub> application to be more precisely directed towards superficial scald control. Van der Merwe *et al.* (1997) showed that, for the cultivars 'Granny Smith' and 'Topred', exposure of fruit to initial low O<sub>2</sub> stress of 0.5 kPa O<sub>2</sub> for 10 days at either -0.5°C or 3°C (after 7 days in air at the same temperatures) could be used as an alternative to DPA for superficial scald control during storage at 1°C and 3 kPa CO<sub>2</sub> with 1 kPa O<sub>2</sub>, since DPA had been banned in some countries. Kupferman (undated) reported that scald was reduced in apples that were well ventilated using a hollow fibre membrane air separator in a purge mode to supply air at 1.5 kPa O<sub>2</sub> + CO<sub>2</sub> below 3 kPa at 0°C or 3°C. The atmosphere was established within 7 days of loading the room and a continuous purging system was used to maintain the atmosphere. Fidler *et al.* (1973) also showed that scald development in storage could be related to the CO<sub>2</sub> and O<sub>2</sub> levels but that the relationship varied between cultivars (Table 6.2). 'Granny Smith' apples were stored at -0.5°C for 6 months in air or for 9 months in 1.5 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> and then ripened at 20°C for 7 days. In air all the apples developed scald but in the CA only a few developed scald (Van Eeden *et al.*, 1992).

Kupferman (undated) also reported differences in susceptibility between apple cultivars and that 'Braeburn', 'Fuji', 'Gala', 'Golden Delicious', 'McIntosh' and 'Spartan' showed a low risk of scald, 'Rome Beauty' a moderate risk, 'Red Delicious' a moderate to high risk and 'Granny Smith' a high risk.

**Table 6.2.** Effects of CA storage conditions during storage at 3.5°C for 5–7 months on the development of superficial scald in three apple cultivars (Fidler *et al.*, 1973).

Storage atmosphere		Cultivar		
CO <sub>2</sub> kPa	O <sub>2</sub> kPa	Wagener	Bramley's Seedling	Edward VII
0	21	100	–	–
0	6	–	89	75
0	5	100	–	–
0	4	–	85	62
0	3	9	30	43
0	2½	–	17	43
0	2	0	–	–
8	13	–	24	3

Lau and Yastremski (1993) found that scald susceptibility in 'Delicious' apples was strain dependent. While storage in 0.7 kPa O<sub>2</sub> effectively reduced scald in 'Starking' and 'Harrold Red' picked over a wide range of maturity stages, it did not adequately reduce scald in 'Starkrimson' during 8 months of storage (Van der Merwe *et al.*, 2003). Yu and Wang (2017) claimed that the current commercial method of controlling superficial scald (ethoxyquin + CA storage) was not adequate for 'd'Anjou' pears produced in hot seasons. They found that a better combination was ethoxyquin at 2700 mg/l + 1-MCP at 0.3 µl/l + ethylene at 0.3 or 0.6 µl/l. This combination also extended storage quality, reduced decay and allowed recovery of ripening capacity after long-term CA storage.

### Vascular streaking

This is a disorder of cassava where the vascular bundles in the tuberous roots turn dark blue to black during storage. The symptoms can develop within a day or so of harvesting and the disorder has been associated with the O<sub>2</sub> level in the atmosphere and other possible causes (Thompson and Arango, 1977). Aracena *et al.* (1993) found that waxed 'Valencia' cassava roots stored at 25°C and 54–56% RH for 3 days had 46% vascular streaking in air and 15% in 1 kPa O<sub>2</sub>. However,

at 25°C and 95–98% RH there was only 1.4% vascular streaking in air or 1 kPa O<sub>2</sub>.

### White core

Arpaia *et al.* (1985) stored the kiwifruit cultivar 'Hayward' for up to 24 weeks in 2 kPa O<sub>2</sub> + 0, 3, 5 or 7 kPa CO<sub>2</sub> at 0°C. The occurrence and severity of white core in CA plus ethylene was highest in 5 kPa CO<sub>2</sub>. Two other physiological disorders were observed (translucency and graininess) and their severity was increased by the combination of high CO<sub>2</sub> and ethylene.

### Fungal Diseases

Reports on the effects of CA storage on the development of diseases of fruit and vegetables have shown mixed results. Disease incidence on cabbages stored in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 2.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> was lower than those stored in air (Prange and Lidster, 1991). *In vitro* studies of Chinese cabbage inoculated with *Phytophthora brassicae* and stored at 1.5°C in 0.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or 3.0 kPa CO<sub>2</sub> + 3.0 kPa CO<sub>2</sub> showed low levels of disease development compared with those stored in air. In contrast, *in vivo* studies showed that the infection caused by *P. brassicae* was significantly higher in the CA atmosphere than in air after 94–97 days of storage (Hermansen and Hoftun, 2005). In celery stored at 8°C, disease suppression was greatest in atmospheres with 7.5–30 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub>, but there was only a slight reduction in 4–16 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or in 1.5–6 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> (Reyes, 1988). A combination of 1 or 2 kPa O<sub>2</sub> + 2 or 4 kPa CO<sub>2</sub> prevented black stem disease development in celery during storage (Smith and Reyes, 1988). 'Golden Delicious' apple losses from rotting were lower in CA storage than at the same temperature in air (Reichel, 1974). 'Cox's Orange Pippin' apples were stored at 2.5°C in air or in 2.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> at 3.5°C by Schulz (1974). There was natural contamination of the fruits on the tree by *Pezizula* spp. and the disease was

shown to be slightly retarded in the CA storage. There were no differences in the occurrence of Botrytis and Penicillium rots between CA and air storage. When both undamaged and damaged fruits were artificially inoculated at harvest, *Penicillium malicorticis* was shown to be more active on injured fruits in CA storage than in those stored in air. 'Hayward' kiwifruit were exposed to 60 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> at 30 or 40°C for 1, 3 or 5 days by Cheah *et al.* (1994) for the control of *Botrytis cinerea* rot. Spore germination and growth of *B. cinerea* were completely inhibited *in vitro* by 60 kPa CO<sub>2</sub> at 40°C and partially suppressed at 30°C. Kiwifruits were inoculated with *B. cinerea* spores, exposed to 60 kPa CO<sub>2</sub> at 30°C or 40°C, and stored in air at 0°C for up to 12 weeks after treatment. Storage in 60 kPa CO<sub>2</sub> at 40°C reduced disease incidence from 85% in air at 20°C to about 50%, but exposure to CO<sub>2</sub> at 40°C for longer than 1 day adversely affected fruit ripening. 'Flame Seedless' and 'Crimson Seedless' grapes that were inoculated with the *B. cinerea* were exposed to 40 kPa CO<sub>2</sub> at 0°C for 48 h and then stored in 12 kPa O<sub>2</sub> + 12 kPa CO<sub>2</sub> also at 0°C. This combined treatment reduced *B. cinerea* incidence from 22% to 0.6% after 4 weeks and from 100% to 7.4% after 7 weeks of storage compared with fruit not treated and stored in air at 0°C (Teles *et al.*, 2014).

It was reported that the growth of the fungi *Rhizopus* spp., *Penicillium* spp., *Phomopsis* spp. and *Sclerotinia* spp. was inhibited by atmospheres of total N<sub>2</sub> but not in atmospheres containing 99 kPa N<sub>2</sub> + 1 kPa O<sub>2</sub> (Ryall, 1963). Kader (1997) indicated that levels of 15–20 kPa CO<sub>2</sub> could retard decay incidence on cherry, blackberry, blueberry, raspberry, strawberry, fig and grape. Storage of strawberries in 10–30 kPa CO<sub>2</sub> or 0.5–2 kPa O<sub>2</sub> are used to reduce disease levels, but 30 kPa CO<sub>2</sub> or less than 2 kPa O<sub>2</sub> was also reported to cause the development of off-flavour in some circumstances (Hardenburg *et al.*, 1990). Harris and Harvey (1973) stored strawberries in atmospheres with 0, 10, 20 or 30 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> at 5°C for 3–5 days to simulate shipping conditions. The fruit were then held at 15.6°C for 1–2 days to simulate distribution. They found that

fruit held in CO<sub>2</sub>-enriched atmospheres had less softening and decay (mostly *B. cinerea*) than those held in air and that 20 or 30 kPa CO<sub>2</sub> was the most effective. However, fruit held in 30 kPa CO<sub>2</sub> developed off-flavours. Differences in decay in fruit held in air and at high CO<sub>2</sub> were greater after subsequent holding in air at 15.6°C than on removal from storage. They also showed that CA storage of strawberries not only reduced disease levels during storage, but also had an additional beneficial effect in disease reduction when it was removed for marketing (Table 6.3). Parsons *et al.* (1974) showed considerable reduction in disease levels on tomatoes in CA storage compared with storage in air. Most of the effect came from the low O<sub>2</sub> levels, with little additional effect from increased CO<sub>2</sub> levels (Table 6.4).

There is evidence of interactions between CA storage and temperature on disease development. After 24 weeks storage of currants (*Ribes* spp.) at -0.5°C and 20 or 25 kPa CO<sub>2</sub>, the average incidence of fungal rots was lowest at about 5%. Rotting increased

to about 35% at 1°C in 25 kPa CO<sub>2</sub> and to about 50% at 1°C and 20 kPa CO<sub>2</sub>. With 0 kPa CO<sub>2</sub> at either temperature, the incidence of rots was about 95% (Roelofs, 1994).

There remains the question of how CA storage actually reduces or controls diseases on fruit and vegetables. Parsons *et al.* (1974) showed that atmospheres containing 3 kPa O<sub>2</sub> reduced decay on stored tomatoes caused by *Rhizopus* or *Alternaria*. However, both genera of fungi grew well *in vitro* in 3 kPa O<sub>2</sub> or less. This led them to hypothesize that the reduction in decay was due to the CA storage conditions acting on the tomato fruit itself so that it developed resistance to the fungi, rather than acting only on the fungi. Probably the answer is a combination of both the reaction of the crop and the susceptibility of the microorganism.

## Bacterial Diseases

Some *in vitro* and *in vivo* studies have indicated that CA may have a negative effect on bacteria. Brooks and McCulloch (1938) reported that storage of lima beans in CO<sub>2</sub> concentrations of 25–35 kPa inhibited bacterial growth without adversely affecting the beans. Parsons and Spalding (1972) inoculated tomatoes with soft rot bacteria and held them for 6 days at 12.8°C in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or in air. Lesions were smaller on fruits stored in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> than on those stored in air, but CA storage did not completely control decay. Amodio *et al.* (2003) stored slices of mushrooms at 0°C either in air or in 3 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> for 24 days and found that there was a slight increase for mesophilic and psychrophilic

**Table 6.3.** The effects of CO<sub>2</sub> concentration during storage for three days at 5°C on the percentage levels of decay in strawberries and the development of decay when removed to ambient conditions of approximately 15°C; modified from Harris and Harvey (1973).

Storage	0 kPa CO <sub>2</sub>	10 kPa CO <sub>2</sub>	20 kPa CO <sub>2</sub>	30 kPa CO <sub>2</sub>
3 days storage at 5°C	11	5	2	1
+ 1 day at 15°C in air	35	9	5	4
+ 2 days at 15°C in air	64	26	11	8

**Table 6.4.** Effects of CA storage conditions on the decay levels of tomatoes harvested at the green-mature stage; modified from Parsons *et al.* (1974).

Storage atmosphere	After removal from 6 weeks at 13°C	Plus 1 week at 15–21°C	Plus 2 weeks at 15–21°C
Air (control)	65.6 %	93.3 %	98.6 %
0 kPa CO <sub>2</sub> + 3 kPa O <sub>2</sub>	2.2 %	4.4 %	16.7 %
3 kPa CO <sub>2</sub> + 3 kPa O <sub>2</sub>	3.3 %	5.6 %	12.2 %
5 kPa CO <sub>2</sub> + 3 kPa O <sub>2</sub>	5.0 %	9.4 %	13.9 %



bacteria in air but no increase or a slight decrease for those in 3 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub>.

Potatoes are cured before long-term storage by exposing them to a higher temperature at the beginning of storage to heal any wounds. Weber (1988) showed that the defence reaction of potatoes to infection by *Erwinia carotovora* subsp. *atroseptica* during the curing period was inhibited by temperatures of less than 10°C, reduced O<sub>2</sub> levels of less than 5 kPa and CO<sub>2</sub> levels of over 20 kPa. CO<sub>2</sub> can retard bacterial growth by increasing the lag phase before they begin to develop. The degree of retardation was shown to increase with increasing concentrations of CO<sub>2</sub>, but Daniels *et al.* (1985) reported that *Clostridium botulinum* may survive even at high CO<sub>2</sub> levels. Sobiczewski *et al.* (1999), in a study of the efficacy of antagonistic bacteria in protection of apples against fungal diseases, found that during storage at 2–3°C there was a tendency for the highest reduction in bacterial population in the atmosphere containing 1.5 kPa CO<sub>2</sub> and 1.5 kPa O<sub>2</sub> compared with air. Zhang *et al.* (2013) stored fresh-cut pineapple at 7°C and found that those packaged in air were not acceptable after 7 days, while those packaged in 50 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> were still acceptable. Also 50 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> retarded the growth of a lactic acid bacterium (*Leucono stoccitreum*) and the two spoilage yeasts tested (*Candida sake* and *C. argentea*).

### Insects

CA storage has been used to control insects. It was suggested that the most promising application for fresh fruit, such as cherries, nectarines, plums, apples, pears, stone fruits, blueberries and strawberries, is to expose them to less than 1 kPa O<sub>2</sub> atmospheres for short periods to replace chemical treatments to meet quarantine requirements (Ke and Kader, 1989, 1992a). However, the levels that are necessary may be phytotoxic to the fruit. They reviewed the insecticidal effects of very low O<sub>2</sub> and/or very high CO<sub>2</sub> atmospheres at various temperatures and

compared these levels with the responses and tolerance of fresh fruit and vegetables to similar CA conditions. They concluded that the time required for 100% mortality was shown to vary with insect species and their developmental stage, temperature, O<sub>2</sub> level, CO<sub>2</sub> level and humidity. Yahia and Kushwaha (1995) reported that 'Hass' avocados, 'Sunrise' papayas and 'Keitt' mangoes tolerated low O<sub>2</sub> ( $\pm 0.5$  kPa) and/or very high CO<sub>2</sub> ( $\pm 50$  kPa) for 1, 2 and 5 days, respectively, at 20°C. They suggested that these treatments could have some potential use as insecticidal atmospheres for quarantine insect control on the basis of fruit tolerance, insect mortality and costs. On stored vegetables, Cantwell *et al.* (1995) also showed that various combinations of O<sub>2</sub>, CO<sub>2</sub> and temperature were effective in achieving complete insect mortality before the development of phytotoxic symptoms. They found that storage in 10–20 kPa CO<sub>2</sub> could control thrips (*Thysanoptera* spp.) in 7 days and the peach potato aphid (*Myzus persicae*) in 10–14 days. In 80–100 kPa CO<sub>2</sub> at 0°C, complete mortality was consistently achieved for both species within 12 h.

Ke *et al.* (1994a) showed that tolerances of peaches and nectarines to CA that were insecticidal were related to the time before occurrence of visual injury and/or off-flavour. The tolerances of 'John Henry' peaches, 'Fantasia' nectarines, 'Fire Red' peaches, 'O'Henry' peaches, 'Royal Giant' nectarine and 'Flamekist' nectarine to 0.25 kPa O<sub>2</sub> + 99.75 kPa N<sub>2</sub> at 20°C were 2.8, 4.0, 4.0, 4.4, 5.1 and 5.2 days, respectively. 'Fairtime' peaches tolerated 0.21 kPa O<sub>2</sub> with 99 kPa CO<sub>2</sub> at 20°C for 3.8 days, 0.21 kPa O<sub>2</sub> with 99 kPa CO<sub>2</sub> at 0°C for 5 days, 0.21 kPa O<sub>2</sub> at 20°C for 6 days, and 0.21 kPa O<sub>2</sub> at 0°C for 19 days. Comparison of fruit tolerance on the time to reach 100% mortality of some insect species suggested that 0.25 kPa O<sub>2</sub> at 20°C is probably not suitable for postharvest insect disinfestation, while 0.21 kPa O<sub>2</sub> with or without 99 kPa CO<sub>2</sub> at 0°C merited further investigation. Kerbel *et al.* (1989) showed that 'Fantasia' nectarines were quite tolerant to exposures to low O<sub>2</sub> and/or high CO<sub>2</sub> for short periods, but may be long enough for insect control. Yahia *et al.* (1992)

showed that insecticidal O<sub>2</sub> concentrations of less than 0.4 kPa O<sub>2</sub> with the balance being N<sub>2</sub> for less than 3 days at 20°C can be used as a quarantine insect control treatment in papaya without the risk of significant fruit injury. Scale insects (*Quadraspidiotus perniciosus*) on stored apples were completely eliminated by storing infested fruit in 2.6–3.0 kPa O<sub>2</sub> + 2.4–2.5 kPa CO<sub>2</sub> or 1.5–1.7 kPa O<sub>2</sub> + 1.0–1.1 kPa CO<sub>2</sub> at 1°C or 3°C for 31–34 weeks, plus an additional week at 20°C and 50–60% RH (Chu, 1992). Leon *et al.* (1997) disinfested ‘Manila’ mangoes by exposing them at 12°C to 1% O<sub>2</sub> combined with either 30% or 50% CO<sub>2</sub> for 3 days. After disinfestation, the mangoes were kept at 12°C for up to 27 days and no quality differences were found between CA-treated and non-treated mangoes. Respiratory activity of the mangoes held at 12°C for 18 days and then allowed to ripen at 25°C was higher than that of those kept at 25°C. Symptoms of chilling injury appeared by 18 days in mangoes that were treated with 50% CO<sub>2</sub>, by 21 days in mangoes treated with 30% CO<sub>2</sub> and by 24 days in those not treated. It appeared that disinfestation by CA increased, to a limited degree, the susceptibility of ‘Manila’ mangoes to chilling injury.

Sweet potatoes were exposed to low O<sub>2</sub> and high CO<sub>2</sub> for 1 week during curing or subsequent storage to evaluate the effect on the weevil *Cylas formicarius elegantulus*. Exposure to levels required for insect control was found not to be feasible during curing, but cured sweet potatoes could tolerate CA combinations that have a potential as quarantine procedures. They tolerated 8 kPa O<sub>2</sub> during curing, but when exposed to 2 or 4 kPa O<sub>2</sub> or to 60 kPa CO<sub>2</sub> + 21 or 8 kPa O<sub>2</sub> they were unusable within 1 week after curing, mainly due to decay. Exposure of cured sweet potatoes to 2 or 4 kPa O<sub>2</sub> + 40 kPa CO<sub>2</sub>, or 4 kPa O<sub>2</sub> + 60 kPa CO<sub>2</sub> for 1 week at 25°C had little effect on postharvest quality but exposure to 2 kPa O<sub>2</sub> + 60 kPa CO<sub>2</sub> resulted in increased decay, a reduction in flavour and more off-flavours (Delate and Brecht, 1989). Sweet potatoes infested with adult and immature stages of *C. f. elegantulus* were exposed to atmospheres containing low O<sub>2</sub> and increased concentrations of

CO<sub>2</sub> for up to 10 days at 25°C and 30°C. Adults were killed within 4–8 days when exposed to 8 kPa O<sub>2</sub> + 40–60 kPa CO<sub>2</sub> at 30°C. At 25°C, exposure to 2 or 4 kPa + 40 or 60 kPa CO<sub>2</sub> at 25°C killed all the adult insects within 2–8 days but exposure to 8 kPa O<sub>2</sub> + 30–60 kPa CO<sub>2</sub> for 1 week at 30°C failed to kill all the weevils. However, no adult weevils emerged from infested roots treated with either 4 kPa O<sub>2</sub> + 60 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 40 or 60 kPa CO<sub>2</sub> for 1 week at 25°C (Delate *et al.*, 1990).

Fumigation of ‘Bing’ cherries with methyl bromide (MB) to control codling moth (*Cydia pomonella*) negatively affected fruit and stem appearance (Retamales *et al.*, 2003). However, they reported that the cherries could be heated to 45°C for 41 min or 47°C for 27 min in an atmosphere of 1 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> + 84 kPa N<sub>2</sub>, which could provide quarantine security against codling moth and western cherry fruit fly (*Rhagoletis cingulata*). Treated cherries had similar incidence of pitting and decay and similar preference ratings after 14 days of storage at 1°C as non-treated or MB-fumigated fruit (Shellie *et al.*, 2001). They also showed that storage in CO<sub>2</sub> levels up to 30 kPa did not markedly influence the quality aspects compared with storage in air, with the exception of decay being significantly reduced. Chervin *et al.* (1999) suggested that exposure to 2 kPa O<sub>2</sub> for 3 days at 28°C approximately halved the time required in cold storage for effective control of late-instar light brown apple moth (*Epiphyas postvittana*) in apples. Preliminary observations suggested that there may not be substantial difference between the resistance of non-diapausing and pre-diapausing codling moth larvae to this treatment followed by cold storage. Consumer panels found that fruit were as acceptable as control fruit. In storage at 0°C the mortality responses of armoured scales, *Hemiberlasia* spp., on New Zealand ‘Hayward’ kiwifruit exposed to storage in air or 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> were equally effective (Whiting, 2003). Mature scales were more tolerant of both storage treatments than immature scales. *Tuta absoluta* is a pest that affects tomatoes and other solanaceous crops in Europe and other Mediterranean areas.

As an alternative to fumigation with MB, Riudavets *et al.* (2016) found that exposure of *T. absoluta*-infected tomatoes to an atmosphere of 95 kPa CO<sub>2</sub> at 25°C for 48 h controlled all stages, but negatively affected fruit quality. However, they found that 40 kPa CO<sub>2</sub> for 72 h gave the same level of control with no negative effects on fruit quality. A cold storage treatment at 1°C for a total of 10 days was also effective in the control of the *T. absoluta* eggs.

*Callosobruchus maculatus* is the principal pest of the stored chickpea. The larvae of this insect feed on the grain, causing considerable damage that produces qualitative and quantitative losses. For control of this species, pesticides such as phosphine and MB have been used. Iturralde-García *et al.* (2016) tested CA storage (50, 70 or 90 kPa CO<sub>2</sub> + 10, 6 or 3 kPa O<sub>2</sub> for up to 120 h) on *C. maculatus*. They found that exposure to 50 kPa CO<sub>2</sub> for 48 h resulted in 100% mortality of adult insects, but third-instar insects had higher resistance than adults, with only

18.9% mortality after 48 h. CA exposure had no significant effects on germination, water absorption, cooking time, texture and colour of the chickpeas. CA has been used in storage of dried cereal and grain legumes to control insect infestation. Tome *et al.* (2000) found that 0, 30, 40, 50 or 60 kPa CO<sub>2</sub> with the balance N<sub>2</sub> did not affect the moisture content, water absorption, cooking time or colour index of dried *Phaseolus vulgaris* beans during 20 days of storage. Brackmann *et al.* (2002b) found that storage of pinto beans (*P. vulgaris*) in 100% N<sub>2</sub> for 19 months had zero losses due to insects, compared with significant losses for those stored in air. The beans stored in total N<sub>2</sub> had a lighter tegument colour and a shorter cooking time than those stored in air. Darkening was a consequence of the oxidation of phenolic compounds. Pereira *et al.* (2007) used 50 ppm ozone to fumigate maize grain for 7 days against insect infestation and found no detrimental effects on the grain during 180 days of subsequent storage.

# 7

## Pre-storage Treatments

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The application of CA storage and MA packaging technologies to the fresh fruit and vegetable industry requires inputs of other technologies, including special treatments of the crop before it is put into store or before it is packaged. In most cases it is advisable to place fruit or vegetables into appropriate storage as quickly as possible after harvest. However, in certain cases it has been shown that some treatments can be beneficial prior to CA storage or MA packaging and these treatments are discussed in this chapter.

### High Temperature

Root crops such as potatoes, yams and sweet potatoes have a cork layer over the surface of their tubers. These cork layers serve as a protection against microorganism infections and reduce water loss, but cork layers can be damaged during harvesting and handling operations. Exposing crops to high temperature and humidity for a few days directly after they have been loaded into the store can facilitate repair of the damaged tissue and reduce postharvest losses.

The term curing is also applied to some fruit, especially citrus fruits. The mechanism in citrus fruits is different to root crops

but it effectively heals wounds and reduces disease levels. Drying is also carried out to aid preservation of onions and garlic, but this simply involves drying their outer layers to reduce microorganism infection and water loss. Various drying recommendations have been published, including exposure to about 30°C and 70% RH for 7–10 days before storage (Thompson *et al.*, 1972a).

Immersing fruit in hot water or brushing with hot water prior to storage is used to control microorganisms that may have infected fruit before storage. The optimum recommended conditions for disease control varies between crop and disease organism. For example, visible appearance of the fungal disease anthracnose (*Colletotrichum gloeosporioides*) during ripening of mangoes was effectively inhibited by hot-water treatments at 46°C for 75 min combined with storage for 2 weeks at 10°C in either 3 kPa O<sub>2</sub> + 97 kPa N<sub>2</sub> or 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> (Kim *et al.*, 2007) (see Fig. 2.2). Hot-water brushing at 56°C for 20 s reduced decay development in grapefruit during storage (Porat *et al.*, 2004). In some cases exposure to temperatures slightly higher than those used in storage have been shown to have beneficial effects. For example, exposure of ‘Granny Smith’ apples to 0.5 kPa O<sub>2</sub> for 10 days at 1°C before CA storage at –0.5°C inhibited the development of superficial scald for fruit

picked at pre-optimum maturity (Van der Merwe *et al.*, 2003). In apples, Gasser and von Arx (2015) found that hot-water treatment did not affect fruit quality or the incidence of physiological disorders during DCA storage. Hot-water treatment of jujube at 40°C increased storage life by slowing ripening and reducing the rate of ethylene evolution (Anuradha and Saleem, 1998).

### Root crops

In root crops, curing involves suberin being deposited in the parenchymatous cells just below the damaged area of the tuber. In sweet potatoes, Stange *et al.* (2001) showed that the interior flesh tissues accumulate antifungal compounds (including 3,5-dicaffeoylquinic acid) under curing conditions of 30°C and 90–95% RH for 24 h, which helps resistance to infection. Sweet potato could be also cured at 32–36°C and 100% RH for 4 days (Thompson, 1972b) or 29–32°C and 85–90% RH for 7–15 days (Lutz, 1952). Curing potatoes has been standard pre-storage practice since the 1920s and can be achieved by exposure to 10°C for 30 days, 15°C for 15 days or 20°C for 10 days, all at high humidity (Burton, 1989). Curing involves suberin being deposited in the parenchymatous cells just below the damaged area of the tuber. Suberin refers to a polymer containing phenolics and long-chain decarboxylic fatty acids, hydroxy fatty acids and, perhaps, epoxy fatty acids, which provides initial protection to the tuber by forming a highly lipophilic coating within and on the surface of tissues. There are some indications that the fatty acids may also act as phytoalexins and the peroxides and epoxides may have toxic effects on infectious microorganisms (Galliard, 1975). Subsequently, below the suberized cells, a meristematic layer of cells is formed which is the cork cambium. This produces new cells that contain lignin, which seal off the damaged area. Both these processes are temperature and humidity sensitive (Burton, 1989). Cassava could be successfully cured by exposure to 25–40°C and 80–85% RH (Booth, 1975) while

cuts of 2 mm deep could be cured, but cuts of 8 mm or deeper showed no curing over a 10-day period (Akhimienho, 1999). In the 1980s a cassava root conditioning protocol was developed in Latin America by scientists of the Centro Internacional de Agricultura Tropical (Wheatley, 1989). It was found that the treatment of cassava roots with the fungicide thiabendazole and their subsequent exposure to conditions of high humidity at ambient temperatures in plastic bags permitted the storage of fresh cassava to be extended from 3 to 21 days without engendering significant postharvest losses or compromising culinary quality. In the 1990s this technology was transferred initially to Ghana and Tanzania by the NRI. On-farm trials and assessments in commercial market environments confirmed that the technology would extend the storage life of fresh cassava. However, due to the reliance on the use of a fungicide and plastic bags, Digges (1995) considered that the cost of the original protocol might militate against its widespread adoption.

A simplified methodology was therefore developed based on the careful out-grading of damaged roots and the temporary storage of water-dipped roots in locally available woven polypropylene sacks which were then held at high humidity under tarpaulins or plastic sheeting (Bancroft and Crentsil, 1995). This approach proved cheaper and was effective at maintaining cassava quality for a reasonable 7–10 days. Socio-economic surveys indicated a high level of satisfaction with this latter process, particularly amongst the snack-food vendors whose profit margins benefited from the extended shelf-life potential of fresh cassava and the reduction in the volume of waste (Bancroft *et al.*, 1998). In common with other root crops, the possibility of curing yams (*Dioscorea* spp.) after harvest in order to extend their storage life and reduce postharvest losses has attracted the attention of researchers for decades (Table 7.1).

In the early 2000s, over a series of consecutive storage seasons, rural farm trials were conducted in the yam-growing regions of Ghana to determine the effects of a range of different low-cost preconditioning and

**Table 7.1.** Optimum curing conditions for yams. Modified from Rees *et al.*, 2012.

Species	Temperature (°C)	Humidity (% RH)	Duration (days)	Reference
<i>D. alata</i>	29–32	90–95	4	González and Rivera (1972)
<i>D. bulbiifera</i>	26–28	high	5–7	Martin (1974)
<i>D. cayenensis</i>	36–40	91–98	1	Thompson <i>et al.</i> (1977)
<i>D. cayenensis</i>	25–40	95–100	1–7	Been <i>et al.</i> (1976)
<i>D. esculenta</i>	26–28	high	5–7	Martin (1974)
<i>D. esculenta</i>	29	84	5	Thompson (1972a)
<i>D. rotundata</i>	25–30	55–82	5	Adesuyi (1973)
<i>D. rotundata</i>	25–40	95–100	1–7	Been <i>et al.</i> (1976)
<i>D. rotundata</i>	26	92	11–15	Nnodu and Nwankiti (1986)
<i>D. rotundata</i>	35–40	95–100	1	Thompson (1972a)

storage regimes on the postharvest quality of local varieties of *Dioscorea rotundata* (white yam). The study suggested that beneficial curing protocols could be deployed in rural communities (Bancroft *et al.*, 2005). For periods of 2–3 months traditional pit storage engendered less rot and moisture loss than storage above ground in barns. Eventually, however, pit-stored yams became prone to nematode infestation and sprouting. Curing in polyethylene bags was not found to be reliable. More successful was the use of temporary ‘curing’ clamps, rooms or enclosures. Under field conditions, curing was enhanced by exposure to temperatures of about 32°C for 5–6 days. Such conditioning helped to reduce the incidence of rots during subsequent storage but did not retard water loss. In contrast, super-optimal curing temperatures favoured the development of rots. Strict grading of the yam tubers prior to storage had a marked impact on the extent of postharvest losses. Although chronic internal infections and disorders can compromise the subsequent marketability of the yams, stored tubers are particularly susceptible to rots that originate at the site where the tuber is cut away from vine.

### Fruit

For crops other than root crops, pre-storage exposure to high temperature is less well established and often still in an experimental stage. Sealing citrus fruits in plastic film

bags and then exposing them to 34–36°C for 3 days resulted in the inhibition of *Penicillium digitatum* infection and reduced decay and blemishes through lignification and an increase in antifungal chemicals in the peel without deleterious effects on the fruit (Ben-Yehoshua *et al.*, 1989). Tariq (1999) found that placing damaged oranges in sealed PE film bags and exposing them to temperatures in the range of 25–35°C for 3–48 h (the lower the temperature, the longer was the exposure time required) completely eliminated the browning and fungal growth. The nature of curing appeared to be related only to temperature and humidity and not O<sub>2</sub> or CO<sub>2</sub>, since where PE bags were used the thickness of the film (30 or 50 µm) had no effect on the process, although it changed the O<sub>2</sub> and CO<sub>2</sub> content around the fruit. Also, curing could be achieved without film packaging if the surrounding atmosphere was maintained at 95–98% RH.

‘Granny Smith’ apples were kept at 46°C for 12 h, 42°C for 24 h, or 38°C for 72 h or 96 h before storage at 0°C for 8 months in 2–3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. The heat-treated fruit were firmer at the end of storage and had a higher TSS:TA ratio and a lower incidence of superficial scald than fruits not heat treated. Pre-storage regimes with longer exposures to high temperatures (46°C for 24 h or 42°C for 48 h) resulted in fruit damage developing during storage (Klein and Lurie, 1992). Brigati and Caccioni (1995) recommended curing kiwifruits in well ventilated warehouses at 15–20°C for 48 h before refrigeration. ‘Niiitaka’ Asian pears were cured

at ambient temperatures of about 18°C for 5, 10 or 15 days followed by step-wise cooling down to 0.5°C, which reduced skin blackening without significantly altering other fruit quality characteristics during storage but affected the physical properties of the fruit's epidermis (Park and Kwon, 1999). Acorn squash tissue held at 28°C for 18 h developed considerable resistance to infection by *Penicillium italicum* (Stange and McDonald, 1999). Phillips and Armstrong (1967) and Purseglove (1968) also recommended curing at 23.9–29.4°C for 2 weeks to harden the shell, but other work claimed that curing was either ineffective or detrimental (Wardlaw, 1938; Lutz and Hardenburg, 1968).

### Irradiation

UV irradiation has been demonstrated to increase the natural resistance of some commodities and has been studied as a direct method for decay control. In many countries irradiation of fresh fruits and vegetables has been approved at specified levels of exposure. UV-C has been shown to reduce postharvest decay on strawberries (Nigro *et al.*, 2000; Pombo *et al.*, 2011) and mangoes (González-Aguilar *et al.*, 2001) and bacterial counts on minimally processed baby spinach leaves (Escalona *et al.*, 2010), watermelon cubes (Artés-Hernández *et al.*, 2010) and white mushrooms (Guan *et al.*, 2012). Guan *et al.* (2012) showed that UV-C doses of 0.45–3.15 kJ/m<sup>2</sup> resulted in 0.67–1.13 log CFU/g reduction of *Escherichia coli* O157:H7 inoculated on mushroom cap surfaces. Sheng *et al.* (2018) exposed table grapes to UV-B or UV-C radiation at 3.6 kJ/m<sup>2</sup> and then stored them at 4°C for 28 days. They found that both UV treatments increased the phenolic compounds of grapes during storage. UV-C irradiation has also been reported to have other effects. Kaewsuksaeng *et al.* (2012) irradiated mature-green limes with UV-B doses at 19.0 kJ/m<sup>2</sup> and then stored them at 25°C in the dark. They found that the irradiation effectively suppressed chlorophyll degradation through the control of the action of the chlorophyll-degrading enzyme and slowed the loss of ascorbic acid. Jiang *et al.* (2010)

reported that UV-C irradiation resulted in maintenance of firmness and enhanced antioxidant capacity in shiitake mushrooms. Kasim and Kasim (2012) found that it prevented yellowing in fresh-cut cress but increased electrolyte leakage due to tissue damage. UV-C irradiation extended postharvest shelf-life of banana by reducing ethylene production and respiration rate compared with those that had not been irradiated (Pongprasert *et al.*, 2011).

The mode of action seems to be related more to its effect on the crop rather than on microorganisms. Pongprasert *et al.* (2011) reported that UV-C could play a role in activation of plant defence mechanisms and antioxidant systems. Nigro *et al.* (2000) indicated that treatment with low UV-C doses produced a reduction in postharvest decay related to induced resistance mechanisms. UV treatment of carrot slices induced production of 6-methoxymellen, which is inhibitory to *Botrytis cinerea* and *Sclerotinia sclerotiorum* (Coates *et al.*, 1995). Sheng *et al.* (2018) found that several key genes involved in phenylpropanoid, flavonoid and stilbenoid pathways, including *PAL*, *CHS*, *F3H*, *LAR*, *ANS* and *STS*, were more expressed during storage of grapes at 4°C for 28 days in response to UV radiation at 3.6 kJ/m<sup>2</sup>, particularly to UV-C.

### Hydrogen

Fumigation of kiwifruit with water with a high hydrogen content (prepared by bubbling hydrogen gas through water until it was saturated) was shown to inhibit senescence of kiwifruit and postharvest production of hydrogen gas was shown to decrease during ripening (Hu *et al.*, 2014a). Hu *et al.* (2018) therefore tested whether fumigation of kiwifruit with hydrogen would affect ripening. They found that softening was delayed, incidence of decay reduced and ethylene production reduced with the down-regulation of the corresponding gene transcripts after fumigation. Al Ubeed *et al.* (2018) reported that hydrogen sulfide can inhibit senescence of fruit and vegetables and reported that its mode of action included

inhibiting the action of ethylene. They fumigated pak choi cabbages with hydrogen sulfide at 250  $\mu\text{l/l}$  and stored them in an ethylene-free atmosphere. They found that the hydrogen sulfide fumigation was effective in inhibiting the loss of green colour and reduced the respiration rate, ion leakage and endogenous ethylene production. In a comparison with 1-MCP they found that both MCP and hydrogen sulfide were equally effective in the parameters they tested, but when ethylene was introduced to the containers 1-MCP was more effective than hydrogen sulfide.

### Waxing and Polishing

Fruit and vegetables can be dipped or sprayed with a range of chemicals postharvest to delay deterioration, control diseases or improve their appearance. There are many treatments that have been tested; some are included in this chapter. Also many treatments have been used in combination with others and therefore may be included under a different heading in this chapter. Some coatings provide a partial barrier to gas exchange between the fruit and its environment that is generally beneficial. However, modification of the atmosphere in and around the fruit cells by the use of coating material can increase disorders associated with high  $\text{CO}_2$  and low  $\text{O}_2$  concentration, such as accumulation of ethanol and alcoholic off-flavours (Smith *et al.*, 1987).

Many fruits can benefit from the application of waxes and subsequent polishing. This is not simply to enhance the appearance, which is of course very important, but also to improve the storage quality of the crop. When the crop is in the field or even during harvesting, transport, washing or grading, it can undergo some scratching or abrasion that not only removes the natural layer of wax but can also scar the epidermis. Washing is essential in the removal of bird faeces, insect marks, chemical residues and general field dirt but can also remove much of the natural wax, especially if used with detergents. The natural layer of

wax reduces moisture loss from the fruit. The appearance of crops such as citrus fruits may be improved by synthetic wax application and some producers may add dyes to the wax in order to modify the colour in addition to providing an enhanced shine. Commercial waxes, mainly used for citrus, are by-products of the petrochemical industry produced from proprietary formulae and include Fruitex, Britex and Seal Britex. There is public interest in development of edible natural biodegradable coatings to replace the synthetic waxes for maintaining postharvest quality of fruit. Natural waxes include sugarcane wax, carnauba wax, shellac and various resins. Carnauba is a wax extracted from the leaves of the palm *Copernicia prunifera*, which is a native of north-eastern Brazil and consists mostly of aliphatic esters, diesters of 4-hydroxycinnamic acid, hydroxycarboxylic acids and fatty alcohols. Lac is a resin secreted by the female lac bug (*Kerria lacca* or *Laccifer lacca*) found on the bark of many species of tree in the forests of India and Thailand. It can be refined and processed into shellac dry flakes or dissolved in ethanol to make liquid shellac. Rosins are obtained from the oleoresins of pine trees after the distillation of turpentine and are used for coating, especially for citrus, primarily to impart a gloss.

These coatings may be applied to the fruit in a foam or liquid bath, a liquid spray or by sponge or brush rollers. Application rates may be governed by legislation in the country where the coating is being applied, or in the country where the fruit is to be exported. It is essential that these regulations, as well as the manufacturers' recommendations, be strictly adhered to. Mechanized spray systems often apply an excess of fluid so that the unused surplus flows through the conveyor back into the holding vessel ready for reapplication. The fruit must be dried after waxing to give good adhesion and retention of the wax. Drying is usually done in a drying tunnel and may also incorporate soft brushes. A blast of warm air is often directed at the fruit as it passes along the conveyor under the fan, or else the heaters are under the fruit and the



fan draws the air through it by suction (Clarke, 1996).

The yeast *Candida guilliermondii* was shown to be antagonistic towards *Penicillium* spp. (McGuire, 1993) and could be successfully incorporated into commercial citrus waxes where it was shown to survive for 2 months at 12°C. Kouassi *et al.* (2012) showed improved disease control when an essential oil was incorporated in shellac or carnauba wax. Coating cassava roots with paraffin wax was shown to keep them in good condition for 1–2 months at room temperature in Bogotá in Colombia (Young *et al.*, 1971) and this coating treatment is commonly applied to cassava roots exported from Costa Rica (Fig. 7.1). Another treatment is dipping the roots in a fungicide (benomyl or thiabendazole) and packing them in plastic film bags directly after harvest (Thompson and Arango, 1977), but there are restrictions on the use of benzimidazole fungicides.

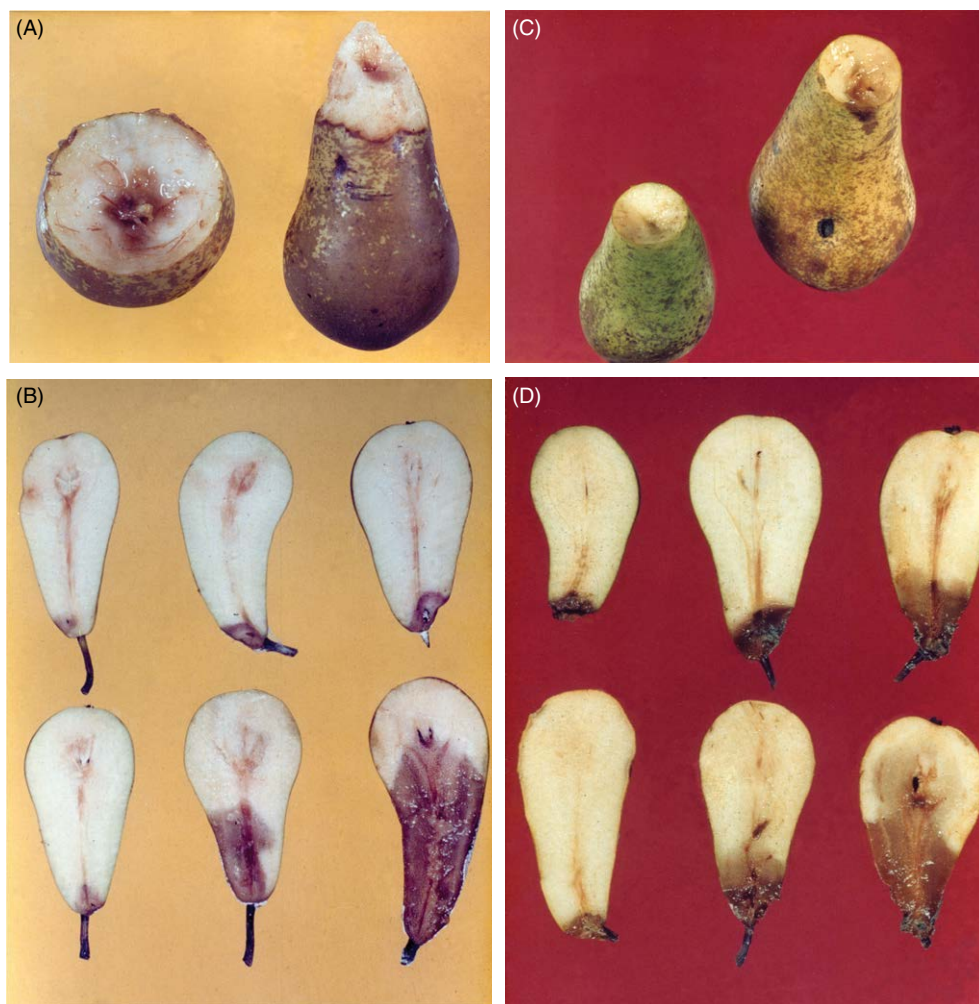
Saberi *et al.* (2018) tested various coatings on storage of Valencia oranges for up to 4 weeks at 20°C or 5°C with an additional storage for 7 days at 20°C. Of the coatings they tested, pea starch and guar gum blended with a lipid mixture of shellac and oleic acid were the best in extending shelf-life and reducing respiration rate, ethylene production, weight loss, softening, peel pitting and decay. Li *et al.* (2018) waxed pineapples directly after harvest and found that waxing delayed their colour change, decreased respiration rate and ethylene production, and reduced internal browning.



**Fig. 7.1.** Cassava coated with paraffin wax in Costa Rica on arrival in UK.

## Carboxymethyl Cellulose

Dan Cutts from Geest Industries and Peter Lowings from Cambridge University introduced a coating in 1982 that consisted of sodium carboxymethyl cellulose, sucrose fatty acid esters and mono-glycerides and di-glycerides. They claimed that it was non-phytotoxic, tasteless and odourless and that it increased the shelf-life of fresh fruit and vegetables. The product was called Tal-Prolong and Lowing supervised two PhD students, Nigel Banks and Roger Bancroft, to investigate the physiological response of different fruits to this preparation. Banks (1984) showed that Tal-Prolong could delay the ripening of bananas. He postulated that the effect was due to the restriction in gas exchange between the fruit and its surrounding atmosphere. This caused an increase of CO<sub>2</sub> and a depletion of O<sub>2</sub> in the cells, thus resulting in an effect similar to that achieved in CA storage. Bancroft (1989, 1995) confirmed these findings and also showed some reduction in levels of disease on top fruit coated with 1.5% Tal-Prolong. Bancroft observed that Tal-Prolong diminished the overall incidence of some common postharvest fungal infections of ‘Conference’ pears held in cold storage. At higher temperatures in both apples and pears, such coatings modified the spatial distribution of rots within the fruit and markedly reduced the rate of spread of lesions. In comparison with experimental controls, Tal-Prolong had a greater impact on rots caused by *Botrytis cinerea*, *Monilinia fructigena* and *Rhizopus nigricans* than those caused by *Alternaria alternata* and *Penicillium expansum* (Fig. 7.2). The efficiency of the treatments in reducing the rate of spread of rots was related to the concentration of Tal-Prolong, the type and cultivar of fruit, the timing of application and the temperature to which the fruit was subsequently exposed. Krishnamurthy (1989) ripened ‘Alphonso’ mangoes at 28–32°C and 30–60% RH coated with various concentrations of Tal-Prolong and found that 1% inhibited the development of yellow peel colour and delayed some of the other ripening changes for 4–5 days. At higher concentrations the fruit



**Fig. 7.2.** 'Conference' pears inoculated with *Monilinia fructigena* or *Penicillium expansum* and held at 20°C in the presence and absence of a surface treatment with 1.5% Tal-Prolong.

(A) The configuration of residual 'healthy' tissues recovered from 'Conference' pears in the presence (right) and absence (left) of Tal-Prolong some 6 days after inoculation with *M. fructigena*.

(B) The extent of lesions caused by *M. fructigena* at 2, 4 and 6 days after inoculation in the presence (top row) and absence (bottom row) of Tal-Prolong.

(C) The configuration of residual 'healthy' tissues recovered from 'Conference' pears in the presence (left) and absence (right) of Tal-Prolong some 6 days after inoculation with *P. expansum*.

(D) The extent of lesions caused by *P. expansum* at 4, 7 and 10 days after inoculation in the presence (top row) and absence (bottom row) of Tal-Prolong.

remained green, but softened, with a high percentage of spoilage.

Tal-Prolong was subsequently reformulated and marketed as Semperfresh. The company that manufactured and marketed both Tal-Prolong and Semperfresh was called Surface Systems International. Al-Zaemey

*et al.* (1993) found that banana crowns coated with Semperfresh showed delayed development of crown rot caused by infection with *Colletotrichum musae*. This effect could be enhanced by the inclusion of organic acids to Semperfresh. Kerbel *et al.* (1989) showed that for 'Granny Smith' apples coated with

Semperfresh (0.75% and 1.5%) and stored at 0°C for 4 or 5 months, ripening was delayed ‘somewhat’ compared with fruit with no coating. The treatments did not reduce levels of bitter pit or superficial scald. Carrillo-Lopez *et al.* (2000) observed ripening delays of several days in ‘Haden’ mangoes coated with Semperfresh. Dang *et al.* (2008) evaluated a combination of Semperfresh + *Aloe vera* gel coatings on ‘Kensington Pride’ mangoes and observed that coated fruit ripened more slowly; however, the coatings also reduced the development of aroma volatiles. Mohamed *et al.* (1992) found that guavas coated with Semperfresh and stored for 2 months at 10°C were in better condition than non-treated fruit, but coating fruit with palm oil was more effective, and cheaper, than Semperfresh. In a comparison between ‘Granny Smith’ apples that had been coated with either 1.5% Semperfresh or 1.5% Nutri-Save, Kader (1988) found that the internal CO<sub>2</sub> level was lower and the internal O<sub>2</sub> level was higher in the Semperfresh-coated fruits than those that had been coated with Nutri-Save. He also showed that coating with Nutri-Save reduced ethylene production and respiration rate to a larger degree than Semperfresh coating. From these results he concluded that Nutri-Save at 1.5% formed a surface barrier on the fruit that was less permeable to gases than 1.5% Semperfresh.

Arnon *et al.* (2014) coated mandarins, oranges and grapefruit with carboxymethyl cellulose (CMC) and chitosan and found that it was equally as effective as polyethylene wax in enhancing fruit gloss. It also slightly delayed softening, but was mostly not effective in reducing weight loss. There were only slight increases in internal CO<sub>2</sub> levels in fruit and ethanol accumulation in juice and flavour after storage was slightly impaired in mandarins but not in oranges and grapefruit. A coating similar to CMC was described by Deng *et al.* (2017). This was cellulose nanofibre-based emulsion (0.3% cellulose nanofibre + 1% oleic acid + 1% sucrose ester fatty acid, w/w on a wet basis), which they found reduced respiration rate and ethylene production in bananas, thus delaying their ripening. The coated bananas ripened normally during 10 days of ambient

storage; the coating simply delayed the ripening processes.

## Starch

Starches that have been used for fruit coatings include oxidized banana starch (Torres *et al.*, 2008), cassava starch (Oriani *et al.*, 2014; Brandelero *et al.*, 2016), corn starch (Peano *et al.*, 2014), rice starch (Kerdchoechuen *et al.*, 2011) and phosphorylated *Swartzia burchelli* seed starch (Gomes *et al.*, 2016). Films for MA packaging have also been prepared from cereal starches, e.g. Mater-Bi®. Kerdchoechuen *et al.* (2011) coated minimally processed pummelos with rice and cassava starches that had been mixed with either 5% or 10% pummelo juice as an anti-browning agent and stored them at 4°C for 21 days. The coatings had no effect on their colour, but the coated pummelos had a lower weight loss, higher sugar content, but lower TA and vitamin C compared with the non-coated pummelos. Strawberries were coated with 2% cassava starch, 1% chitosan, 2% cassava starch + 1% chitosan or left not coated and stored at 10°C packed in plastic hinged boxes. Those coated with cassava starch + chitosan had the lowest weight loss, lowest counts of yeast and psychrophilic microorganisms and the best appearance according to the sensory evaluation (Campos *et al.*, 2011). Oriani *et al.* (2014) coated apple slices with cassava starch at 0%, 2% or 3% (w/v) with or without fennel or cinnamon bark essential oils at 0.05–0.3% (v/v). The coatings reduced weight loss and respiration rate and the coatings that contained 0.3% cinnamon bark oil resulted in increased antioxidant capacity and inhibition of the growth of *Staphylococcus aureus* and *Salmonella choleraesuis*, while coatings containing fennel oil inhibited only *S. aureus*. Gomes *et al.* (2016) tested starch extracted from the seeds of *S. burchelli* that had been phosphorylated with sodium triphosphate. They coated cherry tomatoes and then stored them for up to 21 days at 10 ± 2°C and 80 ± 5% RH. Coated tomatoes retained better appearance with slower ripening and softening than non-treated fruit.

## Carrageenan

Carrageenan is a complex mixture of polysaccharides derived, by solvent extraction, from red seaweed after which the solvent is evaporated, forming a three-dimensional network of polysaccharide double helices (Karbowski *et al.*, 2007). Carrageenan-based coatings carrying antimicrobial or antioxidant compounds have been applied to a variety of foods to reduce microbial damage, oxidation, moisture loss and tissue disintegration (Vargas *et al.*, 2008). Hamzaha *et al.* (2013) tested various combinations of carrageenan films with or without glycerol on papayas. They found that 0.78% (w/v) carrageenan + 0.85% (w/v) glycerol delayed, but did not impede, ripening.

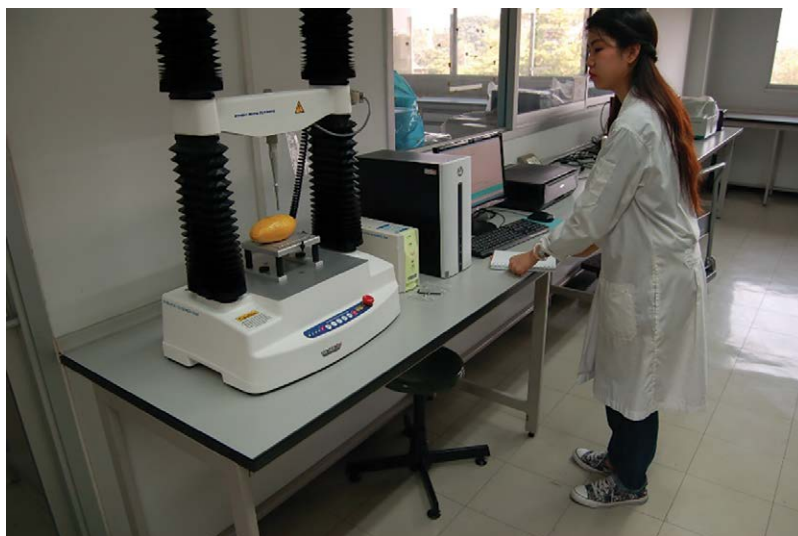
## Chitosan

Chitin is a natural polysaccharide that occurs in many organisms as crystalline microfibrils forming structural components in the exoskeleton of arthropods, the cell walls of fungi and yeast and in other living organisms. Chitosan is produced by partial deacetylation of chitin under alkaline conditions or by enzymatic hydrolysis in the presence of a chitin deacetylase. Chitosan can be used on fruit and vegetables to give a coating that is selectively permeable to water vapour, CO<sub>2</sub>, O<sub>2</sub> and ethylene (Sivakumar *et al.*, 2016). Chitosan coatings have been shown to have fungistatic properties and can modify the internal atmosphere in fruit (No *et al.*, 2007), but they must be dissolved in an acid to activate its antimicrobial and eliciting properties. Chitosan-based products are available commercially, e.g. Nutri-Save, which have shown similar effectiveness as the polymer dissolved in an acid. 'Golden Delicious' apples, coated with Nutri-Save, were kept in storage at 0°C for 6 months and were found to be of superior quality to those stored without coating (Davies *et al.*, 1988). Chitosan has been successfully used as a postharvest coating of peppers and cucumbers to reduce water loss and maintain their quality (El-Ghaouth

*et al.*, 1991) and has been shown to reduce the growth of decay-causing fungi, induce resistance responses in host tissues and produce other effects contributing to reduced decay (Romanazzi *et al.*, 2012). It has been successfully used to control brown rot (*Monilinia fructicola*) on peaches (Li and Yu, 2001), *M. fructicola* infections on nectarines (Casals *et al.*, 2012), grey mould in inoculated grapes (Romanazzi *et al.*, 2006) and reduce the decay of blueberry during storage (Duan *et al.*, 2011). Chitosan has other properties and has been shown to improve the postharvest life of apples (Davies *et al.*, 1988; Kader, 1988), peppers and cucumbers (El-Ghaouth *et al.*, 1991), longans (Jiang and Li, 2001), oyster mushrooms (Cho and Ha, 1998), breadfruit (Worrell *et al.*, 2002) and broccoli (Ansorena *et al.*, 2011). Coating of mangoes with chitosan has been the subject of continuing research in many countries (Fig. 7.3). Zhu *et al.* (2008) coated mangoes with 2% chitosan and found that the coating delayed ripening and reduced disease development in inoculated fruit during subsequent storage. Nongtaodum and Jangchud (2009) found that chitosan coatings (0.5% and 0.8%) could maintain sensory attributes, reduce weight loss and slow changes in TSS, colour and growth of microorganisms. Conversely, chitosan was shown to have no positive effects on papaya (Maharaj and Sankat, 1990) and strawberries (Perdones *et al.*, 2012). Papayas coated in 1.5% Nutri-Save resulted in no significant difference in weight loss, abnormal browning and limited colour development compared with non-treated fruit during storage at 16°C (Maharaj and Sankat, 1990). All the fruits were of the cultivar Tainung Number 1 and were harvested at the colour break stage and placed in hot water at 48°C for 20 min and then benomyl (1.5 g/l) at 52°C for 2 min.

## Essential Oils

Essential oils have been used both as fruit coatings and also successfully as volatiles inside MA packaging (see Chapter 14).



**Fig. 7.3.** Texture of mangoes being tested after coating with chitosan and storage at room temperature at KMITL in Thailand.

Essential oils extracted from the peel of oranges have fungitoxic properties (Sharma and Tripathi, 2006) and essential oils from aromatic plants such as oregano, fennel and lavender have also been reported to control, to a certain degree, postharvest citrus fruit pathogens (Soylu *et al.*, 2005). Kouassi *et al.* (2012) reported good disease control when *Cinnamomum zeylanicum* essential oils were incorporated in shellac or carnauba wax, but were less effective when they were incorporated into paraffin wax or polyethylene wax formulations. Disease control by essential oils appeared to depend on the volume that remained on the fruit skin and probably on the retention of essential oils components. Hashemi *et al.* (2017) coated fresh-cut apricots with basil seed gum containing 6% oregano essential oil and stored them at 4°C for 6 days. The addition of essential oil significantly decreased water vapour permeability of the coating. In comparison with the non-coated apricots, total soluble phenolic and antioxidant activity of oregano essential oil-coated fruit was significantly higher after storage. Taste panels could perceive rosemary and peppermint on grapes, which had been exposed to their essential oils for 24 h, after subsequent storage at 4°C or 20°C for 24 h, but after 48 h

there was no perception (Servili *et al.*, 2017). Both essential oils showed good control of grey mould.

### Alginate

Various forms of alginate gums are used as fruit coatings, including alginic acid, sodium alginate (extracted from the cell walls of brown algae) and potassium alginate (extracted from seaweed). Díaz-Mula *et al.* (2012) coated cherries with sodium alginate (1%, 3% or 5% w/v) and stored them for up to 16 days at 2°C plus 2 days at 20°C. Alginate-coated cherries could be stored, retaining their optimal quality and enhanced antioxidant activity, for 16 days while those that had not been coated could be stored for only 8 days. Valero *et al.* (2013) coated plums with 1% or 3% alginate before storage at 2°C for 7, 14, 21, 28 or 35 days followed by a 3-day period at 20°C. Both treatments, especially 3%, were effective in inhibiting ethylene production, reducing weight loss, improving acidity retention and slowing softening and colour change, but coated fruit had lower anthocyanin and carotenoid contents. Various concentrations of alginate

were combined with citral to coat fresh strawberries and fresh-cut apples. The concentrations used were alginate 1%, alginate 2%, alginate 1% + citral 0.15%, alginate 1% + citral 0.3%, alginate 2% + citral 0.15% and alginate 2% + citral 0.3%, all on a w/v basis. The fruits were then stored for 28 days at 0.5°C (apples) and 14 days at 4°C (strawberries). Antioxidant activity was improved with 1% alginate followed by 1% alginate + 0.3% citral. The highest reductions in microbial spoilage were with 2% alginate with either 0.15% or 0.3% citral concentrations. For strawberries treated with alginate at 2% with citral 0.3%, colour and firmness were better preserved and microbial spoilage was reduced but the TSS, weight loss and antioxidant activity were not affected. Taste panels scored 0.3% citral-coated fruit lowest, therefore alginate 2% with 0.15% citral was the most appropriate among the combinations studied (Guerreiro *et al.*, 2015). Citral is an alkene-type terpene that occurs in the oil of lemon, orange and lemon grass and is used in perfumery and as a flavour. It has strong antimicrobial properties (Onawunmi, 1989; Lawless, 2014).

### Salicylic Acid

Salicylic acid has been used for medical purposes (aspirin) for over a century. It is extracted from willow trees (*Salix* spp.) and also occurs in the fruit of *Ribes*, *Fragaria* and *Rubus* (Hulme, 1970). It has been investigated for its effects on plant material. For example, immersing peaches in salicylic acid at 0.1 g/l delayed their peak of ethylene production, reduced electrolyte leakage (chilling injury) and polyphenol oxidase (PPO) activity during storage compared with fruits that had not been treated (Han *et al.*, 2000). Luo *et al.* (2012) found that treating Japanese plums with salicylic acid suppressed chilling injury, reduced respiration rate and ethylene production and delayed the onset of the climacteric peak. Sayyari *et al.* (2009) also found that dipping fruit in salicylic acid, especially at 2 mM concentration, was highly effective in reducing chilling injury and

electrolyte leakage in the peel of pomegranate, as well as ascorbic acid loss, compared with that observed in non-treated fruit during storage at 2°C for 3 months. Sayyari *et al.* (2012) subsequently reported that dipping fruit in 0.1, 0.5 or 1.0 mM acetyl salicylic acid immediately after harvest reduced chilling injury and helped to maintain quality during storage at 2°C. Fruit that had been treated with acetyl salicylic acid had improved retention of sugars, organic acids, total phenolics and anthocyanins. Pan *et al.* (2007) reported that treatment with hydrogen peroxide, salicylic acid or methyl jasmonate alleviated chilling injury symptoms during storage of bananas. They also found that all these treatments could retard the change of the fruit colour and increase PPO activity, relative electrical conductivity and respiration rate.

Xu and Tian (2008) found that 2 mM salicylic acid did not inhibit *Penicillium expansum* growth *in vitro* but it activated antioxidant defence responses of fruit of the sweet cherry cultivar 'Hongdeng', which may play a role in the resistance against *P. expansum*. However, Yao and Tian (2005) found that 2 mM salicylic acid was toxic to *Monilinia fructicola* on cherries and significantly inhibited mycelial growth and spore germination of the pathogen *in vitro*. Dessalegn *et al.* (2013) treated mangoes with either salicylic acid or potassium phosphate at 1000 mg/l combined with dipping in a sodium bicarbonate solution for 3 min at 51.5°C. Both treatments significantly reduced disease development compared with the sodium bicarbonate hot water treatment alone or those not treated. These combinations kept anthracnose (*Colletotrichum gloeosporioides*) lesion development to below 5% and resulted in 93.3% marketable fruit compared with the mean disease severity on untreated fruit that reached 100% with zero marketability within 12 days of storage.

### Trehalose

Trehalose is sugar that has been tested as a coating material for food preservation (Roser and Colaco, 1993; Abboudi, A.H., 1993,

unpublished results). It is known as mycose or tremalose and is also referred to as mushroom sugar since it is prevalent in some mushrooms, including shiitake (*Lentinula edodes*), maitake (*Grifola fondosa*), nameko (*Pholiota nameko*) and Judas's ear (*Auricularia auricula-judae*), which can contain 1–17% of trehalose. It is a natural  $\alpha$ -linked disaccharide formed by an  $\alpha,\alpha$ -1,1-glucoside bond between two  $\alpha$ -glucose units. This makes trehalose a very stable reducing sugar that is chemically quite inert and biologically non-toxic. It is thought to form a gel phase as cells dehydrate. These properties make it a useful preservative. The use of trehalose in preservation of dehydrated material is covered by the European patent 0 297 887 B1 22 April 1992. Li and Tian (2007) reported that treatment with trehalose could help to retain the viability of the biocontrol yeast *Cryptococcus laurentii* in controlling postharvest diseases of apples. They also found that citric acid could induce accumulation of intracellular trehalose in *C. laurentii*.

### Gum Arabic

Gum arabic is the dried, gummy exudates from the stems and branches of *Acacia* spp. and is used as an edible coating in the food industry. Coating papayas with 10% gum arabic significantly slowed the skin colour changes, retained their firmness and reduced weight loss compared with non-coated fruits (Ali *et al.*, 2010). Ali *et al.* (2010) coated tomatoes with aqueous solutions of 5%, 10%, 15% or 20% gum arabic and then stored them at 20°C and 80–90% RH for 20 days. Coated tomatoes had lower weight loss, were firmer and had higher TA, TSS, AA and slower decay and colour development compared with non-coated fruit, with 10% gum arabic proving optimum. Sensory evaluation tests showed the 10% gum arabic coating had no detrimental effects on quality during storage. Subsequently Ali *et al.* (2013) showed that coated tomatoes had a lower respiration rate and ethylene production and their total antioxidant capacity, lycopene

content, total phenolics and total carotenoids were higher than fruit treated with 5% gum arabic or not coated. Mahfoudhi and Hamdi (2015) stored cherries at 2°C and 90–95% RH for 15 days coated with 10% gum arabic, 10% almond gum or not coated. Both coatings decreased respiration rate and ethylene production and delayed changes in weight, firmness, TA, TSS and colour changes compared with non-coated fruit. Green-mature papayas were coated with either 5% or 10% gum arabic in aqueous solutions. They were then stored at  $13 \pm 1^\circ\text{C}$  by Addai *et al.* (2013). Fruit coated with 10% gum arabic and stored for 7 days had significantly ( $p < 0.05$ ) higher antioxidant capacity (TPC, TFC, FRAP, DPPH and ABTS), physico-chemical properties (colour, pH, TA, moisture and TSS), sensory scores and lower microbial levels than 5% gum arabic and non-coated fruit.

Maqbool *et al.* (2011) dipped bananas in a composite coating of 10% gum arabic + 1% chitosan. They showed that, after 33 days of storage, coated fruit had 24% lower weight loss and 54% lower TSS than those not coated and were firmer, had higher total carbohydrates and reducing sugars than those not coated. Subsequently Maqbool *et al.* (2012) found that postharvest treatment with 0.05% lemongrass oil at 0.05% or cinnamon oil at 0.4% showed fungicidal effects against *C. gloeosporioides*, but the combination of 10% gum arabic with 0.4% cinnamon oil gave even better control.

### Zein

Zein is an alcohol-soluble hydrophobic protein found in the endosperm of maize, where it comprises about 45–50% of its protein. It is biodegradable and used on confectionery products. It has good film-forming properties and is a good barrier to  $\text{O}_2$ . The water vapour permeability of zein films was reported to be about 800 times higher than a 'typical shrink wrapping film' (Aydt *et al.*, 1991). Permeability to  $\text{CO}_2$ ,  $\text{O}_2$  and water vapour was strongly dependent on the concentration of zein in the coating

and the internal CO<sub>2</sub> of 'Gala' apples ranged from 4 to 11 kPa and O<sub>2</sub> 6–19 kPa for zein-coated fruit, with an optimum formulation of 10% zein and 10% propylene glycol (Bai *et al.*, 2003). Coatings that are made from zein may swell and deform in prolonged contact with water, which may 'restrict their use in MA packaging of fruit and vegetables' (Yoshino *et al.*, 2002). Park *et al.* (1994) reported that application of zein coating on tomatoes delayed colour change, weight loss and softening during storage. Chávez-Murillo *et al.* (2015) produced a zein and ethyl cellulose-based coating and tested it on tomatoes. They also found that it decreased their weight loss and respiration rate and delayed changes in colour and softening compared with non-coated fruit. Rakotonirainy *et al.* (2008) reported that broccoli florets packaged in zein films (laminated or coated with tung oil) maintained their original firmness and colour during 6 days of storage. Bai *et al.* (2003) tested zein dissolved in aqueous alcohol with propylene glycol to give a high-gloss coating for apples. Increasing levels of both zein and propylene glycol resulted in increased gloss.

### Methyl Jasmonate

MJ is a plant growth regulator involved in the antioxidant systems used to ameliorate oxidative reactions induced by biotic and abiotic stress. It is volatile and can diffuse through plant membranes and may have a function in mediating intra-plant and inter-plant communications and modulating plant defence responses (Wasternack and Hause, 2013). It appears to up-regulate genes that induce the synthesis of some stress proteins that affect chilling tolerance and resistance to pathogens (Ding *et al.*, 2002; Peng *et al.*, 2014). MJ significantly inhibited lignification of kiwifruit at later stages of storage at 0°C (Li *et al.*, 2017a). Exposure of Sunrise papayas to MJ vapours (10<sup>-5</sup> or 10<sup>-4</sup> M) for 16 hours at 20°C inhibited fungal decay, reduced chilling injury, retained higher levels of organic acids and delayed softening during storage at 10°C for 14–32 days

followed by 4 days shelf-life at 20°C compared with non-treated fruit. MJ-treated fruit also retained higher organic acids (González-Aguilar *et al.*, 2003). It was concluded that the postharvest quality of papayas was significantly enhanced by combining MJ and MA packaging in low-density polyethylene (LDPE) film. Pan *et al.* (2010) reported that treatment of bananas with MJ could alleviate chilling injury symptoms, retard the change of the fruit colour and increase PPO activity and respiration rate. Development of chilling injury in loquats was associated with a reduction in unsaturated:saturated fatty acid ratio. Postharvest treatment with MJ or 1-MCP delayed the decrease in this ratio, thereby increasing chilling resistance (Cao *et al.*, 2009 a, b). MJ (10 µM) reduced hydrogen peroxide in cucumber during storage, suggesting that it alleviated chilling injury by inhibiting hydrogen peroxide generation (Liu *et al.*, 2016b).

### Gibberellins

Gibberellins are a group of plant growth substances that were first isolated from the fungus *Gibberella fujikuroi*. They are identified by numbers, e.g. gibberellic acid = GA<sub>3</sub>. Postharvest treatment with gibberellins has been shown to reduce de-greening in lemons (Mizobutsi *et al.*, 2000) and limes (Sposito *et al.*, 2000), improve retention of vitamin C in limes (Keleg *et al.*, 2003), reduce yellowing and extend storage life of parsley (Lers *et al.*, 1998), reduce chilling injury in persimmon (Besada *et al.*, 2008), decrease cracking in cherries (Yildirim and Koyuncu, 2010) and delay ripening and reduce decay in cactus pear (Schirra *et al.*, 1999). A negative effect of GA<sub>3</sub> was reported by Zoffoli *et al.* (2009), who found that it could induce shatter and predisposed grapes to grey mould (*Botrytis cinerea*).

Biton *et al.* (2014) tested the effects of spraying persimmon plants with a mixture of GA<sub>4</sub>, GA<sub>7</sub> and benzyl adenine, starting 40 days after fruit set, applied once a month for three consecutive months, on subsequent storage at 0°C. This treatment combination



delayed chlorophyll degradation and reduced fruit cuticle cracking and black spot disease (*Alternaria alternata*) susceptibility during the late stages of fruit growth and during storage.

## Bacteria

Biological control of postharvest diseases is becoming of increasing importance. Some bacteria isolated from fruit can be used to control human pathogens that may contaminate minimally processed fruit. For example, the strain CPA-7 of *Pseudomonas graminis*, isolated from fruit, can inhibit the growth of human pathogens including *Salmonella enterica* subsp. *enterica*, *Listeria monocytogenes* and *Escherichia coli* O157:H7 (García-Martin *et al.*, 2018).

Alegre Vilas *et al.* (2013) confirmed previous work that showed *P. graminis* was effective in controlling bacteria and reported that concentrations of *P. graminis* CPA-7 of 105 and 107 cfu/ml extracted from apples avoided *Salmonella* growth at 10°C and lowered *L. monocytogenes* population increases. They also found no effect on fruit quality and the bacteria did not survive to simulated gastric stress. They concluded that it could be used during storage as part of a combination with disinfection, low storage temperature and modified atmosphere packaging (MAP). Collazo *et al.* (2017) confirmed that *P. graminis* CPA-7 treatment had no effect on the fruit quality parameters that were tested and their results suggested that *P. graminis* CPA-7 may trigger the activation of the fruit defence response, thereby mitigating its oxidative damage. Such activation may play a role in the control mechanism against pathogenic bacteria. Previously they had shown that on minimally processed pears *P. graminis* CPA-7 inhibited the growth of *L. monocytogenes* and *S. enterica* subsp. *enterica* by 5.5 and 3.1 log<sub>10</sub>, respectively, after 7 days of storage at 10°C.

## Daminozide

*N*-dimethylaminosuccinamic acid is marketed as DPA, daminozide, Alar, B9 or B995

and is a plant growth regulator that has been used to retard tissue senescence and cell elongation. However, it has been withdrawn from the market in several countries because of suggestions that it might be carcinogenic. In a comparison between pre-harvest and postharvest application of daminozide to 'Cox's Orange Pippin' apples, immersion of fruit in a solution containing 4.25 g/l for 5 min delayed the rise in ethylene production at 15°C by about 2 days, whereas an orchard application of 0.85 g/l resulted in delays of about 3 days. Both modes of application depressed the maximum rate of ethylene production by about 30%. Daminozide-treated apples were shown to be less sensitive to the application of ethylene than non-treated fruit, but this response varied between cultivars (Knee and Looney, 1990). Graell and Recasens (1992) sprayed 'Starking Delicious' apple trees with daminozide (1000 mg/l) in mid-summer in 1987 and 1988. Controls were not sprayed. Fruit was harvested on 12 and 24 September 1987, and 7 and 17 September 1988, and placed in CA storage. Two CA storage rooms were used: a low-ethylene CA room, fitted with a continuous ethylene removal system, with a store ethylene concentration ranging from 4 to 15 µl/l; and a high-ethylene CA room, where ethylene was not scrubbed and store ethylene concentration was > 100 µl/l. In both seasons, after 8 months at 0–1°C in storage in 3 kPa O<sub>2</sub> + 4 kPa CO<sub>2</sub>, with scrubbed ethylene, daminozide-sprayed and earlier-harvested fruits were firmer and more acid after storage than non-sprayed or later-harvested fruits stored in high ethylene CA. Generally, after a 7-day post-storage shelf-life period at 20°C, the differences between treatments were maintained. The TSS content of fruit samples was similar for all treatments.

## 1-Methylcyclopropene

Since the discovery of the properties of 1-MCP as a postharvest ethylene inhibitor, there has been an explosion of relevant publications. 1-MCP is applied to fruit, vegetables and flowers, but mainly apples, as a fumigant for a short period directly after

harvest to preserve quality and reduce losses during subsequent storage. However, Warren *et al.* (2009) applied it as an aqueous dip and Jiang *et al.* (1999) and De Reuck *et al.* (2009) sealed 1-MCP in plastic bags with bananas or litchis. 1-MCP is available as commercial preparations, including Smart-Fresh™ and Ethylbloc™. It can be applied to the store on completion of loading or in containers or gas-tight tents prior to loading into the store, or it can be applied in gas-tight rooms. Gas-tight curtains are available that can be used temporarily to partition part of a store, making it independently gas tight with access through a door closed with a gas-tight zip. This is useful where only a relatively small amount of fruit or vegetables needs to be fumigated. Pallet covers are also available that are sufficiently gas tight for fumigation.

Ma *et al.* (2009) reported that the mode of action of 1-MCP was by inhibiting the activities of ACC and delaying the peaks in the ACC synthase activity and ACC concentration and gene expression of these enzymes and of ethylene receptors at the transcript level. Fu *et al.* (2007) indicated that the beneficial effect of 1-MCP might be due to its ability to increase the antioxidant potential as well as to delay fruit ripening and senescence. They found that it retarded the activities of pectin methylesterase and polygalacturonase during ripening of pears and the activities of the antioxidant enzymes catalase, superoxide dismutase and peroxidase were all significantly higher in 1-MCP-treated pears than in the control fruit. Kesari *et al.* (2010) reported that ethylene exposure of unripe mature bananas induced the expression of the gene *MaPR1a*, which increased with ripening, and 1-MCP treatment prior to ethylene exposure inhibited expression. Liu *et al.* (2016a) suggested that the mode of action of 1-MCP was related to the expression of acid transport genes, including *MdVHA-A*, *MdVHP* and *Ma1*. This was because 1-MCP maintained fruit acidity by regulating the balance between malate biosynthesis and degradation. 1-MCP delayed the postharvest loss of malate and citrate.

De Wild (2001) suggested that short-term CA storage might be replaced by 1-MCP treatment. Watkins and Nock (2004) supported

this view for apples, but, they stated, 'we do not advocate the use of 1-MCP as an alternative to CA storage for medium to long term periods. With increasing storage periods, the two technologies are always more effective when used in combination.' Rizzolo *et al.* (2005) also concluded that 1-MCP could not be a substitute for CA storage but could reinforce the CA effects. Johnson (2008) reported that for apples the greatest benefit from 1-MCP was derived when combined with CA storage, though worthwhile extensions in storage life were achieved in air storage of 1-MCP-treated fruit. 1-MCP is an ethylene inhibitor but in certain circumstances treatment can stimulate ethylene production (Jansasithorn and Kanlavanarat, 2006; McCollum and Maul, 2009). Shinjiro *et al.* (2002) found that treatment of immature, mature-green or ripe pineapples with 1-MCP applied several times during storage resulted in temporary stimulation of ethylene production immediately after each treatment. Most of the reported effects of 1-MCP treatment have been positive, but Ella *et al.* (2003) reported that the use of too low concentrations may lead to some degree of senescence acceleration in leafy vegetables, possibly because of relief from ethylene auto-inhibition. Also in certain circumstances 1-MCP treatment has been associated with decreased quality of fruit (Fabi *et al.*, 2007) and increased rotting (Hofman *et al.*, 2001; Jiang *et al.*, 2001; Bellincontro *et al.*, 2006; Baldwin *et al.*, 2007). Jiang *et al.* (2001) suggested that comparatively low levels of phenolics in strawberries treated with 1-MCP could account for the decreased disease resistance. With 'Cavendish' bananas harvested at the mature-green stage, treatment with 1-MCP delayed peel colour change and fruit softening, extended shelf-life and reduced respiration rate and ethylene production (Jiang *et al.*, 1999). Ripening was delayed when fruits were exposed to 1-MCP at 0.01–1.0 µl/l for 24 h and increasing concentrations of 1-MCP were generally more effective for longer periods of time. Similar results were obtained with fruits sealed in PE bags 0.03 mm thick containing 1-MCP at either 0.5 or 1.0 µl/l, but delays in ripening were longer at about 58 days.

In their review, Li *et al.* (2016b) reported that the main observed effects of 1-MCP on postharvest performance of non-climacteric fruits were that they inhibited:

- senescence, including as rachis browning in grapes, scale senescence in pitaya, and leaf senescence in mandarins marketed with attached leaves;
- the development of physiological disorders, including scald development in pomegranate, pericarp browning in litchi and internal browning in loquat; and
- de-greening and colour change in citrus fruits, strawberry, pitaya, prickly pear and olive.

### Legislation

It may be surprising that permission has been received for the use of 1-MCP as a postharvest treatment when there is considerable public resistance to any chemical treatment of fruit and vegetables, especially postharvest. In the USA (US Federal Register, 2002) it was stated:

This regulation establishes an exemption from the requirement of a tolerance for residues of 1-methylcyclopropene (1-MCP) in or on fruits and vegetables when used as a post harvest plant growth regulator, i.e., for the purpose of inhibiting the effects of ethylene. AgroFresh, Inc. (formerly BioTechnologies for Horticulture) submitted a petition to EPA under the Federal Food, Drug, and Cosmetic Act, as amended by the Food Quality Protection Act of 1996, requesting an exemption from the requirement of a tolerance. This regulation eliminates the need to establish a maximum permissible level for residues of 1-MCP.

The overall conclusion from the evaluation reported on 23 September 2005 by European Commission Health & Consumer Protection Directorate-General (European Commission, 2005) was:

that it may be expected that plant protection products containing 1-methylcyclopropene will fulfil the safety requirements laid down in Article 5(1) (a) and (b) of Directive 91/414/EEC.

As stated by the World Trade Organization (WTO, 2008):

The World Trade Organization does not expect any human health concerns from exposure to residues of 1-MCP when applied or used as directed on the label and in accordance with good agricultural practices. The data submitted by applicant and reviewed by the Agency support the petition for an exemption from the requirement of a tolerance, for 1-MCP on pre-harvested fruits and vegetable, when the product is applied or used as directed on the label. This regulation is effective 9 April 2008.' (WTO, 2008).

### Diseases

The effects of 1-MCP on postharvest disease development are variable. Li *et al.* (2017c) reported that 1-MCP (5 µl/l) suppressed postharvest blue mould (*Penicillium expansum*) of apples by inhibiting the growth of mycelium. Xu *et al.* (2017) showed that 1-MCP suppressed postharvest development of anthracnose disease (*Colletotrichum gloeosporioides*) in mangoes by directly inhibiting spore germination and mycelial growth. However, tomatoes were artificially inoculated with an *Alternaria alternata* conidial suspension then exposed to 1-MCP at 0.6 µl/l for 24 h and stored for 6 days at 10°C and one more week at 20°C. After storage, tomatoes treated with 1-MCP showed significantly more *A. alternata* disease (Estiarte *et al.*, 2016). Jiang *et al.* (2001) found that disease development was quicker in strawberries treated with 500 and 1000 nl/l. 1-MCP inhibited PAL activity and reduced increases in anthocyanin and phenolic content. They suggested that comparatively low levels of phenolics in strawberries treated with 1-MCP could account for their decreased disease resistance.

### Apple

Watkins and Nock (2004), in discussing the beneficial effects of 1-MCP on apples, pointed out that the response can vary

between cultivars as well as harvest maturity, handling practices, time between harvest and treatment and type and duration of storage. On the negative side they found that 1-MCP may increase susceptibility to chilling injury, CO<sub>2</sub> injury and superficial scald. Ethylene production, respiration rate and colour changes in 'Fuji' apples were all inhibited following 1-MCP treatment at 0.45 mmol/m<sup>3</sup> (Fan *et al.*, 1999). The threshold concentration of 1-MCP inhibiting *de novo* ethylene production and action was 1 µl/l. 1-MCP-treated 'McIntosh' and 'Delicious' apples were significantly firmer than non-treated apples following cold storage and post-storage exposure to 20°C for 7–14 days (Rupasinghe *et al.*, 2000). De Wild (2001) made a single application of 1-MCP directly after harvesting to 'Jonagold' and 'Golden Delicious' apples, followed by 6 months storage at 1°C in either 1.2 kPa O<sub>2</sub> + 4 kPa CO<sub>2</sub> or air, followed by shelf-life of 7 days at 18°C. No loss of firmness was found after the use of 1-MCP and yellow discoloration of 'Jonagold' decreased. Apples stored in air softened quicker than in CA, unless treated with 1-MCP. Treated apples did not show significant softening during subsequent shelf-life. Exposure to 1-MCP improved retention of flesh firmness and reduced ethylene production of 'Scarletspur Delicious' and 'Gale Gala' after storage in 0°C both in air and in 2 kPa O<sub>2</sub> + < 2 kPa CO<sub>2</sub> (Drake *et al.*, 2006). In 'McIntosh' apples, 1-MCP treatment reduced internal ethylene concentrations and maintained firmness of fruit during storage compared with the non-treated controls, but the location of the orchard where the apples were grown affected their response to 1-MCP. At one site 'McIntosh' treated with 1-MCP had more internal browning, but there was no effect at another (Robinson *et al.*, 2006). In 'Pink Lady' apples, 1-MCP treatment at 0.5 µl/l at 20°C for 24 h significantly inhibited respiration rate, ethylene production and ACC oxidase activity while it slowed down the decrease in flesh firmness and TA during storage at 0°C (Guo *et al.*, 2007). Internal ethylene concentration in 'Bramley's Seedling' also tended to be higher where treatment was delayed and fruit were softer. In general a lower dose

combined with a shorter treatment time resulted in the highest internal ethylene concentration in 'Cox's Orange Pippin' and 'Bramley's Seedling' and a lower dose in combination with delayed treatment resulted in the highest internal ethylene concentration and lowest firmness in 'Cox's Orange Pippin' (Johnson, 2008). Moya-Leon *et al.* (2007) stored 'Royal Gala' at 0°C and 90–95% RH for 6 months followed by 7 days at ambient. They found that fruit stored in 2.0–2.5 kPa CO<sub>2</sub> + 1.8–2.0 kPa O<sub>2</sub> or those treated with 1-MCP at 625 nl/l and stored in air had reduced ethylene production, softening and acidity loss and were preferred by consumers because their preferred textural characteristics were said to be maintained. However, storage in air gave the highest levels of aromatic volatiles, which were depressed by both CA storage and 1-MCP treatment. 1-MCP at 42 mol/m<sup>3</sup> for 24 h at 20°C inhibited ethylene production regardless of storage temperature over the range of 0–20°C. Results indicated that 1-MCP is an effective means that can be used to delay ripening and to retain fruit quality during transport and retailing at 10°C or 20°C. Fruit stored at 20°C for longer than 40 days showed some shrivelling and decay, regardless of 1-MCP treatment (Argenta *et al.*, 2001).

Combining 1-MCP treatment with DCA has been shown to be effective. Storage of 'Braeburn' at 1°C in DCA at 0.3–0.4 kPa O<sub>2</sub> + 0.7–0.8 kPa CO<sub>2</sub> maintained firmness at levels comparable to 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> plus 1-MCP applied at 625 ppb for 24 h at 3.5°C. Storage was for some 7 months, followed by 7 days of shelf-life at 20°C. DCA storage plus 1-MCP resulted in no softening during storage, but 1-MCP-treated fruit in both 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> and DCA showed a higher incidence of BBD (on average + 53%) than non-treated fruit. DCA storage reduced BBD to 46% compared with fruit stored only in 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>. Fruit susceptibility to low-temperature breakdown and CO<sub>2</sub> injury was significantly higher in DCA storage than in 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> or in air (Lafer, 2008).

The protocol recommended by Johnson (2008) for apples grown in the UK was

1-MCP at 625 nl/l (based on empty store volume) applied for 24 h with minimal delay between harvest and application. He reported that differences in firmness between air and CA stored 'Cox's Orange Pippin' increased with time in store due to the more rapid softening of fruit stored in air. Only storage in 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> gave complete control of scald in 1-MCP-treated 'Bramley's Seedling' apples stored for long periods. Delaying establishment of CA conditions from 10 to 28 days to avoid CO<sub>2</sub> injury in 'Bramley's Seedling' did not affect internal ethylene concentration, scald incidence or firmness in 2002 but in 2003 there was an adverse effect of a 21-day delay on the firmness of fruit stored in 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for 284 days but no effects on scald development. 'Fiesta', 'Jonagold', 'Idared' and 'Braeburn' treated with 1-MCP failed to soften during 90 days of CA storage despite a reduced internal ethylene concentration. 'Meridian' apples treated with 1-MCP were not significantly firmer than non-treated and 'Spartan' did not respond to 1-MCP in terms of either internal ethylene concentration or firmness. Bulens *et al.* (2012) found that when 'Jonagold' were treated with 1-MCP, ethylene biosynthesis was almost completely suppressed throughout their whole postharvest life, with the exception of late-harvested apples that regained some ethylene-forming capacity after 2 weeks of shelf-life.

As reported above, the development of physiological storage disorders has also been associated with 1-MCP treatment. Watkins and Nock (2012b) found that its effects on storage disorders were inconsistent, though they detected slight increases in the risk of CO<sub>2</sub> injury. DeEll *et al.* (2005) reported that 1-MCP resulted in CO<sub>2</sub> injury in 'McIntosh' and 'Empire', and a higher incidence of core browning in 'Empire' and internal browning in 'Gala'. 'Gala' with no 1-MCP in storage at 0°C for 240 days with zero CO<sub>2</sub> resulted in a large reduction in firmness, but 'Gala' in storage in 1 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> had similar respiration rates, ethylene and total volatiles production to those fruits stored in 2.5 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub>. 'Empire' stored either in air at 0°C or in CA of 5.0% CO<sub>2</sub> + 3.0% O<sub>2</sub> at 2.5°C developed large incidences of core browning, which was worse in 1-MCP-

treated fruits. Subsequently DeEll and Ehsani-Moghaddam (2012) suggested that delaying CA storage for 1 or 2 months at 1°C to reduce the development of certain storage disorders in 'Empire' could be utilized in combination with 1-MCP treatment to retard fruit softening during storage at 3°C with 2.5 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub> for 7–9 months.

Li *et al.* (2017c) showed that fumigation with 1-MCP (5 µl/l) inhibited postharvest blue mould (*Penicillium expansum*) of apples during subsequent storage. The mode of action of the 1-MCP was thought to be due to the enhanced oxidative damage to *P. expansum* and destruction of the integrity of plasma membrane of its spores

### Apricot

Fan *et al.* (2000) treated the cultivar 'Perfection' with 1-MCP at 1 µl/l for 4 h at 20°C then stored them at either 0°C or 20°C. The 1-MCP treatment reduced ethylene production and respiration rate and resulted in slower colour change, less firmness loss and loss in TA during storage at both temperatures. There was also delayed production of alcohols and esters during storage at 20°C. They also found a relationship between 1-MCP and maturity of the fruit in that the effects of MCP were less pronounced at more advanced maturity.

### Asian pear

Cold storage of the cultivars 'Gold Nijisseiki' and 'Hosui' plus 1-MCP treatment was effective for long-term storage (Itai and Tanahashi, 2008).

### Avocado

'Ettinger', 'Hass', 'Reed' and 'Fuerte' were treated with various concentrations of 1-MCP for 24 h at 22°C and, after ventilation, were exposed to ethylene at 300 µl/l for 24 h also at 22°C. The fruits were then stored at 22°C in ethylene-free air for ripening assessment.

1-MCP at very low concentrations inhibited ripening. Treatment for 24 h with 1-MCP at 30–70 nl/l delayed ripening by 10–12 days, after which the fruit ripened normally (Feng *et al.*, 2000). Hofman *et al.* (2001) showed that 1-MCP at 25  $\mu$ l/l for 14 h at 20°C increased the ripening time by 4.4 days (40% increase) for ‘Hass’ compared with non-treated fruit. 1-MCP treatment was associated with slightly higher rot severity compared with fruit not treated with 1-MCP. They also showed that 1-MCP treatment could negate the effects of exogenous ethylene and concluded that 1-MCP treatment has the potential to reduce the risk of premature fruit ripening due to accidental exposure to ethylene.

### Banana

Williams *et al.* (2003) found that at 15°C bananas could be stored for 6 weeks in CA, but this was doubled by application of 1-MCP at 300 nl/l for 24 h at 22°C directly after ripening initiation with ethylene. However, 3 nl/l had no effect and 30  $\mu$ l/l slowed ripening ‘excessively’. Eating quality was not affected by 1-MCP at 300 nl/l (Klieber *et al.*, 2003). With ‘Cavendish’ bananas, treatment with 1-MCP delayed peel colour change and softening, extended shelf-life, reduced respiration rate and reduced ethylene production (Jiang *et al.*, 1999). Ripening was delayed when fruits were exposed to 1-MCP at 0.01–1.0  $\mu$ l/l for 24 h and increasing concentrations of 1-MCP were generally more effective for longer periods of time. Similar results were obtained with fruits sealed in PE bags 0.03 mm thick containing 1-MCP at either 0.5 or 1.0  $\mu$ l/l, but delays in ripening were longer at about 58 days. Jiang *et al.* (2004), working on the cultivar ‘Zhonggang’, found that fruits treated with 1-MCP for 24 h at 20°C had significantly delayed peaks of respiration rate and ethylene production, but the peak height of those exposed to ethylene at 50 ml/l was reduced. They also found a reduction in all the changes associated with ripening. The 1-MCP effects on all these changes were higher during low-temperature storage and lower in higher-temperature storage. Bananas treated with

1-MCP were shown to result in uneven de-greening of the peel. De Martino *et al.* (2007) initiated ‘Williams’ to ripen by exposure to ethylene at 200  $\mu$ l/l for 24 h at 20°C. These bananas were then treated with 1-MCP at 200 nl/l at 20°C for 24 h, then stored in  $> 99.9$  kPa  $N_2$  +  $< 0.1$  kPa  $O_2$  in perforated plastic bags at 20°C. No differences in the accumulation of acetaldehyde and ethanol were detected during storage compared with fruits not treated with 1-MCP. Peel de-greening, the decrease in chlorophyll content and chlorophyll fluorescence were delayed after the 1-MCP treatment. There was some general browning throughout the 1-MCP-treated peel in both the green and yellow areas of the ripening peel. They concluded that it appears the 1-MCP-treated peel 24 h after the ethylene treatment may still undertake some normal senescence that occurs during banana ripening. Green-mature ‘Apple’ bananas were treated with 1-MCP at 0, 50, 100, 150 and 200 nl/l for 12 h and then stored at room temperature of  $20 \pm 1^\circ\text{C}$  and  $80 \pm 5\%$  RH. The best concentration in extending the shelf-life based on total sugars, soluble pectins, firmness and external appearance in the end of storage was 1-MCP at 50 nl/l. This concentration delayed the onset of peel colour change by 8 days, while with concentrations at 100, 150 and 200 nl/l it was by 10 days compared with non-treated fruits (Pinheiro *et al.*, 2005). ‘Khai’ bananas were treated with 1-MCP at 50, 100 or 250 ppb for 24 h and then stored at 20°C. 1-MCP delayed ripening and prolonged storage to some 20 days, with the higher concentration being more effective, but bananas treated with 250 ppb 1-MCP had the lowest respiration rate and ethylene production. However, there was no significant difference among treatments for colour changes (Jansasithorn and Kanlavanarat, 2006).

### Blueberry

Changes in the marketable percentage of the cultivars ‘Burlington’ and ‘Coville’ after storage for up to 12 weeks at  $0 \pm 1^\circ\text{C}$  were not affected by pre-treatment for 24 h at 20°C with 1-MCP at rates up to 400 nl/l (DeLong *et al.*, 2003).

### Breadfruit

Treatment with 1-MCP at 100 nl/l for 12 h reduced respiration rate, but treatment at 10, 20 or 50 nl/l did not affect respiration rate or ethylene production. All the treatments tended to delay softening and improve shelf-life. However, at the later stage of ripening, some fruitlets were abnormally hard, causing uneven mushy softening with light brown streaks (Paull *et al.*, 2005). They also reported that fruits treated with 1-MCP at 25 or 50 nl/l followed by storage in perforated PE bags at 17°C had a shelf-life of 10–11 days compared with 4 days for non-treated fruits at the same temperature; and fruits treated with 1-MCP alone also had a 4-day shelf-life.

### Broccoli

Ku and Wills (1999) showed that 1-MCP markedly extended the storage life of the cultivar 'Green Belt' through a delay in the onset of yellowing during storage at either 20°C or 5°C and in development of rotting at 5°C. The beneficial effects at both temperatures were dependent upon concentration of 1-MCP and treatment time. Subsequently Ma *et al.* (2009) also showed that 1-MCP treatment delayed the yellowing of broccoli florets and that exogenous ethylene exposure did not accelerate yellowing in the florets pre-treated with 1-MCP. Xu *et al.* (2016) showed that broccoli florets treated with 1-MCP at 2.5 µl/l decreased the loss of chlorophyll and maintained higher levels of sugars. The expression of genes encoding sucrose transporters (*BoSUC1* and *BoSUC2*) and carbohydrate metabolizing enzymes (*BoINV1*, *BoHK1* and *BoHK2*) was induced by the 1-MCP treatment. Therefore 1-MCP was shown to delay senescence of broccoli florets, which may be attributed to maintaining higher sugar content through regulation of sugar metabolism.

### Capsicum

'Setubal' red peppers were treated with 900 ppb 1-MCP for 24 h at 20°C or left non-treated

and packed in perforated PP bags by Fernández-Trujillo *et al.* (2009). They were stored at either 20°C or 8°C for 4½ days, then at 20°C for 3 days and finally in a domestic refrigerator at 5.6°C for 4½ days to simulate retail distribution. 1-MCP prevented the increase in skin colour and ethylene production but it may have increased susceptibility to shrivelling and weight loss and, to a greater extent, pitting and grey mould.

### Carambola

Generally, submerging fruit in aqueous 1-MCP at 200 µg/l suppressed total volatile production in the cultivar Arkin at both ¼ and ½ yellow harvest maturity or when stored at 5°C for 14 days and transferred to 20°C or stored constantly at 20°C until turning ¼ orange. 1-MCP suppressed volatile production for fruit stored at 20°C where the fruits were harvested at the ¼ and ½ yellow stage and total volatiles were suppressed by 14% and 28%, respectively. Treatment with 1-MCP delayed the time for them to turn ¼ orange by 2–5 days at 20°C (Warren *et al.*, 2009).

### Sudachi (*Citrus sudachi*)

Sudachis were treated with 1-MCP at 0, 0.01, 0.1, 0.5 or 1.0 µl/l for 6, 12 or 24 h at 20°C and then stored at 20°C by Isshiki *et al.* (2005). Treatment with 1-MCP at 0.1 µl/l was the most effective in delaying de-greening and it did not induce peel browning and internal quality degradation. However, at 0.5 and 1.0 µl/l, 1-MCP had little effect on delaying de-greening or inducing peel browning.

### Cucumber

Lima *et al.* (2005) found that exposure of different cultivars to 1-MCP at 1.0 µl/l resulted in significant retention of firmness and surface colour in the cultivar 'Sweet Marketmore' but it had minimal effects on the cultivar Thunder. They found that those

exposed to ethylene at 10  $\mu\text{l/l}$  at 15°C showed rapid and acute cellular breakdown within 4 days; this effect was reduced, but not eliminated, by the 1-MCP treatment. Nilsson (2005) also found some delay in de-greening with 1-MCP, especially when cucumbers were exposed to ethylene. However, it was concluded that they probably showed little benefit from 1-MCP unless exogenous ethylene was present.

### Custard apple

Hofman *et al.* (2001) found that exposure of the *Annona reticulata* cultivar 'Africa Pride' to 1-MCP at 25  $\mu\text{l/l}$  for 14 h at 20°C increased the number of days to ripen by 58% compared with non-treated fruit. 1-MCP treatment was associated with slightly higher severity of external blemishes and slightly higher rot severity compared with non-treated fruit. Since the 1-MCP treatment countered the effects of exposure to exogenous ethylene, they concluded that it had the potential to reduce the risk of premature ripening due to accidental exposure to ethylene.

### Durian

Pulp of the *Durio zibethinus* cultivar 'Mon Thong' was exposed to 0, 50, 100, 200 or 500 ppb of 1-MCP for 12 h at 20°C and then stored in 15  $\mu\text{m}$  PVC film bags at 4°C by Sudto and Uthairatanakij (2007). After 8 days of storage at 20°C, treated fruits were firmer than non-treated ones. The most effective in delaying the accumulation of  $\text{CO}_2$  inside the packages was 50 ppb, but none of the treatments affected ethylene production and 1-MCP treatment had no significant effect on firmness. In contrast, they found that whole fruits that had been exposed to 1-MCP and stored at 20°C were firmer than non-treated ones after 8 days. Treatment with 1-MCP resulted in a reduction of the starch content, an increase of TSS and increased lipoxygenase activity. Also working with the cultivar 'Mon Thong', Amornputti *et al.* (2016) showed that 1-MCP (500  $\text{n}\text{l/l}$  at 25°C) inhibited ethylene biosynthesis during

subsequent storage at 15°C mainly by preventing an increase in ACC oxidase activity in the peel.

### Fig

Fruits of the cultivar 'Bardakci' were harvested at optimum harvest maturity and half were treated with 10 ppb 1-MCP at 20°C for 12 h and the other half left non-treated; both were stored at 0°C with 90–92% RH for 15 days by Gözlekçi *et al.* (2008). They found that the 1-MCP treatment slowed fruit softening but had no effect on TA and TSS.

### Grape

Grapes of the cultivar 'Aleatico' harvested at 20.4–20.6°brix were left non-treated or were treated with ethylene at 500  $\text{mg/l}$  or 1-MCP at 1  $\text{mg/l}$  for 15 h at 20°C and 95–100% RH, then stored at 20°C and 85% RH for 13 days. 1-MCP-treated grapes did not show ethylene production for 2 days after treatment, whereas non-treated grapes produced constant amounts of 3–4  $\mu\text{l/kg/h}$  but after 6 days. There were no significant differences in their respiration rates. 1-MCP-treated grapes showed a significant loss of terpenols and esters, though levels of alcohols were maintained, especially hexan-1-ol. Levels of grey mould (*Botrytis cinerea*) were also higher in 1-MCP treated fruit (Bellincontro *et al.*, 2006).

### Grapefruit

Treatment with 1-MCP at concentrations equal to or greater than 75  $\text{n}\text{l/l}$  inhibited ethylene-induced de-greening but the effect was transient, and increasing 1-MCP concentrations greater than 150  $\text{n}\text{l/l}$  did not cause additional inhibition of de-greening. 1-MCP at both 15 and 75  $\text{n}\text{l/l}$  resulted in a slight suppression of respiration rate while 150 or 300  $\text{n}\text{l/l}$  resulted in higher respiration rates than non-treated fruit. 1-MCP treatment also resulted in an increase in the rate of ethylene production that was dose and time dependent. The effects of 1-MCP on respiration rate



and ethylene production were significantly reduced if fruits were subsequently exposed to ethylene. Rates of ethylene evolution were some 200 nl/kg/h from control and ethylene-treated fruits compared with about 10,000 nl/kg/h from those treated with 1-MCP. Fruits treated with ethylene following 1-MCP treatment had ethylene production rates of about 400 nl/kg/h (McCollum and Maul, 2009).

### Guava

In the cultivar 'Allahabad Safeda', 1-MCP treatment at  $20 \pm 1^\circ\text{C}$  reduced ethylene production, respiration rate, changes in fruit firmness, TSS and TA during storage at both  $10^\circ\text{C}$  and  $25\text{--}29^\circ\text{C}$ , depending upon 1-MCP concentration and duration of exposure. Vitamin C content in 1-MCP-treated fruit was significantly higher and there was a significant reduction in the decay compared with non-treated fruit. Also the development of chilling injury symptoms was ameliorated to a greater extent in 1-MCP-treated fruit. It was reported that 1-MCP at 600 nl/l for 12 h and storage at  $10^\circ\text{C}$  seemed a promising way to extend the storage life while 1-MCP at 300 nl/l for 12 h and 24 h or 600 nl/l for 6 h could provide 4–5 days of extended marketability of fruit under ambient conditions of  $25\text{--}29^\circ\text{C}$  (Singh and Pal, 2008a).

### Kiwifruit

Kiwifruit are particularly susceptible to post-harvest exposure to ethylene. CA storage and 1-MCP treatment delayed kiwifruit softening compared with storage in air or with no 1-MCP treatment (Prencipe *et al.*, 2016). The application of 1-MCP at  $1 \mu\text{l/l}$  for 20 h at room temperature immediately after harvest showed beneficial effects in retaining quality during 3 months of storage at  $1^\circ\text{C}$ . Fruits treated with 1-MCP were firmer and had higher TSS than non-treated fruits. AA was higher in the non-treated fruits but taste panellists did not find significant differences between treatments, except that 1-MCP-treated fruits had better appearance (Antunes *et al.*, 2008). Li *et al.* (2017a) also reported

that treatment with 1-MCP maintained fruit firmness during cold storage, though with enhanced activity of PAL, cinnamyl-alcohol dehydrogenase and peroxidase as well as increased lignification. Lignification occurred after prolonged storage. They found that the expression levels of the lignin-related gene AcPOD1 were increased in fruit core tissue after treatment with 1-MCP, but treatment with MJ or methyl salicylate significantly inhibited lignification.

### Lettuce

The cultivar 'Butterhead' treated with 1-MCP had reduced respiration rates. Ethylene production and russet spotting were both retarded. Retention of AA was much better than in non-treated lettuce during 12 days of storage at either  $2^\circ\text{C}$  or  $6^\circ\text{C}$  (Tay and Perera, 2004).

### Lime

1-MCP conserved the colour of 'Tahiti' limes (*Citrus aurantifolia*) during 30 days of storage at  $10^\circ\text{C}$  (Jomori *et al.*, 2003). Win *et al.* (2006) showed that chlorophyllase and chlorophyll-degrading peroxidase activities in flavedo tissue were delayed in 1-MCP-treated fruit at concentrations of 250 and 500 nl/l. 1-MCP at 250 or 500 nl/l effectively suppressed endogenous ethylene production, but a concentration of 1000 nl/l accelerated yellowing within 9 days, while fruit treated at 750 nl/l turned completely yellow in 15 days. Ethylene production rate of fruits that had been treated with 1-MCP at 1000 nl/l was 1.6 times higher than that of non-treated fruits. Also AA content was reduced in fruits treated with 1-MCP at 1000 nl/l but not in fruits treated with 250, 500 or 750 nl/l.

### Litchi

Sivakumar and Korsten (2007) investigated the possibility of replacing  $\text{SO}_2$  fumigation

as a postharvest treatment with 1-MCP on the cultivar McLean's Red. They simulated export marketing conditions by fumigation with a moderate concentration of 1-MCP for 3 h, then storage in 17 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> for 21 days at 4°C and packaging in MA at 10°C to simulate marketing. They found that with these treatments fungal decay was controlled and the natural pinkish/red colour was retained. They also found that storage in 3 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> retained a deep red colour, showed limited yeast decay and only slightly affected the TSS, TA and AA contents. De Reuck *et al.* (2009) packed the cultivars 'Mauritius' and 'McLean's Red' in bi-oriented polypropylene (BOPP) bags and exposed them to 1-MCP at 300, 500 or 1000 nl/l within the packaging, heat sealed them and stored them at 2°C for up to 21 days. This combination of treatments was effective in preventing browning and retention of the pinkish-red colour in both cultivars. 1-MCP significantly reduced the PPO and peroxidase activity, retained membrane integrity and anthocyanin content and prevented the decline of pericarp colour during storage. At concentrations higher than 300 nl/l, 1-MCP had negative effects on membrane integrity, pericarp browning, PPO and peroxidase activity in both cultivars, but at 1000 nl/l there was a significant suppression of respiration rate and better retention of TSS, TA and firmness.

### Lotus

Li *et al.* (2017d) exposed lotus pods (*Nymphaeas lotus*) to 1-MCP at 0.5 µl/l or 3% lacquer wax and then stored them at 25 ± 1°C and 90–95% RH. Both 1-MCP and lacquer wax reduced browning and retained their quality, but lacquer wax had little influence on the respiration rate. 1-MCP-treated pods were firmer than non-treated pods for only 6 days. The combination of lacquer wax followed by 1-MCP gave equivalent effects on respiration rate and firmness to lacquer wax alone. However, treatment of lotus pods with 1-MCP first, followed by lacquer wax coating, gave better retention of firmness and total phenol concentration, and

reduced respiration rate, accumulation of MDA and browning. This treatment also increased superoxide dismutase (SOD) and catalase activities and decreased PPO and POD activity when compared with non-treated and those in the other treatments.

### Mango

Hofman *et al.* (2001) showed that 1-MCP at 25 µl/l for 14 h at 20°C increased the time to ripening by 5.1 days (37%) in the cultivar 'Kensington Pride' but at least doubled the severity of stem-end rot compared with non-treated fruit. However, they concluded that 1-MCP has the potential to reduce the risk of premature fruit ripening due to accidental exposure to ethylene. Mangoes of the cultivar 'Nam Dok Mai' were treated with 500 ppb 1-MCP and then stored at 13°C in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> by Kramchote *et al.* (2008). They reported that storage in CA was the most effective way to delay fruit yellowing, while 1-MCP was more effective in maintaining firmness and suppressing ethylene production. The production of aroma volatile compounds emitted after 28 days of storage showed the accumulation of high levels of ethanol for fruits in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> while 1-MCP mainly suppressed volatile emission. Wang *et al.* (2007a) harvested the cultivar 'Tainong' at the green-mature stage and treated them with 1-MCP at 1 µl/l for 24 h and stored them at 20°C for up to 16 days. 1-MCP maintained fruit firmness and AA but inhibited the activities of antioxidant enzymes including catalase, SOD and ascorbate peroxidase.

### Melon

Softening, respiration rate and ethylene production were reduced in the cultivar 'Hy-Mark' by 1-MCP, with 300 nl/l being the optimum rate. Treated melons were acceptable for 27 days, whereas the non-treated fruits could be stored for no longer than 7 days. 1-MCP-treated galia melons retained their firmness for up to 30 days and the postharvest

life of treated charentais melons was 15 days at room temperature ( $25 \pm 3^\circ\text{C}$  and  $65 \pm 5\%$  RH). In the cultivar 'Orange Flesh', 1-MCP delayed ripening and controlled decay caused by *Fusarium pallidoroseum* (Alves *et al.*, 2005). Sa *et al.* (2008) stored cantaloupes in X-tend film bags for 14 days at  $3 \pm 2^\circ\text{C}$ , then removed them from their bags and stored them for a further 8 days at room temperature ( $23 \pm 2^\circ\text{C}$  and  $90 \pm 2\%$  RH). The fruits retained their quality in the bags, but pre-treatment with 1-MCP or inclusion of potassium permanganate on vermiculite within the bags gave no additional effects.

### Onion

During storage at  $18^\circ\text{C}$ , exogenous ethylene suppressed sprout growth in the cultivar 'Copra' of both dormant and already-sprouting bulbs by inhibiting leaf blade elongation, but treatment of dormant bulbs with 1-MCP resulted in premature sprouting. The dormancy-breaking effect of 1-MCP indicates a regulatory role of endogenous ethylene in onion bulb dormancy (Bufler, 2009). In contrast, Choep *et al.* (2007) reported that sprout growth was reduced in bulbs treated with 1-MCP at  $1 \mu\text{l/l}$  for 24 h at  $20^\circ\text{C}$  and stored at  $4^\circ\text{C}$  or  $12^\circ\text{C}$ , but not when they were stored at  $20^\circ\text{C}$ . Higher concentrations of sucrose, glucose and fructose were found in 1-MCP-treated bulbs of the cultivar SS1 stored at  $12^\circ\text{C}$  and dry weight was maintained compared with non-treated bulbs.

### Papaya

Hofman *et al.* (2001) showed that exposure to 1-MCP at  $25 \mu\text{l/l}$  for 14 h at  $20^\circ\text{C}$  increased the number of days to ripening by 15.6 days (325%) in the cultivar 'Solo' compared with non-treated fruit. They concluded that 1-MCP treatment has the potential to reduce the risk of premature fruit ripening due to accidental exposure to ethylene. However, Fabi *et al.* (2007) compared non-treated, ethylene-treated or ethylene + 1-MCP-treated 'Golden' papayas and found that 1-MCP

could decrease the quality of fruit and that even the use of ethylene for initiating ripening could result in lower quality when compared with that of fruit allowed to ripen 'naturally'. Hofman *et al.* (2001) also found that 1-MCP could have a negative effect, since treatment was associated with slightly higher severity of external blemishes and slightly higher severity of rots compared with those not treated. Zerpa-Catanho *et al.* (2017) reported that polygalacturonase and endoxylanase correlated negatively with firmness in the hybrid 'Pococi' and that 1-MCP treatment repressed and reduced the expression of genes that controlled the two enzymes, which indicates that treating papaya with 1-MCP could delay softening.

### Parsley

Ella *et al.* (2003) reported that a single application of 1-MCP at  $10 \mu\text{l/l}$  could be used to retard senescence of parsley leaves early in the storage life, but they also found that this treatment could increase ethylene biosynthesis.

### Peach

Treatment with 1-MCP at  $1 \mu\text{l/l}$  at  $20^\circ\text{C}$  for 14 h reduced the rate of softening, loss of sugars and organic acids and inhibited increases in ethylene production and respiration rate during 16 days storage at  $8^\circ\text{C}$  in 90–95% RH for the cultivar 'Mibaekdo' and  $5^\circ\text{C}$  in 90–95% RH for the cultivar 'Hwangdo' (Park *et al.*, 2009). Treating the cultivar 'Jiubao' with 1-MCP at  $0.2 \mu\text{l/l}$  at  $22^\circ\text{C}$  for 24 h slowed the decline in fruit firmness. The minimum concentration of 1-MCP that inhibited fruit softening was  $0.6 \mu\text{l/l}$  and repeated treatment resulted in more effective inhibition of ripening. Changes in TSS, total sugars, TA, soluble pectin and ethylene production were also significantly reduced or delayed by 1-MCP. The activities of PAL, PPO and peroxidase in the inoculated fruit were also enhanced by treatment with 1-MCP (Liu *et al.*, 2005). In the cultivar

'Tardibelle', treatment with 1-MCP was effective in preserving firmness during subsequent cold storage for 21 days in air but had no effect on those in CA (Ortiz and Lara, 2008). Postharvest decay of the cultivar 'Jiubao' that had been inoculated with *Penicillium expansum* was reduced by treatment with 1-MCP and disease progress was reduced during cold storage for 21 days (Liu *et al.*, 2005). Yu *et al.* (2017) found that 1-MCP significantly inhibited chilling injury and reduced softening in the two cultivars they tested during storage at 5°C. Treatment with 1-MCP reduced the activity and expression of enzymes related to sucrose degradation and increased activity and expression of sucrose synthase synthesis, resulting in a lower rate of sucrose degradation and an increase of glucose and fructose content.

### Pear

1-MCP prolonged or enhanced the effects of CA storage of the cultivar 'Conference' (Rizzolo *et al.*, 2005). However, they found that the effects of 1-MCP at 25 nl/l and 50 nl/l declined with duration of storage for up to 22 weeks at -0.5°C (followed by 7 days in air at 20°C) in both CA and air storage. Even re-treating the fruit after 7 and 14 weeks of storage had little additional effect on subsequent ripening. Ethylene production was lower and firmness was higher in fruits treated at 50 nl/l, while those treated at 25 nl/l were similar to non-treated fruit. Non-treated and 25 nl/l-treated fruits reached their best sensory quality after 14 weeks of storage, while 50 nl/l-treated fruits reached the same sensory quality later, retaining a fresh flavour when the quality of non-treated fruits declined and became 'watery or grainy'. The cultivars 'Williams', 'Bosc' and 'Packhams Triumph' were stored for approximately 300 days at -0.5°C in 2.5 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub> or in air by Lafer (2005). The fruits that had been pre-treated with 625 ppb 1-MCP had delayed softening and no loss of TA in all three cultivars, whereas non-treated fruits showed excessive softening and reduction in TA during subsequent shelf-life. The cultivar 'Yali' was exposed to 1-MCP at

0, 0.01, 0.1, 0.2 or 0.5 µl/l for 24 h or 0.2 µl/l for 0, 12, 24 or 48 h and then stored at 20°C and 85–95% RH by Fu *et al.* (2007). Those treated with 1-MCP retained their firmness and TSS content better than those not treated. After 100 days of storage the incidence and index of core browning was reduced by 91% and 97%, respectively, by 1-MCP treatment and the occurrence of black and withered stems was also reduced by 59% by 1-MCP after 32 days of storage. 1-MCP also reduced the ethylene production and respiration rate. Rizzolo *et al.* (2005) showed that the development of superficial scald was not prevented by 1-MCP, but the severity of the symptoms was less in the cultivar 'Conference' after treatment. Isidoro and Almeida (2006) reported that 1-MCP can be used to replace diphenylamine as a postharvest treatment to control scald in the cultivar 'Rocha', though ripening after storage was delayed by relatively high 1-MCP concentrations. Spotts *et al.* (2007) reported that 1-MCP reduced decay (bull's-eye rot, phacidiopycnis rot, stem-end rot and grey mould) in the cultivar 'd'Anjou'. They concluded that a combination of 1-MCP and hexanal at optimal rates may reduce storage decay caused by fungal infection and control superficial scald while allowing normal ripening. Lafer (2005) found that rotting caused by *Penicillium expansum* and *Botrytis cinerea* was the main problem during long-term storage and neither CA nor 1-MCP was effective in preventing fruit rotting. The effect of 1-MCP on fungal decay varied considerably among the cultivars tested and the stage of maturity. Yu and Wang (2017) claimed that the current commercial practice for controlling superficial scald in the cultivar 'd'Anjou' (ethoxyquin and CA storage) was not adequate for the fruit produced in hot seasons. They tested treatment with 1-MCP and found that the combination of ethoxyquin at 2700 mg/l plus 1-MCP at 0.3 µl/l + ethylene at 0.3 or 1.5 µl/l controlled scald, reduced decay and allowed recovery of ripening capacity of 'd'Anjou' after storage at -1.1°C in 1.5 kPa O<sub>2</sub> + < 0.5 kPa CO<sub>2</sub> for 8 months. Escribano *et al.* (2016) treated the cultivar 'Bartlett' after cold storage with 1-MCP at 0.6 µl/l at 0°C for 24 h,

which slowed fruit ripening at 20°C, but also resulted in fully ripened pears with higher sweetness, juiciness and aroma compared with those treated with ethylene at 100 µl/l at 20°C for 24 h and non-treated pears. However, Escribano *et al.* (2017) reported that the response of ‘Bartlett’ to 1-MCP could be variable, due to competition with ethylene. They tested a liquid formulation of 1-MCP (immersion in 500 µg/l for 60 s), which provided a more consistently effective treatment than the fumigation (0.6 µl/l at 0°C for 24 h). The most effective treatment was 1000 µg/l for 60 s, but the extended time to ripen at 20°C after storage ‘could be too long for some commercial applications’.

### Persimmon

Kaynaş *et al.* (2009) reported that the cultivar ‘Fuyu’ treated with 1-MCP at either 625 or 1250 ppb for 24 h at 18–20°C then storage in sealed LDPE bags at 0–1°C and 85–90% RH could be kept for 120 days. Ben Arie *et al.* (2001) reported that there was very little fruit softening during storage at 1°C in an ethylene-free atmosphere and the rates of ethylene evolution and fruit softening declined with the increase in 1-MCP concentration from 0 to 600 ppb. The commercial shelf-life of the fruit was doubled or trebled following exposure to 1-MCP, the effect becoming stronger as the season advanced, and the rate of softening of non-treated fruit accelerated. No adverse effects of the treatment were observed. Black spot (*Alternaria alternata*) was the limiting factor in storage, but 1-MCP had no observable effects on infection levels. ‘Fuyu’ stored by Pinto *et al.* (2007) at 10°C, especially in 1 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> and 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> plus 1-MCP, had a higher percentage of firm fruits and lower decay incidence after 17 days. The cultivar ‘Nathanzy’ was harvested when the orange colour had developed on the peel (commercial maturity) by Ramm (2008) and then treated with 1-MCP at 0.5, 1 or 1.5 µl/l for 24 h at 20°C and stored at 20°C. The non-treated fruits softened within 15 days;

those treated with 1-MCP at 1 or 1.5 µl/l remained firmer for 30 days after harvest and had lower respiration rates, but 0.5 µl/l had a limited inhibitory effect on softening. Changes in TSS, TA and peel colour occurred during storage, but all were significantly delayed by 1-MCP. It was concluded that 1-MCP was effective for quality maintenance and extension of shelf-life in persimmon and could allow harvesting at the orange stage of maturity, at which stage the most desirable organoleptic attributes had been developed on the tree. Luo (2007) treated the cultivar ‘Qiandaowuhe’ with 1-MCP at 3 µl/l for 6 h and stored them at 20°C which greatly extend their postharvest life compared with those not treated. 1-MCP treatment delayed the onset of ethylene production and the respiration rate and delayed the depolymerization of chelator-soluble pectic substances and alkali-soluble pectic substances. These are cell wall hydrolysis enzymes, and the increase in water-soluble pectic substances was reduced compared with non-treated fruit.

### Pineapple

Shinjiro *et al.* (2002) found that the rate of colour change of immature, mature-green or ripe fruits was slower for those that had been treated with 1-MCP. The delaying effect was greater in the immature fruits. They also showed that 1-MCP treatment, several times during storage, actually resulted in temporary stimulation of ethylene production immediately after each treatment but, at the same time, 1-MCP decreased their respiration rate.

### Plum

Manganaris *et al.* (2008) reported that both European (*Prunus domestica*) and Japanese (*P. salicina*) plum responded to 1-MCP by delayed ripening. Postharvest application of 1-MCP at 0, 0.5, 1.0 or 2.0 µl/l for 24 h at 20 ± 1°C to the Japanese cultivar ‘Tegan Blue’ was reported by Khan and Singh (2007) to delay ethylene production significantly and also suppress its magnitude. Also there

was a reduction in the activity of ethylene biosynthesis enzymes and fruit-softening enzymes in the skin and pulp tissues. Subsequently Khan *et al.* (2009) showed similar effects, but additionally they reported reduced TSS, AA, total carotenoids and total antioxidants in 1-MCP-treated 'Tegan Blue' compared with those not treated. They found that 1-MCP at 1  $\mu\text{l/l}$  was effective in extending the storage life for up to 6 weeks in cold storage with minimal loss of quality.

### Potato

Ethylene is used as a sprout suppressant in potatoes. Foukaraki *et al.* (2016) stored potatoes at 6°C for 30 weeks with ethylene applied from harvest or from the first indication of sprouting. In both cases ethylene reduced sprout growth significantly but induced sugar accumulation. However, sugar accumulation was prevented by 1-MCP application prior to storage. Also a rise in AA content in tubers was observed in response to ethylene application while 1-MCP blocked the ethylene-induced AA rise. Prange *et al.* (2005a) concluded that 1-MCP at 1  $\mu\text{l/l}$  for 48 h at 9°C could be used to control fry colour darkening induced by ethylene (4  $\mu\text{l/l}$ ) without affecting ethylene control of tuber sprouting. They found that the number of 1-MCP applications required varied between the two cultivars they tested, in that one application was sufficient in 'Russet Burbank' but not in 'Shepody', which started darkening 4 weeks after exposure in the single ethylene + 1-MCP treatment.

### Sapodilla

In Brazil, *Manilkara achras* fruits treated with 1-MCP at 300  $\text{nl/l}$  for 12 h and then stored in MA packaging at  $25 \pm 2^\circ\text{C}$  and 70  $\pm$  5% RH had delayed softening for 11 days longer than those not treated (Morais *et al.*, 2008). In Mexico, treatment with 1-MCP at 100 or 300  $\text{nl/l}$  resulted in a storage life of 38 days at 14°C (Arevalo-Galarza *et al.*, 2007). 1-MCP treatment at 40 or 80

$\text{nl/l}$  for 24 h at 20°C delayed the increases in respiration rate and ethylene production and resulted in increased PGA activity by 6 days during storage at 20°C. Decreases in AA, TSS, TA and chlorophyll content were also delayed (Zhong *et al.*, 2006). 1-MCP-treated fruit showed an inhibition of cell wall-degrading enzyme activity and less extensive solubilization of polyuronides, hemicellulose and free neutral sugars when compared with non-treated fruit. It was suggested that delayed softening of sapodilla is largely dependent on ethylene synthesis (Morais *et al.*, 2008).

### Spinach

Grozeff *et al.* (2010) found that treatment with 1-MCP at 0.1 or 1.0  $\mu\text{l/l}$  inhibited ethylene sensitivity, which could be successfully used to retain quality and chlorophyll content at 23°C for up to 6 days.

### Strawberry

The cultivar 'Everest' was treated with 1-MCP at various concentrations from 0 to 1000  $\text{nl/l}$  for 2 h at 20°C and then stored for 3 days at 20°C and 95–100% RH. 1-MCP treatment lowered ethylene production and tended to maintain fruit firmness and colour, but disease development was quicker in fruit treated with 500 and 1000  $\text{nl/l}$ . 1-MCP inhibited PAL activity and reduced increases in anthocyanin and phenolic content (Jiang *et al.*, 2001).

### Tomato

1-MCP was applied for 12 h at  $22 \pm 1^\circ\text{C}$  and 80–85% RH. During subsequent storage at  $20 \pm 1^\circ\text{C}$  and 85–95% RH, 1-MCP delayed ripening by 8–11 days for concentrations of 250  $\text{ml/l}$ , 11–13 days for 500  $\text{ml/l}$  and 15–17 days for 1000  $\text{ml/l}$  compared with non-treated tomatoes. After 17 days, fruits treated with 1000  $\text{ml/l}$  were firmer and greener, but total

carotenoids were lower than in non-treated fruits (Moretti *et al.*, undated). Both CA storage at 2 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> and 1 ppm 1-MCP for 24 h significantly delayed colour development and softening. The effect of 1-MCP at 1 ppm was greater than the effect of CA but 1-MCP at 0.5 ppm had no effect (Amodio *et al.*, 2005). Baldwin *et al.* (2007) harvested the cultivar 'Florida 47' at immature-green, green, breaker, turning and pink stages and treated half of them with 1-MCP and stored them all at 18°C. Those treated with 1-MCP had slightly lower levels of most aroma volatiles but ripened more slowly by 3–4 days compared with non-treated fruit,

but at 13°C the delay in ripening of 1-MCP-treated fruit was only 2–3 days. They reported that 1-MCP seemed to 'synchronize' ripening of pink-stage tomatoes.

### Watercress

After 5 days of storage in air there was 25% yellowing at 5°C and 100% at 15°C. 1-MCP applied at 500 or 1000 nl/l for 8 h at 10°C had no significant effect on subsequent storage life at 5°C or 15°C. It was concluded that watercress could be stored for a maximum of 4 days at 5°C (Bron *et al.*, 2005).

# 8

## CA Technology

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It is crucial for the success of CA storage technology that the precise levels of gas are achieved and maintained within the store. Where these are too high, in the case of CO<sub>2</sub>, or too low, for O<sub>2</sub>, then the fruit or vegetable may be irrevocably damaged. The range of manifestations of the symptoms of damage varies with the types of product and the intensity of the effect. The effects of CA storage are not simple. In many cases they are dependent on other environmental factors, including the effects of temperature where CA storage may be less or totally ineffective at certain temperatures on some crops. This effect could be related to crop metabolism and gas exchange at different temperatures. From the limited information available it would appear that the store humidity and CO<sub>2</sub> level in the store could interact, with high CO<sub>2</sub> being more toxic at high than low humidity. The physiological mechanism to explain this effect is difficult to find. CA can reduce or eliminate detrimental effects of ethylene accumulation, possibly by CO<sub>2</sub> competing for sites of ethylene action within the cells of the fruit.

The literature is ambivalent on the quality and rate of deterioration of fruit and vegetables after they have been removed from CA storage. Most of the available information indicates that their storage life is adequate for marketing, but, in certain cases,

the marketable life is even better than freshly harvested produce. The physiological mechanism that could explain this effect has so far not been demonstrated.

The equipment and methods used in the control of the atmosphere inside stores is constantly being developed to provide more accurate control. Use of the physiological responses of the fruit and vegetables to CA storage is increasing, which may eventually replace the methods that were applied during the 20th century. A review of the design, construction, operation and safety considerations of CA stores is given by Bishop (1996). In fresh fruit and vegetable stores the CO<sub>2</sub> and O<sub>2</sub> levels will change naturally through their metabolism because of the reduced or zero gas exchange afforded by the store walls and door. Levels of the respiratory gases in non-CA stores have shown these changes. For example, Kidd and West (1923) showed that the levels of CO<sub>2</sub> and O<sub>2</sub> in the holds of ships from South Africa carrying stone fruit and citrus fruit arriving in UK at 3.9–5°C were 6 kPa CO<sub>2</sub> and 14 kPa O<sub>2</sub> and at 2–3.3°C the levels were 10.0 kPa CO<sub>2</sub> and 10.5 kPa O<sub>2</sub>. The level of CO<sub>2</sub> in a banana ripening room can rise from near zero to 7 kPa in 24 h (Thompson and Seymour, 1984).

Much of the current commercial practice is basically the same as that described



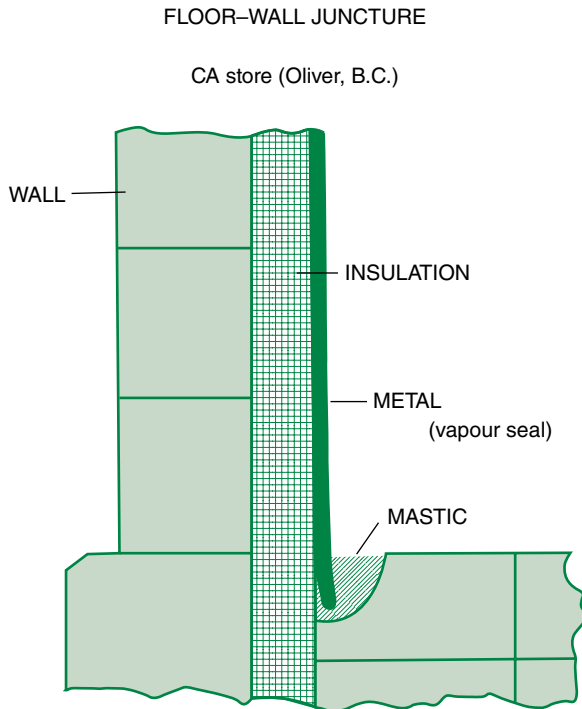
in the early part of the 20th century. The crop is loaded into an insulated storeroom whose walls and door have been made gas tight. The first CA stores were constructed in the same way as refrigerated stores from bricks or concrete blocks with a vapour-proof barrier and an insulation layer that was coated on the inside with plaster. In Australia, Little *et al.* (2000) pointed out the difficulty of converting refrigerated stores to CA stores in the early 1970s because the cork insulation made them so leaky. In order to ensure that the walls were gas tight for CA storage, they were lined with sheets of galvanized steel (Fig. 8.1). The bottoms and tops of the steel sheets were embedded in mastic (a kind of mortar composed of finely

ground oolitic limestone, sand, litherage and linseed oil) and a coating of mastic was also applied where sheets abutted on the walls and ceiling (Fig. 8.2). The walls were insulated using wooden frames with cork, wood shavings or Rockwool insulation. They often had wooden floors, which could result in O<sub>2</sub> leakage into the stores at a faster rate than it was utilized by the fruit giving an equilibrium level of 5–8 kPa O<sub>2</sub>.

Modern CA stores are made from metal-faced insulated panels (usually polyurethane foam and its fire-resistant form, called isophenic foam), which are fitted together with gas-tight patented locking devices (Fig. 8.3). The joints between panels are usually taped with gas-tight tape or painted



**Fig. 8.1.** The inside of a traditional controlled atmosphere store in England showing the internal walls covered with galvanized sheets to render them gas tight. Photograph courtesy of Dr R.O. Sharples.



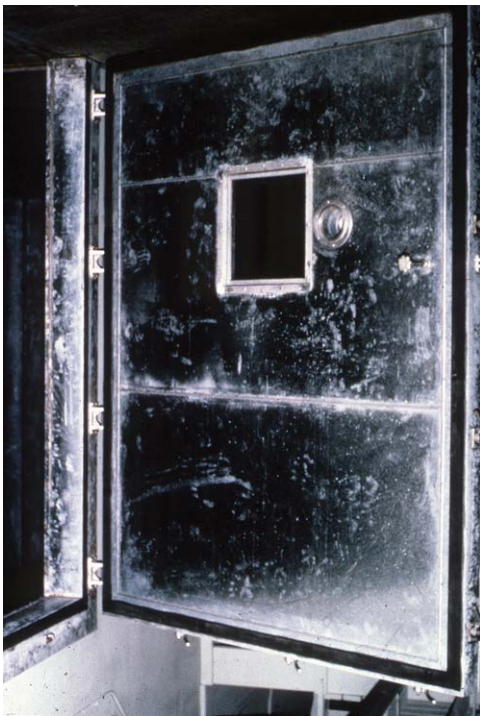
**Fig. 8.2.** Detail of internal wall of a traditional controlled atmosphere store in England covered with galvanized sheets and filled with mastic to render them gas tight. Photograph courtesy of Professor H.D. Tindall.



**Fig. 8.3.** Access corridor of modern CA stores showing the doors with the fruit ready to be loaded. Reproduced with permission from David Bishop, Storage Control Systems Ltd, UK.

with flexible plastic paint or resin sealants to ensure that they are gas tight. Cetiner (2009) described some of the properties of the insulated panels as rigid polyurethane foams having a closed cell structure and natural bonding to steel facers. In China, one company used a thermal insulating board, which is of light structure, high strength, good air-tightness, good temperature-insulation, convenient construction, substantial short construction period and less construction cost. They used ester urethane thermal insulating board and polystyrene heat board for constructing the framework.

Major areas of the store where leaks can occur are the doors. These are usually sealed by having rubber on the jamb or frame (Fig. 8.4) so that when the door is closed the two meet to seal the door. In addition a flexible soft rubber hose may be hammered into

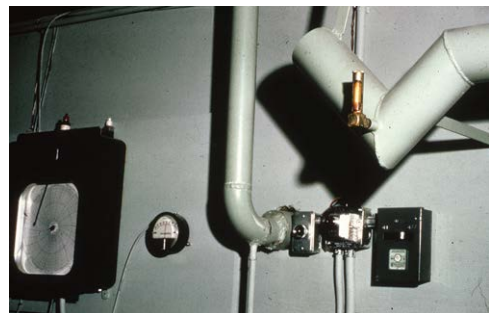


**Fig. 8.4.** Door of a traditional CA store in the UK showing rubber seals on jamb matched by rubber seals on door to ensure that they are gas tight. There is also a circular sampling port and small door to ensure that anyone trapped inside can escape. Photograph courtesy of Dr R.O. Sharples.

the inside between the door and its frame to give a double rubber seal to ensure gas tightness. In other designs screw jacks are spaced around the periphery of the door. Some modern CA stores and fruit ripening rooms have inflatable rubber door seals.

With the constant changes and adjustments of the store temperature and concentrations of the various gases inside the sealed store, variation in pressure can occur. Pressure difference between the store air and the outside air can result in difficulties in retaining the store in a completely gas-tight condition. Stores are therefore fitted with pressure release valves (Fig. 8.5), but these can make the maintenance of the precise gas level difficult, especially the  $O_2$  level in ULO stores. An expansion bag may be fitted to the store to overcome the problem of pressure differences. The bags are gas tight and partially inflated and are placed outside the store, with the inlet to the bag inside the store. If the store air volume increases, this will automatically further inflate the bag; and when the pressure in the store is reduced, air will flow from the bag to the store. The inlet of the expansion bag should be situated before the cooling coils of the refrigeration unit in order to ensure that the air from the expansion bag is cooled before being returned to the store.

Testing the gas tightness of a store is crucial to the accurate maintenance of the required  $CO_2$  and  $O_2$  levels. One way of measuring this was described by MAFF



**Fig. 8.5.** Pressure release valve on CA store to ensure equalization of pressure between the store and externally. Photograph courtesy of Dr R.O. Sharples.

ADAS (1974). A blower or vacuum cleaner is connected to a store ventilator pipe and a manometer connected to another ventilator pipe. It is essential that the manometer is held tightly in the hole so that there are no leaks. A blower then blows air into the store which is maintained until a pressure of 200 Pa has been achieved and then the blower is switched off. The time for the pressure inside the store to fall is then monitored. They recommended that the time taken to fall from 187 to 125 Pa should not be less than 7 min in order to consider the store sufficiently gas tight. If the store does not reach this standard, efforts should be made to locate and seal the leaks.

An example of this CA technology was the largest commercial store in Europe in the late 20th century, which has now gone out of business (East Kent Packers, 1994, personal communication). CA was achieved by sealing the store directly after loading and reducing the temperature to below 10°C. Bags of hydrated lime were loaded into the store prior to sealing as a method of removing respiratory CO<sub>2</sub>. The temperature was then reduced further and when it was below 5°C, N<sub>2</sub> was injected to reduce the O<sub>2</sub> level; for 'Cox's Orange Pippin' apples this was to about 3 kPa. The O<sub>2</sub> level was allowed to decline to about 2 kPa for 7 days through fruit respiration and then progressively to 1.2 kPa over the next 7 days. Headspace monitoring was used that set off an alarm if alcohol fumes were detected.

## Temperature Control

The main way of preserving fruit and vegetables in storage or during long-distance transport is, of course, by refrigeration. CA storage and MA packaging are considered supplements to increase or enhance the effect of refrigeration. There is evidence that the inhabitants of the island of Crete were aware of the importance of temperature in the preservation of food, even as early as 2000 BCE (Koelet, 1992). Mechanical refrigeration was developed in 1755 by a Scotsman, William Cullen, who showed that evaporating

ether under reduced pressure, caused by evacuation, resulted in the temperature of the water in the same vessel being reduced, thus forming ice. Cullen then patented a machine for refrigerating air by the evaporation of water in a vacuum. John Leslie, also a Scot, subsequently developed the technique in 1809 and invented a differential thermometer and a hygrometer, which led him to discover a process of artificial congelation. Leslie's discovery improved upon Cullen's equipment by adding sulfuric acid to absorb the water vapour and he developed the first ice-making machine. In 1834 an American, Jacob Perkins, was granted a British patent (Number 6662) for a vapour compression refrigeration machine. Carle Linde, in Germany, developed an ammonia compression machine. This type of mechanical refrigeration equipment was used in the first sea-freight shipment of chilled beef from Argentina in 1879 and was employed in many of the early experiments in fruit and vegetable stores.

Kidd and West (1927 a, b) produced evidence that CA storage was only successful when applied at low temperatures. Standard refrigeration units are therefore integral components of CA stores. Temperature control consists of pipes containing a refrigerant inside the store. Ammonia or chlorofluorocarbons (CFCs) R-12 and R-22 (the R-number denotes an industry standard specification and is applied to all refrigerants) are common refrigerants, but because of concern about depletion of the ozone layer in the 1980s new refrigerants have been developed that have potentially less detrimental effects on the environment, such as R-410A, which is 50:50 blend of R-32 and R-125 and is used as a substitute for R-22. The pipes containing the refrigerant pass out of the store; the liquid is cooled and passed into the store to reduce the air temperature of the store as it passes over the cooled pipes. A simple refrigeration unit consists of an evaporator, a compressor, a condenser and an expansion valve. The evaporator is the pipe that contains the refrigerant mostly as a liquid at low temperature and low pressure, and is the part of the system that is inside the store. Heat is

absorbed by the evaporator, causing the refrigerant to vaporize. The vapour is drawn along the pipe through the compressor, which is a pump that compresses the gas into a hot high-pressure vapour. This is pumped to the condenser, where the gas is cooled by passing it through a radiator. The radiator is usually a network of pipes open to the outside air. The high-pressure liquid is passed through a series of small-bore pipes, which slows the flow of liquid so that a high pressure builds up. The liquid then passes through an expansion valve, which controls the flow of refrigerant and reduces its pressure. This reduction in pressure results in a reduction in temperature, causing some of the refrigerant to vaporize. This cooled mixture of vapour and liquid refrigerant passes into the evaporator, thus completing the refrigeration cycle. In most stores a fan passes the store air over the coiled pipes containing the refrigerant, which helps to cool the air quickly and distribute it evenly throughout the store.

In commercial practice for long-term CA storage of most crops, the store temperature is initially reduced to 0°C for a week or so, whatever the subsequent storage temperature will be. This would clearly not be applicable to fruit and vegetables that can suffer from chilling injury at relatively high temperatures (10–13°C). Also CA stores are normally designed to a capacity that can be filled in 1 day, so fruit is loaded directly into store and cooled the same day. In the UK the average CA store size was given as about 100 t with variations between 50 and 200 t; in continental Europe it was about 200 t and in North America about 600 t (Bishop, 1996). In the UK the smaller rooms were preferred, because they facilitated the speed of loading and unloading.

### Humidity Control

Most fruit and vegetables require a high humidity when kept in storage. Generally the closer the atmosphere is to saturation the better, as long as moisture does not condense on the fruit or vegetables. The amount of heat absorbed by the cooling coils of the

refrigeration unit is related to the temperature of the refrigerant they contain and the surface area of the coils. Water will condense on the evaporator if the refrigerant temperature is low compared with the store air temperature. Condensate will freeze if the refrigerant temperature is well below freezing, reducing the efficiency of the cooling system. Removal of moisture from the store air results in the stored crop losing moisture by evapotranspiration. In order to reduce crop desiccation, the refrigerant temperature should be kept close to the store air temperature. However, this must be balanced with the removal of the respiratory heat from the crop, temperature leakage through the store insulation and doors and heat generated by fans, otherwise the required crop temperature cannot be maintained. There remains the possibility of having a low temperature differential between the refrigerant and the store air by increasing the area of the cooling surface and this may be helped by adding devices such as fins to the cooling pipes or coiling them into spirals. In a study on evaporator coil refrigerant temperature, the room humidity, cooling rate and mass loss rates were compared in commercial 1200-bin apple storage rooms by Hellickson *et al.* (1995). They reported that rooms in which evaporator coil refrigerant temperatures were dictated by cooling demand required a significantly longer time to achieve desired humidity levels than rooms in which evaporator coil temperatures were controlled by a computer. The overall mass loss rate of apples stored in a room in which the cooling load was dictated by refrigerant temperature was higher than in a room in which refrigerant temperature was maintained at approximately 1°C during the initial cooling period.

A whole range of humidifying devices can also be used to replace the moisture in the air that has been condensed on the cooling coils of refrigeration units. These include spinning disc humidifiers where water is forced at high velocity on to a rapidly spinning disc. The water is broken down into tiny droplets, which are fed into the air circulation system of the store. Sonic humidifiers utilize energy to detach tiny water droplets from a water surface that are fed into the

store's air circulation system. Dijkink *et al.* (2004) described a system that could maintain humidity very precisely in 500 l containers. They were able to maintain  $90.5 \pm 0.1\%$  RH using a hollow fibre membrane contactor that allowed adequate transfer of water vapour between the air in the storage room and a liquid desiccant. The membrane was made of polyetherimide, coated on the inside with a thin non-porous silicone layer. The desiccant was a dilute aqueous glycerol solution, which was pumped through the hollow fibres at a low flow rate.

Another technique that retains high humidity within the store is secondary cooling so that the cooling coils do not come into direct contact with the store air. One such system is the 'jacketed store'. These stores have a metal inner wall inside the store's insulation, with the refrigeration pipes cooling the air space between the inner wall and the outer insulated wall. This means a low temperature can be maintained in the cooling pipes without causing crop desiccation and the whole wall of the store becomes the cooling surface. In a study of export of apples and pears from Australia and New Zealand to the UK, the use of jacketed stores gave a uniform temperature within  $0.5^\circ\text{F}$  ( $-17.5^\circ\text{C}$ ) in a bulk of a stack of 50,000 cases (Food Investigation Board, 1937). Ice bank cooling is also a method of secondary cooling, where the refrigerant pipes are immersed in a tank of water so that the water is frozen. The ice is then used to cool water and the water is converted to a fine mist, which is used to cool and humidify the store air (Neale *et al.*, 1981).

Where crops such as onions are stored under CA conditions, they require an optimum humidity of about 70% RH. This can be achieved by having a large differential between the refrigerant and air temperature of about  $11\text{--}12^\circ\text{C}$  with natural air circulation or  $9\text{--}10^\circ\text{C}$  where air is circulated by a fan.

## Gas Control Equipment

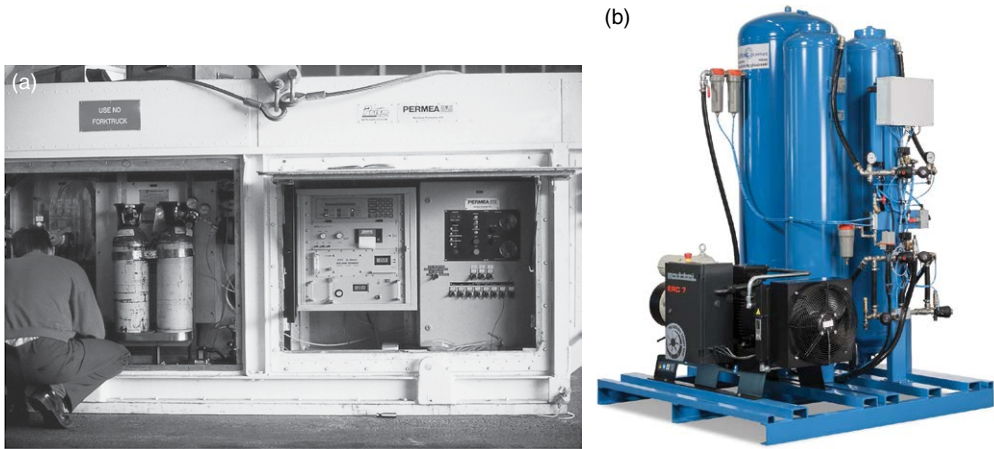
Burton (1982) indicated that in the very early gas stores (CA stores) the levels of  $\text{CO}_2$  and  $\text{O}_2$  were controlled by making the storeroom gas tight with a controlled air leak. This resulted in the two gases being

maintained at approximately equal levels of about 10%. Even in more recent times the  $\text{CO}_2$  level in the store was controlled while the level of  $\text{O}_2$  was not and was calculated as  $\text{O}_2 + \text{CO}_2 = 21$  kPa (Fidler and Mann, 1972). Kidd and West (1927a, b) recognized that it was desirable to have more precise control over the level of gases in the store and to be able to control them independently. In subsequent work the levels of  $\text{CO}_2$  and  $\text{O}_2$  were monitored and adjusted by hand and considerable variation in the levels could occur. Kidd and West (1923) showed that the levels monitored in a store varied between 1 kPa and 23 kPa for  $\text{CO}_2$  and between 4 kPa and 21 kPa for  $\text{O}_2$  over a 6-month storage period.

In the 1940s R.E. Smock at Cornell University introduced the addition of a sodium hydroxide solution to the system for removal of  $\text{CO}_2$ . The use of hydrated lime for removing  $\text{CO}_2$  was subsequently developed by Charles Eaves in Canada. The Whirlpool Corporation in the USA built 'Tectrol' units for apple CA stores. Tectrol and other types used the combustion of propane or natural gas to develop an atmosphere low in  $\text{O}_2$ . In the USA, Donald Dewey and George Mattus developed some automatic controls for CA storage (Mattus, 1963).  $\text{N}_2$  generators using selective gas permeable membranes (Fig. 8.6) and  $\text{N}_2$  cylinders are now more commonly used.

## Generating equipment

In early work the method used to achieve the required  $\text{O}_2$  and  $\text{CO}_2$  levels in CA rooms was product generated. This obviously resulted in a delay in achieving the required condition which, in turn, could reduce the maximum storage life or increase losses during storage. One method of reducing  $\text{O}_2$  rapidly was to introduce air into the store that had first been passed over a gas burner. The room was ventilated to prevent the increase in pressure; therefore the method was referred to as gas flushing. This gave a rapid reduction in  $\text{O}_2$  but many of the products of burning gas, such as carbon monoxide, can act like ethylene and have a negative effect



**Fig. 8.6.** Nitrogen generation. (a) Nitrogen generator in use at a commercial controlled atmosphere store in Kent in England in the 1990s. (b) Modern Pressure Swing Adsorption nitrogen generator. Reproduced with permission from Eric van der Zwet, Besseling Group BV, The Netherlands.

on maximum storage life of the stored crop. Natural or propane gas was combined with outside air and burned, which lowered the  $O_2$  and increased  $CO_2$ . Most of the  $CO_2$  was removed before the atmosphere was fed into the CA room. An Italian method used an 'Isolcell' unit where the effluent was passed through a catalyst to ensure complete oxidation of any partially oxidized compounds to  $CO_2$ . Activated carbon was used to scrub out most of the  $CO_2$  before the generated atmosphere entered the store room. A different type of atmosphere generator was developed in The Netherlands in the early 1980s, called 'Oxydrain'. This utilized ammonia rather than propane. The ammonia was cracked, forming  $N_2$  and  $H_2$ , and the  $H_2$  was burned with the  $O_2$  from the room so that no  $CO_2$  was produced. The 'Arcat' system recirculated the air from the room through the natural gas or propane burner and there was a separate system to scrub out the  $CO_2$  produced. The use of a carbon molecular sieve  $N_2$  generator in CA storage was described by Zhou *et al.* (1992b).

### Oxygen control

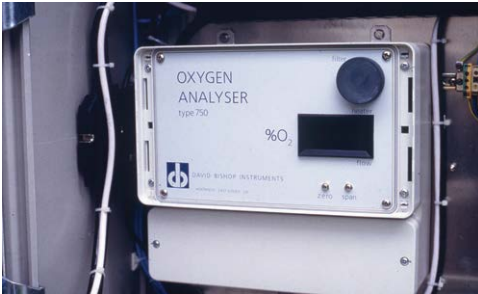
In traditional CA storage systems, when the  $O_2$  had reached the level required for the

particular crop it was maintained at that level by frequently introducing fresh air from outside the store. Sharples and Stow (1986) reported that tolerance limits were set at  $\pm 0.15$  kPa for  $O_2$  levels below 2 kPa, and  $\pm 0.3$  kPa for  $O_2$  levels of 2 kPa and above. This method is still in use. In early CA stores the samples were analysed for both  $O_2$  and  $CO_2$  levels using an Orsat gas analyser, which measured the levels of the two gases by absorbing them and noting the changes in volume. With continuing equipment development, the precision with which the set levels of  $CO_2$  and  $O_2$  can be maintained is constantly increasing. With recent developments in the control systems used in CA stores, it is possible to control  $O_2$  levels close to the theoretical minimum. This is because modern systems can achieve a much lower fluctuation in gas levels and ULO storage (levels around 1 kPa or less) is now common. Modern stores use electrochemical cells (Fig. 8.7) or sometimes a paramagnetic analyser (Fig. 8.8), which are monitored and controlled by a computer. Instruments have been developed that combine analysis of both  $CO_2$  and  $O_2$  from the same sample. These are used for sampling gases in MA packages but can be adapted for CA storage (Fig. 8.9).

The benefits of  $O_2$  levels as low as 1 kPa, or less, have been shown to be optimum in



**Fig. 8.7.** Electrochemical O<sub>2</sub> sensor with typical resolution down to 0.001 kPa O<sub>2</sub>. Reproduced with permission from Storage Control Systems Ltd.



**Fig. 8.8.** Paramagnetic O<sub>2</sub> analyser installed at Cranfield University, UK, in the 1980s for CA storage research.



**Fig. 8.9.** Monitoring for gas levels. A small gas sample is taken from the package or the air and injected into the apparatus that measures and displays oxygen and CO<sub>2</sub> concentration. The analyser uses zirconia ceramic technology for O<sub>2</sub> and non-dispersive infrared spectroscopy for CO<sub>2</sub>.

the extension of the storage of some fruit. These precise O<sub>2</sub> levels can also be used to control physiological disorders. For example, in apples O<sub>2</sub> levels down to 0.6 kPa were found to control BBD and with Bramley's Seedling O<sub>2</sub> levels down to 0.4 kPa were shown to maintain firmness and gave control of scab (Rees, 2014).

At these very low O<sub>2</sub> levels precise control is vital in order not to damage the fruit. Methods that have been developed are based on approaches to the physiology of the fruit. There are three main approaches: one based on respiratory quotient (RQ), one based on ethanol biosynthesis and one based on chlorophyll fluorescence (CF). These monitoring and control methods have been called dynamic controlled atmosphere (DCA) storage while monitoring the O<sub>2</sub> and CO<sub>2</sub> predetermined levels in the store air has been referred to as static controlled atmosphere storage.

### Carbon dioxide control

There are many different types of scrubbers that can be used to remove CO<sub>2</sub> from CA stores, but they can basically be divided into two types. One, sometimes called 'passive scrubbing', uses a chemical that reacts with CO<sub>2</sub> and thus removes it from the store; and the other is a renewable system sometimes called 'active scrubbing'. Sharples and Stow (1986) stated that the CO<sub>2</sub> level in the store should be maintained at  $\pm 0.5$  kPa of the recommended level. Burdon *et al.* (2005) reported that the method of CO<sub>2</sub> control may affect the volatile composition of the room atmosphere, which in turn may affect the volatile synthesis of fruit. They compared activated carbon scrubbing, hydrated lime scrubbing, N<sub>2</sub> purging and storage in air on 'Hayward' kiwifruits at 0°C in 2 kPa O<sub>2</sub> + 5 kPaCO<sub>2</sub>. After storage the fruits were allowed to ripen at 20°C and although volatile profiles differed between CA-stored and air-stored fruits, and also among fruits from the different CO<sub>2</sub> scrubbing systems, the different CO<sub>2</sub> scrubbing systems did result in measurable differences in ripe fruit volatile profiles. They also reported (Lallu



*et al.*, 2005) that there was little effect of the CO<sub>2</sub> removal system on fruit softening and the incidence of rots, but N<sub>2</sub> flushing resulted in kiwifruit with the lowest incidence of physiological pitting. They concluded that different CO<sub>2</sub> removal systems altered room volatile profiles but did not consistently affect the quality of kiwifruit.

Dual-bed scrubbers are where one bed is removing CO<sub>2</sub> while the other is being renewed and the two beds are constantly switched to provide continuity of CO<sub>2</sub> removal. Hermann A. Zinnen, Anil R. Oroskar and Chin-Hsiung Chang invented CO<sub>2</sub> removal using aminated carbon molecular sieves, a method that was patented by Allied-Signal Inc. in 1989 (US4810266 A). Carbon molecular sieves are prepared from O<sub>2</sub>-free precursors carbonized in the absence of O<sub>2</sub>. The pores of the material are enlarged by high-temperature steaming and alcohol amines are used to impart an amine functionality to the sieves. The store air is passed through the sieve, with the absorbed CO<sub>2</sub> being released by heating to a moderate temperature.

Occasionally N<sub>2</sub> purging is used for flushing out CO<sub>2</sub>. Measurement of CO<sub>2</sub> levels forms part of the measurement used in some modern DCA systems for calculating RQ. Some commercial stores still use bags of hydrated calcium hydroxide powder, which reacts with CO<sub>2</sub>; this is used solely, or as a supplement to the activated carbon, especially when the fruit or vegetable is very sensitive to CO<sub>2</sub>. Some of the scrubbing systems used commercially are reviewed below.

### *Passive scrubbing*

The simplest method is to place bags or pallets of a CO<sub>2</sub>-absorbing chemical (usually calcium hydroxide) inside the store, which can keep CO<sub>2</sub> levels low (usually about 1 kPa). This method is referred to as 'product generated', since the gas levels are produced by the crop's metabolism. Calcium hydroxide reacts irreversibly with CO<sub>2</sub> to produce calcium carbonate, water and heat. The disadvantage of this method is the space occupied by the lime and the inefficiency of CO<sub>2</sub> absorption, because less than 20% of the

lime in the bags is used under normal stacking procedures (Olsen, 1986). In a store of 55 t capacity containing hydrated lime, the CO<sub>2</sub> concentration fell to less than 1 kPa while the lime was fresh, then rose gradually; the average CO<sub>2</sub> concentration during the 128-day storage period was 2.2 kPa. Blank (1973) found that the apple cultivars 'Cox's Orange Pippin', 'Ingrid Marie', 'Boskoop' and 'Finkenwerder' stored well under these conditions and the annual total cost was reduced to between one-half and one-quarter of that for stores with activated carbon scrubbers. This simple method of passive scrubbing is thought suitable for stores holding up to 100 t.

For greater control the bags or pallets of lime may be placed in a separate airtight room. When the CO<sub>2</sub> level in the store is above that which is required, a fan draws the store atmosphere through the room containing the bags of lime (Fig. 8.10) until the



**Fig. 8.10.** A simple lime scrubber that was commonly used in CA stores in UK in the mid to late 20th century. Photograph courtesy of Dr R.O. Sharples.

required level is reached. After scrubbing, the air should re-enter the store just before it passes over the cooling coils of the refrigeration unit. There may also be some dehydration of the air as it is passed through the lime and the air may need humidification before being reintroduced to the store. The amount of lime required depends on the type and variety of the crop and the storage temperature. For example, Koelet (1992) stated that for 1 t of apples, 7.5 kg of 'high calcium lime' is needed every 6–10 weeks for most cultivars. Bishop (1996) also calculated the amount of lime required to absorb CO<sub>2</sub> from CA stores on the theoretical basis. He calculated that 1 kg of lime will absorb 0.59 kg of CO<sub>2</sub>, but for a practical capacity he estimated 0.4 kg of CO<sub>2</sub> per 1 kg of lime. On that basis the requirement of lime for 'Cox's Orange Pippin' apples stored in < 1 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> was 50 kg of lime per tonne of fruit. For other apple cultivars, storage regimes and periods, Bishop indicated that probably less lime would be required. Thompson (1996) quoted a general requirement for apples of only 25 kg for 6 months. Olsen (1986) found that a total of 23 kg of lime per tonne of apples was required for a 128-day storage period given in two portions, with one change during storage.

With passive scrubbers the time taken for the levels of these two gases to reach the optimum (especially for the O<sub>2</sub> to fall from the 21 kPa in fresh air) can reduce the maximum storage life of the crop. It is common, therefore, to fill the store with the crop, seal it and inject N<sub>2</sub> gas until the O<sub>2</sub> has reached the required level and then maintain it in the way described above. The N<sub>2</sub> may be obtained from large liquid N<sub>2</sub> cylinders or from N<sub>2</sub> generators (see Fig. 8.6).

### *Active scrubbing*

Active scrubbing consists of two containers of material that can absorb CO<sub>2</sub>. Store air is passed through one of these containers when it is required to reduce the CO<sub>2</sub> in the store atmosphere while the CO<sub>2</sub> is being removed from the other one. When one is near-saturated with CO<sub>2</sub> the containers are reversed and the CO<sub>2</sub> is removed from the

first while the second absorbs CO<sub>2</sub> from the store. Active scrubbers have the advantage of being compact and are particularly suitable for use in CA transport systems. Koelet (1992) described such a system where a ventilator sucks air from the store through a plastic tube to an active carbon filter scrubber. When the active carbon becomes saturated with CO<sub>2</sub> the tube is disconnected from the store and outside air is passed over it to release the CO<sub>2</sub>. When this process is complete the active carbon scrubber is connected to a 'lung', which is a balloon of at least 5 m<sup>3</sup> capacity (depending on the size of the equipment) containing an atmosphere with low O<sub>2</sub>. This process is to ensure that when the active carbon scrubber is reconnected to the store it does not contain high O<sub>2</sub> levels. Baumann (1989) described a simple active scrubbing system that could be used in stores to remove both CO<sub>2</sub> and ethylene using activated charcoal. A chart was also presented that showed the amount of activated charcoal required in relation to the CO<sub>2</sub> levels required and ethylene output of the fruit in the store.

Pressure swing adsorption (PSA) uses a compressor that passes air under pressure through a molecular sieve, made from alumino-silicate minerals called zeolites, which separates the O<sub>2</sub> and the N<sub>2</sub> in the air. PSA is a dual-circuit system so that when one circuit is providing N<sub>2</sub> for the store, the other circuit is being renewed.

Hollow fibre technology is similar to PSA but hollow fibre technology is also used to generate N<sub>2</sub> to inject into the store. The walls of these fibres are differentially permeable to O<sub>2</sub> and N<sub>2</sub>. Compressed air is introduced into the fibres and by varying the pressure it is possible to regulate the purity of the N<sub>2</sub> coming out of the equipment and produce an output of almost pure N<sub>2</sub>. The O<sub>2</sub> content at the output of the equipment will vary with the throughput. For a small machine with a throughput of 5 nm<sup>3</sup>/h the O<sub>2</sub> content will be 0.1 kPa, at 10 nm<sup>3</sup>/h it will be 1 kPa and at 13 nm<sup>3</sup>/h it will be 2 kPa.

Molecular sieves and activated carbon can hold CO<sub>2</sub> and organic molecules such as ethylene. When fresh air is passed through

these substances the molecules are released. This means that they can be used in a two-stage system where the store air is being passed through the substance to absorb the ethylene while the other stage is being cleared by the passage of fresh air. After an appropriate period the two stages are reversed. Hydrated aluminium silicate or aluminium calcium silicate are used. The regeneration of the molecular sieve beds can be achieved when they are warmed to 100°C to drive off the CO<sub>2</sub> and ethylene. This system of regeneration is referred to as 'temperature swing', where the gases are absorbed at low temperature and released at high temperature. Regeneration can also be achieved by reducing the pressure around the molecular sieve, which is called 'pressure swing'. During the regeneration cycle the trapped gases are usually ventilated to the outside, but they can be directed back into the container if this is required.

In China, a carbon molecular sieve N<sub>2</sub> generator was developed (Zhou and Yu, 1989) that was made from fine coal powder, which was refined and formed to provide apertures similar in size to N<sub>2</sub> (3.17 Å) but smaller than O<sub>2</sub> (3.7 Å). When air was passed through it under high pressure, the O<sub>2</sub> molecules were absorbed on to the fine coal powder and the air passing through had an enriched N<sub>2</sub> level.

A 40% solution of ethanolamine can be used in temperature swing scrubbers, which can be regenerated by heating the chemical to 110°C. However, ethanolamine is corrosive to metals and is not commonly used for CA stores.

Pressurized water scrubbers work by passing the store air through a water tower which is pumped outside the store, where it is ventilated with fresh air to remove the CO<sub>2</sub> dissolved in it inside the store. However, O<sub>2</sub> is dissolved in the water while it is being ventilated and will therefore provide a store atmosphere of about 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. A pressurized water CO<sub>2</sub> scrubber was described by Vigneault and Raghavan (1991) which could reduce the time required to reduce the O<sub>2</sub> content in a CA store from 417 h to 140 h.

Sodium hydroxide and potassium hydroxide scrubbers work when the store air is bubbled through a saturated solution, at about 50 l/min, when there is excess CO<sub>2</sub> in the store. This method requires about 14 kg sodium hydroxide for each 9 m<sup>3</sup> of apples per week and, since the chemical reaction which absorbs CO<sub>2</sub> is not reversible, the method can be expensive.

Selective diffusion membrane scrubbers have been used commercially (Marcelin and LeTeinturier, 1966) but they require a large membrane surface area in order to maintain the appropriate gaseous content around the fruit.

### Nitrogen cylinders

This is a simple system where cylinders or a tank containing N<sub>2</sub> are connected to a vaporizer and pipes leading to the CA store. There is a sampling pump and meters to monitor O<sub>2</sub> and CO<sub>2</sub> levels in the storeroom and an air pump to inject outside air. As the O<sub>2</sub> is depleted fresh air is introduced and as CO<sub>2</sub> accumulates N<sub>2</sub> is introduced. N<sub>2</sub> gas pressurized to about 40 pounds per square inch (psi) (about 276 kPa) from liquid N<sub>2</sub> tanks can be fed into the room through a vaporizer, which can establish the required O<sub>2</sub> levels in the store quickly with purge rates of up to 20,000 cubic feet (566 m<sup>3</sup>) per hour. Care must be taken to allow adequate ventilation to avoid a potentially catastrophic over-pressurization of the room. Additionally, very high purge rates may let purge gas escape through the vent before it is completely mixed with the store air. Introducing N<sub>2</sub> gas behind the refrigeration coils and venting low on the return wall assures high mixing effectiveness (Olsen, 1986).

### CO<sub>2</sub> addition

For crops that have only a very short marketable life, such as strawberries, the CO<sub>2</sub> level may also be increased to the required level by direct injection of CO<sub>2</sub> from a pressurized gas cylinder. This is commonly only used in transport of fruit in CA containers.

## Ethylene control

Ethylene is synthesized by plant cells. In climacteric fruit its synthesis initiates ripening. Ethylene can also have negative effects on fresh fruit and vegetables, especially when accumulated in stores. For example, Wills *et al.* (1999) showed that the postharvest life of some non-climacteric fruit and vegetables could be extended by up to 60% when stored in an atmosphere containing less than 5 ppb ethylene compared with those stored in 100 ppb. There are various ways in which ethylene can be removed from stores, including absorption, reaction, ozone scrubbers and catalytic converters.

### Absorption

Molecular sieves and activated carbon can hold organic molecules such as ethylene. When fresh air is passed through these substances the organic molecules are released. So two-stage systems have been developed where one stage is absorbing ethylene while the other is ventilating to fresh air. The system automatically switches when one side is judged to be saturated.

### Reaction

Ethylene can be oxidized at room temperatures when it comes into contact with potassium permanganate. Various commercial systems are available, such as Ethisorb<sup>®</sup>, Purafil<sup>®</sup> Bi-On 4<sup>®</sup> and Sorbilen<sup>®</sup>, which can be placed inside the CA store or even inside the actual package containing the crop.

### Ozone scrubbers

Ozone is a powerful oxidizing agent and reacts with ethylene to produce CO<sub>2</sub> and water. It can be easily generated from O<sub>2</sub> by ultraviolet (UV) radiation or electrical discharge. Scott and Wills (1973) described a UV lamp giving out radiation at 184 and 254 nanometres which produced both ozone and atomic oxygen. The store atmosphere was passed through the reaction chamber with a fan and any ethylene it contained

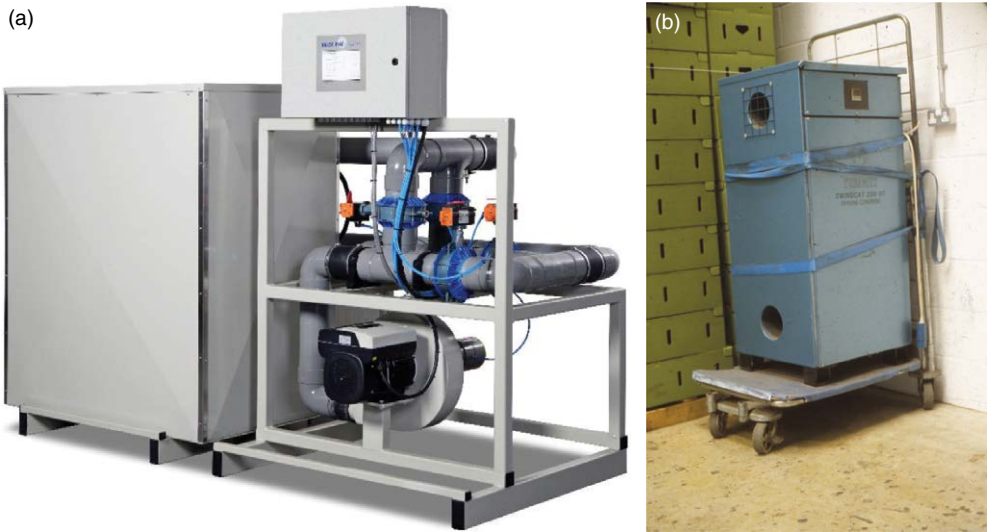
was rapidly oxidized. The outlet of the reaction chamber contained rusty steel wool, which would react with any excess ozone and prevent it entering the store.

### Catalytic converters

Air from the store is passed through a device where it is heated to over 200°C (usually 230–260°C) in the presence of an appropriate catalyst, usually platinum, that oxidizes ethylene and any other unsaturated hydrocarbon to CO<sub>2</sub> and water. A catalytic converter was marketed by Tubamet AG of Vaduz in Liechtenstein in 1993 and called 'Swingcat'. Another is produced by the French company Absoger (2016) who claim that it can maintain store ethylene levels below 20 parts per billion. Huang *et al.* (2002) described a scrubber that maintained ethylene concentration below 100 ppb in a sealed plastic tent containing cabbages. Sfakiotakis *et al.* (1993) reported that an ethylene scrubber retained ethylene levels below 300 ppb in an apple CA store. A two-column ethylene converter is available in which the store air passes through a catalyst bed where the ethylene is broken down and then to the second catalyst bed, where the remaining ethylene is broken down and the air cooled (Fig. 8.11a). It is claimed that this system can reduce the ethylene level to about 1 ppb and its heat recovery and temperature control mean that it consumes minimal energy (Eric van der Zwet, 2018, personal communication). Ethylene scrubbers can be incorporated in the gas control system or portable ethylene scrubbers are available that can be placed in a store when required (Fig. 8.11b).

## Plastic Tents and Membranes

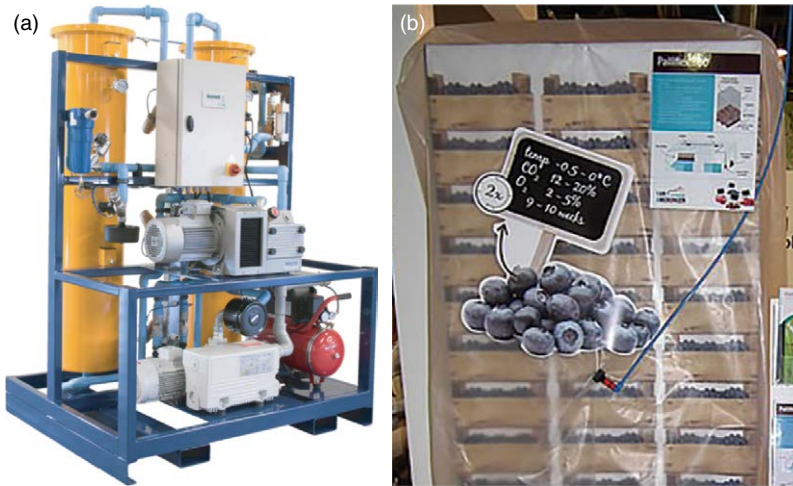
Xie *et al.* (1990) compared storage of apples for 117 days in small plastic bags and large plastic tents. After storage the proportion of fruits of good quality among those stored non-packed was 54%, or more than 98% for those in small plastic bags and 87–89% for those in the large plastic tents. Fruit firmness and vitamin C were highest and acidity



**Fig. 8.11.** (a) A modern Besseling ethylene converter (Reproduced with permission from Eric van der Zwet, Besseling Group BV, The Netherlands). (b) A portable Tubamet Swingcat ethylene scrubber in use in a wholesale packhouse in UK in the 1990s.

lowest in small plastic bags, followed by plastic tents and then non-packed. Leyte and Forney (1999) designed and constructed a tent for the CA storage of small quantities of fruit and vegetables suspended from pallet racking in a cold room. The tents could hold two standard pallets stacked 1.8 m high with fruit or vegetables and were sealed with two airtight zips and a small water trough, resulting in an airtight chamber that successfully maintained a CA storage environment. Huang *et al.* (2002) described studies using a sealed plastic tent inside a cold storage room. A  $2 \times 2 \times 2$  m tent made of LDPE was set up inside a cold room maintained at  $0\text{--}1^\circ\text{C}$ . The composition of the atmosphere within the tent was maintained at 2–6 kPa  $\text{O}_2$  (mostly at 3 kPa) and 3–5 kPa  $\text{CO}_2$  (mostly at 5 kPa). Ethylene concentration was controlled at below 0.1 ppm using an ethylene scrubber. The temperature inside the tent was always about  $0.7^\circ\text{C}$  higher than the temperature in the rest of the room. In 1997 a company developed a system that involved enclosing a large trolley in a six-sided insulating blanket (a material with the trade name of ‘Tempro’) that was closed with heavy-duty Velcro. This was shown to maintain the product

temperature within  $1^\circ\text{C}$ , at a starting temperature of  $0^\circ\text{C}$ , during a 22 h period. The inner walls of the insulating blanket had pockets in which dry ice could be placed to help to maintain product temperature and presumably increase the  $\text{CO}_2$  content. The Palliflex storage system was developed for CA storage by Van Amerongen CA Technology, USA, in about 2002. The system can be used to set individual CA conditions on individual pallets, which means that different types of fruit and vegetables requiring the same temperature but different atmospheres can be stored in the same room. The system consists of a single plastic pallet of fruit or vegetables with a transparent gas-tight plastic film cover made from polyamide/polyethylene (PA/PE) or polyurethane (PU) film (Fig. 8.12). The sealed pallet is connected to a regulating system that has built-in  $\text{O}_2$  and  $\text{CO}_2$  analysers and an air pump and is connected to  $\text{CO}_2$  and  $\text{N}_2$  cylinders and an  $\text{O}_2$  scrubber (Fig. 8.12). The sealed pallets can be linked together so that one monitoring and control unit can be used for many pallets or just one. Doäyan and Erkan (2014) described experiments with the Palliflex storage system that had been successfully completed in Turkey and it was found to be



**Fig. 8.12.** Palliflex storage system: (a) control unit; (b) pallet of fruit sealed in plastic film. Reproduced with permission from Ramzi Amari, Van Amerongen CA Technology BV, The Netherlands.

advantageous for use in storage and transport of horticultural commodities. Sahin *et al.* (2015) compared storage of ‘Red Globe’ grapes at 0°C with 95% RH for 90 days in different CA conditions using a Palliflex storage system and Selcuk and Erkan (2015) used the Palliflex storage system for experiments on CA storage of medlar fruit.

Besseling have also developed a similar system that they called the ‘Pallet Fresh System’ (Fig. 8.13). Each pallet is covered with a polyethylene film cover that can be sealed and can be used for individual pallets or groups of pallets. The covered pallets are placed in a cold store and the atmosphere inside the pallet cover can be monitored and controlled by an external CO<sub>2</sub> cylinder, O<sub>2</sub> air compressor and PSA N<sub>2</sub> generator (Eric van der Zwet, 2018, personal communication). Another Dutch company, Fotein BV, supplies a modified atmosphere system for pallets called ‘Ship it Fresh!’, which consists of a nine-layer foil comprising four different types of plastic providing tensile and flexible strength, impact resistance, gas tightness and ease of sealing. Covers are equipped with a gas exchange membrane so that the gas conditions in the cover remain virtually unchanged during transport. The unit, which is 1 m wide and 1.2 m deep, consists of a preformed top cover 1.80 m high

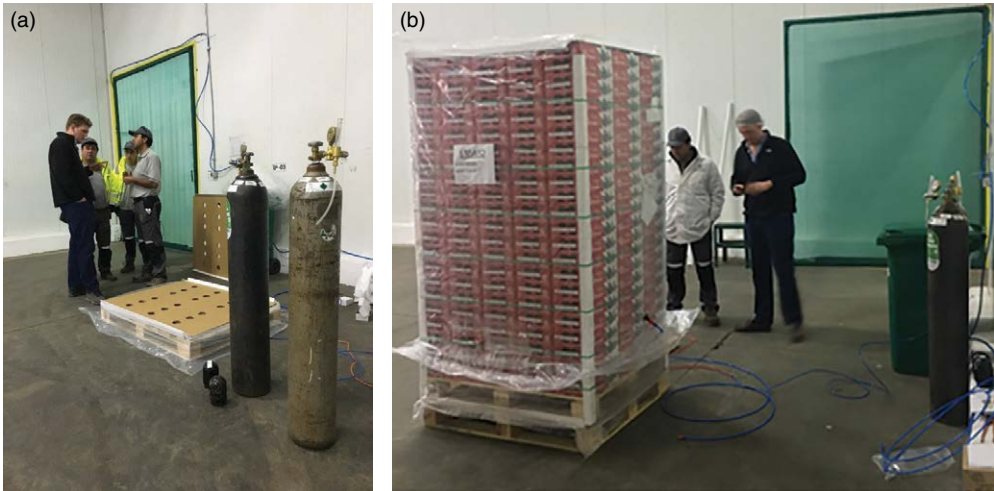
and a bottom cover 30 cm high. The unit has a maximum height of 2.25 m (Fig. 8.14).

Browne *et al.* (1984) tested PE-covered pallets of ‘Cambridge Favourite’ strawberries sealed with a patented mechanism and injected with CO<sub>2</sub> and found after transport at 2°C that the pallets contained between 3 and 16 kPa CO<sub>2</sub>, most containing < 10 kPa. In subsequent experiments they found that 10 kPa CO<sub>2</sub> did not consistently retard spoilage of fruit by *Botrytis cinerea* or *Mucor piriformis*, as they had hoped, and also caused persistent off-flavours. However, in spite of these results they recommended that PE covers be used on pre-cooled strawberries to ‘help keep the fruit cool and reduce moisture loss during distribution’.

Marcellin and LeTeinturier (1966) reported a system of controlling the gases in store by a silicone rubber diffuser. In trials they reported that they were able to maintain an atmosphere of 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> + 94 kPa N<sub>2</sub> with the simple operation of only gas analysis. In later work by Gariepy *et al.* (1984) a silicone membrane system was shown to maintain 3.5–5 kPa CO<sub>2</sub> + 1.5–3 kPa O<sub>2</sub>. They found that, in this system, total mass loss was 14% after 198 days of storage of cabbage compared with 40% in air. On a commercial-scale experiment (472 t), an ‘Atmolysair System’ CA room was compared



**Fig. 8.13.** The ‘Pallet Fresh System’. Reproduced with permission from Eric van der Zwet, Besseling Group BV, The Netherlands.



**Fig. 8.14.** A ‘Ship it Fresh’ modified atmosphere pallet: (a) being assembled; and (b) loaded with fruit for shipment. Reproduced with permission from Rob Veltman, Managing Director, Fotein ([www.fotein.com](http://www.fotein.com)).

with a conventional cold room for storing ‘Winter Green’ cabbages for 32 weeks at 1.3°C. The atmosphere in the Atmolsair System room was 5–6 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> +

92 kPa N<sub>2</sub> and traces of other gases. This was compared with a cold room maintained at 0.3°C. The average trimming losses were less than 10% for the Atmolsair System

room but exceeded 30% in the cold room (Raghavan *et al.*, 1984). Rukavishnikov *et al.* (1984) described a 100 t store with the crop sealed inside 150 or 300  $\mu\text{m}$  PE films, with windows of membranes selective for  $\text{O}_2$  and  $\text{CO}_2$  permeability made of polyvinyl trimethyl silane or silicon-organic polymers. Apples and pears stored at 1–4°C under the covers had 93% and 94% sound fruit, respectively, after 6–7 months of storage, whereas the fruit storage in air at similar temperatures had less than 50% sound fruit after 5 months. A design procedure for a silicon-membrane CA store, based on parametric relationships, was suggested by Garipey *et al.* (1988) for selecting the silicone membrane area for long-term CA storage of leeks and celery. A ‘modified gaseous components system’ for CA storage using a gas separation membrane was designed, constructed and tested by Kawagoe *et al.* (1991). The membrane was permeable to  $\text{CO}_2$ ,  $\text{O}_2$  and  $\text{N}_2$  but to different degrees, thus the levels of the three gases varied. A distinctive feature of the system was the application of gas circuit selection to obtain the desired modified gaseous components. It was shown to be effective in decreasing  $\text{CO}_2$  content and increasing  $\text{O}_2$  content, but  $\text{N}_2$  introduction to the chamber was not included in the study.

## Ripening Rooms

### Design

Ripening rooms are designed for climacteric fruit that are harvested before they are ripened and placed in temperature-controlled rooms under CA conditions in order to initiate and control ripening. Also some citrus, which are non-climacteric fruit, are placed in temperature and CA rooms simply to change the colour of the peel. These are called degreening rooms and ethylene is commonly used. During the 1990s there was an increasing demand for all the fruit being offered for sale in a supermarket to be of exactly the same stage of ripeness so that it has an acceptable and predictable shelf-life. This led to the development of a system called ‘pressure ripening’, which is used mainly for bananas but is applicable to any climacteric

fruit. The system involves direction of the circulating air in the ripening room by channels through boxes of fruit so that exogenous ethylene gas, which initiates ripening, is in contact equally with all the fruit in the room. At the same time the  $\text{CO}_2$ , which can impede ripening initiation, is not allowed to concentrate around the fruit (Fig. 8.15).

The primary requirements for ripening rooms (Fig. 8.16) are that they should:

- have a good temperature control system;
- have good and effective air circulation;
- be gas tight; and
- have a good system for introducing fresh air.

Air circulation around the fruit is important to prevent local accumulations of the  $\text{CO}_2$  given out by the fruit and to ensure good contact between the fruit and the ethylene gas applied to initiate ripening. In the case of bananas and some other types of fruit, the boxes are lined with polyethylene film and usually transported to ripening rooms stacked on pallets. Several systems have been used. One is to remove each box from the pallet, pull back the plastic film and re-stack them on pallets so that there is a space between boxes (Fig. 8.17). In other cases the pallets are stacked so that one handhold in each box is facing outwards. As the fruit are loaded into the rooms, the plastic just inside each handhold is torn to facilitate air exchange. This is especially important where fruit have been vacuum packed.

Air circulation systems are usually largely convective. Air is blown through the cooler and then across the top of the store just below the ceiling. The cooled air falls by convection through the boxes of fruit and is taken up at floor level for recirculation. Many modern ripening rooms have air channels in the floor through which air is circulated at high pressure. This forces it up through the pallets of boxes and should give better air circulation. Special devices such as inflatable air bags placed between pallets are now used to ensure better air circulation and, therefore, more even fruit ripening.

Good ventilation to enable fresh air to be introduced is very important for successful fruit ripening. During the period of initiation of ripening, which is usually 24 h, no fresh



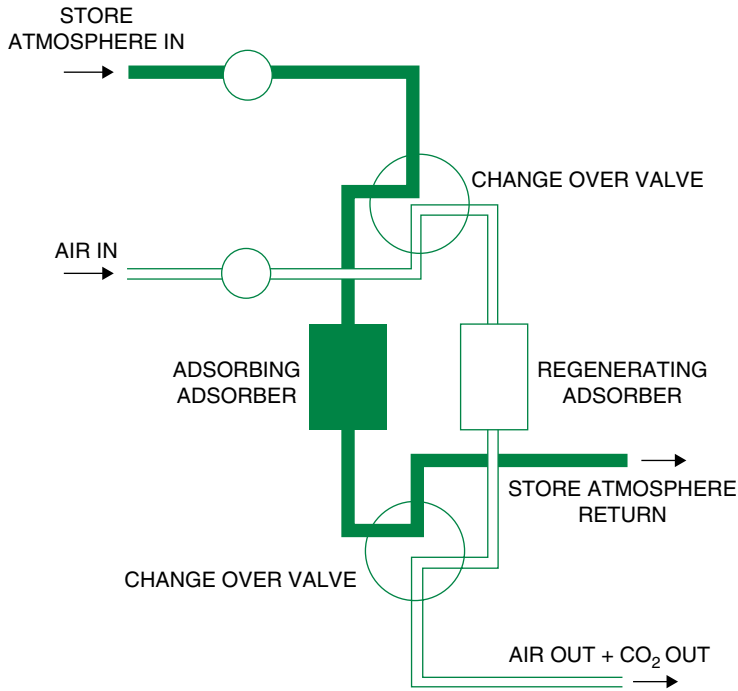


Fig. 8.15. Schematic of CO<sub>2</sub> scrubber.

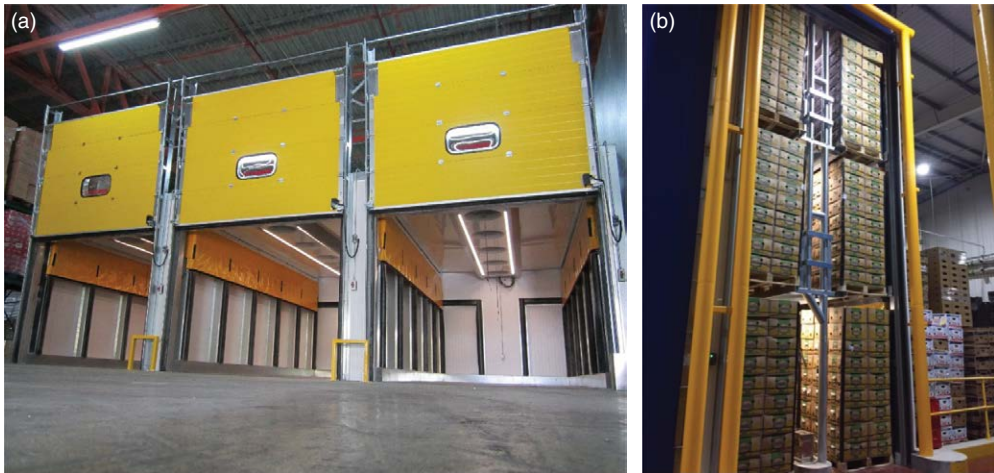


Fig. 8.16. Banana ripening room in UK in the 1990s.

air is introduced into the rooms. This is the period when ethylene is introduced. Directly after this period the rooms must be thoroughly ventilated. Setting up a ripening system for bananas in Ecuador, Thompson and Seymour (1984) found that the CO<sub>2</sub> level of ripening rooms had gone up from about zero to 7 kPa during this initial 24 h initiation period. Even with a good fan extraction system it took 40 min of ventilation to bring the CO<sub>2</sub> levels to below 1 kPa. This ventilation with fresh air must be repeated every 24 h during subsequent ripening to prevent levels of CO<sub>2</sub> becoming too high. If rooms are not frequently ventilated, ripening can be delayed, or abnormal ripening can occur.

Ripening rooms need to be gas tight in order to ensure that threshold levels of ethylene are maintained around the fruit during the initiation period. As in CA stores, the most common place where leakage occurs is around the doors.

It is advisable to have high humidity of 90–95% RH in ripening rooms. To this end many rooms are fitted with some humidification device such as a spinning disc



**Fig. 8.17.** (a) Pressure ripening rooms. (b) Pallets of bananas loaded into a pressure ripening room. Reproduced with permission from Dave Rodden, Advanced Ripening Technologies Ltd, UK.

humidifier. However, if the rooms are full of fruit and the refrigerant in the cooling coils used to maintain the room temperature is regulated to within a few degrees of the required room temperature, this should be sufficient to keep the humidity high.

### Ethylene generators

Ethylene can be applied in a liquid formulation as Ethrel. For example, recommended ripening conditions of Otaheite apple (*Spondias cytherea*) were 2–4 days at ambient conditions or 20–22°C and 90–95% RH with 500 ppm Ethrel for 1–1½ days followed by 7–8°C and 90–95% RH, since fully ripe fruit were found to be less sensitive to chilling injury (Mohammed *et al.*, 2011). In commercial ripening rooms in the Yemen, buckets of sodium hydroxide were placed throughout the ripening rooms. When all these were in place, measured amounts of Ethrel were added, which gave an instant release of ethylene gas into the room (Thompson, 1985). Steel cylinders containing ethylene are also used (Fig. 8.18), as are calcium carbide and other chemicals that release ethylene analogues.

However, ethylene generators are currently used almost exclusively. These are

devices that are placed in ripening rooms. A liquid is poured into them and they are connected to an electrical power source and they produce ethylene over a protracted period (Fig. 8.19). The manufacturers of these generators do not provide information on exactly what the liquid is that they supply for use in the generators or the process by which the ethylene is generated. A possible way of generating ethylene would be to heat ethanol in a controlled way in the presence of a copper catalyst. Care should be exercised in doing this because of the inflammability of the alcohol.

The way that generators are used is to calculate the volume of the store and place the correct number of generators in the store to provide the required ethylene concentration. This method has the advantage of supplying ethylene to the store over 16 h rather than applying it in one dose from cylinders. This means that there is a better chance of achieving the desired ripening or de-greening effect where there are problems of the room being perfectly gas tight. Gull (1981) used this method for ripening tomatoes. Blankenship and Sisler (1991) reported that effluent produced by the catalytic generators had a distinctly different odour than that produced by ethylene from cylinders. When they compared the two methods for ripening tomatoes, they found inconclusive



**Fig. 8.18.** Ethylene being injected into a banana ripening room at Geest, UK, in the 1970s.

results in that taste panellists could detect a difference between the fruit ripened by the two methods, but they did not express a preference for either treatment. There tended to be less variation in colour between fruits ripened with ethylene from cylinders compared to those ripened from ethylene generators.

Bananas are sensitive to physiological levels of ethylene as low as  $0.3\text{--}0.5\ \mu\text{l/l}$  if the  $\text{O}_2$  and  $\text{CO}_2$  levels are similar to those found in outside fresh air (Peacock, 1972). The three main factors affecting response to external ethylene are: fruit maturity; time after harvest when ethylene exposure began; and the length of exposure to ethylene.

### Modelling

Modelling has been used for CA storage and considerable work has been reported at the



**Fig. 8.19.** An ethylene generator being filled and prepared to be used in a banana ripening room. Reproduced with permission from Greg Akins, Catalytic Generators Inc., Norfolk, Virginia, USA.

Catholic University of Leuven in Belgium led by Dr Bart Nicolă. However, in DCA storage Yearsley *et al.* (1996) experienced problems finding an acceptable mathematical method to estimate low  $\text{O}_2$  using ethanol and had to resort to subjective visual estimation by a human panel. Lee *et al.* (1991, 2003a) developed a model that fitted a polynomial equation to the respiration rate of different fruits to determine optimum CA storage conditions, and successfully tested it on cherry tomatoes. Nahor *et al.* (2003) described a model for simulation of the dynamics of heat, moisture and gas exchange in CA stores consisting of three interacting systems: the cooled space, the refrigeration unit and the gas handling unit. They reported that model predictions were usually found to have a good agreement with the experimental results. Due to the inclusion of product respiration rate and product quality models, the advantage of the simulation model was that the optimization of CA plant performance with respect to the final

quality of the stored fruit or vegetable could be carried out. Owing to the hierarchical and object-oriented approach, greater flexibility with respect to model modification and reusability was achieved. Nahor *et al.* (2005) later reported that the Michaelis–Menten type gas exchange models were appropriate for storage of apples and pears, as opposed to the empirical ones, due to their generic behaviour. They recommended that the kinetic parameters of the model should be expressed as a function of the physiological age of the crop to include the influence of age when the models were to be used to simulate gas exchange of bulk stored fruit for longer storage period. Gwanpua *et al.* (2012) proposed kinetic equations for a model to describe softening of ‘Braeburn’ apples in CA storage. They measured changes in pectin-degrading enzymes, pectin degradation and ethylene production and found that their model could explain most of the total variance of the measured data. Gwanpua *et al.* (2018) also used the Michaelis–Menten constant for a mathematical model to describe autocatalytic ethylene production on softening and colour changes during storage. They calibrated, and successfully validated, their model on avocados stored at 5°C, 10°C or 20°C in air with the addition of ethylene at 0, 1 or 10 µl/l. CO<sub>2</sub> evolution, O<sub>2</sub> consumption and ethanol production were fitted to a polynomial equation used in ‘response surface methodology’ and tested on cherry tomatoes in order to determine optimum CA storage conditions. The regression equations fitted well with the experimental data of the respiration characteristics having a coefficient of determination  $R^2 > 0.8$  and the effective CA conditions for cherry tomatoes were estimated at 4–6 kPa O<sub>2</sub> + 3.5–10 kPa CO<sub>2</sub>. Ho *et al.* (2016) evaluated gas exchange in pears during CA storage using a continuum diffusion respiration model. Simulation results showed that internal O<sub>2</sub> and CO<sub>2</sub> variation in the gas profiles in the pears were affected by variable diffusion, maximum respiration rate and the ‘three dimensional morphology’ of the fruit. They also modelled the onset of fermentation during low O<sub>2</sub> levels in ULO storage. Their model was used to show that

harvest maturity and fruit size were important criteria to be taken into account prior to CA storage.

Peppelenbos *et al.* (1996) used a Michaelis–Menten kinetic model to describe the O<sub>2</sub> uptake and the CO<sub>2</sub> production depending on the O<sub>2</sub> partial pressure and validated the models by respiration measurements. Castellanos *et al.* (2016) successfully developed models to represent changes in the fruit firmness and colour during storage of feijoa fruits in MA packaging with  $R^2$  values of 0.95 for firmness and 0.94–0.95 for colour. They were able to estimate, in advance, the equilibrium concentration of the gases in the packaging headspace for each MA package they tested, using a previously developed model. Pereira *et al.* (2017) developed a model for defining optimum packaging and storage conditions. They tested the model on minimally processed rocket leaves during 10 days of storage in MA packages at 0°C, 5°C or 10°C factorially combined with 2.5, 5, 10 and 20 kPa O<sub>2</sub>. The RQ was found to range between 0.6 and 1.3 and the temperature contributed more than 80% of the Michaelis–Menten model variance, while O<sub>2</sub> pressure contributed only with 13%.

## Safety

The use of CA storage has health and safety implications. One factor that should be taken into account is that the gases in the atmosphere could possibly have a stimulating effect on microorganisms. Also the levels of these gases could have detrimental effects on the workers operating the stores. In Britain the Health and Safety Executive (HSE, 1991) showed that work in confined spaces could be potentially dangerous and entry must be strictly controlled, preferably through some permit system. Also it is recommended that anyone entering such an area should have proper training and instruction in the precautions and emergency breathing apparatus. Stringent procedures need to be in place with a person on watch outside and the formulation of a rescue plan. MAFF ADAS (1974) also indicated

that when a store is sealed anyone entering it must wear breathing apparatus. Warning notices should be placed at all entrances to CA stores and an alarm switch located near the door inside the chamber in the event that someone may be shut in. There should be a release mechanism so that the door can be opened from the inside. When the produce is to be unloaded from a store, the main doors should be opened and the circulating fans run at full speed for at least an hour before unloading is commenced. In the UK the areas around the store chamber must be kept free of impedimenta in compliance with the appropriate Agricultural Safety Regulations.

### Ethylene

Ethylene is a colourless gas with a sweetish odour and taste, has asphyxiate and anaesthetic properties and is flammable. Its flammable limits in air are 3.1–32% volume for volume and its auto ignition temperature is 543°C. Care must be taken when the gas is used for fruit ripening to ensure that levels in the atmosphere do not reach 3.1%. As added precautions, all electrical fitting must be of a 'spark-free' type and warning notices relating to smoking and fire hazards must be displayed around the rooms. For the USA, Crisosto (1997) said that 'All electrical equipment, including lights, fan motors and switches, should comply with the National Electric Codes for Class 1, Group D equipment and installation.'

### Oxygen

O<sub>2</sub> levels in the atmosphere of 12–16 kPa can affect human muscular coordination, increase respiration rate, and affect a person's ability to think clearly. At lower levels vomiting and further impediment to coordination and thinking can occur. At levels below 6 kPa human beings rapidly lose consciousness, and breathing and heart beating stop (Bishop, 1996). O<sub>2</sub> levels of over 21 kPa can create an explosive hazard in the store

and great care needs to be taken when hyperbaric storage or high O<sub>2</sub> MA packaging is used.

### Carbon dioxide

Being in a room with higher than ambient CO<sub>2</sub> levels can be hazardous to health. The limits for CO<sub>2</sub> levels in rooms for human occupation were quoted by Bishop (1996) from the Health and Safety Executive, publication EH40-95 *Occupational Exposure Limits*, as 0.5 kPa CO<sub>2</sub> for continuous exposure and 1.5 kPa CO<sub>2</sub> for 10 min exposure.

### Microorganisms

Microorganisms, especially fungi, can grow on fruit and vegetables during storage and can produce toxic chemicals. One example is the mycotoxin patulin (4-hydroxy-4H-furo [3,2-c], pyran-2(6H)-one). The EU (European Commission, 2006) has established patulin limits of 50 µg/kg for apple juice, 25 µg/kg for solid apple products and 10 µg/kg for baby foods. Patulin can occur in apples and pears that have been contaminated with *Penicillium patulum*, *P. expansum*, *P. urticae*, *Aspergillus clavatus* or *Byssoschlamys nivea* and stored for long periods. Morales *et al.* (2007) inoculated the apple cultivar 'Golden' with *P. expansum* and stored them at 1°C for up to 2½ months in either low O<sub>2</sub> or ultra-low O<sub>2</sub>. Generally, bigger lesions were observed under ULO than under low O<sub>2</sub>, but no patulin was detected in either condition. There was an increase in lesion size when fruits were subsequently stored at 20°C and patulin was detected, but there was no difference in patulin content between ULO and low O<sub>2</sub>. Dos Santos *et al.* (2018) found that healthy fruits were not contaminated with patulin, even when stored together with infected decayed apples. Application of 1-MCP increased the percentage of fruit with decay and patulin contamination in 'Galaxy' apples. Neither CA nor DCA storage (DCA-CF or DCA-RQ 1.3) prevented patulin production in either 'Galaxy' or 'Fuji Kiku' apples.

# 9

## Dynamic CA Storage

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For optimum conditions for CA storage, it is essential that the precise levels of gases in the store are known and maintained. As has been described in Chapter 8, these levels have traditionally been determined by experimentation. Basically the lower the O<sub>2</sub> content in the CA store, the better are the storage conditions for the crop, down to a critical O<sub>2</sub> level, called the anaerobic compensation point (ACP), when the effects become negative. Dynamic controlled atmosphere (DCA) storage uses ultra-low oxygen (ULO) levels and monitors these by measuring physiological responses (respiration rate, chlorophyll fluorescence, fermentation) of the stored crop and controls the levels through a computer. DCA therefore uses the responses of the actual crop being stored as monitors of the store atmosphere, instead of an arbitrary setting of the O<sub>2</sub> and CO<sub>2</sub> levels in the store based on previous experience and experimentation. In other words, CA conditions are adjusted dynamically between and within seasons to optimize the product's response to CA. CA storage has therefore involved two phases. In the first phase, the dynamic nature of the CA conditions was solely determined from empirical data. In the second and current phase, bio-sensing of the product physiology is employed to reset the CA

conditions to reflect the changing metabolic condition of the stored product.

### History of DCA

#### Phase 1: DCA using empirical data

Alique and De la Plaza (1982) are believed to be the first to use the term 'dynamic controlled atmosphere'. Taking advantage of research showing that apples varied over storage time in their susceptibility to CO<sub>2</sub> damage, they proposed to increase the CO<sub>2</sub> in the store after 1 month from 0 to 2 kPa and after 2 months to 5 kPa, at constant 3 kPa O<sub>2</sub>, successfully demonstrating its effect on 'Golden Delicious' apples (Alique and De la Plaza, 1982) and 'Blanquilla' pears (Alique *et al.*, 1984). Qi *et al.* (1989) introduced the term 'two-dimensional dynamic controlled atmosphere storage' (TDCA) in which the first dimension is a change in storage temperature from 10–15°C to 0°C and the second dimension is a change in storage CO<sub>2</sub> from an initial 10–15 kPa to about 6 kPa with a constant 3 kPa O<sub>2</sub>. Mattheis *et al.* (1998) used the term 'dynamic atmosphere storage' in which temperature and CO<sub>2</sub> are kept constant and O<sub>2</sub> is varied from 1 kPa to

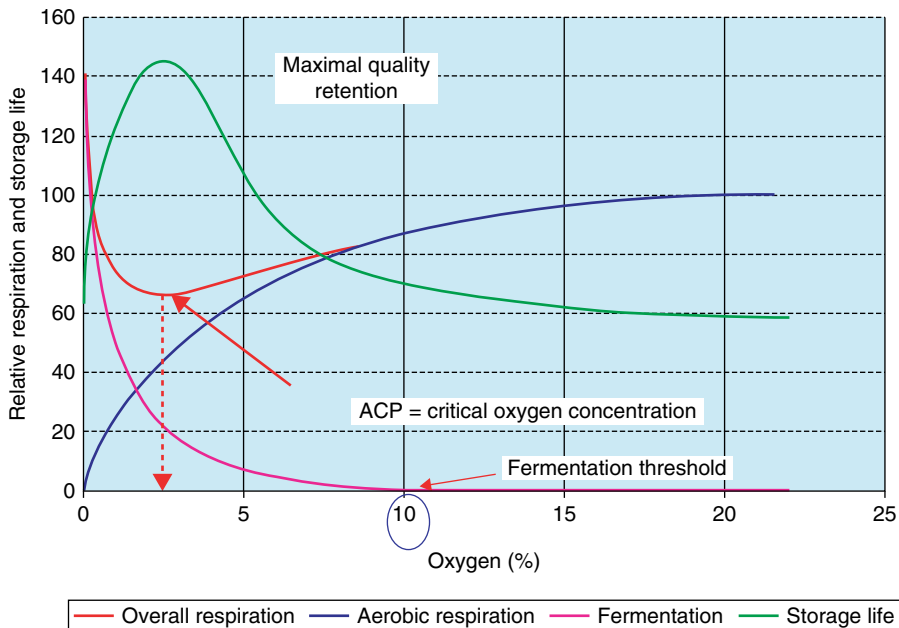
ambient air for periods of 1, 2 or 3 days and then returned to 1 kPa  $O_2$ .

### Phase 2: DCA using product physiology

DCA storage using product physiology, rather than static general recommendations, was described conceptually by Wollin *et al.* (1985) at the 4th National Controlled Atmosphere Research Conference. They speculated that the respiratory quotient (RQ) may be the method to determine the lowest  $O_2$  level, but identified several constraints to its use outside carefully controlled laboratory conditions. The concept of Wollin *et al.* (1985) was used to devise an automated test system that sought out the optimum combination of temperature,  $O_2$  and  $CO_2$  to achieve minimum respiration (Wolfe *et al.*, 1993). Jameson (1993) described a similar research project in the UK but no data or

evidence of success was provided by either group. The minimum point of respiration rate they sought was the ACP, which is the  $O_2$  concentration that produces the lowest  $CO_2$  production. The assumption was that this was the  $O_2$  concentration where the respiration rate changes from primarily aerobic (above the ACP) to primarily anaerobic (below the ACP) and fresh fruit or vegetables held exactly at or just above the ACP would have the maximum storage life (Fig. 9.1).

Yearsley *et al.* (1996) attempted to integrate terminology by introducing the general term lower oxygen limit (LOL) and stated that the optimum storage atmosphere occurs just above the LOL at which aerobic respiration is at the lowest level that can be achieved without development of anaerobic metabolism. They stated that, from a physiological perspective, there are two approaches to determining the LOL: (i) the ACP that is the  $O_2$  concentration at which  $CO_2$  production is minimal; and



**Fig. 9.1.** It is generally accepted that maximal storage life and quality retention (green line) occurs when the product is stored at a storage  $O_2$  concentration (% or kPa) exactly at or just above the Anaerobic Compensation Point (ACP) but not below the ACP. The blue line (aerobic respiration) is measured as  $O_2$  consumed and the red line (overall respiration) is measured as total  $CO_2$  produced. The ACP occurs at the lowest total  $CO_2$  production. The Fermentation threshold can also be called the Extinction Point (EP). The ACP can also be called the lower oxygen limit (LOL).

(ii) measures of the onset of ethanol production or respiratory quotient breakpoint. Using experimental data on 'Cox's Orange Pippin' apples, they identified how these two approaches of determining the LOL produce different LOL values and how internal gas concentrations are more relevant than external gas (headspace) measurements. At that time of writing (1996) the use of chlorophyll fluorescence to determine LOL was not known.

### Respiratory Quotient

RQ is the measure of moles CO<sub>2</sub> evolved to moles O<sub>2</sub> absorbed in plant cells. RQ = 1 when the substrate is carbohydrate, but it is lower for lipids and proteins. Burton (1952) measured RQ in potatoes stored at 10°C in 5–7 kPa CO<sub>2</sub> for up to 14 weeks. The increased CO<sub>2</sub> reduced both O<sub>2</sub> uptake and CO<sub>2</sub> output by about 25–30%, but the RQ was unaffected and remained close to 1. Wollin *et al.* (1985) discussed the possibility that RQ may be used to calculate the lowest O<sub>2</sub> level that can be tolerated in fruit during storage to be incorporated in an automated CA system. Yearsley *et al.* (1996) also discussed the possibility of using RQ in CA stores and considered that the fermentation threshold could best be represented by RQ. They reported that RQ gave the safest estimate of the true lower O<sub>2</sub> limit for optimizing storage atmospheres. The methods used for research in the Federal University of Santa Maria in Brazil were described by dos Santos *et al.* (2018) as:

DCA-RQ conditions were measured twice a week during nine months of storage. For RQ determination, O<sub>2</sub> and CO<sub>2</sub> partial pressure were measured immediately after the chambers were closed. After 13 h, the RQ was calculated as the ratio between CO<sub>2</sub> produced and O<sub>2</sub> consumed within this period. The O<sub>2</sub> partial pressure was controlled by RQ variation. If the RQ calculated was higher than the pre-established value (RQ 1.3), O<sub>2</sub> levels were increased by ± 0.02 kPa to reduce anaerobic metabolism and RQ. If the calculated RQ was less than

pre-established RQ, O<sub>2</sub> partial pressure was decreased by ± 0.02 kPa to increase anaerobic metabolism.

Van Schaik *et al.* (2015) stored 'Elstar' apples for 5 months at 1.8°C in 1.2 kPa O<sub>2</sub> + 2% CO<sub>2</sub>, using the DCR system, and measured RQ and ethanol. In all the rooms the final O<sub>2</sub> levels were set by the measurement of RQ. Their results indicated that 'the DCR system is a promising interactive storage system' and that it also 'reduced respiratory heat production up to a factor of 2 compared to standard ULO, resulting in significant energy savings'. Bessemans *et al.* (2016) found that for 'Granny Smith' apples in storage containers, the DCA-RQ system controlled O<sub>2</sub> and CO<sub>2</sub> partial pressures 'in an autonomous way', with the RQ breakpoint varying between 0.25 kPa and 0.4 kPa O<sub>2</sub>. They reported some difficulties in obtaining reliable RQ measurement at very low O<sub>2</sub> levels, due to the low respiration rate of the fruit. Another issue was reported to be the variation in the estimated level of the CO<sub>2</sub> production rate (Gasser *et al.*, 2003). The storage of 'Galaxy' apples under DCA-RQ 1.3 showed lower ethylene production, lower respiration rates, lower mealiness and higher flesh firmness in comparison with static CA-stored fruit (Thewes *et al.*, 2017c). As the storage temperature increased from 0.5°C to 1.0°C, the incidence of low O<sub>2</sub> injury and flesh breakdown was reduced in 'Royal Gala' apples stored in DCA-RQ (Weber *et al.*, 2015). The meaning of DCA-RQ 1.3, 1.5 and 2.0 is that the numbers refer to the RQ ratio that is set during storage. To achieve this, the O<sub>2</sub> partial pressures are changed during the entire storage period to obtain these RQ levels. Thus, the fruit stored in DCA-RQ 1.5 had a higher anaerobic metabolism (lower O<sub>2</sub> partial pressures) compared with the ones stored in DCA-RQ 1.3 (F.R. Thewes, 2017, personal communication). Bessemans *et al.* (2018) tested a method to correct real-time measurements of RQ for gas leakage in an empty storage container. The model was successfully implemented in an automated DCA-RQ control system under 0.4 kPa O<sub>2</sub> and 1.3 kPa CO<sub>2</sub> using 'Braeburn' apples, which showed that the RQ estimates without a



leakage correction were heavily biased when leakage of atmospheric air into the storage container occurred and would lead to erroneous control of the storage atmosphere composition. Gasser *et al.* (2003) assessed the possibility of adopting this strategy of using RQ by examining, in carefully controlled laboratory conditions, the effect of decreasing  $O_2$  levels on the  $CO_2$  production and ACP of 'Idared' apples. They also evaluated the possible influence of harvest date, storage duration and storage conditions for the sample fruit prior to the measurements. They detected an ACP in all the samples, though the  $CO_2$  production below the ACP was dependent on rate of  $O_2$  reduction, degree of maturity, storage duration of the sample fruit and the previous storage conditions. More specifically, as storage duration increased, apples stored in CA, compared with refrigerated air, showed less increase in  $CO_2$  production below the ACP.

### Safepod™

A commercial system, which measures the  $CO_2$  given out and the  $O_2$  taken in by fruit and calculates RQ, was developed by International Controlled Atmosphere Ltd in the UK and Storage Control Systems in the USA. They called the system Safepod™

and the patent for Safepod™ is held jointly by David Bishop and Jim Schaefer (US Patent No. 8739694, Canadian Patent No. CA2746152) (David Bishop, Storage Control Systems, 2018, personal communication). The Safepod™ container (Fig. 9.2) sits in the CA store and thus has the same temperature, humidity, pressure and atmosphere as the store. Periodically the valves are closed and the  $CO_2$  and  $O_2$  are measured and the RQ is calculated. Safepod™ is based on stress detection on a representative sample of fruit (some 70 kg) and automatic measurement of respiration rate and RQ, which can then be used to automatically adjust the store atmosphere to its optimum  $O_2$  and  $CO_2$  levels. The Safepod™ is a hermetically sealed enclosure with a stainless steel base and a clear moulded cover that sits in a water trough for sealing (Fig. 9.3). Motorized valves and a circulation fan are normally open to maintain the atmosphere within the Safepod™ at the same conditions as the surrounding CA room. DeEll and Lum (2017) used Safepod™ to store 'Empire' apples at 0.6 kPa  $O_2$  + 0.5 kPa  $CO_2$  for 8 months at 1.5°C and 3°C and found that they were firmer after storage than those stored under CA at 2.5 kPa  $O_2$  + 2 kPa  $CO_2$  or 1.5 kPa  $O_2$  + 1.2 kPa  $CO_2$ , but their results 'were from a single season, and further research is needed to confirm these observations'. There are about 70 Safepod™



**Fig. 9.2.** (a) Safepod™ unit with (b) close-up of visual controls. Reproduced with permission from Storage Control Systems Ltd ([www.storagecontrol.com](http://www.storagecontrol.com)).



**Fig. 9.3.** Safepod™ unit, containing apples, on a pallet ready for loading into DCA apple store. Reproduced with permission from Storage Control Systems Ltd ([www.storagecontrol.com](http://www.storagecontrol.com)).

systems in the UK and about 150 in the USA (David Bishop, personal communication, 2018).

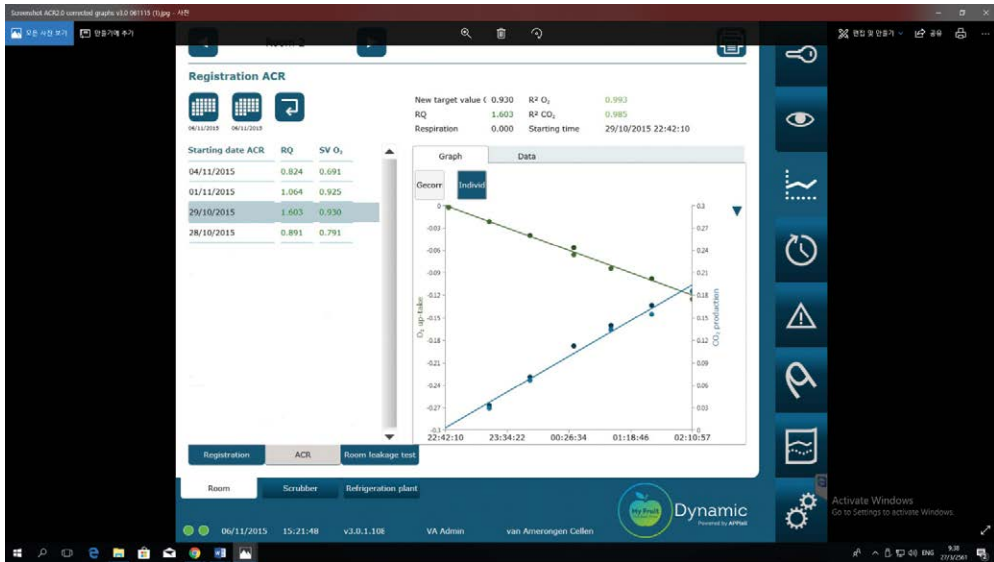
### Advanced Control of Respiration

Another commercial DCA-RQ system is marketed as Advanced Control of Respiration (ACR) by Van Amerongen CA Technology BV (2018). The company has been producing equipment for CA storage of fruit and vegetables since 1969, with equipment installed in more than 50 countries throughout the world. At the time of writing, there are 70 rooms equipped with ACR technology in South Africa, though not all rooms are being used for low  $O_2$  storage (R. Hurdall, H. Ackermann, personal communication), plus 550 worldwide (Rob Veltman, 2018, personal communication).

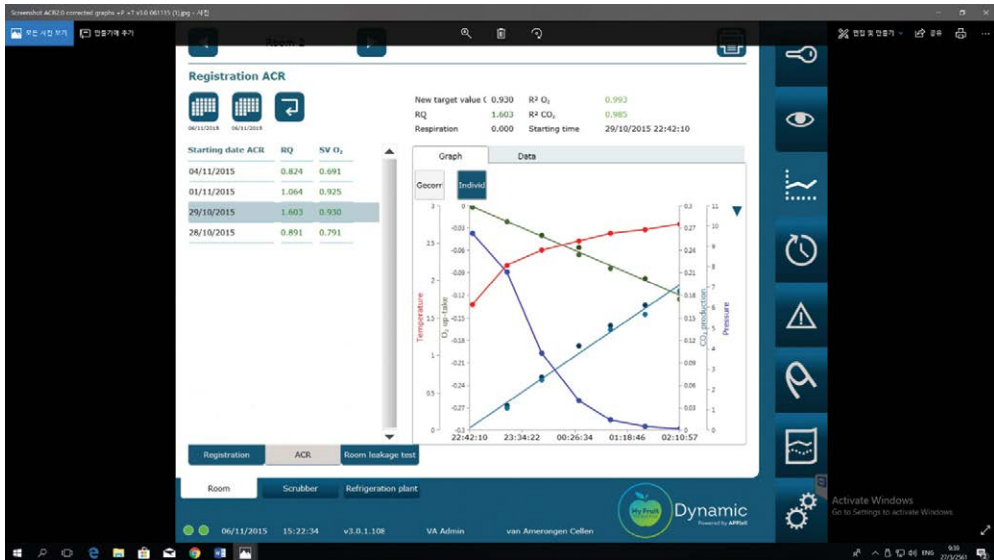
The following information is taken from an unpublished paper supplied by Rob Veltman of Van Amerongen. The development of ACR was initially guided by the need to control scald in some cultivars of pome fruit during long-term storage and to avoid fermentation. The DCA system was therefore designed to continuously check

for fermentation and search for the lowest  $O_2$  level possible throughout the entire storage period. It was considered vital to determine RQ values of the total cold store load, avoiding compartments with samples of fruit because these samples might not be representative of those in the cold store and there could be a microclimate effect in a container. The principle of ACR is based on ‘ACR runs’ during which the respiration rate of the fruit is determined by bringing the store under a slight over-pressure, and preventing any  $O_2$  entering the store during the run. Then the partial pressure of  $O_2$  and  $CO_2$  is measured with an interval time of 30–60 min. Based on these measurements, the control system calculates the respiration rate in terms of  $O_2$  uptake and  $CO_2$  production, calculates the RQ and adjusts the  $O_2$  set point automatically by using a cultivar specific algorithm (Figs 9.4 and 9.5).

Van Amerongen, in collaboration with Wageningen University, showed that during Dynamic Control System (DCS) storage of ‘Elstar’ apples,  $O_2$  partial pressures were adjusted based on the first traces of alcohol in the storage atmosphere. In containers with about 200 kg of apples this led to storage at  $O_2$  partial pressures much lower



**Fig. 9.4.** Screen shot of ACR measuring  $CO_2$  output,  $O_2$  uptake and RQ calculation. Reproduced with permission from Rob Veltman, Managing Director, Van Amerongen BV ([www.van-amerongen.com](http://www.van-amerongen.com)).



**Fig. 9.5.** Screen shot of ACR measuring  $CO_2$  output,  $O_2$  uptake, temperature pressure and RQ calculation. Reproduced with permission from Rob Veltman, Managing Director, Van Amerongen BV ([www.van-amerongen.com](http://www.van-amerongen.com)).

than 1 kPa, even as low as 0.5 kPa. Scaling up DCS to cold stores of typically 150 t showed that ethanol in the storage air was not always a reliable marker for fermentation. Variation in, amongst others, growing

location and harvest date appeared to cause re-metabolization of alcohol. Alcohol produced by fermenting apples seems to be consumed by apples that still respire normally. Re-metabolization is thought to cause

a build-up of ethanol in a part of the apple population in a cold store, while ethanol in the storage air is zero or not detectable.

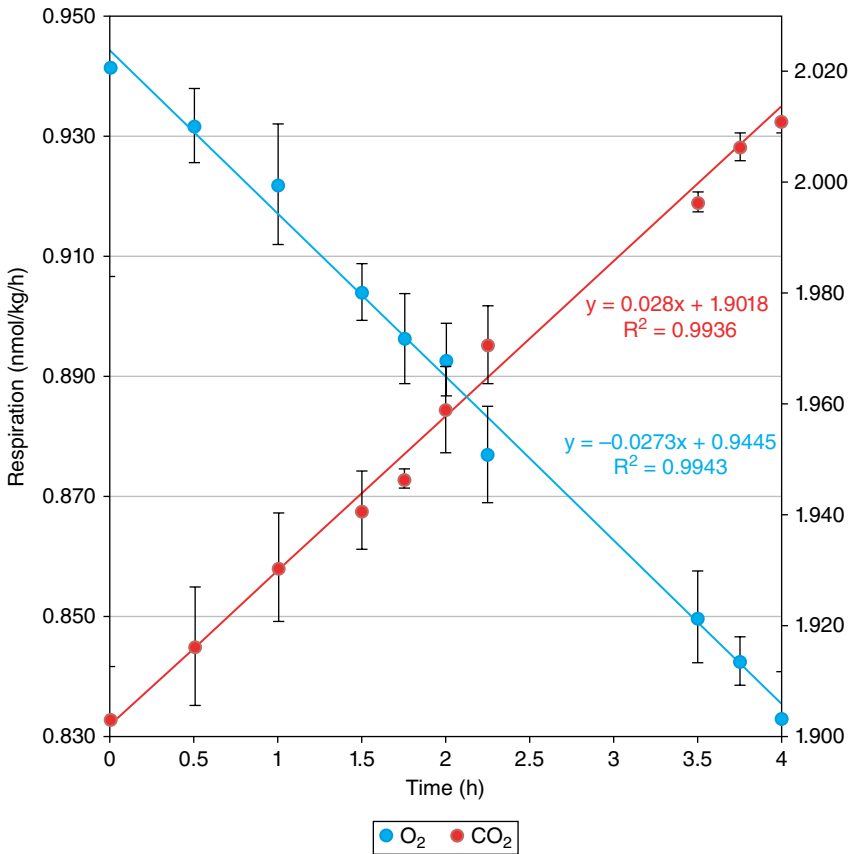
The next step in the trials was trying to avoid re-metabolization. In experiments a compartment was used in the storeroom with a selection of apples from the different sources. Ethanol measurements were carried out on this 20–40 kg sample. However, the compartment introduced several new issues. One of them was the question of which fruits should be selected for the measurements. Next, the compartment was warmer inside, which led to condensation and finally increased rotting. Rotting fruit also produce alcohol, and this made the interpretation of the measurements difficult. Finally, ethanol is not easy to detect. It requires special, expensive meters, but above all, ethanol sticks to tubing and it therefore may not be detected centrally as is the case for O<sub>2</sub> and CO<sub>2</sub>. In 2006 there were seriously doubts about whether alcohol was the right practical indicator to detect product response. After successful experiments in several projects with Wageningen University (see Fig. 1.3) a new dynamic CA system was developed by Van Amerongen called ACR (Veltman, 2013). Up to 2015 the system was called Dynamic Control of Respiration (DCR) and the company was partnered, for a while, with Agrofresh. The name was changed to the preferred Advanced Control of Respiration (ACR). Some of the challenges that needed to be overcome in applying ACR were as follows.

- The cooling system had to be interrupted during an ‘ACR run’ (the period the respiration rate was determined).
- The same for the activity of all CA equipment, such as the CO<sub>2</sub> scrubber, the N<sub>2</sub> generator and aeration.
- A cold store should be gas-tight enough. To overcome slight leakages that would let in O<sub>2</sub> and disturb the measurement of O<sub>2</sub> uptake, the cold store was brought under a slight overpressure at the start of an ACR run by injecting N<sub>2</sub> (10 mm water column).
- The sensitivity of the O<sub>2</sub> and CO<sub>2</sub> meters could be an issue, and calibration of the meters needed to be reconsidered.

Experiments were successfully carried out at two local apple-packing houses. The lines representing the respiration rates in Fig. 9.6 had  $R^2$  levels that were always higher than 97%. ACR has been patented (Veltman, 2013) and a newer version has been developed where ethylene production is also monitored.

## Ethanol

Another approach to using product physiology to control the O<sub>2</sub> content in CA stores is the measurement of ethanol content in the store room, based on the assumption that the appearance of ethanol indicates the LOL and onset of anaerobic respiration (fermentation) in the stored crop. Measuring atmospheric ethanol is further supported by the observation that there is a stable ratio between the ethanol content in the tissue and the atmosphere around the crop (North and Cockburn, 1975). However, they warned that this stable ratio is suitable for stores equipped with lime CO<sub>2</sub> scrubbers but not for stores equipped with activated carbon scrubbers. North and Cockburn (1975) described detection for ethanol in the stores using indicator tubes called ‘Alcotest’ that contained crystals, which changed colour from yellow to green in the presence of alcohol. An approximate quantitative assessment of the alcohol level was achieved by measuring the quantity of store air required to achieve this colour change. For use in commercial CA stores, the alcohol detecting device could be connected to an alarm system so that when fermentation was just beginning the O<sub>2</sub> level could be slightly increased to stop fermentation levels, so that the cell tissues would return to aerobic respiration and the crop would not be damaged. For the apple cultivars studied by Fidler *et al.* (1973) the fruit could fully recover from alcohol contents of less than 40 mg/kg when subsequently aerated. The next report of O<sub>2</sub> control using ethanol was 20 years later by Schouten (1995), who concluded that a DCS for the O<sub>2</sub> content in CA storage rooms was possible, based on his results with ‘Jonagold’ apple and Brussels



**Fig. 9.6.** Respiration of Elstar apples stored at 0.9% O<sub>2</sub> in a 125 t cold store. There were four measuring points that were averaged. There was minimal variation in the O<sub>2</sub> and CO<sub>2</sub> values at the four different measuring points. Reproduced with permission from Rob Veltman, Managing Director, Van Amerongen BV ([www.van-amerongen.com](http://www.van-amerongen.com)).

sprouts. Schouten *et al.* (1997) confirmed that DCS storage of ‘Elstar’ apple, in which O<sub>2</sub> is adjusted to maintain ethanol below 1 ppm in the headspace (0.3–0.7 kPa O<sub>2</sub> + < 0.5 kPa CO<sub>2</sub>), maintained fruit quality better than ULO-stored fruit (1.2 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>). See also the previous section, ‘Advanced Control of Respiration’, where some of the challenges in measuring ethanol levels in CA stores are discussed by Veltman.

A closely related approach, ILOS+, was introduced during the same period. Initial Low Oxygen Stress (ILOS) is a technique, first described by Eaves *et al.* (1969a, b), which uses short (up to 2 weeks) low O<sub>2</sub> stress treatments, i.e. typically but not always ≤ 0.5 kPa. ILOS+ is repeated applications of

low O<sub>2</sub> stress at various times in the storage period with the addition of destructive ethanol tissue analysis used to determine when to stop treatment.

### Dynamic Control System

DCS was developed and introduced by Food & Biobased Research at Wageningen Agricultural University in The Netherlands in 1997. It was first licensed to Van Amerongen (VA), which eventually abandoned the DCS method and switched to the ACR method. The DCS system carries out a pulp analysis for the presence of ethanol. When

no ethanol is detected, the O<sub>2</sub> level in the CA room can be decreased, but when ethanol is detected the O<sub>2</sub> can be set to increase automatically. DCS uses small sample chambers integrated into the main store enabling low-level measurements of ethanol from the fruit or vegetable samples when the ACP is reached. Veltman *et al.* (2003) described DCS as 'the oxygen concentration is slowly lowered until the product emits a stress-signal. More specifically, DCS uses ethanol, the final product of fermentation, to establish the lowest possible oxygen concentration under which a product can be stored.' They described storage of 'Elstar' apples and showed that they were firmer after DCS storage, especially after shelf-life, with better colour retention and less skin spots compared with traditional CA storage. Paillart (2013) reported DCS being successfully used for long storage of white cabbage where the minimum O<sub>2</sub> concentration was adjusted by controlling the ethanol concentration within the storage facilities. He also claimed that the energy consumption of storage under DCS was lower than other ULO systems. DCS is marketed by Storex BV and is used in more than 200 storerooms (Storex, 2018), mainly for 'Elstar' and 'Jonagold' apples in The Netherlands (Prange *et al.*, 2013) plus at least 8000 t of stored apples (cultivars not specified) in other countries, e.g. Poland.

### Initial Low Oxygen Stress

1-MCP, CA, DCA and ILOS have all been shown to have potential to inhibit superficial scald and increase the storage life of apples and pears (Robatscher *et al.*, 2012). ILOS is essentially a pre-storage treatment where fruit are exposed to anaerobic conditions; commonly 7 days in 100% N<sub>2</sub> (Eaves *et al.*, 1969a) or 0.5 kPa O<sub>2</sub> (Little *et al.*, 1982) who showed that the optimum for scald control in 'Granny Smith' apples was 0.5 kPa O<sub>2</sub> rather than 0 kPa O<sub>2</sub>, for 9–14 days followed by CA storage. ILOS exposure of 'Granny Smith' to 0.5 kPa O<sub>2</sub> reduced scald compared with continuous storage at 1 kPa

in Australia, but this did not positively affect 'Red Delicious' in the same trial. In another trial, ILOS (10 days at 0.5 kPa O<sub>2</sub>) followed by CA (1 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> at 1°C) without DPA controlled scald on 'Granny Smith' and 'Red Delicious' apples in South Africa. ILOS also markedly reduced scald in 'Granny Smith', 'Rome' and 'Red Delicious' in Michigan trials (Kupferman 2001a). Sabban-Amin *et al.* (2011) also found that apples stored under ILO showed better post-storage quality, due to lower superficial scald, higher flesh firmness and greener skin colour. ILOS+ (ILOS with ethanol monitoring) was reported to be used on about 19,000 t of stored apples (cultivars not specified) in the South Tyrol region of Italy (Robert Prange, 2018, personal communication).

### Chlorophyll Fluorescence

The above methods rely on measuring changes in ACP, RQ or ethanol in the atmosphere, or in tissue, that could signal that the product is at or near its LOL. The third system to determine the LOL is the use of chlorophyll fluorescence (CF), which does not need to measure ethanol or other fermentative products. Beginning in the 1980s, CF was being examined as method to detect various stresses (Prange, 1986; Prange *et al.*, 1990). Prange and research associates in his laboratory continued research into the 1990s on the effect of postharvest stresses on CF, culminating in the discovery in 1999–2000 of a sudden and reversible 'spike' in CF when O<sub>2</sub> is lowered below the product's apparent LOL. They concluded that it is a very sensitive, non-destructive method of dynamically controlling the O<sub>2</sub>, and possibly the CO<sub>2</sub> environment, according to the unique requirements of each product.

Prange *et al.* (2003b) described the commercial version of this technology, which measures CF on a group of fruit or vegetables using a periodic low irradiance. It was first adopted commercially in the 2003–2004 storage season in Washington State, USA, and South Tyrol, Italy. This system is a pulse frequency modulated (PFM)

proprietary technology that uses extrapolation to produce a theoretical estimate of the minimum fluorescence ( $F_o$ ) parameter for which they coined a new term,  $F_{\alpha}$ . With DCA, one can detect changes in the LOL and immediately alter the  $O_2$  level in the storeroom. The LOL varies with the product, cultivar, season, growing regions and harvests, as well as time in storage (Table 9.1).

In the example in Table 9.1, three of the four apple cultivars had a drop in the LOL of about 0.4 kPa  $O_2$  and in the fourth cultivar, 'Empire', there was no change in LOL. Subsequent research shows that this CF system can detect not only the LOL but other stresses experienced by stored product, e.g.  $CO_2$ , low temperature (chilling), 1-MCP application, and the presence of toxic ammonia, refrigeration gas and water loss (Prange *et al.*, 2010, 2012). These stresses, except for water loss, result in an immediate increase in  $F_{\alpha}$ , similar to the increase caused by  $O_2$  below the LOL.

When it was first introduced, the application of the CF method in DCA was described as dynamic low- $O_2$  controlled atmosphere (DLOCA) (DeLong *et al.*, 2004). Another term used by a few researchers was fluorescence CA (FCA) (Vanoli *et al.*, 2007). Other users have embraced neither of these terms. Although DCA did not initially refer solely to the DCA-CF method, many researchers and users referred to the two DCA-ethanol methodologies above as DCS and ILOS+, respectively, and shortened DCA-CF to just DCA (e.g. Zanella *et al.*, 2005, 2008; Raffo *et al.*, 2009; Gasser *et al.*, 2010; Streif *et al.*, 2010). Since 2012, in order to avoid confusion, especially with the RQ methods, Prange and Zanella began using DCA-CF to distinguish it from DCA-RQ.

**Table 9.1.** The effects of time in storage on the LOL (kPa  $O_2$ ) detected by DCA-CF on four apple cultivars (Prange *et al.*, 2013).

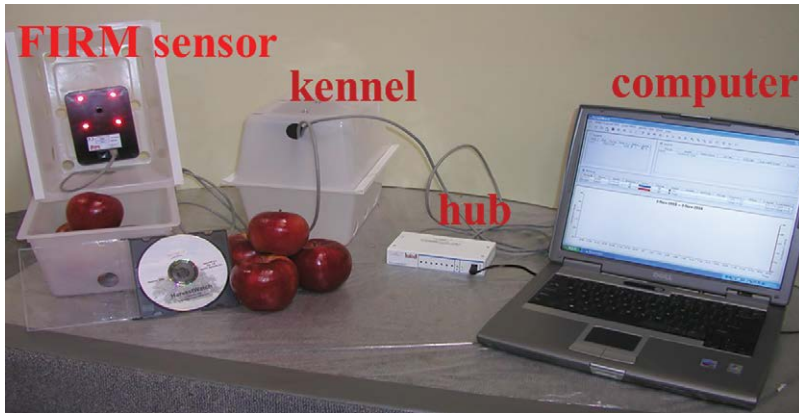
Apple cultivar	10–19 October	1–4 December
Delicious	0.85	0.47
Golden Delicious	0.92	0.45
Honeycrisp	0.90	0.50
Empire	0.90	0.88

### HarvestWatch™

The discovery of a sudden and reversible 'spike' in CF when  $O_2$  was lowered below the product's apparent LOL resulted in a commercial system, using this discovery, being first presented at the ISHS CA symposium in Rotterdam, The Netherlands in 2001, and subsequently published (Prange *et al.*, 2002, 2003b). The patented DCA system is globally marketed as HarvestWatch™ and in Italy it is marketed by Isolcell S.p.A. as IsoStore® (Isolcell Italia, 2018) and used around the world, mainly in commercial apple (Prange *et al.*, 2010) and pear storage (Prange *et al.*, 2011) (Figs 9.7, 9.8 and 9.9). The current estimate of its usage at the end of 2017 was about 1979 DCA-CF rooms of > 20 cultivars of apples and pears in > 26 countries supported by research in > 19 research laboratories. The growth in DCA-CF was initially in Italy (Fig. 9.10) and after 2013 adoption outside Italy became more dominant, with a steady growth rate overall. Robert Prange (2018, personal communication) reported that DCA-CF accounts for about 85–90% of the DCA technology usage with the remaining 10–15% being ethanol and RQ-related technologies.

### Fruit Observer™

A DCA-CF system was developed by Besseling Group BV and their partners in Germany and Brazil to monitor and control fruit during DCA storage. The system continuously measures CF,  $O_2$ ,  $CO_2$ , temperature and relative humidity in CA rooms and shows a reaction of the chlorophyll when the LOL has been reached. The Fruit Observer™ software provides continuous information about the physiological condition of the observed fruit (Fig. 9.11). The initial screen of the software is split into two graphs showing: (1) actions – levels of  $O_2$ ,  $CO_2$ , temperature and relative humidity; and (2) reactions – chlorophyll response. A description of the system was given in an article in a commercial fruit journal by Pullano (2017).



**Fig. 9.7.** A DCA-CF (HarvestWatch™) system consists of multiple FIRM sensors, each mounted within a kennel and connected using CAT 6/7 cables to a hub and then to a monitoring computer with HarvestWatch™ software.



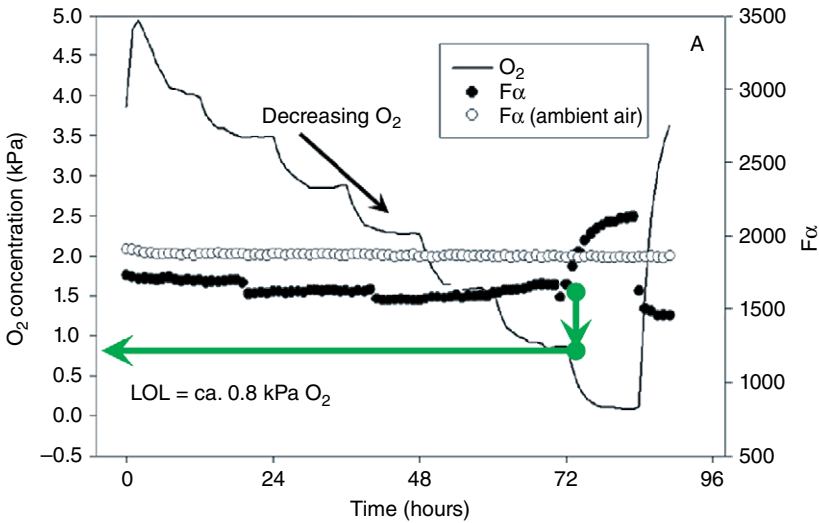
**Fig. 9.8.** Typical arrangement of six DCA-CF (HarvestWatch™) FIRM kennels connected to one hub in a commercial CA room.

### Comparison of Ethanol, RQ and CF Systems

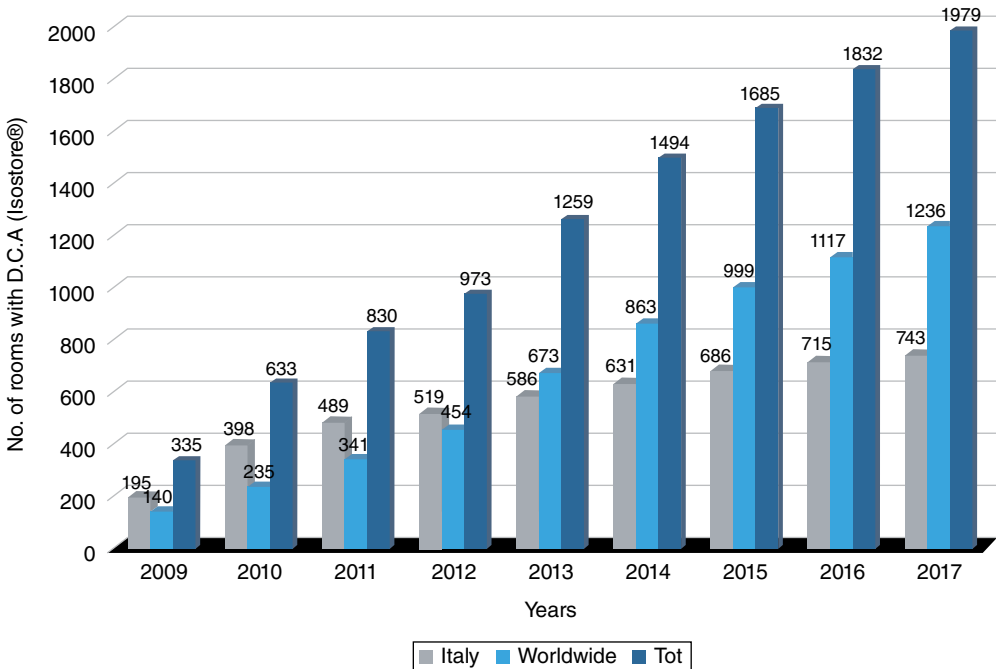
Although it is not technically possible to compare concurrently the RQ, ethanol and CF commercial systems, one can simultaneously measure ethanol production, CO<sub>2</sub> production and CF and their accuracy in identifying the LOL. Hourly values of CF from ‘Honeycrisp’ and ‘Delicious’ apples, compared with CO<sub>2</sub> production (ACP), are presented in Fig. 9.12 and ethanol production

in Fig. 9.13. They show that lowering O<sub>2</sub> had measurable effects on all three responses, i.e. ethanol production, CO<sub>2</sub> production and CF. The difference was mostly in the ease of interpretation. Chlorophyll response was easily interpreted, whereas the ethanol and CO<sub>2</sub> measurements were less precise, resulting in uncertainty in the LOL value, compared with CF. The CF LOL, which was the O<sub>2</sub> concentration at which the CF value did not return to the baseline, corresponds with the ACP (lowest CO<sub>2</sub> production) (Fig. 9.12).





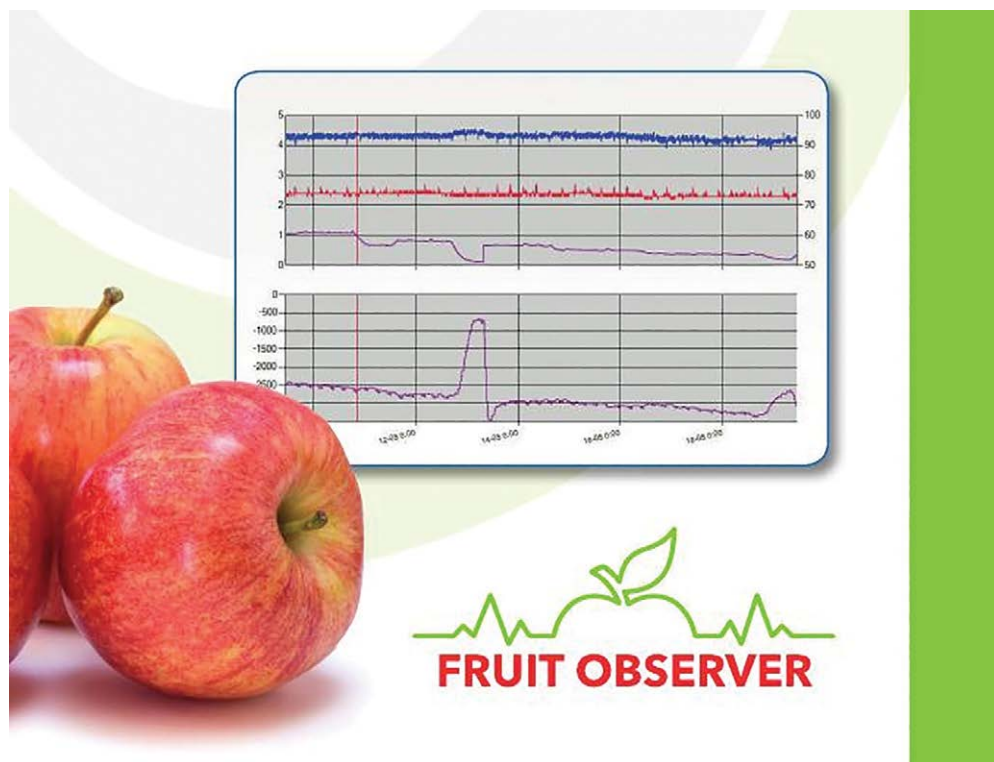
**Fig. 9.9.** Using HarvestWatch™ to determine LOL in an apple sample. The O<sub>2</sub> is gradually lowered and the chlorophyll fluorescence signal (F<sub>α</sub>, solid dots) suddenly increases at ca. 72 h. The LOL is the O<sub>2</sub> value (0.8 kPa) when the increase (spike) begins. The O<sub>2</sub> is increased above the LOL at 84 h and the F<sub>α</sub> signal immediately returns to its baseline.



**Fig. 9.10.** Number of storerooms equipped with DCA-CF between 2006 and 2017.

Furthermore, the CF returns to the baseline after anoxia is terminated, whereas the CO<sub>2</sub> production does not return to its normoxia values and is highly variable. Ethanol

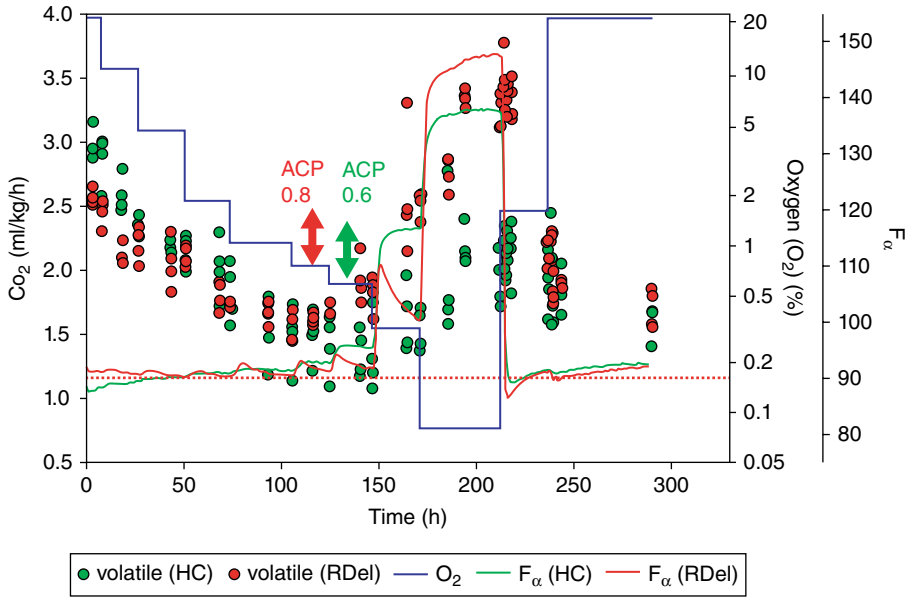
production (Fig. 9.13) was even more difficult to interpret. ‘Honeycrisp’ ethanol production occurs in normoxia and after anoxia, ethanol production continues in normoxia.



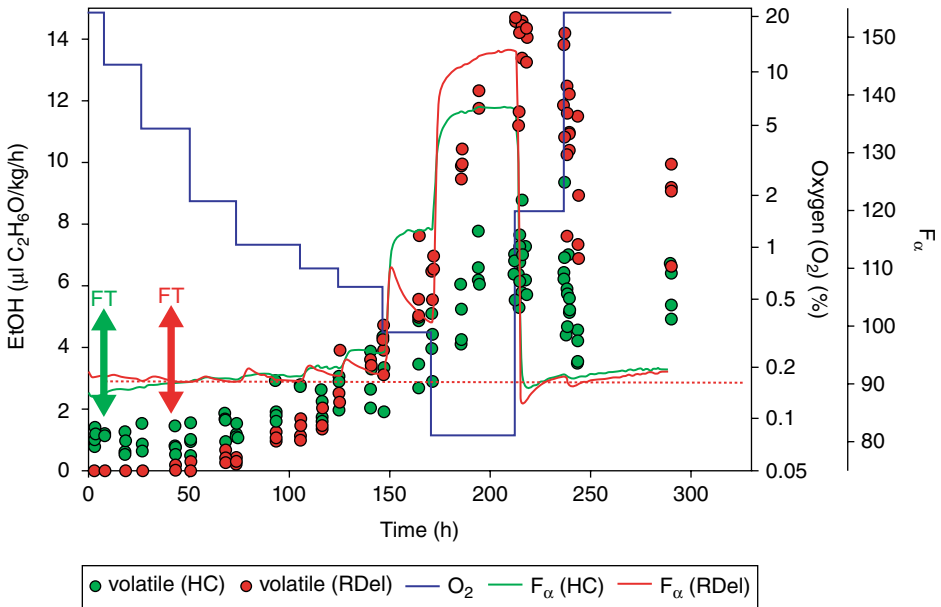
**Fig. 9.11.** Fruit Observer™. The upper graph shows which parameter triggered the reaction of chlorophyll, which is simultaneously displayed in the lower graph. The upper graph shows the actions: blue is humidity; red is temperature; and purple is O<sub>2</sub> level. The lower graph shows the reaction on the actions. The O<sub>2</sub> is too low, which caused a peak. The O<sub>2</sub> was then slowly lowered and again there was a peak at exactly the same LOL point. This shows that the chlorophyll recovers as long as the O<sub>2</sub> is set at the right level. Reproduced with permission from Eric van der Zwet of Besseling Group BV, The Netherlands.

Weber *et al.* (2015) compared static CA with DCA-RQ 2 and DCA-CF on ‘Royal Gala’ apples. Flesh firmness, ethylene production and respiration rates were lower in apples stored in DCA-RQ and DCA-CF than in CA. They concluded that storage in DCA-RQ 2 at 1°C or DCA-CF at 0.5°C maintained fruit quality during 8 months of storage. In a comparison between CA, DCA-CF and DCA-RQ on ‘Fuji Suprema’ apples, Thewes *et al.* (2017a) found that storage under both DCA methods had lower ethylene production and retained higher flesh firmness compared with those under more conventional CA. The fruit stored under CA had the highest butyl acetate, 2-methylbutyl acetate and hexyl acetate. DCA-RQ resulted in lower decay incidence when the atmosphere was established immediately, compared with a

30-day delay in establishment. However, the apples stored in DCA-RQ had the highest total ester concentration and the volatile compounds that are characteristic to ‘Fuji’ apples (ethyl 2-methyl butanoate, ethyl butanoate and ethyl hexanoate). The storage under DCA-CF resulted in the lowest production of volatile compounds. Thewes *et al.* (2017b) stored ‘Galaxy’ apples in CA at 1.2 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>, DCA-RQ 1.3 + 1.2 kPa CO<sub>2</sub> or DCA-RQ 1.5 + 1.2 kPa CO<sub>2</sub> for 9 months plus 7 days of shelf-life at 20°C. DCA-RQ suppressed internal ethylene concentration, ethylene production and respiration rate and the highest ester concentration occurred in overripe fruit stored at DCA-RQ 1.3 and DCA-RQ 1.5, while those stored in DCA-RQ 1.5 had higher concentrations of total esters and characteristic aroma volatile



**Fig. 9.12.** Effect of O<sub>2</sub> reduction, prolonged anoxia and return to air on CO<sub>2</sub> production at 3.5°C in six ‘Delicious’ and six ‘Honeycrisp’ apples. RDel = ‘Delicious’ (Red); HC = ‘Honeycrisp’ (Green); ACP = anaerobic compensation point = LOL; dotted line = chlorophyll fluorescence baseline.



**Fig. 9.13.** Effect of O<sub>2</sub> reduction, prolonged anoxia and return to air on ethanol (EtOH) production at 3.5°C in six ‘Delicious’ and six ‘Honeycrisp’ apples. RDel = ‘Delicious’ (Red); HC = ‘Honeycrisp’ (Green); FT = Fermentation Threshold; dotted line = chlorophyll fluorescence baseline.

compounds. DCA storage reduced ethylene production and respiration rate of ‘Royal Gala’ apples stored for 9 months at 1.0°C plus 7 days at 20°C, resulting in higher flesh firmness, TA and less physiological disorders. Storage in static CA, DCA-RQ 1.5 and DCA-RQ 2.0 had higher levels of butyl acetate, 2-methylbutyl acetate and hexyl acetate compared with fruit stored in DCA-CF. Fruit stored in DCA-RQ 1.5 and DCA-RQ 2.0 had higher increments of ethanol and ethyl acetate, but far below the odour threshold (Both *et al.*, 2017). Patulin is a mycotoxin produced postharvest by some fungi infecting apples (see Chapter 8). Dos Santos *et al.* (2018) tested whether DCA storage could reduce patulin production in apples. They stored ‘Galaxy’ at  $2.0 \pm 0.1^\circ\text{C}$  and ‘Fuji Kiku’ at  $0.5 \pm 0.1^\circ\text{C}$  for 9 months plus 7 days shelf-life at 20°C in CA (1.0 kPa O<sub>2</sub> + 0.8 kPa CO<sub>2</sub>), DCA-CF + 1.2 kPa CO<sub>2</sub> or DCA-RQ 1.3 + 1.2 kPa CO<sub>2</sub>. They found that none of these storage conditions prevented patulin production.

### Current Situation

Most of the published DCA research has been conducted on apples and pears (Table 9.2) but there are research and commercial trial reports on a variety of other fruit and vegetables. The first two regions to adopt commercial DCA-CF were South Tyrol (Italy) and Washington State (USA) in 2003, after a trial season. In the South Tyrol, 88% of the apple crop was stored in CA in 2016–2017 (Fig. 9.14). About 21.8% of this amount was stored in DCA: 19.4% in DCA-CF + 2.4% in ILOS+, a DCA-ethanol technology. In Washington State, Stemilt Growers first used DCA-CF commercially in 2003 on organic ‘Delicious’ apples and there has been a steady increase in DCA since then. In a survey in November 2017 that covered about 50% of the apple storage in Washington State (Table 9.3), 88% of the apples were in CA, which is similar to the South Tyrol, but only 3% were in DCA, compared with 21.8% in the South Tyrol. However,

the planned construction of new storages will be 29% DCA.

Even though ethanol and RQ methods were known before 2001, significant commercial adoption of DCA only began with the availability of DCA-CF in 2001. The explanation is a combination of factors. Firstly, DCA-CF offered a package of technical features that were appealing (Prange *et al.*, 2013):

- accurate and fast detection of the internal flesh ACP (= LOL);
- durable and reliable (never need to calibrate or replace components);
- non-destructive repeat measurements on large surface areas;
- rapid and frequent measurement;
- non-chemical;
- real-time monitoring of produce allowing for on-site or remote monitoring and archiving of data for future reference; and
- detection of changes due to senescence, decay or incorrect storage conditions, i.e. temperature, unwanted toxic gases such as ammonia refrigerant.

Secondly, independent research and commercial trials around the world showed that using DCA-CF to hold O<sub>2</sub> to just above the ACP not only preserved quality in apples and pears, it also controlled several disorders, especially superficial scald. This bolstered the commercial adoption, especially in South Tyrol, and in 2010–2011 the South Tyrol industry decided to rely solely on DCA-CF and/or 1-MCP for superficial scald control rather than the postharvest chemical DPA (Zanella and Stürz, 2013). The third factor occurred in 2012 when the EU decided that DPA and ethoxyquin were no longer acceptable in the EU. This meant no apples or pears stored in the EU or imported into the EU could be treated with these two chemicals. The significant increase in demand for organic fruit has also been a major factor, especially in Washington State where organic apple production is approaching 20%. DCA is an acceptable postharvest technology for organic fruit, but 1-MCP, DPA and ethoxyquin are not acceptable.

**Table 9.2.** List of some publications providing information on DCA use in fruits and vegetables.

Commodity	Cultivar	Publications	
<b>Fruits</b>			
Apple	Braeburn	Gasser <i>et al.</i> , 2008; Lafer, 2008; Lafer, 2009; Poldervaart, 2010a; Prange <i>et al.</i> , 2012; Withnall, 2008	
	Cortland	DeLong <i>et al.</i> , 2004, 2007; Prange <i>et al.</i> , 2012; Wright <i>et al.</i> , 2012	
	Cripp's Pink (Pink Lady)	Withnall, 2008; Prange <i>et al.</i> , 2012	
	Delicious (Red)	DeLong <i>et al.</i> , 2004, 2007; Lafer, 2009; Prange <i>et al.</i> , 2012; Stephens and Tanner, 2005; Withnall, 2008; Wright <i>et al.</i> , 2012; Zanella and Stürz, 2013	
	Elstar	Gasser <i>et al.</i> , 2008; Köpcke, 2009; Poldervaart, 2010a; Prange <i>et al.</i> , 2012	
	Fuji	Withnall, 2008; Zanella <i>et al.</i> , 2008	
	Gala	Withnall, 2008; Zanella <i>et al.</i> , 2008	
	Golden Delicious	DeLong <i>et al.</i> , 2004; Gasser <i>et al.</i> , 2008; Poldervaart, 2010a; Prange <i>et al.</i> , 2012; Withnall, 2008; Zanella <i>et al.</i> , 2008	
	Honeycrisp	DeLong <i>et al.</i> , 2004; Prange <i>et al.</i> , 2010; 2012; Wright <i>et al.</i> , 2008, 2010, 2011, 2012	
	Granny Smith	Lafer, 2009; Prange <i>et al.</i> , 2012; Wright <i>et al.</i> , 2012; Withnall, 2008; Zanella <i>et al.</i> , 2005	
	Idared	Poldervaart, 2010a; Prange <i>et al.</i> , 2012	
	Jonagold	DeLong <i>et al.</i> , 2004; Prange <i>et al.</i> , 2012	
	Maigold	Gasser <i>et al.</i> , 2008; Poldervaart, 2010a	
	McIntosh	DeLong <i>et al.</i> , 2004; Prange <i>et al.</i> , 2002; 2003	
	Northern Spy	Prange <i>et al.</i> , 2005	
	Pinova (Piñata)	Raffo <i>et al.</i> , 2009	
	Topaz	Lafer, 2009; Poldervaart, 2010a,b	
	Avocado	–	Prange <i>et al.</i> , 2003; Yearsley <i>et al.</i> , 2003
		Hass	Burdon, 2009; Burdon and Lallu, 2008; Burdon <i>et al.</i> , 2008; 2010; Prange <i>et al.</i> , 2002; Yearsley <i>et al.</i> , 2003
Banana	–	Prange <i>et al.</i> , 2002; 2003	
Kiwifruit	–	Prange <i>et al.</i> , 2002; 2003	
	Hayward	Lallu and Burdon, 2007	
Mango	–	Prange <i>et al.</i> , 2002; 2003	
Pear	–	Prange <i>et al.</i> , 2003	
	Abbé Fétel (Abaté Fétel)	Rizzolo <i>et al.</i> , 2008; Vanoli <i>et al.</i> , 2007, 2010a,b; Zerbini and Grassi, 2010	
	Bartlett (Williams Bon Chrétien)	Mattheis, 2007; Prange <i>et al.</i> , 2002, 2011; 2012	
	Conference	Zerbini and Grassi, 2010	
	d'Anjou	Mattheis, 2007; Mattheis and Ruddell, 2011; Prange <i>et al.</i> , 2011; 2012	
	Forelle	Prange <i>et al.</i> , 2011	
	Packham's Triumph	Prange <i>et al.</i> , 2011, 2012	
	Spadona	Prange <i>et al.</i> , 2011	
Uta	Lafer, 2011		
<b>Vegetables</b>			
Asparagus	–	Prange <i>et al.</i> , 2005	
Broccoli	–	Prange <i>et al.</i> , 2005	
Cabbage	–	Prange <i>et al.</i> , 2003	
Pepper (green)	–	Prange <i>et al.</i> , 2003	
Lettuce	Iceberg	Prange <i>et al.</i> , 2003, 2005	
	Romaine	Prange <i>et al.</i> , 2003	
Spinach	–	Prange <i>et al.</i> , 2012; Wright <i>et al.</i> , 2011, 2012	

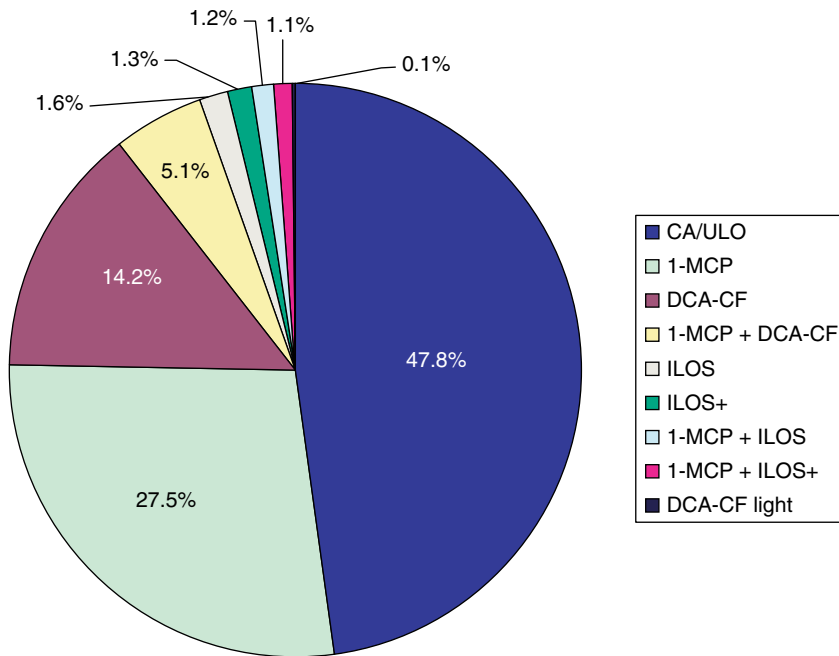
### Future Trends

Two benefits of DCA that were not initially anticipated but are becoming more evident are energy conservation and flavour enhancement. DCA results in a substantial reduction

**Table 9.3.** Current apple storage and plans for the future for apples in Washington State apple industry (adapted from R. Blakey, 2018, personal communication).

Storage type	Current (2017)		Plans for next 2 years	
	Bins	%	Bins	%
Refrigerated Air (RA)	450,000	12	0	0
Controlled Atmosphere (CA)	3.2 million	85	332,000	71
Dynamic CA (DCA)	101,000	3	133,000	29

in fruit respiration, with the result that higher storage temperatures are possible, thus saving energy (J. Streif, as quoted in Poldervaart, 2010a). In New Zealand, experiments have been conducted using a combination of DCA and high temperature storage to achieve energy savings during storage, and/or avoid chilling-related storage disorders in ‘Royal Gala’ and ‘Pink Lady’ apples (N. Lallu, personal communication). Preliminary results show that, at 5°C, storage savings of 35% were possible during cooling and 15% during storage. In addition, flesh browning in ‘Pink Lady’ could be avoided by storage at 3°C in DCA. Energy saving and quality benefits have also been reported by Kitterman *et al.* (2015a) and Neuwald *et al.* (2016) on ‘Golden Delicious’, ‘Jonagold’ and ‘Pinova’ apples in Germany. In yellow-fleshed kiwifruit, chilling injury was avoided by storage at 7°C in DCA whilst maintaining the same firmness as fruit stored at 0°C in air (N. Lallu, personal communication). Thus, with more research



**Fig. 9.14.** CA storage technologies used in the apple industry in South Tyrol, Italy, in 2016–2017. Total CA storage = 764,315 t. An additional 100,000 t were stored in Regular Atmosphere (RA). Initial Low Oxygen Stress (ILOS) technology is not a DCA technology since it does not use fruit physiology to control the CA conditions. DCA-CF light uses fewer sensors per store and a higher O<sub>2</sub> safety margin. Total DCA-CF (DCA-CF, 1-MCP + DCA-CF, DCA-CF light) = 19.2%. Total ILOS+ (ILOS+, 1-MCP + ILOS+) = 2.4%. Adapted from Zanella (personal communication).

and commercial trials, it is likely that the use of DCA will allow the use of warmer storage temperatures.

After its introduction, commercial users of DCA have reported anecdotally that DCA enhances flavour of apple cultivars such as 'Pink Lady' and 'Granny Smith', compared with other storage technologies being used. These reports were confirmed by Zanella *et al.* (2005) and Raffo *et al.* (2009), who concluded that DCA storage technology, besides avoiding any chemical treatment, can preserve

apple flavour/aroma compounds better than 1-MCP + ULO during long-term storage. Raffo *et al.* (2009) observed that DCA favoured the production of branched-chain esters over straight-chain esters. In the case of apples, the flavour enhancement is adding demand and value to the product in the market place. This is an unforeseen benefit of DCA that may extend beyond apple cultivars and suggests that DCA may be able to alter flavours and aromas in a predictable way and/or increase flavour-life of the product.

# 10

## Hyperbaric and High-oxygen Storage

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Hyperbaric storage was recently reviewed in more detail by Thompson (2015) and much of the information contained here has been updated from that book. Increasing the pressure (hyperbaric conditions) in a sealed room or cabinet can increase the partial pressure of O<sub>2</sub>. Hyperbaric conditions have been used in therapy to control human medical conditions, usually in a specially designed pressure chamber. Hyperbaric O<sub>2</sub> therapy enables the patient to breathe pure O<sub>2</sub> while under increased atmospheric pressure. Its use has included the treatment of decompression sickness, gas gangrene and carbon monoxide poisoning as well as being tested for the control of cerebral palsy and multiple sclerosis (Mathieu, 2006).

High pressures are used in food processing and were perhaps first proposed by Royer in 1895 to kill bacteria (Naik *et al.*, 2013). Ahmed and Ramaswamy (2006) and Baba and Ikeda (2003) stated that hyperbaric treatment of fresh fruit and vegetables is different from high-pressure treatment used in processing foods where O<sub>2</sub> pressures of between 400 and 1200 MPa are used. Goyette *et al.* (2011) defined hyperbaric storage as 'exposing fruit or vegetable to compressed air in a range lower than 10 atmospheres'. Hite *et al.* (1914) reported that exposure of fruit or vegetables to 680 MPa for 10 min at room temperature gave a 5–6 log cycle reduction in

microorganisms. The first reports of commercial hyperbaric processing were probably in the early 1990s in Japan for acid foods to be stored in chilled conditions. Such high pressures for fresh fruit and vegetables can cause irreversible damage to cell structure. Goyette *et al.* (2007) stated that it may be possible to use much lower hyperbaric pressures for fruit and vegetables than for processed products and pressures greater than 100 MPa may be above the threshold for irreversible tissue damage, thus causing substantial injuries (Goyette *et al.*, 2012). High-pressure inactivation of yeast and moulds has also been reported, for example in citrus juices that had been exposed to 400 MPa for 10 min at 40°C. Juice treated in this way did not spoil during storage for up to 3 months (Olsson, 1995). Hyperbaric conditions have also been used successfully in increasing the storage time of meat and eggs at room temperature for several days where compressed air, at 15 to 44 atmospheres (atm), was used. It was reported that bacteria, yeasts and moulds were killed in foods by this treatment (Bert, 1878 quoted by Kader and Ben-Yehoshua, 2000). Charm *et al.* (1977) reported that at high pressure the storage life of Atlantic cod fish was greatly extended compared with storage at atmospheric pressure. After 30 days the fish held at –3°C and 238 atm had significantly lower bacterial counts and a higher sensory



evaluation than those stored  $-25^{\circ}\text{C}$  under atmospheric pressure. Moreira *et al.* (2015) compared storage of soup at different pressures and temperatures. They concluded that 100 MPa compared with 150 MPa and 4 h exposure compared with 8 h resulted in more pronounced microbial growth inhibition and microbial inactivation. Aerobic mesophiles showed less susceptibility to hyperbaric storage compared with Enterobacteriaceae and yeast and moulds. Hyperbaric storage at  $25^{\circ}\text{C}$  or  $30^{\circ}\text{C}$  generally maintained the physicochemical parameters at levels similar to refrigeration at  $4^{\circ}\text{C}$  under normal atmospheric pressures. High pressure was reported to have potential as a treatment to control quarantine insect pests in fresh or minimally processed fruit and vegetables (Butz *et al.*, 2004).

A novel way of increasing pressure is storage in sealed containers underwater. Storage of cereals underwater at a depth of 30 m maintained a constant low temperature as well as increased pressure and successful experiments were carried out in Japan using this technique (Mitsuda *et al.*, 1972). Saraiva (2014) observed that food that had been recovered from a sunken submarine in the 1990s was still in a consumable condition. The submarine had been sunk 10 months earlier to a depth of 1540 m where the pressure was about 15 MPa and the temperature  $3\text{--}4^{\circ}\text{C}$ .

Hyperbaric conditions, even at variable room temperatures of up to  $37^{\circ}\text{C}$ , have been shown to preserve foods and thus achieve significant energy savings (Fernandes *et al.*, 2015). Hyperbaric storage, at room temperature, could be more energy efficient than refrigeration, since the only energy costs are during compression and no additional energy is required to subsequently maintain the product under pressure. Liplap *et al.* (2012) showed that hyperbaric storage of avocados in ambient conditions used some 3% of the energy required for commercial refrigerated storage at  $5^{\circ}\text{C}$  but could have similar effects in delaying ripening. The capital costs of high-pressure equipment are high (Saraiva, 2014), but Balasubramaniam *et al.* (2008) commented that the expansion of high-pressure food processing should help to reduce equipment costs by increasing the scale.

## Technology

Hyperbaric tanks and chambers are used mainly in the medical and diving industries. As indicated above, they have also been used to improve the postharvest life of some fruit and vegetables. For example, Robitaille and Badenhop (1981) described a completely autonomous storage system with  $\text{CO}_2$  removal and automatic  $\text{O}_2$  replenishment for hyperbaric storage that was successfully used to store mushrooms. Goyette (2010) described the small stainless steel test chamber he used as consisting of an outer chamber and inner chamber. The purpose of the outer chamber was to resist pressures up to 15 atm. The inlet of the outer chamber was connected to a cylinder of compressed air with a manometer to regulate the pressure. The airtight vessel could be pressurized up to 7 atm and instrumented to automatically monitor gas concentration using computer controlled valves, flow meter and  $\text{CO}_2$  gas analyser. Liplap *et al.* (2014c) developed a method to determine the metabolic respiration rate of fruit or vegetables during the transient period at the beginning of a hyperbaric treatment for the correction in the apparent respiration rate by considering the dilution effect of flushing the system and the error associated with gas solubilization as the gas partial pressure varied. They simulated the dilution process by using the general equation for exhaust ventilation, thus allowing for the elimination of the dilution effect during the calculation of the net respiration rate. See also Chapter 13 for a description of a hyperbaric container.

## Effects of High Pressure

As indicated above, the effect of high-pressure processing is mainly to control microorganisms but there is little or no detrimental effect on flavour, as can happen in thermal processing (Hill, 1997). However, Hill quoted work where some fruit juices were affected by high-pressure processing; for example, in grapefruit juice, which had good retention of vitamin C and many consumers

found the juice more acceptable after high-pressure processing because it was less bitter. Small molecules, which were reported to be the characteristics of flavouring, and nutritional components typically remained unchanged when fruit or vegetables were exposed to high pressure (Horie *et al.*, 1991). Ludikhuyze *et al.* (2002) and Yordanov and Angelova (2010) reviewed the effects of high pressures on fruit and vegetable processing and reported that vegetative cells were inactivated at about 300 MPa at ambient temperature, while spore inactivation required 600 MPa or more in combination with a temperature of 60–70°C. Previously, Larson *et al.* (1918, quoted by Ludikhuyze *et al.*, 2002) had observed that pressures up to 1800 MPa at room temperature were not sufficient to achieve commercial sterility of food products. A combination of pressure with temperatures of 60°C and higher was required for extensive inactivation of spores, and the lower the pressure applied, the higher was the temperature required to achieve inactivation (Sale *et al.*, 1970). At temperatures below 60°C in combination with a pressure of about 400 MPa, there was a maximal three log-cycle reduction of spores of *Bacillus coagulans* (Roberts and Hoover, 1996) and *Clostridium sporogenes* (Mills *et al.*, 1998). Generally, it was found that Gram-positive bacteria were more resistant to pressure than Gram-negative bacteria, while fungi, yeasts and *Enterococcus hirae* were more resistant to high pressure than *Listeria monocytogenes* and *Aeromonas hydrophila*. The most resistant to high pressure were bacterial spores (Fonberg-Broczek *et al.*, 2005).

In addition to its effect on bacteria, it was noted that high-pressure treatment helped to preserve colour in fruit and vegetable products, including tomato juice (Poretta *et al.*, 1995), orange juice (Ferrari and Di Matteo, 1996), guava purée (Yen and Lin, 1996) and avocado purée (Lopez-Malo *et al.*, 1998). The decrease in vitamin C content in strawberry purée and guava purée during storage after pressure treatment of 400–600 MPa for 15–30 min was much lower than loss of vitamin C content in non-pressure-treated purée (Sancho *et al.*, 1999). Luscher *et al.*

(2005) used 250 MPa during the freeze processing at –27°C of potatoes and found that after thawing there was a considerable improvement in texture, colour and visual appearance compared with freezing at atmospheric pressure.

There are various effects of high O<sub>2</sub> storage on the chemical changes and enzyme activity of some fruit and vegetables, which would be expected to be reproduced by hyperbaric conditions. The effects of hyperbaric storage appear to be mainly on cell metabolism of the fruit or vegetables and on microorganisms that infect them. Under hyperbaric pressures, a large change in the respiration rate was observed immediately after the pressure was applied and its amplitude decreased during the initial period of the hyperbaric treatment, which was described as an unsteady or transient state (Goyette *et al.*, 2012). Eggleston and Tanner (2005) found that at pressure of 600 MPa the respiration rate of carrot sticks decreased and this effect was greater the longer they were exposed over periods of 2–10 min. Liplap *et al.* (2012, 2013c, 2014a) described the effects of hyperbaric conditions on the respiration rate of lettuce, maize and avocados. From these studies, it can be concluded that the method used to measure respiration rate was limited, as it only allowed the respiration rate to be determined after the system reached equilibrium, which might require several hours. It was also proposed that gas dilution, solubilization and/or de-solubilization processes could take place under hyperbaric pressure treatment. It is these processes that may be responsible for the apparent large changes in the respiration rate of the produce exposed to hyperbaric conditions. Liplap *et al.* (2013b) found that the trend in antioxidant activity observed from both O<sub>2</sub> radical absorbance capacity and trolox equivalent antioxidant capacity assays was generally similar. In storage of tomatoes at 13°C and 20°C there was no significant effect observed of hyperbaric exposure on lipophilic antioxidant and hydrophilic antioxidant compared with tomatoes under atmospheric pressure.

Liplap *et al.* (2014b) showed that bacterial growth was affected by hyperbaric

pressure at 20°C under 100, 200, 400, 625 and 850 kPa. As hyperbaric pressure increased, the bacterial growth significantly decreased, but the effect varied with species. The maximum growth at 850 kPa was reduced by 56% for *Pseudomonas cichorii*, by 43% for *Pectobacterium carotovorum* and by 71% for *Pseudomonas marginalis*. In a review, San Martín *et al.* (2002) reported inhibitory effects of hyperbaric pressure on the growth of several microorganisms but very high pressure was required to kill or inactivate their growth. Liplap *et al.* (2014a) exposed lettuce at 20°C to pressures ranging from 100 to 850 kPa for 5 days and found that the development of decay was delayed under hyperbaric pressure, especially at 850 kPa, in comparison with those under normal atmospheric pressure. They considered the mode of action of hyperbaric pressure on the growth of microorganisms could be due to the direct impact of elevated pressure itself on the microorganisms, as had been demonstrated on infections in strawberry juice (Segovia-Bravo *et al.*, 2012). Another explanation could be that the elevated O<sub>2</sub> caused toxicity to bacteria, yeasts and moulds or the enhancement of the host pathogen defence compound synthesis, induced by mild stress. This latter effect of stress has been demonstrated for tomatoes by Lu *et al.* (2010). High CO<sub>2</sub> can also affect bacterial growth and Enfors and Molin (1980) found that the growth of *Bacillus cereus* was completely inhibited at 3 atm of CO<sub>2</sub> and *Streptococcus cremoris* at 11 atm of CO<sub>2</sub>. Vigneault *et al.* (2012) commented that exposure to hyperbaric pressures could be an alternative to chemical treatment for preserving postharvest quality of fruit and vegetables.

In addition to the control of microorganisms and some reductions in susceptibility to pathogens (Baba *et al.*, 1999; Romanazzi *et al.*, 2008) the apparent effects of hyperbaric conditions on fresh fruit and vegetables, as well as decreased respiration rate, include decreased ethylene production, slowing of the ripening processes and the possible extension of the synthesis of certain chemicals (Baba and Ikeda, 2003; Eggleston and Tanner, 2005; Goyette *et al.*, 2012). Other effects of hyperbaric exposure

described by Baba and Ikeda (2003) and Goyette *et al.* (2012) include reduced weight loss, maintenance of peel colour and TSS:TA ratio, improved lycopene synthesis and some evidence of reduction in chilling injury in tomatoes. There was some indication that hyperbaric storage could improve the retention of flavour in stored fruit compared with storage under atmospheric pressure. Yang *et al.* (2009) found that the composition of volatile compounds produced by peaches was higher after storage under hyperbaric pressure compared with CA storage or storage in air at atmospheric pressure.

Hendrickx *et al.* (1998) reported that exposure to pressures of 100 MPa or higher can induce structural rearrangements in enzymes, which can cause their activation or their partial or total inactivation in a reversible or irreversible manner. The specific effect of pressure depends on several factors, including the structure of the enzyme, its origin, the medium composition, pH, or the temperature and the pressure levels applied. For example, pectin methylesterase (PME) from peppers, tomatoes, white grapefruit, plums and carrots has been shown to be resistant to exposure to high pressure, since pressures higher than 700 MPa are usually required to induce short-term inactivation at room temperature (Segovia-Bravo *et al.*, 2012). With tomatoes, polygalacturonase (PG) is much more pressure labile than PME and almost complete PG inactivation was shown to occur in cherry tomatoes at 500 MPa in ambient temperature (Tangwongchai *et al.*, 2000). Verlent *et al.* (2004) found that the optimal temperature for tomato PME activity at atmospheric pressure was about 45°C at pH 8.0 and about 35°C at pH 4.4, but at both pH levels the optimal temperature increased as pressure was increased over the range of 0.1 to 600 MPa. Also at both pH levels the catalytic activity of tomato PME was higher at elevated pressure than at atmospheric pressure.

### Oxidation

One effect of hyperbaric conditions is to increase the partial pressure of O<sub>2</sub> above that

found in the air at normal atmospheric pressure. It has been shown by many workers that high O<sub>2</sub> levels in storage can affect various postharvest processes in whole fresh fruit and vegetables as well as those that have been minimally processed. As well as oxidative processes, the effects of the increased O<sub>2</sub> levels include respiration rate, ethylene production, volatile compounds, chlorophyll degradation, softening, pigments, nutrient content, sprouting, free radicals, diseases and physiological disorders. González-Roncero and Day (1998) reported that high O<sub>2</sub> MAP had beneficial effects on the degree of lipid oxidation. Shredded lettuce in MAP stored at 5°C for 10 days showed browning, but those in 80 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> did not (Heimdal *et al.*, 1995). Sliced apples stored at 1°C for 2 weeks had less browning in 100 kPa O<sub>2</sub> than in air (Lu and Toivonen, 2000). Jiang (2013) coated button mushrooms (*Agaricus bisporus*) with 2% alginate and stored them in jars continuously ventilated with 100 kPa O<sub>2</sub> at 4°C for up to 16 days and found that this treatment delayed browning. Since the browning reaction is oxidative, it is counter-intuitive that levels should be lower in high O<sub>2</sub>. Day (1996) reported that there were highly positive effects of storing minimally processed fruit and vegetables in 70 and 80 kPa O<sub>2</sub> in plastic film bags. He indicated that it inhibited undesirable fermentation reactions and delayed enzymic browning and that the O<sub>2</sub> levels of > 65 kPa inhibited both aerobic and anaerobic microbial growth. He also showed that the cut-surface browning of slices of apples was inhibited during storage in 100 kPa O<sub>2</sub> compared with those that had been stored in air. He explained this effect by suggesting that the high O<sub>2</sub> levels may cause substrate inhibition of PPO or that, alternatively, high levels of colourless quinones formed may cause feedback inhibition of PPO.

### Respiration rate

Kidd and West (1934) found that storage of apples in 100 kPa O<sub>2</sub> accelerated the onset of the climacteric rise in respiration rate and Biale and Young (1947) reported a similar increase in respiration rate in lemons

exposed to 34.1, 67.5 or 99.2 kPa O<sub>2</sub>. With avocados, Biale (1946) found that there was only a small acceleration in the time of the onset of the climacteric rise in respiration rate when they were exposed to 50 or 100 kPa O<sub>2</sub>. Exposure of potato tubers for 'some hours' to 100 kPa O<sub>2</sub> had little or no effect on their respiration rate, but prolonged exposure led, at first, to an increase in their respiration rate, over a period of some 2 weeks, but thereafter the effects of 'O<sub>2</sub> poisoning' became apparent and after 5–6 weeks the effect on respiration rate was negligible (Barker and Mapson, 1955). Cherries or apricots exposed to 30, 50, 75, and 100 kPa O<sub>2</sub> showed no effect on their respiration rate, but in plums the respiration rate was stimulated in proportion to the increasing O<sub>2</sub> concentration (Claypool and Allen, 1951). However, Escalona *et al.* (2006a and b) concluded that 80 kPa O<sub>2</sub> must be used in MAP of fresh-cut lettuce in combination with 10–20 kPa CO<sub>2</sub> to reduce their respiration rate and avoid fermentation. Zheng *et al.* (2008) found that zucchini squash exposed to 100 kPa O<sub>2</sub> had the lowest respiration rate compared with storage in air or 60 kPa O<sub>2</sub>.

### Ethylene

Effects on increased O<sub>2</sub> levels during storage have been reported to have various effects on ethylene biosynthesis. 'Russet Burbank' potatoes stored at 7°C had a higher ethylene production rate in 80 kPa O<sub>2</sub> + 12 kPa CO<sub>2</sub> compared with storage in air (Creech *et al.*, 1973) and 'Bartlett' pears in storage at 20°C had a higher ethylene production rate in 100 kPa O<sub>2</sub> than in air (Frenkel, 1975). Similar results were reported by Morris and Kader (1977) for mature-green and breaker tomatoes stored at 20°C when they were exposed to 30 or 50 kPa O<sub>2</sub> but exposure to 80 or 100 kPa O<sub>2</sub> reduced ethylene production rates and muskmelons stored in 100 kPa O<sub>2</sub> at 20°C had similar ethylene production levels as those stored in air (Altman and Corey, 1987). Zheng *et al.* (2008) found that zucchini squash exposed to 60 kPa O<sub>2</sub> had the lowest ethylene production compared with those stored in 100 kPa O<sub>2</sub> or in air.

### Volatile compounds

It appears that exposure to high O<sub>2</sub> does not affect the volatile content of fruit. For example, Rosenfeld *et al.* (1999) found that blueberries in MAP stored at 4°C or 12°C for up to 17 days had similar sensory quality whether flushed with air or 40 kPa O<sub>2</sub>. Yahia (1989) found that apples that had been stored at 3.3°C in 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> for up to 9 months did not have increased volatile formation when subsequently exposed to 100 kPa O<sub>2</sub> at 3.3°C for up to 4 weeks.

### Texture

Jiang (2013) coated button mushrooms with 2% alginate and stored them in jars continuously ventilated with 100 kPa O<sub>2</sub> at 4°C for up to 16 days and found that this treatment maintained their firmness and delayed cap opening. Additionally the treatments delayed changes in the TSS, total sugars and AA and inhibited the activity of PPO and POD throughout storage. ‘Bartlett’ pears kept at 20°C in 100 kPa O<sub>2</sub> had higher rates of softening than those kept in air (Frenkel, 1975). Day (1996) reported that softening of slices of apples was inhibited during storage in 100 kPa O<sub>2</sub> compared with those stored in air.

### Pigments

‘Bartlett’ pears stored at 20°C in 100 kPa O<sub>2</sub> had higher rates of chlorophyll degradation than those stored in air (Frenkel, 1975). Bi-ale and Young (1947) reported that exposure of lemon to 99.2 kPa O<sub>2</sub> accelerated de-greening and ‘Hamlin’ oranges stored in 50 kPa O<sub>2</sub> had an increased rate of de-greening (Jahn *et al.*, 1969). Aharoni and Houck (1980) exposed oranges for 4 weeks at 15°C to 40 or 80 kPa O<sub>2</sub>, followed by 2 weeks in air. They found that fruits kept in 80 kPa O<sub>2</sub> had the palest coloured peel, but their endocarp and juice were the deepest orange. The response was intermediate for oranges kept in 40 kPa O<sub>2</sub>. Aharoni and Houck (1982) reported that

storage in 40 or 80 kPa O<sub>2</sub> increased anthocyanin synthesis of flesh and juice of blood oranges. Li *et al.* (1973) reported that ripening of tomatoes at 12–13°C was accelerated in 40–50 kPa O<sub>2</sub> compared with air; and Frenkel and Garrison (1976) found that lycopene synthesis in *rin* tomatoes was stimulated in storage in 60 or 100 kPa O<sub>2</sub> in the presence of ethylene at 10 ml/l. Brown spots on the skins of bananas is a normal stage of ripening that usually occurs when the skin has turned from green to yellow or fully yellow, depending on the variety. Maneenuam *et al.* (2007) compared the effect of storage in different O<sub>2</sub> partial pressures at 25°C and 90% RH on peel spotting in the variety ‘Sucrier’ (*Musa AA*). The fruits had first been initiated to ripen and were turning yellow. They were then transferred to atmospheres containing either 90 kPa O<sub>2</sub> or 18 kPa O<sub>2</sub> in gas-tight chambers. Peel spotting and a decrease in dopamine levels (a free phenolic compound) were quicker in fruit in 90 kPa O<sub>2</sub>, indicating that dopamine might be a substrate for the browning reaction. The browning reaction is related to the activity of PAL that converts phenylalanine to free phenolic substances that form the substrate that is converted to quinones by PPO.

### Nutrition

Barker and Mapson (1955) reported that AA content of potato tubers kept in 100 kPa O<sub>2</sub> was lower than in those stored in air. Day *et al.* (1998) reported that high O<sub>2</sub> MAP had beneficial effects on the retention of AA and degree of lipid oxidation. They also stated that high O<sub>2</sub> in MAP of minimally processed lettuce did not decrease antioxidant levels in comparison with low O<sub>2</sub> MAP.

### Sprouting

Abdel-Rahman and Isenberg (1974) found that exposure of carrots to 40 kPa O<sub>2</sub> increased sprouting and rooting during storage at 0°C compared with those stored in air.

### Free radicals

A free radical is an element or compound that remains unaltered during its ordinary chemical changes. Increased O<sub>2</sub> concentrations around and within fruit or vegetables were shown by Fridovich (1986) to result in higher levels of free radicals that can damage plant tissues.

### Chilling injury

Zheng *et al.* (2008) reported that there was an indication of low chilling injury in zucchini squash exposed to high O<sub>2</sub> and that the O<sub>2</sub> radical absorbance capacity and total phenolic levels in the skin were both induced by cold storage and further enhanced by storage under 60 kPa O<sub>2</sub>. The enhanced antioxidative enzyme activities and the O<sub>2</sub> radical absorbance capacity and phenolic levels appeared to correlate with the reduced chilling injury.

### Decay

Day (1996) stated that high O<sub>2</sub> levels could influence both aerobic and anaerobic growth of microorganisms. Amanatidou *et al.* (1999) found that exposure to 80–90 kPa O<sub>2</sub> generally did not inhibit microbial growth strongly, but caused a significant reduction in the growth rate of some of the microorganisms they tested, including *Salmonella enteritidis*, *S. typhimurium* and *Candida guilliermondii* (a yeast used in biological control). Among the ten microbial species studied, growth of some was even stimulated by high O<sub>2</sub>. The combined application of 80–90 kPa O<sub>2</sub> plus either 20 or 10 kPa CO<sub>2</sub> had an inhibitory effect on the growth of all the microorganisms they tested. They concluded that when high O<sub>2</sub> or high CO<sub>2</sub> was applied alone, the inhibitory effect on microbial growth was highly variable, but stronger and more consistent inhibition of microbial growth occurred when the two gases were used in combination. Wszelaki and Mitcham (1999, 2000)

found that 80–100 kPa O<sub>2</sub> inhibited the *in vitro* growth of *Botrytis cinerea* on strawberries. However, only 100 kPa O<sub>2</sub> inhibited growth of *B. cinerea* more than 15 kPa CO<sub>2</sub> in air and then only after exposure for 14 days. No residual effect on *in vitro* fungal growth was observed upon transfer to air. In *in vivo* studies, they found that *B. cinerea* on strawberries was reduced during 15 days of storage at 5°C in 80–100 kPa O<sub>2</sub>, but there was some fermentation in the fruit. However, storage in 40 kPa O<sub>2</sub> was effective and gave good control of rotting without detrimental effects on fruit quality during storage at 5°C for 7 days. Deng *et al.* (2005) also reported that high O<sub>2</sub> storage conditions significantly reduced fruit decay in strawberries. High O<sub>2</sub> has been combined with other treatments; for example, strawberries were treated with a solution of 1% chitosan, packaged in modified atmosphere film packages with either 80 kPa O<sub>2</sub> or 5 kPa O<sub>2</sub>, with the balance N<sub>2</sub>, and then stored at 4°C, 8°C, 12°C and 15°C. The coating inhibited the growth of microorganisms at all temperatures, especially when combined with high O<sub>2</sub> that also seemed to help in retaining their colour (Tamer and Çopur, 2010).

High O<sub>2</sub> levels have been tested on minimally processed fruit and vegetables. Day (1996) suggested that high O<sub>2</sub> atmospheres could be advantageous for MAP by directly inhibiting the decay-causing organisms, particularly fungi on soft fruits. Day *et al.* (1998) reported that 99 kPa O<sub>2</sub> alone did not prevent the growth of *Pseudomonas fragi*, *Aeromonas hydrophila*, *Yersinia enterocolitica* and *Listeria monocytogenes* but growth of *P. fragi* was reduced by 14% and *A. hydrophila* by 15%. The combination of 80 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> was more effective in inhibiting growth of all the organisms tested at 8°C than either 80 kPa O<sub>2</sub> or 20 kPa CO<sub>2</sub> alone. Amanatidou *et al.* (2000) found that a combination of 50 kPa O<sub>2</sub> + 30 kPa CO<sub>2</sub> prolonged the shelf-life of sliced carrots by 2–3 days compared with storage in air. In *in vitro* studies, Caldwell (1965) reported that exposure to 10 atm pressure of pure O<sub>2</sub> completely suppressed growth in some fungi and two species of bacteria. However, unlike

the tissues of higher plants, the fungi, when removed from the pressure and returned to air, recovered and began to grow apparently quite normally if the period of exposure was not too long. In these cases a period of some days normally elapsed before the growth of the colonies in air began again.

### Physiological disorders

Kidd and West (1934) showed that storage of 'Bramley's Seedling' apples at 4°C for 4 months in 100 kPa O<sub>2</sub> resulted in disorders whose symptoms included mealy flesh and browning of skin and flesh. Solomos *et al.* (1997) reported that the apple cultivars 'Gala' and 'Granny Smith' exposed to 100 kPa O<sub>2</sub> developed extensive peel injury compared with that which occurred under lower O<sub>2</sub> atmospheres, especially in 1 kPa O<sub>2</sub>. However, Lurie *et al.* (1991) found that storage of 'Granny Smith' apples at 0°C in 70 kPa O<sub>2</sub> for 1 month did not accelerate the severity of the disorder caused by sunscald. Production of  $\alpha$ -farnesene and trienol, related to development of storage scald, increased in apples stored at 0°C in 100 kPa O<sub>2</sub> atmospheres for up to 3 months. 'Granny Smith' apples stored in 100 kPa O<sub>2</sub> were completely 'bronzed' after 3 months and contained high ethanol concentrations (Whitaker *et al.*, 1998). An atmosphere of 100 kPa O<sub>2</sub> potentiated the effect of ethylene at 0.5 ml/l on isocoumarin formation in carrots, resulting in a fivefold increase over that found in carrots treated with ethylene in air (Lafuente *et al.*, 1996). Super-atmospheric O<sub>2</sub> levels increased ethylene production and the incidence and severity of pink rib and ethylene-induced russet spotting on lettuce (Klaustermeyer and Morris, 1975).

## Effects on Selected Crops

### Avocado

Liplap *et al.* (2012) stored avocados for 7 days at ambient temperature using pressure levels of 1, 3, 5, 7 or 9 atm compared with those

stored in commercial cold storage conditions of 5°C under atmospheric pressure. Hyperbaric pressure decreased ripening, resulting in an extension of the storage life, and the respiration rates were inversely proportional to the pressure applied. They concluded that hyperbaric storage had potential for extending the storage life and maintaining the quality of avocado while using little energy.

### Cherry

Romanazzi *et al.* (2008) exposed 'Ferrovia' cherries for 24 h under approximately 1.5 atm or 1 atm for 4 h, then stored them at 0 ± 1°C for 14 days followed by 7 days at 20 ± 1°C under atmospheric pressure. Cherries that had been exposed to 1.5 atm had reduced incidence of brown rot, grey mould and blue mould, compared with those exposed to 1 atm. They concluded that induced resistance from the hyperbaric conditions was likely to be responsible for the decay reductions.

### Grape

Romanazzi *et al.* (2008) artificially wounded 'Italia' grapes and inoculated the wounds with 20  $\mu$ l of a *Botrytis cinerea* conidial suspension (5 × 10<sup>4</sup> spores/ml) and then exposed them for 24 h under either approximately 1.5 atm or 1 atm. They were then removed from the hypobaric chambers and stored at 2 ± 1°C for 3 days under atmospheric pressure. Hyperbaric storage resulted in a significant reduction of grey mould lesion diameter and percentage of *B. cinerea* infections on the fruit. As with cherries, they concluded that induced resistance was likely to be responsible for the decay reductions.

### Lettuce

Liplap *et al.* (2013a) stored lettuces in a range of pressures from 100 to 850 kPa at 20°C

and 100 kPa at 4°C. The hyperbaric storage at 20°C resulted in noticeable changes in sensory quality, but they were still considered marketable after 3 days, while those stored at atmospheric pressure and 4°C showed little degradation even during 7 days storage. After 5 days of storage the respiration rate of those under 625 and 850 kPa remained fairly stable, while the respiration rate began to increase in those in lower pressures, which they assumed was an indication of the initiation of decay. At 20°C those under 850 kPa showed better quality than those under atmospheric pressure. They concluded that 'overall, hyperbaric treatment has the potential of being used as an alternative technique for short-term storage of lettuce without refrigeration'.

### Mango

Apelbaum and Barkai-Golan (1977) tested the effect of hyperbaric pressure storage on mangoes for 16 days and found that pressures from 250 to 700 kPa did not result in a shelf-life longer than commercial storage in air at atmospheric pressure.

### Melon juice

Queirós *et al.* (2014) stored melon juice for 8 h at 25°C, 30°C or 37°C, under atmospheric pressure and under hyperbaric pressures within the range of 25–150 MPa. These were compared with storage at 4°C under atmospheric pressure. They found that juice under hyperbaric storage of 50 and 75 MPa had similar or lower microbial counts (total aerobic mesophiles, Enterobacteriaceae, and yeasts and moulds) compared to juice stored at 4°C, while at 100 and 150 MPa the counts were lower for all the tested temperatures, indicating an additional microbial inactivation effect. At 25 MPa no microbial inhibition was observed. Juice stored under hyperbaric conditions had similar pH, TA, TSS, browning and cloudiness levels to juice that had been in storage at 4°C.

### Mume

At an average temperature of 22°C, the shelf-life of mume fruit (*Prunus mume*) was reported to be only 2 or 3 days (Miyazaki, 1983). Mume fruits were subjected to pressures of 0.5–5 MPa (500–5000 kPa) for 10 min and maintained at 0.5 MPa (500 kPa) for 5 days by Baba and Ikeda (2003). Treatment at 0.5 MPa decreased the respiration rate, ethylene production and weight loss during storage and showed reduced chilling injury symptoms. During storage at 0°C or 5°C, chilling injury occurred as surface pitting and/or peel browning (Goto *et al.*, 1988).

### Mushroom

Robitaille and Badenhop (1981) stored mushrooms under 35 atm, which did not affect their respiration rate, but significantly reduced moisture loss and cap browning compared with storage at normal pressure. Neither pressurization nor gradual depressurization over 6 h injured the mushrooms.

### Peach

The composition of volatile compounds emanating from peach fruit varied quantitatively and qualitatively during 4 weeks of storage. Yang *et al.* (2009) identified 21 compounds prior to storage and 59 after storage. Storage under hyperbaric pressures contributed most to the concentration of total volatile compounds compared with CA storage and storage in air.

### Tomato

Romanazzi *et al.* (2008) observed that hyperbaric storage had been shown to have variable effects on the shelf-life of tomatoes. In their review, Ahmed and Ramaswamy (2006) reported that at 20°C hyperbaric exposure at  $\geq 0.3$  MPa resulted in respiration



rates equal to or higher than those in fruit stored in normal atmospheric pressure. Liplap *et al.* (2013a) subjected early-breaker tomatoes to normal atmospheric pressure, 0.3, 0.5, 0.7 or 0.9 MPa at 20°C or 13°C for 4 days, followed by ripening at 20°C for up to 10 days at atmospheric pressure. Hyperbaric treatment initially inhibited lycopene synthesis but then enhanced lycopene accumulation during exposure and subsequent ripening. All antioxidants were found in lower concentrations in tomatoes subjected to atmospheric pressure at 13°C. They concluded that, overall, hyperbaric treatment at 20°C had potential to extend tomato shelf-life during short treatment durations without adverse impact on quality during ripening. They showed that the only consistent effect of hyperbaric treatment at 0.5, 0.7 and 0.9 MPa was a reduction in weight loss and enhanced firmness retention for up to 5 days of ripening after treatment. Hyperbaric storage at 0.5, 0.7 and 0.9 MPa significantly reduced weight loss and retained colour, firmness, TSS, TA and TSS:TA ratio at similar levels to the tomatoes treated at 13°C and 0.1 MPa. Firmness after treatment was highest for fruit from 0.1 MPa at 13°C and from 0.5, 0.7 and 0.9 MPa at 20°C. The higher firmness advantage declined during the 5 days of ripening after treatment, with higher firmness only being retained for fruit at 0.9 MPa at 20°C and at 0.1 MPa at 13°C. After 10 days of ripening, firmness was similar for all treatments. The lowest respiration rate was in those stored at 0.1 MPa at 13°C. They also exposed early breaker-stage tomatoes to pressures of 1, 3, 5, 7 or 9 atm for 5, 10 or 15 days at 13°C, followed by a storage at 20°C for 12 days (Goyette *et al.*, 2012). Based on firmness values, those that had been stored in normal atmospheric pressure were no longer acceptable for consumption after 12 days of subsequent storage at 20°C. Those that had been stored at 7 and 9 atm for 15 days had irreversible physiological damage, while those exposed to 3, 5 or 7 atmospheres for 10 days or 5 atm for 5 days maintained marketable

firmness. Lycopene content was improved in all the fruit that had been stored under hyperbaric pressures followed by 12 days of maturation compared with those that had been under atmospheric pressure, with the highest lycopene content (28% more than in those from atmospheric pressure) from fruit that had been stored under 5 atm for 10 or 15 days. Goyette (2010) reported that hyperbaric exposure of tomatoes showed that their respiration rate was inversely proportional to the pressure applied. Respiration rate was reduced by 20% under 9 atm compared with those under 1 atm. At the onset of hypobaric storage the RQ was low and increased to reach a value of approximately 1 within 120 h. Low RQ values were caused by solubilization of CO<sub>2</sub> in the tomato cells at the beginning of the process. Liplap *et al.* (2013b) concluded that, overall, hyperbaric exposure at 20°C had potential to extend tomato shelf-life during short-duration treatment without adverse impact on quality during ripening.

### Watermelon juice

Fidalgo *et al.* (2014) compared the preservation of watermelon juice at room temperature and 5°C at atmospheric pressure with preservation under 100 MPa at room temperature. After 8 h exposure to 100 MPa, the initial microbial loads of the watermelon juice were reduced by 1 log-unit, for total aerobic mesophiles, to levels of about 3 log-units and 1–2 log-units for Enterobacteriaceae, yeasts and moulds that were below the detection limit. These levels remained unchanged for up to 60 h. Similar results were obtained at 30°C under 100 MPa after 8 h. At atmospheric pressure for 24 h at room temperature and for 8 h at 30°C, microbial levels were above quantification limits and unacceptable for consumption. Storage at 5°C after the hyperbaric exposure gave an extended shelf-life.

## Hypobaric and Low-oxygen Storage

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Hypobaric storage has also been referred to as low-pressure storage and sub-atmospheric pressure storage. With hypobaric storage the control of the O<sub>2</sub> level in the store can be very accurately and easily measured and achieved by measuring the pressure inside the store with a vacuum gauge. Hypobaric storage was recently reviewed in more detail by Thompson (2015) and much of the information contained here has been modified from that book. The application of hypobaric conditions to CA storage is that the reduced pressure reduces the partial pressure of O<sub>2</sub> and thus O<sub>2</sub> availability to fruit or vegetables in the store. The reduction in the partial pressure of the O<sub>2</sub> is proportional to the reduction in pressure in the store. The humidity in the store atmosphere has to be taken into account when calculating the partial pressure of O<sub>2</sub> in the store, using a psychometric chart. Also as air pressure is reduced, so the boiling point of water is reduced, thus increasing water loss from the stored fruit or vegetable. Store humidity, therefore, must be kept high.

Burg (2004) reviewed the considerable literature on the effects of hypobaric conditions on the storage and transport of meat and horticultural produce as well as fruit, vegetables and flowers. An example of meat transport was given by Sharp (1985), who reported on the successful transport of lamb carcasses at 0.006 atmosphere (atm) ( $\approx$  4.6 mm

Hg; 1 atm = 101.325 kPa) in a hypobaric container with an inner wall temperature of about  $-1.5^{\circ}\text{C}$  and found that the carcasses were still in good condition with a mean weight loss of only 2.5% during the 40 days of transport. Burg (1967) took out a patent in the USA for the application of hypobaric storage and he gave a detailed evaluation and justification of the award of the patent in Burg (2004). Stanley P. Burg was listed as the inventor and Grumman Allied Industries Incorporated as the original assignee. In support of the application several examples of the effects on horticultural products, meat and fish were given, including those in [Table 11.1](#).

Burg (2004) refers to legal actions in which he was involved, related to 'LP patents' and tax implications of those patents, that were eventually resolved. Most of the research on hypobaric storage and its commercial application has been carried out in the USA, particularly by Stanley P. Burg and many other workers, including David Dille, but research work has also been carried out in the UK (Langridge and Sharples, 1972; Ward, 1975; Hughes *et al.*, 1981), Germany (Bangerth, 1973, 1974, 1984; Bangerth and Streif, 1987), Spain (Alvarez, 1980), Canada (Loughheed *et al.*, 1974, 1977), Israel (Apelbaum and Barkai-Golan, 1977; Apelbaum *et al.*, 1977a, b; Aharoni *et al.*, 1986), China (Chang, 2001; Cao, 2005) and Italy (Romanazzi *et al.*, 2001, 2008). Burg

**Table 11.1.** Effects of hypobaric storage on the postharvest life of some fruit and vegetables. Taken with modifications from Burg, S.P. 1977 Patent US4061483A, Low temperature hypobaric storage of metabolically active matter. Available at <https://patents.google.com/patent/US4061483A/en?q=Patent+US4061483A+> (accessed September 2018).

Crop	Temperature	Pressure	Storage time
Bartlett, Clapp and Comice pears	-1 to 1°C	Atmospheric 60 mm Hg	1½–3 months 4–6 months
McIntosh, Red Delicious, Golden Delicious and Jonathan apples	-1 to 2°C	Atmospheric 60 mm Hg	2–4 months 6 months
Waldin avocados	10°C	Atmospheric 60–80 mm Hg	12–16 days 30 days
Lula avocados	8°C	Atmospheric 40 and 80 mm Hg	23–30 days 75–100 days
Booth 8 avocados	8°C	Atmospheric 40 and 80 mm Hg	8–12 days About 45 days
Fresh green onions (scallions)	0–3°C	Atmospheric 60–80 mm Hg	2–3 days More than 3 weeks
Green peppers (capsicums)	8–13°C	Atmospheric 60–80 mm Hg	16–18 days 46 days
Snap-beans	5–8°C	Atmospheric 60 mm Hg	7–10 days 26 days
Cucumbers	10°C	Atmospheric 80 mm Hg	10–14 days 49 days
Pole-beans	8°C	Atmospheric 60 mm Hg	10–13 days 30 days
Tioga and Florida 90 strawberries	0–2°C	Atmospheric 80–200 mm Hg	5–7 days 4–5 weeks
Blueberries	0–1°C	Atmospheric 80–200 mm Hg	4 weeks At least 6 weeks
Iceberg lettuce	0–4°C	Atmospheric 150–200 mm Hg	2 weeks About 4 weeks
Ruby Red grapefruit	6°C	Atmospheric 80–150 mm Hg	4–6 weeks 90 days
Mature-green tomatoes	13°C	Atmospheric 80 mm Hg	2 weeks 8 weeks

(2014) evaluated the reasons why hypobaric technology has had only limited impact on the food industry. His contentions included criticisms of the research techniques and also included:

- its high cost;
- experimental error caused by humidifying air at 1 atm pressure instead of at the low storage pressure;
- insufficient air changes so that the stored commodity consumed all available O<sub>2</sub> and suffered anaerobic damage;
- laboratory apparatus installed in cold-rooms that had a non-uniform air distribution pattern that produced a 'cold spot' on the vacuum chamber's surface (this

created an evaporation/condensation cycle between the commodity and chamber cold spot); and

- the commodity being stored in a sealed system that accumulated an active ethylene concentration between occasional ventings.

Burg (2014) also commented that 'experimental errors by academics and other concerns have prevented hypobaric storage from achieving more widespread adoption'.

The literature reports several effects of hypobaric conditions on fresh fruit and vegetables that can lead to extension in post-harvest life. Workman *et al.* (1957) found that the respiration rate of tomatoes was

reduced when they were stored at 20°C under 88 mm Hg compared with those stored at 20°C under atmospheric pressure (note that 1 mm Hg = 0.133 kPa). At the same time Stoddard and Hummel (1957) stored several types of fruit and vegetables in household refrigerators and found that those stored under 658–709 mm Hg had increased postharvest lives of 20–92%, depending on the crop, compared with those stored in the same refrigerators at atmospheric pressure. Burg and Burg (1966b) showed delays in ripening of bananas when they were stored in 125–360 mm Hg compared with those in atmospheric pressure. They also showed that limes stored at 15°C took about 10 days for 50% of them to turn from green to yellow in 760 mm Hg, while this took up to 56 days in storage in 152 mm Hg at the same temperature. Burg (1975) reported that ‘Lula’ avocados remained firm for 3½ months under hypobaric conditions and then ripened normally upon removal. This length of storage is almost twice that reported for Lula avocados held under CA storage at normal atmospheric pressure (Hatton and Reeder, 1965).

It has been well established that the reduced O<sub>2</sub> in CA storage is a major factor in extending postharvest life. However, in addition to the effects of reduced O<sub>2</sub> on respiration rate, hypobaric conditions have other effects. Since hypobaric chambers are constantly ventilated and air is removed, ethylene produced by the fruit or vegetable will be constantly removed. Ethylene has also been shown to be removed more quickly from the plant cells where it is being synthesized. This ethylene removal would therefore reduce its effect in the ripening process and also reduce the effects on the development of some physiological disorders associated with ethylene accumulation. Chen *et al.* (2013b) reported that fresh bamboo shoots stored at 2°C in 50 mm Hg for 35 days had reduced ethylene production, compared with those stored in atmospheric pressure, which delayed their softening. Burg (2004) reviewed several publications that reported inhibition of growth and sporulation of pathogenic fungi in hypobaric conditions that then resumed growth and

sporulation when they were removed to atmospheric pressure, which may also be related to reduced O<sub>2</sub>. There is also strong evidence that hypobaric conditions can help to control insect infestation in fruit. For example, Davenport *et al.* (2006) reported that all the eggs and larvae of Caribbean fruit fly (*Anastrepha suspensa*) in mangoes stored at 13°C in 15–20 mm Hg with ≥ 98% RH were killed within 11 days, while a substantial number of eggs survived for the 14-day storage period under atmospheric pressure. Hypobaric conditions may have other effects on fruit physiology. For example, Wang *et al.* (2015) concluded that the increased shelf-life of honey peach (*Prunus persica*) in hypobaric storage could be due to increased energy status, enhanced antioxidant ability and less membrane damage.

## Technology

As indicated above, hypobaric storage has hardly been taken up commercially. Probably the only commercial application was the Grumman Corporation in the USA in the 1970s. The company developed and constructed a hypobaric container which it called ‘Dormovac’ (similar to the one in Fig. 11.1). It was operated commercially at 2.2–2.8°C and a pressure of 15 mm Hg, but they were unable to make it profitable, resulting in eventual losses of some \$50 million (Grumman Park, undated). However, as indicated above, there has been considerable experimentation on hypobaric storage.

Since fresh fruit and vegetables in hypobaric store are constantly respiring, it is essential that the store atmosphere is constantly being changed in order to maintain the required partial pressure, especially the desired O<sub>2</sub> level. This is achieved by a vacuum pump evacuating the air from the store and constantly replenishing fresh air from the outside. The air inlet and the air evacuation from the store are balanced in such a way as to achieve the required reduced pressure within the store. There are two important considerations in developing and applying this technology to crop storage. The first is that the store needs to be designed



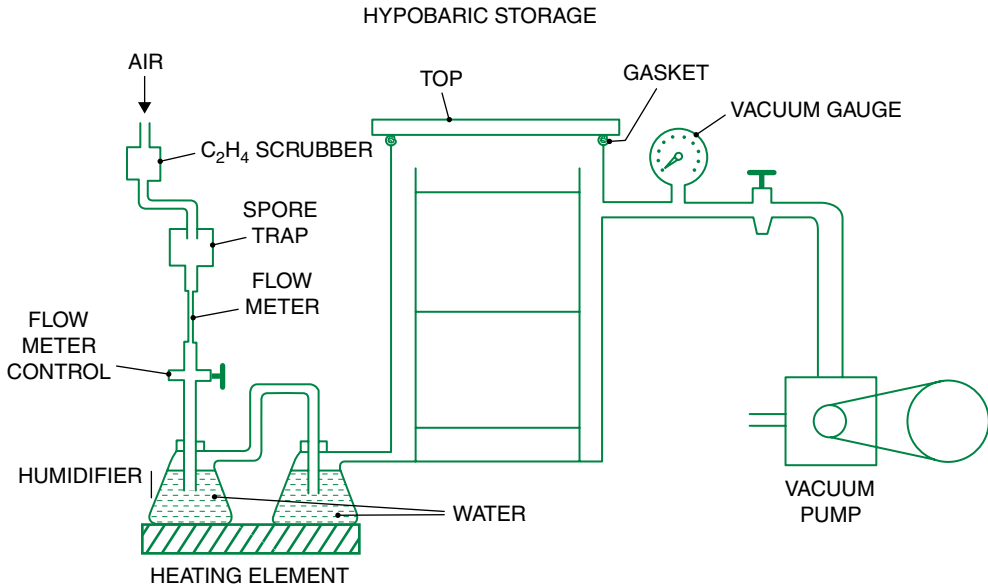
**Fig. 11.1.** Reefer container that has been tested for hypobaric storage.

to withstand low pressures without imploding. The second is that the reduced pressure inside the store can result in rapid water loss from the crops. To overcome the first, stores have to be strongly constructed; for example, with thick steel plate with a curved interior. For the second, the air being introduced into the store must be as close as possible to saturated (100% RH). If it is less than this, serious dehydration of the crop can occur.

Various systems have been developed for experimental work to test hypobaric storage, but most are similar to the one described by Dilley (1977). This type of system was also used by Hughes *et al.* (1981) where specially constructed 100 l steel barrels were used that had transparent Perspex lids 4 cm thick through which the produce could be observed during storage. Air was introduced through an inlet that had been bubbled through water in order to saturate it (though no measurement of the relative

humidity was made). A vacuum pump and vacuum gauge were attached to the outlet and the flow rate was adjusted on the pump to give a flow rate of 5 l/h (Fig. 11.2).

The systems were kept in temperature-controlled cabinets that were adjusted to  $\pm 1^\circ\text{C}$ . Romanazzi *et al.* (2001) used a vacuum pump in 64 l gas-proof tanks in which the fruit was placed at atmospheric pressure as controls or a vacuum was applied with a vacuum pump and measured with an external vacuum gauge. Al-Qurashi *et al.* (2005) used quart-sized pressure cookers that were continuously evacuated by a belt-drive pump. The inlet air was humidified by bubbling it through water in 5-gallon (22.7 l) containers then through a filter to prevent water from getting into the pressure cookers from the ventilated air. The air filter was quarter-filled with water and cellulose pads were inserted to increase the humidity of the air flowing to the pressure cookers. The pressure cookers were also sealed at the lids



**Fig. 11.2.** Schematic diagram of the experimental hypobaric storage system set up at the Tropical Products Institute in London in the late 1970s.

with ‘Play-Doh’, allowing an air tight seal. Valve regulators, located between the filter and the pressure cookers, were used to maintain the desired pressures by admitting air at the proper rate. The pressure within the pressure cookers was monitored with pressure gauges placed at the top of the pressure cookers. Jiao *et al.* (2012) used aluminium chambers ( $0.61 \times 0.43 \times 0.58$  m) with a two-stage rotary vacuum pump regulated by a compact proportional solenoid valve controlled by a proportional/integral/derivative computer control system. Chamber pressure was monitored with a digital pressure gauge. A rotameter was used to adjust the air exchange rate and the ingoing rarefied air was passed through a humidifier before entering the hypobaric chamber, in order to keep the humidity near saturation. The relative humidity was calculated by measuring wet-bulb and dry-bulb temperatures having relatively high accuracy ( $\pm 0.1\%$ ). They experimented with various modifications and showed that added foam covering the chambers maintained the temperature of the inside air to within  $\pm 0.1^\circ\text{C}$ . The regulating system kept pressure to within  $\pm 1\%$  of the set point and maintained humidity at

$> 98\%$  RH under various air exchange rates and pressures (the measurements inside the cabinet varied between 98.40 and 99.35 RH) with a chamber leakage rate of 0.009 kPa/h and hypobaric system leakage rate of 0.48 kPa/h. Spalding and Reeder (1976) used pure  $\text{CO}_2$  and  $\text{O}_2$  from gas cylinders that were metered into a 150 ml glass mixing chamber and their flow rates regulated to supply the required mixture of  $\text{O}_2$  and  $\text{CO}_2$ . Flow into the chamber was controlled at 110 ml/min (1 air change per hour). Chamber pressures were maintained at about 2 mm Hg and the humidity at 98–100% RH as determined by a humidity-sensing element. Jamieson (1980) used vacuum desiccators for the fruit or vegetables to be tested. A vacuum pump sucked air from the desiccators and air was allowed in by first passing it through a flask containing water that was constantly heated on an electric plate to maintain near-saturation humidity. Air flow and pressure inside the desiccators were measured and controlled with a flow meter, a vacuum gauge, a pressure regulator and a needle valve. Burg (2004) described equipment used in laboratory-scale hypobaric experiments with the fruit in desiccators

supplied with air passed through water in a beaker with a vacuum pump attached to the outlet of the desiccators and a manometer for monitoring the pressure inside the desiccators.

Jiao *et al.* (2012) described a hypobaric chamber called 'VivaFresh' that had been made in 2007 by Atlas Technologies (Port Townsend, Washington, USA). The VivaFresh system comprised aluminium chambers (0.61 m long × 0.43 m wide × 0.58 m high) with a two-stage rotary vacuum pump regulated by a compact proportional solenoid valve controlled by a proportional/integral/derivative computer control system. Chamber pressures were monitored with a digital pressure gauge and a rotameter was used to adjust the air exchange rate. The ingoing air passed through a humidifier before entering the hypobaric chamber in order to keep the humidity near saturation. The temperature inside the chamber and the exterior chamber wall and humidity was calculated by measuring wet-bulb and dry-bulb temperatures using calibrated YSI 55000 Series GEM thermistors with a relatively high accuracy of ± 0.1%. Temperature variation of the chamber wall was controlled to within ± 0.2°C and the inside air to within ± 0.1°C. Humidity measured inside the chambers varied between 98.4% and 99.35% RH. The pressure was within 1% of the set point and O<sub>2</sub> concentration could be controlled at less than 0.6% when the pressure was less than 3.3 kPa. The leakage rate of the chamber was given as 0.01 kPa/h.

A different hypobaric system was described by Knee and Aggarwal (2000) who used plastic containers, capable of being evacuated to 380 mm Hg with a vacuum pump, for their experiments. They found that this required 18 strokes of the pump for a container with a nominal volume of 500 ml capacity and 24 strokes for a 750 ml capacity container. These were placed in a refrigerator run at either 4°C or 8°C, depending on the product to be stored. Overall the vacuum containers showed little advantage over conventional plastic containers for the types of produce tested. The multinational company VacuFresh also makes small plastic containers of 1.2–2.8 l capacity that are sold together with a hand pump. They are used for storing

fruit and vegetables under hypobaric conditions in a domestic refrigerator. The company claims in its advertisements, 'Keeps food fresh up to 5 times longer. Vacuum locks in freshness and prevents premature spoilage and food decay.' In the mid-1970s Prodesarrollo in Colombia (A.K. Thompson, unpublished) investigated the possibility of storing vegetables (potatoes, carrots and cabbages) in a brick-built store room at a café/ski centre at an altitude of about 4800 m in Colombia that would have a barometric pressure of about 57 kPa (430 mm Hg), where the temperature was also low. The trial was on the slopes of the volcano Nevado Del Ruiz (near the town of Manzales in the Departamento de Caldas) whose peak is 5321 m above sea level. The trial was unsuccessful, due to rapid desiccation of the vegetables, and because of transport problems it was not repeated.

### Adding Carbon Dioxide

Burg (2004) observed that hypobaric conditions, which can decrease ambient and intercellular CO<sub>2</sub>, are an important advantage, providing benefits that cannot be duplicated by increasing CO<sub>2</sub> levels given in some controlled atmosphere storage recommendations. Spalding and Reeder (1976) concluded that high CO<sub>2</sub> was necessary for the successful storage of Waldin avocados, even in a hypobaric system. Yahia (2011) commented that CO<sub>2</sub> was considered essential in controlling decay and ameliorating chilling injury in avocados and CO<sub>2</sub> cannot be added in a low pressure system. Burg (2004) contended that the effects of hypobaric conditions on removing CO<sub>2</sub>, produced by respiration, from cells and intercellular spaces may result in reduced bacterial and fungal growth, better AA retention, inactivation of the ethylene-forming enzyme and the prevention of succinate formation. In confirmation of the effect of CO<sub>2</sub> on bacterial growth, Wells (1974) reported that the postharvest pathogens *Erwinia carotovora*, *E. atroseptica* and *Pseudomonas fluorescens* were unable to multiply in the very low CO<sub>2</sub> levels, as well as the low O<sub>2</sub> levels, that are to be found in the cells of vegetables stored

under hypobaric conditions. However, Enfors and Molin (1980) found that when *P. fragi* was grown at O<sub>2</sub> limitation (0.0025 atm O<sub>2</sub>) and exposed to 0.99 atm CO<sub>2</sub>, the inhibiting effect of the CO<sub>2</sub> was added to that of the O<sub>2</sub> limitation. They did not note any indications of a synergistic effect between CO<sub>2</sub> inhibition and O<sub>2</sub> limitation. Laurin *et al.* (2006) reported that cucumbers exposed to hypobaric conditions of 532 mm Hg for only 6 h could exhibit an indirect stress response that occurred only when they were transferred to atmospheric pressure, preventing closure of stomata. They explained that this residual effect may have been due to the possibility that hypobaric conditions enhanced outward diffusion of CO<sub>2</sub>, reducing intercellular CO<sub>2</sub> concentration and causing stomata to open. When the cucumbers were transferred to atmospheric pressure, stomata could still have remained open to restore the CO<sub>2</sub> concentration.

### Pre-storage Treatments

As with any other technique, storage under hypobaric conditions has been successfully used in combination with other treatments. For example, Romanazzi *et al.* (2003) found that the combination of spraying with chitosan at 0.1%, 0.5% or 1.0% 7 days before harvest and exposure to 380 mm Hg for 4 h directly after harvest effectively controlled fungal decay of sweet cherries during 14 days of storage at 0 ± 1°C, followed by a 7-day shelf-life. Fungi associated with rots included brown rot (*Monilinia* sp.), grey mould (*Botrytis cinerea*), blue mould (*Penicillium expansum*), alternaria rot (*Alternaria* sp.) and rhizopus rot (*Rhizopus* sp.). Kashimura *et al.* (2010) found that applying 1-MCP to 'Jonagold' and 'Fuji' apples and 'Shinsei' and 'Shinsui' Japanese pears under 152 mm Hg reduced the exposure time required to have the same effect as applying 1-MCP at atmospheric pressure. Dong *et al.* (2013) concluded that high efficacy of 1-MCP applied to tomatoes under hypobaric conditions was due to rapid ingress and accumulation of the gaseous 1-MCP. They had found that mid-climacteric tomatoes exposed to 1-MCP at

500 nl/l under 76 or 16 mm Hg for 1 h showed acute disturbance of ripening. Spalding and Reeder (1976) found that limes coated with wax containing 0.1% of either of the fungicides thiabendazole or benomyl remained green and suitable for marketing after 3–4 weeks under hypobaric storage of 170 mm Hg at 21.1°C.

Burg (2013, personal communication) described a method for generating hypochlorous acid vapour into a commodity packed for shipment in a hypobaric intermodal container. Within 1 h the vapour was 100% effective in killing bacteria, fungi and viruses both on the plant surfaces and within its interior, without injuring the commodity. The US Food and Drug Administration was reported to have given clearance for use of hypochlorous acid vapour in a hypobaric system and Burg has filed a new patent on hypobaric storage.

### Effects on Physiology, Quality and Deterioration

#### Volatiles

Hypobaric storage has the advantage of constantly removing ethylene from the store. This constant removal would have the effect of reducing the detrimental effects of ethylene on postharvest life. Burg and Kosson (1983) reported that hypobaric storage lowers the internal equilibrium content of volatiles, including ethylene. Hypobaric conditions have also been shown by Goszczynska and Ryszard (1988) to accelerate the outward diffusion of gases from the internal tissues of horticultural crops during storage. Removal of volatiles may also have detrimental effects. Bangerth (1984) reported that apples stored for protracted periods under 50–75 mm Hg did not develop normal aroma and flavour when they were subsequently ripened in atmospheric pressure. There is some indication that this may be due to the low partial pressure of O<sub>2</sub>, since the same effect has been reported for apples in CA storage (Burg, 2004). Wang and Dilley (2000) proposed that hypobaric ventilation



removes a scald-related volatile substance that otherwise accumulates and partitions into the epicuticular wax of fruit stored under air at atmospheric pressure. They provided evidence that  $\alpha$ -farnesene and 6-methyl-5-hepten-2-one (MHO) accumulation in the epicuticular wax associated with hypobaric storage may be involved. Burg (2010) reported that:

The high gaseous diffusion rate at a low pressure eliminates the commodity's surface to centre O<sub>2</sub> gradient created by respiratory O<sub>2</sub> consumption, causing different commodity types to have nearly identical low O<sub>2</sub> tolerances, near 0.1%.

Under hypobaric conditions, Burg (2004) reported that fermentation is not induced at O<sub>2</sub> levels as low as 0.06–0.15 kPa, while fermentation can occur under atmospheric pressure. Burton (1989) and Mapson and Burton (1962) reported that gaseous diffusion through the periderm of mature potato tubers was entirely or almost entirely through the lenticels. They reported that permeability of tubers ranged from about 0.7 to about 2.5 mm<sup>3</sup>/cm<sup>2</sup>/h/kPa depending on maturity, time in storage and cultivar, and O<sub>2</sub> diffusion would occur with an O<sub>2</sub> deficit of about 0.4–0.5 kPa.

### Respiration rate

Hypobaric storage has been shown to reduce respiration rate compared with storage under atmospheric pressure. This has been shown on many fruit and vegetables, including sections of oat leaves (Veierskov and Kirk, 1986), oranges (Min and Oogaki, 1986), tomatoes (Workman *et al.*, 1957), apples (Bubb, 1975b), asparagus spears (Li *et al.*, 2006) and cranberries (Lougheed *et al.*, 1978). Cytochrome oxidase, the final electron-transferring enzyme of the respiratory chain, has a great affinity to O<sub>2</sub> but cytochrome oxidase is still able to operate under low O<sub>2</sub> pressure, such as 0.01 atm, without its activity being altered (Mapson and Burton, 1962; Burton, 1989). However, the reduction of respiration rate, during hypobaric storage, was attributed to the malfunctioning of

oxidases such as PPO or AA oxidase. Hence, under the hypobaric condition of 0.70 atm (532 mm Hg), it is expected that O<sub>2</sub> partial pressure is not sufficiently reduced to cause the respiration rate to decrease. In the tissues of higher plants, the effect of exposure to pure O<sub>2</sub> was to stop CO<sub>2</sub> output, possibly the result of the inactivation of the associated enzyme systems (Caldwell, 1965).

### Chilling injury

Hypobaric storage can reduce susceptibility to chilling injury. Chen *et al.* (2013a) reported that MDA content in Chinese bayberry fruit (*Myrica rubra*) was related to chilling injury since it is considered to be an indicator of membrane lipid peroxidation caused by oxidative stress. Electrolyte leakages, as well as MDA content, are indicators of cell membrane damage. MDA content and electrolyte leakage are used to indicate lipid peroxidation of membrane lipids and membrane permeability, respectively, which increase during low temperature storage (Zhao *et al.*, 2006). MDA content of the Chinese bayberry fruit under normal atmospheric pressure conditions increased gradually during storage, while storage for 15 days under hypobaric pressures of 85 ± 5, 55 ± 5 and 15 ± 5 kPa all inhibited the accumulation of MDA. Similar results had been obtained by Li *et al.* (2006), who reported that hypobaric storage could reduce MDA accumulation in asparagus.

### Chlorosis

Nilsen and Hodges (1983) exposed bean leaves (*Phaseolus vulgaris*) to ethylene by dipping them in 30 ppm Ethephon and storing them at 26°C, which resulted in them becoming chlorotic more rapidly (reaching peak levels within 6 h) than those not treated. However, when the Ethephon-treated leaves were stored under hypobaric conditions (200 millibars, with O<sub>2</sub> and CO<sub>2</sub> compositions set to approximate normal atmospheric partial pressures), chlorophyll loss was prevented.

## Desiccation

Reduced pressure inside the store can result in rapid water loss from the crop, since the boiling point of water reduces from 100°C at atmospheric pressure to 0°C at 4.6 mm Hg. Therefore there is a clear tendency to desiccation under hypobaric storage and fruit and vegetables need to be retained at humidity as close as possible to saturation in order to limit weight loss. The particular hypobaric pressures used for fruit and vegetable storage can also affect their weight loss. Apelbaum *et al.* (1977b) tested the effect of hypobaric storage on mango fruit and observed that at pressure below 50 mm Hg, mangoes were severely desiccated. Cicale and Jamieson (1978, quoted by Burg 2004) reported that their best results were storage of avocados at 61 mm Hg, since lower storage pressure resulted in higher desiccation. Patterson and Melsted (1977) reported that cherries under hypobaric conditions resulted in some problems with desiccation, especially at 41 mm Hg. Bubb (1975b) found that apples under hypobaric storage (35–40 mm Hg and 70–80 mm Hg) had higher weight losses than the apples under atmospheric pressure or controlled atmosphere storage. Hughes *et al.* (1981) also reported increased desiccation of capsicums during hypobaric storage compared with those stored under atmospheric pressure. An *et al.* (2009) reported that curled lettuce had high moisture loss during storage under both 190 and 380 mm Hg compared with those under atmospheric pressure. Spalding and Reeder (1976) reported that average weight loss and shrivelling of limes were higher during storage under hypobaric conditions compared with those stored at atmospheric pressure. Conversely, Burg (2004) found that at 0–3°C spring onions, in general, lost less weight during hypobaric storage than during storage in air at atmospheric pressure. He also claimed that the VacuFresh system of hypobaric storage had a very slow removal of air and therefore there was no desiccation problem of the fruit and vegetables transported in these containers. In general, the weight loss of radishes during storage at 1°C under 56 mm Hg and near-saturation

humidity was less than that which occurred during storage under atmospheric pressure (McKeown and Lougheed, 1981). Cicale and Jamieson (1978, quoted by Burg 2004) found that avocados lost 1.2% in weight at various pressures ranging from 61 to 203 mm Hg compared with 5.7% under atmospheric pressure during storage at 6°C for 35 days. Spalding and Reeder (1976) reported that humidity did not appear to be a factor in the storage life of avocados, since the acceptability of avocados stored under hypobaric storage at 80–85% RH and 98–100% RH was not significantly different. All successful hypobaric systems have some form of humidification of the air as it enters the chambers, but Burg (2004) commented that in many cases these may have been insufficient.

In order to minimize desiccation, McKeown and Lougheed (1981) reported that the highest weight loss of asparagus was while the pressure was decreasing and humidity was not near saturation and they sprayed asparagus spears with water before hypobaric storage to reduce desiccation. Laurin *et al.* (2006) reported that water can be sprayed on cucumbers to resolve the problem of insufficient relative humidity, which causes desiccation during hypobaric storage.

In his hypobaric patent application, Burg (1976) states:

Though relative humidities of 80% are usefully permissible, the preferable relative humidity of the air in the storage chamber should be higher than approximately 90% for the storage of foodstuffs such as fruits and vegetables.

Burg and Kosson (1983) commented that reduction in air pressure surrounding plant tissue will reduce the cellular hydrostatic pressure. This reduction can lead to decreased cellular water potential, but cellular activity will only be slightly altered by this pressure reduction and changing this by 20 atm will only change the water activity by 3% or 4%. Burton (1989) described the vapour pressure gradient between the surface of fruit or vegetables and the air in which they are in contact. A proportion of the water in fresh fruit, vegetables and flowers will pass from cell to cell and eventually to the

surface, where it evaporates using latent heat from the commodity. This principle is used in vacuum cooling of fruit and vegetables. Some fruit and vegetables are coated with a waxy cuticle or, in the case of some root crops, a periderm that restricts this water loss. The permeability of fruit and some vegetable tissue to gas exchange is also affected by intercellular spaces and cuticle composition as well as the presence of stomata, lenticels and hydrothodes. Laurin *et al.* (2006) commented that it is likely that desiccation of fruit and vegetables under hypobaric storage is related to an increase in transpiration rate enhanced by the properties and action of stomata, lenticels, cuticle and epidermal cells. In grapes the epidermis does not contain a significant number of functional stomata; therefore water loss occurs mostly through the cuticle, which in turn restricts water loss. Stomata have been shown to be affected by storage conditions. For example, after 96 h of storage, cucumbers under hypobaric conditions had significantly more open stomata than those under atmospheric pressure. They also commented that it is likely that desiccation under hypobaric conditions is due to an increase in transpiration rate enhanced by the properties and action of stomata as well as moisture being more volatile at reduced atmospheric pressure.

### Fungal infection

Although the high humidity maintained in hypobaric stores is generally suitable for fungal growth and decay development, there are several reports that show hypobaric conditions can reduce decay. Many authors have pointed out that hypobaric conditions can retard or limit pathogen growth (Burg and Kosson, 1983; Goszczynska and Ryszard, 1988; Loughheed *et al.*, 1978; Chau and Alvarez, 1983). Inhibition of the germination and growth of fungal spore at low O<sub>2</sub> levels was demonstrated in the early studies of Brown (1922). Couey *et al.* (1966) showed that postharvest decay in strawberry fruit was reduced at O<sub>2</sub> levels of 0.5% or less and they demonstrated a direct effect of such low O<sub>2</sub> environments on growth of

mycelium and sporulation. Less pathogenic breakdown was observed in cranberries under hypobaric conditions of 76 mm Hg than in fruit stored under atmospheric pressure. Similarly, Chau and Alvarez (1983) reported that papaya inoculated with *Colletotrichum gloeosporioides* developed less infection when stored under about 15 mm Hg than when stored under normal atmospheric pressure. Apelbaum and Barkai-Golan (1977) showed that the degree of inhibition of fungal growth in hypobaric stores increased with the reduction in pressure below 150 mm Hg. They also reported that hypobaric pressure had direct fungistatic effects on spore germination and mycelium growth of various storage fungi. In a review, the control of postharvest diseases by maintaining O<sub>2</sub> partial pressure in the region of 0.1–0.25 ± 0.008 kPa was reported by Burg (2004). However, these low O<sub>2</sub> levels could damage the fruit. In contrast, studies by Bangerth (1974) with various fruit and vegetables, including tomatoes, peppers and cucumbers, found a high incidence of decay after storage at hypobaric pressure. In order to suppress decay, he recommended a combination of hypobaric storage with postharvest fungicidal treatments. Barkai Golan and Phillips (1991) reported that in *in vitro* studies, different fungi responded differently to reduced pressures. For example, storage under 100 mm Hg inhibited spore germination of *Penicillium digitatum* compared with 760 mm Hg, while inhibition of spore germination of *Botrytis cinerea* and *Alternaria alternata* occurred only at 50 mm Hg. However, 50 mm Hg had no effect on the germination of *Geotrichum candidum* spores, but reducing the pressure to 25 mm Hg totally prevented spore germination of *P. digitatum*, *B. cinerea* and *A. alternata* though it had almost no effect on the germination of *G. candidum*. Transfer of inhibited cultures from hypobaric to atmospheric pressure resulted in renewed growth, suggesting that there was no irreversible damage to the fungi (Alvarez, 1980; Alvarez and Nishijima, 1987). In contrast, the effectiveness of short hypobaric treatments against postharvest diseases was investigated by Romanazzi *et al.* (2001) who

found that it reduced fungal infections in sweet cherries, strawberries and table grapes. Adams *et al.* (1976) investigated the effects of a range of pressures (760–122 mm Hg) on the growth of *Penicillium expansum* and *P. patulum* and their production of the toxin patulin. They demonstrated that the amount of sporulation decreased with reduction in pressure, but mycelial growth was similar for 456 and 357 mm Hg and patulin production was lower at the lower pressures.

### Insects

Disinfestation of fruit and vegetables by exposure to hypobaric conditions during export has been described by Burg (2010), Chen *et al.* (2005), Davenport *et al.* (2006), Johnson and Zettler (2009), Mbata and Philips (2001) and Navarro *et al.* (2001, 2007). Effective control of insects has been observed at O<sub>2</sub> concentrations of less than 6.6 kPa, and especially at 0.15–0.30 kPa (Burg, 2004). Burg (2004) and Aharoni *et al.* (1986) gave the optimal condition for transporting many tropical fruits at 13°C and where the pressure in the container could be reduced to 15–20 mm Hg this would kill most insects infesting the fruit. They claimed that in these conditions 98% of fruit fly eggs and larvae were killed within 1 week and all of them by the 11th day, and all the green peach aphids on wrapped head lettuce in 2½ days at 2°C. Insect mortality under hypobaric storage is predominantly caused by low O<sub>2</sub> concentrations (Navarro and Calderon, 1979), though the low humidity that could be generated in a hypobaric system could also enhance its lethal effect on insects (Navarro, 1978; Jiao *et al.*, 2012). When insects were placed into a hypoxic environment for a sufficient duration, adenosine triphosphate production was reduced, resulting in increasing membrane phospholipid hydrolysis (Herreid, 1980). Cell and mitochondrial membranes then become permeable, causing cell damage or death (Mitcham *et al.*, 2006). A patent was taken out in the USA in 2005 by Timothy K. Essert and Manuel C. Lagunas-Solar of the University of California (PCT/US2004/013225).

### Contamination

Hypobaric conditions have been shown to be effective in enabling different crops to be stored together without mutual contamination. For example, carrots exposed to ethylene can synthesize isocoumarin, which gives them a bitter flavour (Lafuente *et al.*, 1989). Keeping apples, cabbages and carrots together in a store at 2°C and 60 mm Hg resulted in no isocoumarin detected in the carrots despite the assumed presence of ethylene from the apples (McKeown and Loughheed, 1981). A sensory panel compared these carrots with those stored in O<sub>2</sub> levels below 2 kPa and bags of slaked lime to absorb CO<sub>2</sub> and found that those from the hypobaric storage were superior.

### Effects on Selected Crops

#### Apple

Optimum hypobaric conditions appear to vary with cultivars. In Korea, Kim *et al.* (1969, quoted by Ryall and Pentzer, 1974) reported that optimum conditions for ‘Summer Pearmain’ were 200 mm Hg and for ‘Jonathan’ it was 100 mm Hg. Bubb and Langridge (1974) found no extension in the storage life of ‘Cox’s Orange Pippin’ at 3.3°C under 380 mm Hg (0.5 atm). However, they found that storage of ‘Tydeman’s Late Orange’ at 3.3°C had reduced respiration rates and ethylene production under 76 mm Hg compared with those stored under atmospheric pressure or under 380 mm Hg. Bubb (1975b) harvested ‘Cox’s Orange Pippin’ apples in September and compared the following storage conditions in 3.3°C: air at atmospheric pressure; CA of 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> at atmospheric pressure; and hypobaric conditions of 35–40 mm Hg or 70–80 mm Hg. They found that the onset of ethylene biosynthesis was delayed by 50 days in the CA stored fruit, by 70 days in the hypobaric stored fruit at 70–80 mm Hg and by 100 days under the 35–40 mm Hg, all compared with those stored in air in atmospheric pressure.

Respiration rate was about half the level in fruit that had been stored under hypobaric conditions and slightly more in fruit that had been stored under CA compared with those stored in air under atmospheric pressure. This effect continued when fruit was removed to 10°C in air for 3 weeks. Those under hypobaric storage had higher weight losses than the other fruit, but those that had been stored under 35–40 mm Hg were firmer than those in air or CA storage. In a subsequent experiment, weight loss was reduced to 0.3% per month by improving the humidification system. The apples from both hypobaric conditions were assessed as having poorer flavour and a 'denser texture' than those that had been stored in air or CA conditions.

Bubb (1975a) also compared the storage of 'Cox's Orange Pippin' apples at 3.3°C in air with CA storage under 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> and 1 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> at atmospheric pressure and under hypobaric storage in 25–30 mm Hg and 50–60 mm Hg. He found that respiration rate of those under hypobaric storage was some 50% higher than those stored under CA storage; ethylene production was lower in those stored under 25–30 mm Hg compared with those under CA storage in 1 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>, but ethylene production was inhibited in fruit under 50–60 mm Hg only until mid-December, after which the production rate rose sharply. The ethylene production rate of the fruit under 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> was similar to the latter but the rise in ethylene production rate was much slower. Little physiological damage was observed on any fruit until the end of April, when all samples showed lenticel rotting, with fruit under 50–60 mm Hg being the worst; and slight core flush and breakdown, with 25–30 mm Hg being the worst for both disorders. Sound fruit from all treatments was transferred to 12°C at that time and the apples that had been stored under 25–30 mm Hg continued to show depression in ethylene production for 2–3 weeks.

Sharples and Langridge (1973) found that 'Cox's Orange Pippin' apples, stored at 3.3°C with airflow of 15 l/h, had more lenticel blotch pitting for those stored under 380 mm Hg compared with those at atmospheric

pressure. However, they reported that there was only 20% breakdown in the fruit stored under 380 mm Hg compared with 50% in those under atmospheric pressure in apples from poor keeping-quality orchards and they commented that this effect may have been related to higher weight loss from the fruit under hypobaric conditions. Bangerth (1984) reported that apples stored under hypobaric pressure of 51 mm Hg never produced autocatalytic ethylene or developed a respiratory climacteric during 11 months of storage. Only a slight decrease in fruit firmness was measured during that time. When ethylene was continuously supplied into the hypobaric containers, a considerable response was observed at the beginning of the storage period, but later the effect of ethylene was only marginal. He also found that there was no diminished response to ethylene in storage under 51 mm Hg, whatever the storage temperature tested. There was a similar decrease in sensitivity to ethylene in terms of respiration rate, softening and volatile flavour substances after shelf-life evaluation with the fruits that had been stored under hypobaric pressures for 2½, 5, 7 or 10 months.

Storing apples under hypobaric conditions resulted in delayed softening, control of physiological disorders and reduced decay development, which resulted in extended shelf-life after removal from storage (Laugheed *et al.*, 1978; Dilley *et al.*, 1982). Wang and Dilley (2000) also found that hypobaric conditions could prevent the development of scald. They reported that apples of the cultivars 'Law Rome' and 'Granny Smith' that were placed under hypobaric conditions of 38 mm Hg within 1 month of harvest did not develop scald during storage at 1°C, but scald developed if there was a delay of 3 months in establishing hypobaric conditions, as it did on fruit stored under atmospheric pressure throughout.

Jiao *et al.* (2013) suggested that exposure to 10 mm Hg at 10°C and greater than 98% RH had potential as an alternative disinfestation treatment against codling moth in apples and 15 days of exposure of the cultivar 'Red Delicious' had no detrimental effect on fruit quality. They studied eggs, 2nd to 3rd instar larvae, 5th instar larvae

and pupae and found that the 5th instar larvae were the most tolerant stage for codling moth exposed to the treatment.

### Asparagus

At 0°C and 20, 40 or 80 mm Hg, Dilley (1990) found that spears could be kept in marketable condition for 4–6 weeks with better AA retention at 20 mm Hg than at the other hypobaric conditions. Li and Zhang (2006) reported an extension in postharvest life of green asparagus in storage under 112.5 ± 37.5 mm Hg. Li *et al.* (2006) found that storage life of asparagus at room temperature was 6 days, in refrigerated storage it was 25 days and in refrigerated storage with hypobaric conditions it was up to 50 days. They also reported that the spears under hypobaric storage had a lower respiration rate, lower losses of chlorophyll, AA, TA and TSS, reduced MDA accumulation and improved sensory quality. McKeown and Loughheed (1981) found that in storage at 3°C and near-saturation humidity under 61 mm Hg and also under 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>, CA-stored asparagus spears remained firm and green for 42 days while those in air were senescent. However, spears from all three treatments showed a disorder that resembled chilling injury. They concluded that there appears to be limited potential for the storage of asparagus at 3°C. The storage life of fresh green asparagus under hypobaric pressure of 266–342 mm Hg was extended to 50 days compared with 25 days refrigerated storage and only 6 days under room temperature, both at atmospheric pressure. Furthermore, the respiration rate was lower and it prevented loss of chlorophyll, vitamin C and acidity, improved sensory qualities and delayed postharvest senescence when stored under hypobaric conditions (Li *et al.*, 2006). Li (2006) stored the asparagus cultivar UC800 at -2°C and 85 ± 5% RH under hyperbaric conditions, starting at 10 kPa then transferred to 20 kPa and finally to 35 kPa. This combination was compared with storage under atmospheric pressure or constant reduced pressure levels. This three-stage hypobaric storage

slowed the reduction of de-greening, sugar, soluble protein, AA and TA, compared with the normal atmosphere and constant hypobaric levels. It also reduced their respiration rate and ethylene emission and increased the activities of superoxide dismutase and catalase.

### Avocado

Burg (2004) summarized his work over many years on the effects of hypobaric storage on avocados. The cultivar 'Choquette' stored at 14.4°C under atmospheric pressure started to ripen in 8–9 days and the fruits were fully ripe in 14 days. Softening of those under 40–101 mm Hg began after 25 days and when they were transferred to 20°C under atmospheric pressure all fruit developed normal taste with no internal blackening or decay. He subsequently found that in storage at 12.8°C, hypobaric conditions of 101–152 mm Hg were better than at 40–81 mm Hg; and in later work he reported that 21 mm Hg was optimal at 10°C. With the cultivar 'Waldin', he reported that in storage at 10°C their postharvest life was improved as the pressure was lowered from 101–152 mm Hg down to 60–81 mm Hg, with the fruit remaining firm for 30 days at 61–81 mm Hg compared with 12–16 days at atmospheric pressure. He reported similar results for avocados in storage at 12°C, but all fruit ripened quicker.

Spalding and Reeder (1976) compared storage of 'Waldin' at 7.2°C and 98–100% RH for 25 days at atmospheric pressure in air with CA storage under 2 kPa O<sub>2</sub> and 10 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> and 0 kPa CO<sub>2</sub> and two hypobaric storage conditions in 91 mm Hg, one with added CO<sub>2</sub> at 10 kPa. After storage, all the fruit was ripened at 21.1°C. They found that 92% of the fruits stored in the CA of 2 kPa O<sub>2</sub> and 10 kPa CO<sub>2</sub> were acceptable and all those in the hypobaric conditions of 91 mm Hg plus 10 kPa CO<sub>2</sub>, while none of those in the other treatments were acceptable. The factors that affected acceptability were anthracnose disease (*Colletotrichum gloeosporioides*) and chilling injury, both of which were completely absent in fruit stored under 91 mm Hg plus 10 kPa

CO<sub>2</sub>. They defined acceptable fruit as having good appearance, free of moderate or severe decay and chilling injury and having no off-flavours. They also found no stem-end rot (*Diplodia natalensis*) directly after storage; and after ripening at 21.1°C again no stem-end rot was detected, except at low levels on those that had been stored under 91 mm Hg and higher levels in those that had been stored under 2 kPa O<sub>2</sub> and 10 kPa CO<sub>2</sub>:

Black pitted areas developed in lenticels during softening of avocados stored at atmospheric pressure or hypobaric plus 10 kPa CO<sub>2</sub>. However, pitting was slight and was not considered to be objectionable to the average consumer. Tissue from the infected areas contained *Pestalotia* spp. fungus.

From this they concluded that high CO<sub>2</sub> was necessary for the successful storage of avocados which is borne out by many other workers. However, Burdon *et al.* (2008) found that the inclusion of CO<sub>2</sub> at 5 kPa under CA storage retarded fruit ripening but stimulated rot expression and they concluded that it should not be used for CA storage of New Zealand-grown 'Hass'. With the cultivar 'Lula', Burg (2004) reported that in storage at 7.2°C the fruits began to soften within 21 days and all were soft after 41 days under atmospheric pressure, while under 81–122 mm Hg it took 88 days for them to begin to soften. In another experiment at 8°C and 40–81 mm Hg, the fruits remained firm for 75–100 days compared with 23–30 days under atmospheric pressure. When they were stored at 61 mm Hg for 102 days, they became eating-ripe within 3–4 days when they were transferred to 26.7°C at atmospheric pressure (Burg, 2004). Spalding and Reeder (1976) compared storage of 'Lula' at 10°C and 98–100% RH for 6 weeks at atmospheric pressure in air, CA storage under 2 kPa O<sub>2</sub> and 10 kPa CO<sub>2</sub> and hypobaric storage under 76 and 152 mm Hg. After storage, all the fruit was ripened at 21.1°C and they found that 70% of the fruits stored under CA were acceptable and none of those in the other treatments were acceptable, which was mainly due to chilling injury symptoms and decay due to anthracnose. 'Lula' avocados stored under 2 kPa O<sub>2</sub> with 10 kPa CO<sub>2</sub>

under atmospheric pressure were acceptable after softening and also this CA mixture inhibited the development of decay and chilling injury, confirming the finding that CO<sub>2</sub> is essential under CA storage (Spalding and Reeder, 1975).

With the cultivar 'Booth 8', Burg (2004) reported that after storage at 4.4°C under 40, 61, 81 or 122 mm Hg for 30 days they ripened in 2–3 days when transferred to 20°C under atmospheric pressure but had a poor flavour. In another experiment, waxed 'Booth 8' were stored at 7.8–10°C where they began to ripen in 8–22 days under atmospheric pressure, while under 40–81 mm Hg they did not soften during 50 days of storage but ripened without skin darkening when they were transferred to 23.9°C under atmospheric pressure. After 64 days of storage at 7.8–10°C under 40–81 mm Hg, fruits were still firm but they did not ripen to an acceptable quality when transferred to 23.9°C under atmospheric pressure.

With the cultivar 'Hass' it was reported that in storage at 5°C under atmospheric pressure fruits softened within 30 days, while under 15 to 40 mm Hg they were still hard and almost half of those under 61 mm Hg had begun to ripen (Cicale and Jamieson, 1978, quoted by Burg, 2004). In storage under 40 mm Hg, fruits began to soften after 38–45 days but under 15–21 kPa they remained firm and when removed to ambient conditions they ripened in 5.4 days, which was the same period as freshly harvested fruit. In another trial at 6°C for 35 days, fruit under atmospheric pressure lost 1.2% in weight at pressures ranging from 61 to 203 mm Hg compared with 5.7% under atmospheric pressure. After 70 days of storage the weight losses were 1.7% for 81 and 101 mm Hg and 3% for 61 mm Hg. They commented that their best results were storage at 61 mm Hg, since this retarded softening, and fruits ripened normally when transferred to 14°C under atmospheric pressure and lower storage pressures resulted in higher desiccation.

Spores of *Glomerella cingulata* (which causes anthracnose) germinate on the surface of avocado fruit in the field and form appressoria. The fungus then remains quiescent until antifungal dienes in the skin

of fruit break down due to degradation by lipoxygenase activity. Breakdown of the dienes has been shown to be delayed by various treatments, including hypobaric storage (Prusky *et al.*, 1983, 1995). Previous studies on hypobaric storage of avocados suggested that atmospheres both low in O<sub>2</sub> and high in CO<sub>2</sub> are required for successful suppression of anthracnose development (Spalding and Reeder, 1976).

### Apricot

Salunkhe and Wu (1973b) and Haard and Salunkhe (1975) found that storage life of apricots could be extended from 53 days in cold storage to 90 days in cold storage combined with reduced pressure of 102 mm Hg. They found that hypobaric storage delayed carotenoid production, but after storage carotenoid, sugar and acid levels were the same as for those that had been in cold storage at atmospheric pressure.

### Bamboo shoots

Chen *et al.* (2013b) stored freshly harvested bamboo shoots (*Phyllostachys violascens*) at 2 ± 1°C under various hypobaric conditions (101, 75, 50 and 25 kPa) for 35 days. They found that under 50 kPa there were reduced accumulations of lignin and cellulose in the shoots' cells. They also found that it inhibited ethylene production, reduced the rate of accumulation of MDA and hydrogen peroxide, and maintained significantly higher activities of superoxide dismutase (SOD), catalase and ascorbate peroxidase, but restrained the activities of PAL and POD. They therefore concluded that the delay in flesh lignification was due to maintenance of higher antioxidant enzyme activities and reduced ethylene production.

### Banana

In his review, Burg (2004) reported that at 13.3–14.4°C the banana varieties 'Valery'

and 'Gros Michel' ripened in 10 days under atmospheric pressure; they remained green for 40–50 days at 152 mm Hg, but mould could develop. He also reported that 'Valery' at 13.3°C remained green for more than 105 days in storage under 6.4, 9.5 or 12.7 kPa and ripened normally when they were transferred to atmospheric pressure and exposed to exogenous ethylene. There were no deleterious effects on flavour or aroma. Burg (2004) stored bananas under 49, 61, 76, 101, 122 and 167 mm Hg for 30 days and found they lost 1.1–3.6% in weight, with the higher losses at the higher pressures because of their higher respiration rates. All the fruits were still green and there were no apparent differences between the different hypobaric conditions. Bangerth (1984) found that there was no diminished response to ethylene in storage under 51 mm Hg, whatever the storage temperature tested. Apelbaum *et al.* (1977a) stored the variety 'Dwarf Cavendish' at 14°C at 81 and 253 mm Hg and atmospheric pressure and found that they began to turn yellow after 30 days under atmospheric pressure, 60 days under 253 mm Hg and were still dark green under 81 mm Hg. When they were subsequently transferred to atmospheric pressure at 20°C and exposed to exogenous ethylene, they all ripened to a good flavour, texture and aroma. Bangerth (1984) successfully stored 'Cavendish' at 14°C under 51 mm Hg for 12 weeks and found that when they were subsequently ripened in exogenous ethylene at 50 µl/l they were the same quality as freshly harvested fruit or those that had been stored under atmospheric pressure.

Hypobaric storage has also been shown to delay the speed of ripening of bananas that have been initiated to ripen. Quazi and Freebairn (1970) found that fruit that had been initiated to ripen by exposure to exogenous ethylene for 2–5 h longer than required to initiate ripening did not ripen during hypobaric storage, but began to ripen within 1–2 days after transfer to ambient atmospheric pressure and had good eating quality and texture. However, fruit that had been initiated to ripen by exposure to ethylene with exposure time of 16 h longer than required did ripen during hypobaric storage,



again with good eating quality and texture. Liu (1976b) initiated 'Dwarf Cavendish' to ripen by exposure to ethylene at 10  $\mu\text{l/l}$  at 21°C. They then stored them at 14°C for 28 days under hypobaric storage of 51 and 79 mm Hg or CA storage of 1 kPa O<sub>2</sub> + 99 kPa N<sub>2</sub>. All fruits remained green and firm and continued to ripen normally after they had been removed to ripening temperature in atmospheric pressure. Quazi and Freebairn (1970) showed that high CO<sub>2</sub> and low O<sub>2</sub> delayed the increased production of ethylene associated with the initiation of ripening in bananas, but the application of exogenous ethylene was shown to reverse this effect. Wade (1974) showed that bananas could be ripened in atmospheres of reduced O<sub>2</sub>, even as low as 1 kPa, but the peel failed to de-green, which resulted in ripe fruit that were still green. Similar effects were shown at O<sub>2</sub> levels as high as 15 kPa. Since the de-greening process in Cavendish bananas is entirely due to chlorophyll degradation (Seymour *et al.*, 1987b; Blackbourn *et al.*, 1990), the CA storage effect was presumably due to suppression of this process. Hesselman and Freebairn (1969) showed that ripening of bananas that had already been initiated to ripen by ethylene was slowed in low O<sub>2</sub> atmospheres.

### Bean

Green bean pods (*Phaseolus vulgaris*) could be kept in good condition in storage for 30 days at reduced pressure compared with 10–13 days in cold storage alone (Haard and Salunkhe, 1975). Spalding (1980) reported that the cultivars 'McCaslan 42' pole beans and 'Sprite' bush beans stored better at 7°C under 76 and 152 mm Hg for 2 weeks than similar beans stored at 760 mm Hg. Burg (1975) found that beans stored at 7.2°C and 60 mm Hg were in excellent condition after 26 days compared with those stored at the same temperature at 760 mm Hg, which were shrivelled and in poor condition. Knee and Aggarwal (2000) found that water-soaked lesions appeared on green beans kept in vacuum containers at 380 mm Hg; however, this effect is commonly associated with water condensation (Thompson, 2015).

Hypobaric storage has been shown to be effective on dried beans. Berrios *et al.* (1999) reported that the combined effect of refrigeration and hypobaric storage demonstrated potential for maintaining the fresh quality of black beans (*P. vulgaris*) in storage for up to 2 years. Black beans stored at 4.5°C and 50–60% RH and hypobaric pressure of 125 mm Hg exhibited quality factors characteristic of fresh beans, such as shorter cooking time, smaller quantities of solids loss, lower leaching of electrolytes and lower percentage of hard-shell than beans stored at 23–25°C and 30–50% RH. At 4.5°C, beans stored under 125 mm Hg had a germination rate of 93% while the rate for those stored at atmospheric pressure was 72%. Beans stored in ambient conditions exhibited hard-to-cook, which is a problem in the processing industry and means that their cooking time is considerably extended

### Beet

In general, the weight loss after storage at 1°C under 61 mm Hg and near-saturation humidity and CA storage under 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> was less than that which occurred after air storage. The beets also retained a fresh appearance after holding under 61 mm Hg and sensory evaluation indicated that those held under 61 mm Hg were similar to those held in air, while beets held under 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> had a lower rating and off-flavours (McKeown and Loughed, 1981).

### Blueberry

Al-Qurashi *et al.* (2005) found that the cultivar 'Rabbiteye' stored under 13.5 mm Hg lost less weight, were firm, developed less decay and did not show any shrivelling during storage at 4°C for 28 days, compared with those stored at atmospheric pressure as a control. Borecka and Pliszka (1985) observed that blueberries stored under 38 mm Hg tasted good, contained less acid and had lower total soluble solid than other treatments. Burg (2004) reported that in storage

at 0–2°C those under atmospheric pressure spoiled within 4 weeks while those under hypobaric storage ranging from 81 to 203 mm Hg were limited to 6 weeks. This was reported to be because of mould growth. He also quoted Dilley *et al.* (1989), who reported similar results on reduction in mould on the cultivar ‘Jersey’ during 44 days of storage, with 87% decay for those under atmospheric pressure and only some 11% for those under 21 mm Hg.

### Broccoli

Broccoli heads were kept for 4 days at 1.1°C and then at 0°C and 10 mm Hg or atmospheric pressure for 21 days. They were assessed for quality after a shelf-life at 10°C. The ones that had been stored under atmospheric pressure had 60–90% yellowing and those that been in hypobaric storage had 40% yellowing with no off odours or off-flavours (Burg, 2004).

### Brussels sprouts

Burg (2004) recommended that storage of Brussels sprouts under 10 mm Hg should be tested, since it could prevent the growth of microorganisms that could result in postharvest diseases, although Ward (1975) found no improvement in storage under 76 mm Hg and high humidity.

### Cabbage

Cabbages, like other leaf vegetables, become chlorotic when exposed to ethylene. It was reported that white cabbages were successfully stored with apples at 0°C and 60 mm Hg (McKeown and Loughheed, 1981), presumably by limiting the effect of ethylene produced by the apples or removing the ethylene from the container before it could have an effect. Onoda *et al.* (1989) found that cabbages retained better appearance and had lower weight loss when stored in cycles between 100 and 300 mm Hg (with no

humidification) than those stored at a constant atmospheric pressure.

### Capsicum

Kopec (1980) compared storage of capsicum at atmospheric pressure with storage under 77 mm Hg and found that hypobaric storage retarded ripening, but once ripening started it progressed at the same rate as in atmospheric pressure. However, capsicums are classified as non-climacteric (Bosland and Votava, 2000) and so it is difficult to interpret these results, since climacteric fruits do not ripen. Perhaps what is being referred to is change of colour from green. Hughes *et al.* (1981) experimented with capsicums in hypobaric storage at 8.6–9.0°C and 82–90% RH under 38, 76 or 150 mm Hg. They found that they had a significantly higher weight loss during storage than those stored at normal pressure (Table 11.2). This was undoubtedly due to the comparatively low humidity, which should have been as close as possible to saturation. However, all had similar levels of ‘sound’ fruit after 20 days within the range of 60–72%. Hypobaric stored capsicums did not have an increased subsequent storage life compared with those stored under atmospheric pressure at the same temperature (Table 11.2). Burg (1975) found that capsicums could be kept for 7 weeks at 7.2°C under hypobaric storage without loss of colour or ‘crispness’. Burg (2004) reported that under storage at 7.2°C and 80 mm Hg capsicums were in excellent condition after 28 days, while those stored under atmospheric pressure began to deteriorate after 16 days and were in poor condition after 21 days. Even after 46 days, capsicums under hypobaric storage were considered to be marketable except for a trace of mould on the cut stem. It was concluded that decay was the limiting factor in hypobaric storage. Jamieson (1980, quoted by Burg, 2004) found that after 50 days storage 87%, 80%, 47% and 0% were ‘saleable’ from 80, 40, 15 and 760 mm Hg, respectively, in experiments in the Grumman Allied Industries laboratory. Bangerth (1973) stored the cultivar ‘Neusiedler Ideal’ at 10–12°C for 23 days and found that those under 75 mm Hg were firmer, greener and had

**Table 11.2.** The effects of hypobaric conditions during storage and after removal on the mean percentage weight loss of fruit as a percentage of their original weight and on the mean percentage of sound marketable fruit. Modified from Hughes *et al.* (1981).

Pressure	20 days hypobaric storage at 8.8°C		7 days subsequent shelf-life in 760 mm Hg pressure at 20°C	
	Weight loss %	Marketable %	Weight loss %	Marketable %
760 mm Hg	0.03	68	0.87	43
152 mm Hg	0.16	72	0.76	46
76 mm Hg	0.22	68	1.00	36
38 mm Hg	0.15	60	1.10	42

slightly higher AA content and lower ethylene production than those stored under atmospheric pressure.

### Cauliflower

Cauliflowers were kept for 4 days at 1.1°C then 0°C and 10, 20 or 40 mm Hg or atmospheric pressure for 21 days. They were assessed for quality after a shelf-life at 10°C by Burg (2004). At the end of the shelf-life test the ones that had been stored under atmospheric pressure had leaves that were yellow and dry and easily abscised with minimal handling. Those that been in hypobaric storage had some yellowing but remained firmly attached when handled and had a superior appearance, especially those that had been stored at 10 or 20 mm Hg. However, Ward (1975) found no improvement in storage under 76 mm Hg and high humidity.

### Cherry

The cultivar 'Bing' stored well for 93 days in experimental hypobaric chambers at 102 mm Hg at 0°C. This was up to 33 days longer than those stored at 0°C under atmospheric pressure. However, their pedicels stayed green for only 60 days and pedicel browning can affect marketability. Hypobaric conditions delayed chlorophyll and starch breakdown in fruit as well as carotenoid formation and the decrease in sugars and total acidity (Salunkhe and Wu, 1973b). Patterson and Melsted (1977) reported that cherries could be stored for 6–10 weeks (depending on their condition

at harvest) at 41–203 mm Hg with good retention of colour and brightness and delayed disease development. They also found that they stored equally well under CA storage with high CO<sub>2</sub>, but under hypobaric conditions there were some problems with desiccation, especially at 41 mm Hg. CA storage recommendations included: –1°C to –0.6°C with 20–25 kPa CO<sub>2</sub> and 0.5–2 kPa O<sub>2</sub> which helped to retain fruit firmness, green pedicels and bright fruit colour (Hardenburg *et al.*, 1990); –1.1°C with 20–25 kPa CO<sub>2</sub> and 10–20 kPa O<sub>2</sub> (SeaLand, 1991); 20–30 kPa CO<sub>2</sub> reduced decay (Haard and Salunkhe, 1975); and 0–5°C with 10–12 kPa CO<sub>2</sub> and 3–10 kPa O<sub>2</sub> (Kader, 1989).

The effectiveness of short hypobaric treatments against postharvest rots was investigated by Romanazzi *et al.* (2001), who found that the cultivar Ferrovia exposed to 0.50 atm for 4 h had the lowest incidence of *Botrytis cinerea*, *Monilinia laxa* and total rots. Romanazzi *et al.* (2003) also found that the combination of spraying with chitosan at 0.1%, 0.5% or 1.0% 7 days before harvest and storage under 0.50 atm for 4 h directly after harvest effectively controlled fungal decay of sweet cherries during 14 days of storage at 0 ± 1°C, followed by a 7-day shelf-life. Fungi associated with rots included brown rot (*M. laxa*), grey mould (*B. cinerea*), blue mould (*Penicillium expansum*), *Alternaria* rot (*Alternaria* sp.) and *Rhizopus* rot (*Rhizopus* sp.).

### Cranberry

Storing cranberries (*Vaccinium macrocarpon*) under 80 mm Hg resulted in a lower

respiration rate and ethylene production as well as an extended shelf-life compared with those stored under atmospheric pressure (Pelter, 1975 quoted by Al-Qurashi *et al.*, 2005). Lougheed *et al.* (1978) explained that ethylene and respiration rate of cranberries stored under 80 mm Hg decreased compared with fruit stored under atmospheric pressure.

### Cucumber

Hypobaric storage of cucumbers in 76 mm Hg extended the storage life to 7 weeks, compared with 3–4 weeks in cold storage (Bangerth, 1974), and reduced respiration rate by 67–75%. Burg (2004) reported that storage at 7.2°C resulted in chilling injury and reducing the pressure to 100, 120 or 160 mm Hg reduced chilling injury. However, under 80 mm Hg there were no symptoms after 7 weeks but chilling injury occurred within 1–2 days when they were removed to ambient conditions. Cucumbers were placed at 71 mm Hg for 6 h in the dark at 20°C to simulate air flight transportation and placed subsequently in cold storage facilities at 101 mm Hg at 20°C and 70% RH for 7 days. Results showed that cucumbers exposed to the hypobaric conditions had significantly more open stomata, compared with those at atmospheric pressure, after 96 h of subsequent storage (Laurin *et al.*, 2006).

### Currants

Bangerth (1973) stored blackcurrants, redcurrants and white currants (all *Ribes sativum*) at about 3°C either in atmospheric pressure or under 76 mm Hg for up to 38 days. Those under the hypobaric conditions retained their AA content better than those under atmospheric pressure and also had higher sugar content, less decay and a better taste. Percentages of spoiled berries after 38 days of storage were 21–37% for those under atmospheric pressure and only 0.5–5% for those under hypobaric conditions.

### Grape

Burg (1975) found that hypobaric storage extended the postharvest life of grapes from 14 days in cold storage in air to as much as 60–90 days. Burg (2004) subsequently reported that the cultivar Red Emperor could be stored for 90 days under hypobaric pressures of 30, 65, 80 or 160 mm Hg at 1.7–4.4°C, compared with 21–56 days at –0.5°C to 0°C at atmospheric pressure. Hypochlorous acid vapour was included to control diseases during hypobaric storage. Exposure to an air flow that had been in contact with hypochlorous acid at 0.25%, 1.5% or 5.25% had no mould after 60 days at 1.7–4.4°C. The effectiveness of short hypobaric exposure against postharvest rots had also been investigated by Romanazzi *et al.* (2001). They found that exposure of bunches of the cultivar Italia to 190 mm Hg for 24 h significantly reduced the incidence of grey mould during subsequent storage. Grapes were also wounded and inoculated after hypobaric treatment and it was determined that this treatment decreased infection and diameter of lesions significantly compared with the untreated fruits. The combination of pre-storage treatment with essential oils and storage under 50 kPa (380 mm Hg) gave good control of grey mould in grapes (Servili *et al.*, 2017).

### Grapefruit

Haard and Salunkhe (1975) mentioned two reports that suggested that hypobaric storage could extend the postharvest life of grapefruit compared with cold storage alone. In one report the increase was from 20 days at normal atmospheric pressure to 3–4 months under hypobaric conditions and in the other from 30–40 days at normal atmospheric pressure to 90–120 days under hypobaric storage. Hypobaric storage at 380 mm (the lower limit of the experimental equipment) and 4.5°C had no effect on the incidence of chilling injury (Grierson, 1971).

### Kohlrabi

Bangerth (1974) reported better leaf retention in kohlrabi in storage at 2–4°C under 75 mm Hg compared with storage under atmospheric pressure at the same temperature.

### Leek

Ward (1975) found no improvement in storage of leeks under 76 mm Hg and high humidity.

### Lettuce

It was claimed that hypobaric storage increased the storage life of lettuce from 14 days in conventional cold stores up to 40–50 days (Haard and Salunkhe, 1975). However, Ward (1975) found no improvement in storage under 76 mm Hg and high humidity and Bangerth (1973) found no improvement in storage under 75 mm Hg. Burg (2004) reported an increase in the physiological disorder, pink rib, when they were exposed to 30 or 80 mm Hg at 0–1°C for 4 weeks. Conversely Jamieson (1980, quoted by Burg, 2004) found that storage at 0.5°C and 90–95% RH under 10, 20, 40 or 80 mm Hg resulted in the lettuce remaining in excellent condition after 63 days of storage. The lettuce that had been stored in the same conditions at atmospheric pressure had significant incidence of pink rib, black heart and decay after only 37 days storage. Burg (2004) also reported that the manufacturer of hypobaric storage containers, Grumman Allied Industries, investigated marginal and pink discoloration and found that during storage at 5 mm Hg over 35 days the disorder was ‘eliminated’. However, it was ‘progressively accentuated’ at 10–40 mm Hg and ‘then decreased in frequency as the pressure was increased from 80 to 160 mm Hg’. It was concluded that the best storage condition was 2°C and 5 mm Hg for 21 days, but with storage for 36 days there was a high incidence of russet spotting. An *et al.* (2009) studied packaging of curled lettuce in small rigid containers

under different hypobaric conditions. In both 190 and 380 mm Hg the lettuce had high moisture loss without any observable benefit in keeping quality compared with atmospheric pressure storage.

### Lime

Spalding and Reeder (1976) reported that limes coated with wax containing 0.1% of either of the fungicides thiabendazole or benomyl remained green and suitable for marketing after 3–4 weeks under hypobaric storage of 170 mm Hg at 21.1°C. They also found that hypobaric storage did not affect chilling injury, but Pantastico (1975) reported that storage in 7kPa O<sub>2</sub> reduced the symptoms of chilling injury compared with storage in air. Haard and Salunkhe (1975) stated that ‘Tahiti’ limes could be stored for 14–35 days in cold storage, but this was extended to 60–90 days when it was combined with hypobaric conditions. Spalding and Reeder (1974a) stored ‘Tahiti’ limes under 152 mm Hg pressure for 6 weeks at 10°C. They showed only small changes in colour, rind thickness, juice content, TSS, total acids and AA. Decay averaged 7.8% compared with 8.3% for those stored in air. Those stored at 228 mm Hg maintained acceptable green colour, but were a slightly lighter green than limes at 152 mm Hg. Those stored under 76 mm Hg maintained acceptable green colour, but had low juice content, thick rinds and a high incidence of decay. Limes from all treatments had acceptable flavour. In subsequent work Spalding and Reeder (1976) found that ‘Tahiti’ limes retained green colour, juice content and flavour acceptable for marketing and had a low incidence of decay during storage at a pressure of 170 mm Hg for up to 6 weeks at 10°C or 15.6°C with 98–100% RH. Fruits stored under atmospheric pressure turned yellow within 3 weeks. They concluded that hypobaric storage, but not CA storage, could be used to extend the storage life of limes, which may provide an advantage for hypobaric storage. They also reported higher weight loss and shrivelling of limes during storage under hypobaric conditions compared with those stored at atmospheric pressure.

### Loquat

Hypobaric storage at 40–50 mm Hg reduced decay by 87% and also reduced fruit browning, flesh ‘leatheriness’, respiration rate and ethylene production of loquats compared with those stored under atmospheric pressure at 2–4°C for 49 days. Peroxidase and PPO activities increased and then reached the highest values after 14 days in air storage. Hypobaric storage reduced the increase in PPO and POD activity compared with those under atmospheric pressure storage and delayed the onset of their peak activity (Gao *et al.*, 2006).

### Jujube

Considerable work has been reported from China on the hypobaric storage of jujube (*Zizyphus jujuba*). Chang (2001) stored the cultivars ‘LiZao’ and ‘DongZao’ under hypobaric conditions. They found that during storage the decrease of firmness and the rate of browning were both significantly ( $p = 0.01$ ) inhibited in storage under 154 mm Hg compared with those not under hypobaric conditions. Weight loss under hypobaric storage was less than 2.5%. It was also reported that the fruits turned red slowly under 154, 385 and 616 mm Hg. Wang *et al.* (2007b) found that during hypobaric storage of the jujube cultivar ‘DongZao’ the fruits softened more slowly and had better retention of AA and other organic acids than those stored at normal pressure. Cao (2005) found that hypobaric storage significantly retained firmness and vitamin C content, reduced acetaldehyde and ethanol contents in pulp and respiratory rate, inhibited AA oxidase and alcohol dehydrogenase activities and slowed down the rate of ethylene production in fruit. Hypobaric storage significantly reduced ‘number of climacteric peak’, but did not postpone ‘climacteric peak’. The work was on the cultivars ‘Huanghua’, ‘Zhanhua’, ‘Zhanhua 2’, ‘Shandong Wudi’ and ‘Dagang’ harvested at different stages of maturity. Cui (2008) reported that hypobaric storage, under 20.3, 50.7 or 101.3 kPa, decreased the respiration rate, delayed loss of AA, decreased the rate of

superoxide anion production and prolonged the postharvest life of the cultivar ‘Lizao’. There were no significant differences between 20.3 kPa and 50.7 kPa; therefore 50.7 kPa was recommended, to reduce the cost. Jin *et al.* (2006) compared storage in different hypobaric conditions and found that 55.7 kPa resulted in retention of the best quality.

### Mango

Ilangantileke *et al.* (1989) tested storage of the cultivar ‘Okrang’ at 13 °C and 60–100 mm Hg and found that the fruit kept well for 4 weeks. Apelbaum *et al.* (1977b) observed that at pressures below 0.05 MPa (50 mm Hg) mangoes underwent desiccation and, even upon ripening, did not develop their natural red and orange colour. However, fruits submitted to pressures of 0.013 and 0.01 MPa (10 and 13 mm Hg) had an extended storage life of 25 and 35 days, respectively. Several Florida varieties of mango that were ripened in air at ambient temperatures showed less anthracnose (*Colletotrichum gloeosporioides*) and stem-end rot (*Diplodia natalensis*) when they had previously been stored under hypobaric conditions at 13°C. The reduction in decay coincided with a retardation in fruit ripening, permitting prolonged storage at 13°C (Spalding and Reeder, 1977). It is possible that this effect on anthracnose and stem-end rot may have been a reflection of the delay in ripening, since postharvest diseases of mangoes, particularly anthracnose, develop as the fruit ripens (Thompson, 2015).

### Okra

Knee and Aggarwal (2000) found that okra showed less darkening of seeds in vacuum containers at 380 mm Hg than in containers without a vacuum.

### Onion

McKeown and Lougheed (1981), Hardenburg *et al.* (1990) and Burg (2004) reported

that bulb onions stored at 26°C and 58% RH under 61 mm Hg for 28 days lost 12.2% in weight. McKeown and Loughheed (1981) reported that there was potential for the use of hypobaric conditions of 61 mm Hg in 26°C and low humidity for the curing of onions.

Burg (2004) found that at 0–3°C spring onions could be stored for almost 3 weeks under 50–60 mm Hg, compared with less than 6 days under atmospheric pressure. Also, storage under 100–150 mm Hg had only a slight beneficial effect compared with atmospheric pressure. Storage at 1°C under 61 mm Hg and near-saturation humidity had little effect on the content of leaf chlorophyll of spring onions after 21 days. In general, their weight loss after hypobaric storage or CA storage under 2kPa O<sub>2</sub> + 0kPa CO<sub>2</sub> was less than that which occurred after air storage. The spring onions also retained a fresh appearance after hypobaric storage, but there was no difference among treatments in the sensory evaluation (McKeown and Loughheed, 1981). They also found better chlorophyll retention in storage at 1°C under 55–60 mm Hg than under atmospheric pressure or CA storage. However, Ward (1975) found no improvement in storage under 76 mm Hg and high humidity.

### Orange

Min and Oogaki (1986) reported a reduction in the respiration rate of oranges stored under hypobaric conditions compared with those stored under atmospheric pressure. The respiration rate remained lower even after the fruits were transferred to atmospheric pressure. The orange cultivar 'Fukuhara' had lower rates of ethylene production and respiration rate after hypobaric storage compared with those under atmospheric pressure throughout. Oranges stored at 190 mm Hg and 98% RH decayed more rapidly after return to atmospheric pressure than those stored at 190 mm Hg and 75% RH or atmospheric pressure and 86% RH throughout (Min and Oogaki, 1986). However, the fruits stored at these humidity levels and low pressure would probably have been severely desiccated.

### Papaya

Papaya stored at 10°C and 98% RH and a pressure of 20 mm Hg ripened more slowly and had less disease development than fruit at atmospheric pressure in refrigerated containers (Alvarez, 1980). Alvarez and Nishijima (1987) also reported that hypobaric storage appeared to suppress postharvest disease development in papaya fruit.

### Parsley

It was reported that storage at 3°C under hypobaric conditions of 75 mm Hg extended the storage life of parsley from 5 weeks in atmospheric pressure to 8 weeks without appreciable losses in protein, AA or chlorophyll content (Bangerth, 1974).

### Peach

Hypobaric storage prolonged the postharvest life of peaches from 7 days at atmospheric pressure to 27 days. It delayed chlorophyll and starch breakdown, carotenoid synthesis and the decrease in sugars and acidity (Salunkhe and Wu, 1973a). Storage at 11–12°C in an aerated hypobaric system at either 190 or 76 mm Hg retarded ripening, compared with storage at the same temperature under atmospheric pressure (Kondou *et al.*, 1983). In the cultivar 'Okubo' that had been stored under hypobaric conditions of 108 mm Hg, the ripening rate after storage was slower than that of fruit that had been stored in air at atmospheric pressure throughout. The physiological disorder known as mealy breakdown was reduced under hypobaric storage, but no effect was found on flesh browning or on abnormal peeling (Kajiura, 1975). S.W. Porritt and R.E. Woodruff (quoted by Pattie and Loughheed, 1974) suggested that 'LPS (low pressure storage) has not proven suitable for peaches'. They recommended that additional research was needed to assess further the value of hypobaric storage to extend storage life of peaches. Wang *et al.* (2015) stored 'Honey' peach

under 101, 10–20, 40–50 or 70–80 kPa for 30 days at 0°C and at 85–90% RH followed by 4 days at 25°C and 80–85% RH. They found that storage at 10–20 kPa delayed decay rates, maintained overall quality and extended the shelf-life. They showed that it effectively delayed increases in both O<sub>2</sub> radical dot production rate and ozone content, enhanced activities of catalase and superoxide dismutase and increased contents of adenosine triphosphate and adenosine diphosphate, while reducing lipoxygenase activity and adenosine monophosphate content during the storage period as well as the following shelf-life.

### Pear

Hypobaric conditions prolonged the storage life of pears by 1.5–4.5 months compared with cold storage alone. It delayed chlorophyll and starch breakdown, carotenoid formation and the decrease in sugars and acidity (Salunkhe and Wu, 1973a).

### Pineapple

Staby (1976) stated that storage of pineapple under hypobaric condition could extend the storage life for up to 30–40 days.

### Plum

Prune (*Prunus domestica*) cultivars that grown mainly for dried fruit production (prune is the French word for plum) were successfully stored under 40–101 mm Hg by Patterson and Melsted (1978).

### Potato

Greening and glycoalkaloids synthesis, including solanine, can occur in stored potato tubers when they are exposed to strong light (Mori and Kozukue, 1995). Burg (2004) reported that greening in light was inhibited

in potato tubers stored at 15°C and 126 mm Hg but there was no effect on their solanine content. He also reported similar results during storage at 2.5°C, 5°C, 10°C or 20°C and 76–95 mm Hg and under CA storage with 2–2.5 kPa O<sub>2</sub>.

### Radish

McKeown and Lougheed (1981) reported that during storage at 1°C under 55–60 mm Hg for 7–28 days the weight losses were similar to the levels reported for CA storage with 2 kPa O<sub>2</sub> and slaked lime to control CO<sub>2</sub>. Off-flavours developed in the CA stored radishes but not in those under hypobaric storage. Bangerth (1974) found that in storage at 2.2–2.8°C and 95% RH there was decreased losses of protein, AA and chlorophyll losses under 75 mm Hg compared with storage in the same conditions under atmospheric pressure. In general, the weight loss during storage at 1°C under 61 mm Hg and near-saturation humidity and CA storage in 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> was less than that which occurred during storage under atmospheric pressure. Radishes also retained a fresh appearance after storage under 61 mm Hg and their sensory evaluation indicated that they were similar to those held in air, while those held under 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> atmosphere had a lower rating and off-flavours (McKeown and Lougheed, 1981). However, Ward (1975) found no improvement in storage under 76 mm Hg and high humidity.

### Spinach

Bangerth (1973) found that storage of spinach at 3°C and 95% RH with 75 mm Hg retained the green colour, vitamin C and total protein content for almost 7 weeks.

### Squash

Burg (2004) reported that storage of 'Yellow Crookneck' squash at 7.2°C and 65, 80 or 150 mm Hg had no beneficial effects



compared with storage at atmospheric pressure. McKeown and Lougheed (1981) carried out limited trials with 'Acorn' squash at 10°C and at low humidity under 61 mm Hg for 42 days after curing, which 'showed some potential'.

### Strawberry

It was reported by Haard and Salunkhe (1975) that the strawberry cultivars 'Tioga' and 'Florida 90' in cold storage could be stored for 21 days under hypobaric pressure, compared with 5–7 days at atmospheric pressure. Bubb (1975c) compared storage of the cultivar 'Cambridge Favourite' at 3.3°C under atmospheric pressure with storage under 25 mm Hg and 101 mm Hg and found better flavour retention under hypobaric storage. When he compared this with storage in O<sub>2</sub> partial pressures similar to those obtained under the hypobaric conditions, he did not observe similar flavour retention. There was some evidence of slight tainting of the flavour after hypobaric storage, but after subsequent shelf-life at 18.5°C he reported that the tainting of the flavour of the fruit became objectionable. After 10 days of storage none of the samples exceeded 2% weight loss, but there were high levels of rotting in all samples, though no consistent effect on fruit firmness. Knee and Aggarwal (2000) found a higher proportion of strawberry fruits were infected by fungi in vacuum containers at 380 mm Hg than in containers with no vacuum. An *et al.* (2009) also studied hypobaric packaging in rigid small containers during storage at 3°C with intermittent fresh-air flushing and vacuum treatment to avoid creating an anoxic environment. Bacterial growth was slightly reduced in fruit in hypobaric packaging, but there was an operational drawback with the 25.3 kPa conditions in that fresh-air flushing and repetitive vacuum application were required too frequently for the densely packed strawberry packages.

The effectiveness of short hypobaric treatments against postharvest rots was investigated by Romanazzi *et al.* (2001). On 'Pajaro', the greatest reductions of *B. cinerea* and *Rhizopus stolonifer* rot were observed on

fruits treated for 4 h at 0.25 and 0.50 atm, respectively. Hashmi *et al.* (2013a) exposed strawberries to hypobaric pressures of 190, 380 and 570 mm Hg for 4 h at 20°C and subsequently stored them at 20°C or 5°C. They found that exposure to 380 mm Hg consistently delayed rot development at both 20°C and 5°C and did not affect their weight loss and firmness, but an initial increase in respiration rate was observed. They also found (Hashmi *et al.*, 2013b) that exposure of strawberries to 380 mm Hg for 4 h reduced rot incidence from natural infection during subsequent storage for 4 days at 20°C. The same treatment had the same effect when the fruit had been inoculated with *B. cinerea* or *R. stolonifer* spores. They found that activities of defence-related enzymes were increased after exposure to hypobaric conditions. PAL and chitinase peaked 12 h after exposure, while POD increased immediately after exposure. PPO activity remained unaffected during subsequent storage for 48 h at 20°C. However, they found that exposure to atmospheric pressure for 4 h did not influence rot development.

### Sweetcorn

Haard and Salunkhe (1975) found that hypobaric storage of sweetcorn could increase storage life to 21 days from only 4–8 days in a conventional cold storage at atmospheric pressure. Burg (2004) stored the cultivar 'Wintergreen' at  $1.7 \pm 1^\circ\text{C}$  at pressures over the range of 10–760 mm Hg and found that the respiration rate decreased with decreasing pressure. After 7 days, a sensory evaluation panel determined that those stored under 20 mm Hg had the highest scores for sweetness and flavour, while after 11 days those that had been stored under 50 mm Hg had the best flavour score.

### Tomato

Storage of mature-green tomatoes under hypobaric conditions of 646, 471, 278 or 102 mm Hg at 12.7°C resulted in a reduction in fruit respiration, especially under 102 mm Hg

(Wu and Salunkhe, 1972). Wu *et al.* (1972) stored tomatoes for 100 days under 102 mm Hg and then transferred them to 646 mm Hg, both at 12.8°C and 90–95% RH, where they ripened normally based on their lycopene, chlorophyll, starch, sugar and  $\beta$ -carotene content. However, their aroma was inferior to those that had been stored under atmospheric pressure. Dong *et al.* (2013) demonstrated that mid-climacteric tomatoes harvested at the breaker or pink stages exposed to 76 mm Hg for 6 h showed transient increased sensitivity to 1-MCP. Subsequently they found that mid-climacteric fruit exposed to 1-MCP at 500 nl/l under 76 or 16 mm Hg for 1 h showed acute disturbance of ripening. They concluded that high efficacy of 1-MCP applied under hypobaric conditions was due to rapid ingress and accumulation of internal gaseous 1-MCP.

### Turnip

Onoda *et al.* (1989) found that turnips retained better appearance and lower weight loss when stored in cycles of hypobaric pressures (between 100 and 300 mm Hg with no humidification) than those stored at a constant atmospheric pressure.

### Watercress

Hypobaric storage of watercress at 3°C and 75 mm Hg for 12 days resulted in them retaining their AA content better than those stored under atmospheric pressure. There was also better retention of protein and chlorophyll content under hypobaric storage (Bangerth, 1974).

# 12

## Recommended CA Conditions

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Recommended storage conditions vary considerably even for the same species. The factors that can affect storage conditions are dealt with in earlier chapters, but the definition of exactly what is meant by maximum or optimum storage period will also vary. So recommendations may be dependent on the subjective judgement of the experimenter or store operator. They usually include the assumption that the crop is in excellent condition before being placed in storage and may also assume that pre-storage treatments such as fungicides and pre-cooling have been applied, that there has been no delay between harvesting the crop and placing it in storage and that the fruit or vegetable had been harvested at an appropriate stage of maturity. In this chapter, recommended conditions are given by species and sometimes by cultivar within a species. It was considered that it could be confusing if the author gave his subjective judgement and so it is for the reader to make their own judgement on the basis of sometimes conflicting recommendations, depending on their own experience and objectives.

### Apple (*Malus domestica*, *M. pumila*)

There are over 7500 known cultivars of apple (Wikipedia, 2018) and each growing

region varies in cultivars grown and their recommended storage conditions. CA storage recommendations for each region are constantly evolving with changes in cultivars, clones of cultivars and new technologies, which means that recommendations over 10 years old could be of limited value to current store managers. Examples of well done regional storage recommendations are those of the UK Department for Environment, Food and Rural Affairs (Defra, 2017; Defra AHDB, 2017), a Swiss publication (Gasser and Siegrist, 2011) and two French publications (Mathieu-Hurtiger *et al.*, 2010, 2014). The following recommendations will refer primarily to static CA storage. DCA storage may be included as well, where available.

The following cultivar discussion is restricted to the most popular cultivars worldwide. In 2015, in terms of worldwide production, the top ten cultivars (including clones) were 'Fuji', 'Delicious', 'Golden Delicious', 'Gala', 'Granny Smith', 'Idared', 'Jonagold', 'Cripps Pink' (Pink Lady®), 'Braeburn' and 'Elstar'. The next ten cultivars were 'McIntosh', 'Honeycrisp' (Honeycrunch®), 'Jonathan', 'Rome Beauty', 'Gloster', 'Empire', 'Orin', 'Reinette du Canada', 'Melrose' and 'Tsugaru' (modified from O'Rourke, 2017). Between 2015 and 2025, amongst other cultivars, only four are expected to increase their proportion of total production:

‘Scifresh’ (Jazz®), ‘Pinova’ (Piñata®, Evelina®), ‘Ambrosia’ and ‘Nicoter’ (Kanzi®) (O’Rourke, 2017). Clearly the CA storage conditions for many more cultivars have been published and reviewed in previous editions of this book (Thompson, 1998, 2010) and elsewhere. In this edition we focus on the most recent literature of the 24 most important cultivars in terms of production. Also individual ‘Delicious’ strains would be removed and covered in the ‘Delicious’ section.

### Ambrosia

‘Ambrosia’ is regarded as a low ethylene producer (DeLong *et al.*, 2016). There is no consistent evidence indicating whether it is CO<sub>2</sub> sensitive or not. DeEll (2012) recommended two CA options:

- For 6–7 months storage, 0°C in 2.5 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>.
- For 6–8 months storage, 0–1°C in 1.7 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>.

Since some research indicates that ‘Ambrosia’ has a tendency to produce low-temperature disorders, such as soft scald and core flush, a more recent recommendation is a storage temperature of 3°C with 2.0–2.5 kPa O<sub>2</sub> and CO<sub>2</sub> (John DeLong, personal communication, 2018).

### Braeburn

‘Braeburn’ is considered to be CO<sub>2</sub> intolerant (Kupferman, 2003). Thus, CO<sub>2</sub> should remain well below the O<sub>2</sub> level at all times and temperatures should be held slightly elevated during CA storage (> 0°C), especially in cooler regions. This is less critical in warmer growing regions (e.g. Brazil, South Africa). Since the application of 1-MCP significantly increases CO<sub>2</sub> intolerance in ‘Braeburn’ fruit, 1-MCP is not recommended.

The recommendation for UK-grown ‘Braeburn’, for storage until March, is 1.5–2.0°C with 1.2–2.0 kPa O<sub>2</sub> and < 1.0 kPa CO<sub>2</sub>, with a delay of 3 weeks before establishing CA (Defra AHDB, 2017). The recommendation

for Switzerland is 1.5°C with 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> (Gasser and Siegrist, 2011). In France, the ‘Braeburn’ recommendation is to store at 0.5–1.0°C with a choice of three CA recommendations: (i) CA at 2–3 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> (March); (ii) ULO at 1.5–1.8 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> (April); or (iii) XLO (extreme low oxygen) at 1 kPa O<sub>2</sub> + ≤ 0.8 kPa CO<sub>2</sub> (April to May) (Mathieu-Hurtiger *et al.*, 2010, 2014). In South Tyrol, Italy, the recommendation is 1.3°C with 1.5 kPa O<sub>2</sub> and 1.0 kPa CO<sub>2</sub> with the O<sub>2</sub> pulldown delayed by 2 weeks (Zanella and Rossi, 2015). The recommendation for ‘Braeburn’ in Washington State is 0–1°C with 1.5 kPa O<sub>2</sub> and 0.5 kPa CO<sub>2</sub>; for New Zealand 0.5°C with 3.0 kPa O<sub>2</sub> and 0.5–1.0 kPa CO<sub>2</sub>; and for South Africa –0.5°C with 1.5 kPa O<sub>2</sub> and 1.5 kPa CO<sub>2</sub> (Kupferman, 2003). In Patagonia, Argentina, the recommendation is 0.5–1°C with 1.2–3.0 kPa O<sub>2</sub> and 0.5–1.2 kPa CO<sub>2</sub> (Benítez, 2001). In Australia, Little *et al.* (2000) recommended step-wise cooling to a final temperature of 1.0°C with 2.5 kPa O<sub>2</sub> and 0.5 kPa CO<sub>2</sub>. In commercial practice in Washington State in 2017, the store temperature was 1.7°C rather than 0–1°C as recommended by Kupferman (2003) (Helgard Ackermann, personal communication). Storage of ‘Braeburn’ at 1°C in DCA, which was set at 0.3–0.4 kPa O<sub>2</sub> + 0.7–0.8 kPa CO<sub>2</sub>, maintained firmness during some 7 months of storage followed by a shelf-life period of 7 days at 20°C. DCA storage reduced ‘Braeburn’ browning disorder (BBD), compared with fruit stored in 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>. The susceptibility to low temperature breakdown (soft scald, +7.6%) and external CO<sub>2</sub> injury (+3.6%) was significantly higher in DCA storage than storage in 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> (Lafer, 2008). Both of these disorders can be controlled by higher storage temperature and Gasser *et al.* (2008) successfully stored ‘Braeburn’ at 3°C for 200 days in either DCA-CF or DCA-RQ without any physiological disorders, while fruit firmness of DCA-stored fruit was, in general, significantly higher than in ULO-stored control fruit. Zanella and Rossi (2015) also reported greater firmness retention in DCA-CF-stored ‘Braeburn’ fruit stored at 1.3°C, compared with the recommended CA (1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>). Since ‘Braeburn’ is prone to greater storage disorders if treated

with 1-MCP, DCA has become a popular storage technology for 'Braeburn'. For example, by 2011–2012, over 40% of the 'Braeburn' apples in South Tyrol, Italy, were stored in DCA-CF (Zanella and Stürz, 2013).

### Cripps Pink (including clones)

'Cripps Pink' is also marketed as 'Pink Lady' or by clone name, e.g. Rosy Glow, Lady in Red. 'Cripps Pink' originated from a cross between 'Lady Williams' and 'Golden Delicious' made by J.E.L. Cripps at the Stoneville Horticultural Research Station, near Perth in Western Australia, in 1973. The aim of the cross was to combine the sweet, superficial-scald-free fruit of 'Golden Delicious' with the firm, long-storing fruit of 'Lady Williams' (Cripps *et al.*, 1993). Unfortunately, superficial scald can appear in 'Cripps Pink' if the fruit are harvested early and stored in air, rather than in CA (Cripps *et al.*, 1993; Kupferman, 2003). 'Cripps Pink' is a firm apple with a very low water permeance, regardless of harvest date (Maguire *et al.*, 1997), which is a factor that may affect its CA storage requirements. 'Cripps Pink' is considered to be CO<sub>2</sub> intolerant (Jobling *et al.*, 2004). Thus, CO<sub>2</sub> should remain well below the oxygen level at all times and temperatures should be held slightly elevated during CA storage (> 0°C), especially in cooler growing regions (James *et al.*, 2010). This is less critical in warmer growing regions, e.g. Australia. In Australia, Cripps *et al.* (1993) stated that the storage life can be 8–9 months or longer if held at 0–1°C in CA (O<sub>2</sub> and CO<sub>2</sub> not specified) and Little *et al.* (2000) recommended that Australian 'Cripps Pink' be stored at 0.5°C with 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> and suggested that, where possible, ethylene should be scrubbed during loading and for the first 10 days. In Argentina, Benítez (2001) recommended 0°C with 1.2–2.5 kPa O<sub>2</sub> + 0.8–1.5 kPa CO<sub>2</sub>. In France, Mathieu-Hurtiger *et al.* (2010, 2014) stated that 'Cripps Pink' is sensitive to internal browning, caused by late harvests, the storage temperature being too low, high CO<sub>2</sub> levels and superficial scald (as a result of early harvesting). They gave three

recommendations for 'Cripps Pink' with increasing storage durations: (i) CA at 1.5–2.5°C with 2–3 kPa O<sub>2</sub> + < 1.5 kPa CO<sub>2</sub>; (ii) ULO at 2.0–2.5°C with 1.5–1.8 kPa O<sub>2</sub> + < 1.0 kPa CO<sub>2</sub>; or (iii) XLO at 2.0–2.5°C with 1.0 kPa O<sub>2</sub> + ≤ 0.8 kPa CO<sub>2</sub>. In South Tyrol in Italy, stores for 'Cripps Pink' are filled at 4.5–5.0°C, the temperature is decreased over 2 weeks to 2.5°C, at a maximum rate of 1.5°C each week, then CA is established at 1.8 kPa O<sub>2</sub> + 1.3 kPa CO<sub>2</sub> (Zanella and Rossi, 2015). A significant quantity of 'Cripps Pink' is stored commercially in DCA-CF in South Tyrol, primarily to achieve some quality benefits, including control of superficial scald (Zanella and Stürz, 2013). Zanella and Rossi (2015) showed that both DCA-CF and 1-MCP improve firmness retention in 'Cripps Pink', compared with CA at 1.8 kPa O<sub>2</sub>.

### Delicious (including clones, also known as strains)

There are several strains of 'Delicious' (also known as 'Red Delicious'), such as Starking, Starkrimson, Starkspur, Ultrared, Oregon Spur, Early Red One, Topred and Classic. Some publications will only identify the 'Delicious' apple by the clone name, omitting the fact that they are strains of 'Delicious'. Kupferman (2003) described 'Delicious' as somewhat CO<sub>2</sub> tolerant and also tolerant of rapid CA. In South Tyrol, Italy, Zanella and Rossi (2015) recommended two CA options, both at 1.3°C: (i) 1.5 kPa O<sub>2</sub> + 1.3 kPa CO<sub>2</sub>; or (ii) 1.0 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>. In Patagonia, Argentina, Benítez (2001) recommended –0.5°C to +0.5°C in 2.0 kPa O<sub>2</sub> + 1.5–2.0 kPa CO<sub>2</sub>. DeEll (2012) recommended two CA options for Ontario, Canada: (i) for 7–9 months of storage, 0°C in 2.5 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>; or (ii) for 8–10 months, 0°C in 0.7–1.5 kPa O<sub>2</sub> + 1.0–1.5 kPa CO<sub>2</sub>. Little *et al.* (2000) recommended that Australian-grown 'Delicious' (all strains) be stored at 0.5°C in 1.8 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>. In France, the 'Delicious' (labelled as 'Rouge Américaines') recommendation is to store at 0.5–1.0°C with a choice of three CA recommendations (Mathieu-Hurtiger *et al.*, 2010, 2014) with

increasing storage durations: (i) CA at 2–3 kPa  $O_2$  + 3.0 kPa  $CO_2$ ; (ii) ULO at 1.5–1.8 kPa  $O_2$  + < 2.0 kPa  $CO_2$ ; or (iii) XLO at 0.8–1.0 kPa  $O_2$  + 0.8 kPa  $CO_2$ .

DeLong *et al.* (2007) found that at 0°C for up to 8 months the HarvestWatch chlorophyll fluorescence system (DCA-CF) retained firmness and quality to a greater extent than 0.9–1.1 kPa  $O_2$  + 1.2–1.3 kPa  $CO_2$ . ‘Delicious’ was one of the first cultivars to be stored commercially in DCA. ‘Delicious’ is prone to superficial scald and DCA controls superficial scald as well as maintaining firmness (DeLong *et al.*, 2004). As a result, by 2011–2012, over 40% of the ‘Delicious’ apples in South Tyrol, Italy, were stored in DCA-CF (Zanella and Stürz, 2013).

### Elstar

‘Elstar’, like ‘Golden Delicious’ (Kupferman, 2003), one of its parents, can be characterized as  $CO_2$  tolerant and also tolerant of rapid CA. Thus,  $CO_2$  can remain above the oxygen level at all times. Kupferman (2003) reported that ‘Elstar’ in Nova Scotia, Canada, can be stored at 0–0.5°C in 2.5 kPa  $O_2$  + 4.5 kPa  $CO_2$  and Netherlands ‘Elstar’ can be held at 1.8°C in 1–1.2 kPa  $O_2$  + 2.5 kPa  $CO_2$ . In Switzerland, there are two options, both at 0.5°C: (i) 2.0 kPa  $O_2$  + 3 kPa  $CO_2$ ; or (ii) 1.0 kPa  $O_2$  + 3.0 kPa  $CO_2$  (Gasser and Siegrist, 2011). In France, the ‘Elstar’ recommendation is to store at 0.5–2.0°C with a choice of two CA recommendations with increasing storage durations (Mathieu-Hurtiger *et al.*, 2010, 2014): (i) CA at 2–3 kPa  $O_2$  + 2–3 kPa  $CO_2$ ; or (ii) ULO at 1.5–1.8 kPa  $O_2$  + 1.5 kPa  $CO_2$ , with insufficient data for an XLO recommendation.

‘Elstar’ grown in some parts of Europe is prone to having skin flecking on the fruit in storage that can be exacerbated by 1-MCP. The influence of DCA-CF storage with step-wise  $O_2$  reduction, in comparison with storage in 1.4 kPa  $O_2$ , was examined by Hennecke *et al.* (2008). The results showed significant improvements in flesh firmness after removal from DCA storage and also post-storage for 3 weeks under cold storage condition compared with 1.4 kPa  $O_2$  storage. The occurrence of skin spots was also significantly

reduced by DCA storage. Veltman *et al.* (2003) used a DCA storage based on ethanol production to establish the lowest possible  $O_2$  concentration under which the fruit could be stored. They found that fruit was firmer, tended to develop less skin spots and had a better colour retention after DCA storage, especially after shelf-life, compared with traditional CA storage. Schouten (1997) compared storage at 1–2°C in either 2.5 kPa  $CO_2$  + 1.2 kPa  $O_2$  or < 0.5 kPa  $CO_2$  + 0.3–0.7 kPa  $O_2$ . The fruit stored in 0.5 kPa  $CO_2$  + 0.3–0.7 kPa  $O_2$  was shown to be of better quality both directly after storage and after a 10-day shelf-life period in air at 18°C. Gasser *et al.* (2008) successfully stored ‘Elstar’ at 3°C for 200 days in either DCA-CF or DCA-RQ without any physiological disorders, while fruit firmness of DCA-stored fruit was similar to ULO-stored control fruit. Köpcke (2009) examined the influence of different  $O_2$  concentrations in CA storage on the skin spots occurrence on ‘Elstar’ apples with and without previous SmartFresh (1-MCP) treatments and concluded that higher  $O_2$  levels and previous SmartFresh treatments increased the disorder, whereas almost no skin spots were detected on fruit stored under DCA-CF. Köpcke (2015) examined the possibility of increasing the store temperature (to save energy) from the recommended 2.0°C in ULO of 1.4 kPa  $O_2$  + 2.6 kPa  $CO_2$  to higher temperatures of either 3.5°C or ≤ 10°C in ULO or DCA-CF (with and without 1-MCP, always with 2.6 kPa  $CO_2$ ). The results showed that increasing the store temperature from 2.0°C to 3.5°C maintained most, but not all, quality measurements when using DCA-CF or 1-MCP alone, but combining DCA-CF and 1-MCP showed no loss in quality.

### Empire

Recent research shows that ‘Empire’ should be considered a chilling and  $CO_2$ -sensitive cultivar. Wang *et al.* (2000c) reported a  $CO_2$ -linked disorder of ‘Empire’ with symptoms resembling superficial scald. Symptoms were effectively controlled by conditioning fruit at 3°C (but not at 0°C) for 3–4 weeks at

1.5–3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> prior to storage in 1.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. Reduction of the disorder was also achieved by storage at 3°C in air, but excessive ripening and associated loss of flesh firmness occurred during subsequent CA storage. Defra AHDB (2017) recommended that UK ‘Empire’ fruit be stored until February at 3–4°C in 2.0 kPa O<sub>2</sub> + < 1.0 kPa CO<sub>2</sub>. DeEll (2012) recommended four CA options for Ontario (Canada) ‘Empire’ fruit, all at 2°C: (i) stored for 5–7 months, 2.5 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>; (ii) for > 7 months, use 1-MCP and store in 2.5 kPa O<sub>2</sub> + < 0.5 kPa CO<sub>2</sub> (for the first 6 weeks) and then 2.0 kPa CO<sub>2</sub>; (iii) for > 7 months, use 1-MCP and DPA (to control CO<sub>2</sub> damage) and store in 2.5 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>; or (iv) for 6–8 months, store in 1.5 kPa O<sub>2</sub> + 1.0–1.5 kPa CO<sub>2</sub>.

### Fuji

‘Fuji’ is a low ethylene producer (Benítez, 2001). It is considered to be CO<sub>2</sub> intolerant (Kupferman, 2003). Thus, CO<sub>2</sub> should remain well below the oxygen level at all times and temperatures should be held slightly elevated during CA storage (> 0°C), especially in cooler growing regions. This is less critical in warmer growing regions (e.g. Brazil). Little *et al.* (2000) recommend that Australian ‘Fuji’ be stored at 0.5°C with 2.5 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>. The recommendation for Washington State ‘Fuji’ is 1°C with 2.0 kPa O<sub>2</sub> + 0.5 kPa CO<sub>2</sub> (Kupferman, 2003). The recommended CA condition for ‘Fuji’ apple in the South Tyrol, Italy, is 1.3°C with 1.5 kPa O<sub>2</sub> + 1.3 kPa CO<sub>2</sub> (Zanella and Rossi, 2015). In Ontario, Canada, the recommendation is 0–1°C with 2.5 kPa O<sub>2</sub> + 1.0–2.0 kPa CO<sub>2</sub> (DeEll, 2012). In Patagonia, Argentina, the recommendation is 0.5–1°C with 1.2–2.5 kPa O<sub>2</sub> + 0.8–1.3 kPa CO<sub>2</sub> (Benítez, 2001). In France, the ‘Fuji’ recommendation is to store at 0.5–1.0°C with a choice of two CA recommendations with increasing storage durations: (i) CA at 2–3 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub>; or (ii) ULO at 1.5–1.8 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>, with no XLO recommendation (Mathieu-Hurtiger *et al.*, 2010, 2014). A significant amount of ‘Fuji’ is stored commercially in DCA-CF in South Tyrol, Italy, primarily to achieve some quality

benefits, including superficial scald control (Zanella and Stürz, 2013). ‘Fuji’ is noted for having good firmness retention, perhaps due to its low ethylene production, and this was confirmed by Zanella and Rossi (2015), who showed that both DCA-CF and 1-MCP did little to improve firmness retention in ‘Fuji’, compared with CA (1.5 kPa O<sub>2</sub>).

### Gala (including clones)

There are numerous ‘Gala’ clones including Mondial Gala, Must Gala, Royal Gala, Galaxy Gala and Ruby Gala that have arisen by bud mutation, some of which have been patented. Kupferman (2003) considered ‘Gala’ (like ‘Golden Delicious’) to be tolerant of low storage temperature, CO<sub>2</sub> and rapid CA. Based on his survey of various recommendations he concluded that ‘Gala’ can be stored as low as 1.0 kPa O<sub>2</sub> with CO<sub>2</sub> levels up to 2.5 kPa at 1°C and that the O<sub>2</sub> should be raised as the temperature is lowered below this point. Defra AHDB (2017) recommended that UK ‘Gala’ be stored at +0.5°C in various combinations of O<sub>2</sub> and CO<sub>2</sub>, depending on storage duration: (i) storage until late December in 2 kPa O<sub>2</sub> + < 1.0 kPa CO<sub>2</sub>; (ii) storage until mid-February in 1.2 kPa O<sub>2</sub> + < 1.0 kPa CO<sub>2</sub>; or (iii) storage until late March in 1 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. Gasser and Siegrist (2011) recommended that Swiss ‘Gala’ be stored in CA at +0.5°C in either 2 kPa O<sub>2</sub> + 2–3 kPa CO<sub>2</sub> or 1 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. In France, the ‘Gala’ recommendation is to store at 0.5–1.0°C with a choice of three CA recommendations with increasing storage durations: (i) CA at 2–3 kPa O<sub>2</sub> + 2–3 kPa CO<sub>2</sub>; (ii) ULO at 1.5–1.8 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub>; or (iii) XLO at 0.8–1.0 kPa O<sub>2</sub> + ≤ 0.8 kPa CO<sub>2</sub> (Mathieu-Hurtiger *et al.*, 2010, 2014). In Australia, Little *et al.* (2000) recommend storage at +0.5°C in 1.8 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>. In Brazil, Weber *et al.* (2011) recommended CA storage conditions for ‘Gala’ of 1°C (rather than 0°C or 0.5°C) in 1.0 kPa O<sub>2</sub> (rather than 0.6 or 0.8 kPa O<sub>2</sub>) and 2.0 kPa CO<sub>2</sub> (rather than 1.0 or 1.5 kPa CO<sub>2</sub>) for up to 7 months. In Argentina, CA-stored ‘Gala’ are sensitive to epidermal spotting at low temperatures; therefore, the pulp temperature is

maintained between +0.5°C and +0.75°C (Benítez, 2001). Even though Argentina ‘Gala’ have shown some sensitivity to CO<sub>2</sub>, its presence is very important due to its fungistatic effect. Some combinations used successfully are as follows: 1.5 kPa O<sub>2</sub> + 0.5–1.0 kPa CO<sub>2</sub>, or 1.2 kPa O<sub>2</sub> + 1.2 kPa CO<sub>2</sub>, or 3 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub>. Some authors propose levels of up to 2.5 kPa CO<sub>2</sub>, provided that they work with higher levels of 1.5 kPa O<sub>2</sub>.

‘Gala’ stores well in DCA storage, and regardless of region and growing season, ‘Gala’ has probably the lowest LOL of any apple cultivar, consistently being lower than 0.4 kPa O<sub>2</sub> and as low as 0.06 kPa, as measured by DCA-CF (e.g. Brackmann *et al.*, 2015; Thewes *et al.*, 2015; Weber *et al.*, 2015; Nock *et al.*, 2016). There are several disorders on ‘Gala’, such as core browning and stem-end browning, that are eliminated or reduced in DCA-CF (Nock *et al.*, 2016, 2018). Zanella and Rossi (2015), in their examination of postharvest firmness retention, found that ‘Gala’ firmness was only slightly increased in DCA-CF, in comparison with ULO – CA of 1.3°C in 1.0 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>. There has been extensive research on DCA-RQ on ‘Royal Gala’ and ‘Galaxy Gala’ in Brazil (Brackmann *et al.*, 2015; Weber *et al.*, 2015; Thewes *et al.*, 2017c, 2018). Weber *et al.* (2015) used RQs of 2, 4 and 6 on ‘Royal Gala’ and concluded that an RQ of 2 was the best. Both Brackmann *et al.* (2015) and Thewes *et al.* (2018) preferred an RQ of 1.5 for ‘Galaxy Gala’, possibly based on the statement in Brackmann *et al.* (2015) that all ‘Gala’ clones store best at RQ = 1.5. However, Thewes *et al.* (2017c) compared RQ = 1.3 and 1.5 on ‘Galaxy Gala’ and concluded that RQ = 1.3 was better than RQ = 1.5.

### Gloster (Gloster 69)

Knee and Tsantili (1988) state that ‘Gloster’ apples are unusual because they do not accumulate ethylene during storage at 2 kPa O<sub>2</sub> (+ 0 kPa CO<sub>2</sub>) at 1.5°C or 3.5°C with continuous ethylene removal. Ethylene production remained low for up to 200 days, and increased on transfer of fruit to 15°C. In spite of this evidence that fruit remained

prelimacteric, some softening and production of soluble pectin and volatile esters occurred at 3.5°C. These processes were suppressed at 1.5°C, but chlorophyll, starch, malate and sucrose losses and increases in glucose and fructose occurred at both temperatures. Hansen (1977) recommended 0–1°C with 3 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub>, Meheriuk (1993) 0.5–2°C in 1–3 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> for 6–8 months and Van Schaik (1985) 1–2°C in air for 4 months, or in scrubbed CA storage at 1–2°C in 0–1 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> for 7½ months with a shelf-life of 20–22 days. Konopacka and Plochanski (2002) reported a positive effect on firmness at 3°C in 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> compared with 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, although this difference decreased during the shelf-life for 14 days at 18°C. Sharples and Stow (1986) recommended 1.5–2°C with < 1 kPa CO<sub>2</sub> and 2 kPa O<sub>2</sub>. Older general recommendations have been summarized by Herregods (1993, personal communication) and Meheriuk (1993) from a variety of countries as follows:

	Temperature °C	kPa CO <sub>2</sub>	kPa O <sub>2</sub>
Belgium	0.8	2	2
Canada	0	4.5	2.5
(Nova Scotia)			
Canada	0	1.5	1.5
(Nova Scotia)			
Denmark	0	2.5–3	3–3.5
France	1–2	1–1.5	1.5
Germany	2	1.7–1.9	1.3–1.5
(Saxony)			
Germany	1–2	2	2
(Westphalia)			
Holland	1	3	1.2
Slovenia	1	3	1
Slovenia	1	3	3
Spain	0.5	2	3
Switzerland	2–4	3–4	2–3

Köpcke (2015) examined the possibility of increasing the store temperature for ‘Gloster’ (to save energy) from the previously recommended 2.0°C in ULO of 1.4 kPa O<sub>2</sub> + 1.8 kPa CO<sub>2</sub> to higher temperatures of either 3.5°C or ≤ 10°C in ULO or DCA-CF (with and without 1-MCP, always with 2.6 kPa CO<sub>2</sub>). The results showed that increasing the store temperature from 2.0°C to 3.5°C



or  $\leq 10^{\circ}\text{C}$  maintained some quality measurements when using DCA-CF or 1-MCP alone, but combining DCA-CF and 1-MCP was more beneficial at the higher temperature. Watercore in ‘Gloster’ dissipated faster with increasing temperature and DCA-CF prevented the transformation of watercore to flesh browning in 1-MCP-treated apples.

### Golden Delicious

Kupferman (2003) considered ‘Golden Delicious’ (like ‘Gala’) to be tolerant of low storage temperature,  $\text{CO}_2$  and rapid CA. Based on his survey of various recommendations, he concluded that ‘Golden Delicious’ can be stored as low as 1.0 kPa  $\text{O}_2$  with  $\text{CO}_2$  levels up to 2.5 kPa at  $1^{\circ}\text{C}$  and the  $\text{O}_2$  should be raised as the temperature is lowered below this point. Defra AHDB (2017) recommended that UK ‘Golden Delicious’ should be stored at  $1.5\text{--}3.5^{\circ}\text{C}$  in 3 kPa  $\text{O}_2$  + 5 kPa  $\text{CO}_2$ . Gasser and Siegrist (2011) recommended that Swiss ‘Golden Delicious’ should be stored in CA at  $2^{\circ}\text{C}$  in either 2 kPa  $\text{O}_2$  + 4 kPa  $\text{CO}_2$  or 1 kPa  $\text{O}_2$  + 3 kPa  $\text{CO}_2$ . In France, the ‘Golden Delicious’ recommendation is to store at  $0.5\text{--}1.0^{\circ}\text{C}$  with a choice of three CA recommendations with increasing storage durations: (i) CA at 2–3 kPa  $\text{O}_2$  + 3–5 kPa  $\text{CO}_2$ ; (ii) ULO at 1.5–1.8 kPa  $\text{O}_2$  + 2–3 kPa  $\text{CO}_2$ ; or (iii) XLO at 0.8–1.0 kPa  $\text{O}_2$  + 0.8–1.0 kPa  $\text{CO}_2$  (Mathieu-Hurtiger *et al.*, 2010, 2014). In Australia, Little *et al.* (2000) recommended storage at  $+0.5^{\circ}\text{C}$  in 1.8 kPa  $\text{O}_2$  + 3.0 kPa  $\text{CO}_2$ . In Argentina, CA-stored ‘Golden Delicious’ are held at  $0\text{--}0.5^{\circ}\text{C}$  in either (i) conventional CA of 2–3 kPa  $\text{O}_2$  + 3–5 kPa  $\text{CO}_2$ , or (ii) 1–1.5 kPa  $\text{O}_2$  + < 3 kPa  $\text{CO}_2$ . At the INTA High Valley Experimental Station they have obtained good results with 1.5 kPa  $\text{O}_2$  + 1.5–2.0 kPa  $\text{CO}_2$  (Benítez, 2001). In Ontario, Canada, the recommendation is  $0^{\circ}\text{C}$  in either (i) conventional CA of 2.5 kPa  $\text{O}_2$  + 2.5 kPa  $\text{CO}_2$ , or (ii) 1.5 kPa  $\text{O}_2$  + 1.5 kPa  $\text{CO}_2$  (DeEll, 2012).

Kitemann *et al.* (2015a) tested the possibility of storing ‘Golden Delicious’ for 7 months using (i) ULO at a high storage temperature,  $5^{\circ}\text{C}$ , in 1 kPa  $\text{O}_2$  + 2.5 kPa  $\text{CO}_2$  + 1-MCP, compared with (ii) ULO at  $1^{\circ}\text{C}$ , in

1 kPa  $\text{O}_2$  + 2.5 kPa  $\text{CO}_2$ , or (iii) DCA at  $1^{\circ}\text{C}$ ,  $\cong 0.7$  kPa  $\text{O}_2$  + 1.5 kPa  $\text{CO}_2$ . In addition, ‘reverse samples’ of 1-MCP-treated apples were stored at  $1^{\circ}\text{C}$  (non 1-MCP-treated ULO room) and apples without 1-MCP treatment were stored at  $5^{\circ}\text{C}$  (1-MCP-treated room). The firmness and rot incidence of ‘Golden Delicious’ did differ between treatments but apples stored at  $5^{\circ}\text{C}$  showed a lower weight loss, compared with those at  $1^{\circ}\text{C}$ .

The value of ‘Golden Delicious’ in some markets is determined by its skin colour and this can be influenced by CA conditions. Little *et al.* (2000) stated that 8.0 kPa  $\text{CO}_2$  can be used for the first 10 days to improve firmness and green colour after storage. Similarly, Mathieu-Hurtiger *et al.* (2014) suggested that in ULO, the  $\text{CO}_2$  concentration can be lower than 2–3 kPa if a more yellow skin is desired. ‘Golden Delicious’ responds well to DCA storage conditions. For example, by 2011–2012, ‘Golden Delicious’ was c. 10% of the DCA-stored apples in the South Tyrol industry (Prange *et al.*, 2013).

### Granny Smith

This variety is considered to be  $\text{CO}_2$  intolerant (Kupferman, 2003). Thus,  $\text{CO}_2$  should remain well below the oxygen level at all times and temperatures should be held slightly elevated during CA storage ( $> 0^{\circ}\text{C}$ ), especially in cooler growing regions. This is less critical in warmer growing regions, e.g. South Africa. Gasser and Siegrist (2011) recommended that Swiss ‘Granny Smith’ should be stored in CA at  $0.5^{\circ}\text{C}$  in 1 kPa  $\text{O}_2$  + 2 kPa  $\text{CO}_2$ . In France, the ‘Granny Smith’ recommendation is to store at  $0.5\text{--}1.0^{\circ}\text{C}$  with a choice of three CA recommendations with increasing storage durations: (i) CA – 2–3 kPa  $\text{O}_2$  + 1.5–2.0 kPa  $\text{CO}_2$ ; (ii) ULO – 1.5–1.8 kPa  $\text{O}_2$  + < 1.2 kPa  $\text{CO}_2$ ; or (iii) XLO – 0.8–1.0 kPa  $\text{O}_2$  + < 0.8 kPa  $\text{CO}_2$  (Mathieu-Hurtiger *et al.*, 2010, 2014). In Australia, Little *et al.* (2000) recommended storage at  $1.0^{\circ}\text{C}$  in 1.8 kPa  $\text{O}_2$  + 1.0 kPa  $\text{CO}_2$ . In Argentina, CA-stored ‘Granny Smith’ are held at  $0\text{--}0.5^{\circ}\text{C}$  in either (i) conventional CA at 2 kPa  $\text{O}_2$  + 1.5–2.0 kPa

CO<sub>2</sub> or (ii) ULO at 0.8–1.2 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub>. Negative temperatures increase the risk of CO<sub>2</sub> damage, especially when using the ULO conditions (Benítez, 2001).

‘Granny Smith’ is very susceptible to superficial scald. Exposure to initial low oxygen stress (ILOS, 0.5 kPa O<sub>2</sub> for 10 days at 1°C) followed immediately by 20 weeks (pre-optimally harvested) or 12 weeks (optimally harvested) of CA at –0.5°C in 1.5 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub> or 1.5 kPa O<sub>2</sub> + 3.0 kPa CO<sub>2</sub> significantly inhibited the development of superficial scald, but only after 8 weeks for pre-optimally and 2 weeks for optimally harvested ‘Granny Smith’ (Van der Merwe *et al.*, 2003). Zanella *et al.* (2005) confirmed that the development of superficial scald on ‘Granny Smith’ was completely controlled by means of dynamic CA (DCA-CF) on a commercial scale during 6 months of storage, even after a subsequent shelf-life period of 7 or 14 days at 20°C. Mditshwa *et al.* (2017) found that intermittent DCA-CF treatment, alternating with RA, provided complete superficial scald control for ‘Granny Smith’ but only for fruit stored for less than 70 days.

### Honeycrisp (Honeycrunch® in EU)

‘Honeycrisp’ is a high-demand, high-value cultivar that will rapidly deteriorate after harvest if its storage requirements are not provided. It is intolerant of rapid CA, low storage temperature and CO<sub>2</sub>. In North American regions there is a difference of opinion on whether ‘Honeycrisp’ should be stored in CA (Beaudry *et al.*, 2009, 2014; Watkins and Nock, 2012b; Nichols *et al.*, 2008). Prior to either RA or CA, ‘Honeycrisp’ has to be pre-conditioned in RA for a period of time at a high temperature (5–7 days at 10–20°C, depending on region) in order to avoid soft scald, internal breakdown disorders and external CO<sub>2</sub> damage (Prange *et al.*, 2003a; DeLong *et al.*, 2004; Beaudry *et al.*, 2009). ‘Honeycrisp’ is a firm apple with a very low water permeance (Schotsmans, 2006) and so a preconditioning treatment generally does not reduce

quality, if limited to 1 week or less. Nova Scotia ‘Honeycrisp’ can be stored in CA at 3–5°C in 2.0–2.5 kPa O<sub>2</sub> + 1.0–1.5 kPa CO<sub>2</sub> (Beaudry *et al.*, 2009; DeLong *et al.*, 2004). In Michigan, the current recommendation is 3.5°C in 3.0 kPa O<sub>2</sub> plus keeping CO<sub>2</sub> ‘low in the first month, in much the same way as for “Empire”, and then allow the CO<sub>2</sub> levels to increase’ (Beaudry *et al.*, 2014). In Washington State, USA, the current recommendation is 3–4°C in 2–4 kPa O<sub>2</sub> + 0.5–1.0 kPa CO<sub>2</sub> (Hanrahan and Blakey, 2017). The Washington State recommendation also states that for lots with bitter pit as a major concern, CA can be initiated during the preconditioning period to decrease the incidence of bitter pit in storage. Nova Scotia ‘Honeycrisp’ fruit have been successfully stored experimentally in DCA-CF without injury for 6 months (DeLong *et al.*, 2004) and 9 months (Nichols *et al.*, 2008).

### Idared

The recommended CA storage temperature is colder in North America, c. 0–3°C, compared with 3–4°C in Europe. ‘Idared’ is somewhat tolerant to CO<sub>2</sub> with the recommended CO<sub>2</sub> equal to O<sub>2</sub> or slightly higher. In Nova Scotia, the recommendation is 0–3°C in 2.0 kPa O<sub>2</sub> + 0.5–1.5 kPa CO<sub>2</sub> and in Michigan the recommendation is 0°C in 1.5 kPa O<sub>2</sub> + < 3.0 kPa CO<sub>2</sub> (Kupferman, 2003). In Ontario, Canada, the recommendation is 0°C in 2.5 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> or in 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> (DeEll, 2012). In the UK, the recommendation is 3–4°C in 3.0 kPa O<sub>2</sub> + 5.0 kPa CO<sub>2</sub> or in 1.0 kPa O<sub>2</sub> + < 1.0 kPa CO<sub>2</sub>. Further south, Gasser and Siegrist (2011) recommended that Swiss-grown ‘Idared’ be stored at 4°C in 2.0 kPa O<sub>2</sub> + 3.0 kPa CO<sub>2</sub>, or in 1.0 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub>. In France, the recommendation is 2–4°C in (i) CA at 2–3 kPa O<sub>2</sub> + 3–4 kPa CO<sub>2</sub>; or (ii) ULO at 1.5–1.8 kPa O<sub>2</sub> + 1.8–2.2 kPa CO<sub>2</sub>.

‘Idared’ has been examined under DCA conditions. Gasser *et al.* (2003), using a laboratory-based DCA-RQ system, measured ACP in ‘Idared’ and found that the amount of CO<sub>2</sub> production depended on the rate of

O<sub>2</sub> reduction, fruit maturity, storage duration and previous storage conditions. After 5–9 months of storage at 4°C, the ACP increased in regular atmosphere (RA) but did not change in low-O<sub>2</sub> CA (1.0 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub>).

Gasser *et al.* (2005) found that the concentration at the ACP of 'Idared' was 0.37 kPa O<sub>2</sub> just after harvest, with a similar level 5 months later, but it decreased to 0.17 kPa O<sub>2</sub> after 8 months of storage. Fruits held in 4 kPa CO<sub>2</sub> and at 1°C showed the same or a slightly higher respiration rate compared with fruit held at 4°C. Levels of organic volatiles increased 2–4 times and ethanol up to 10 times at or below the anaerobic compensation point. Gasser *et al.* (2008) successfully stored 'Idared' at 3°C for 200 days in either DCA-CF or DCA-RQ without any physiological disorders, while fruit firmness of DCA-stored fruit was, in general, significantly higher than in ULO-stored control fruit.

### Jonagold (including clones)

There are many 'Jonagold' clones that are sometimes marketed by clone name, e.g. Jonagored, Red Prince. The storage requirements of 'Jonagold' are very similar to 'Golden Delicious' and 'Gala'. 'Jonagold' can be considered to be tolerant of low storage temperature, CO<sub>2</sub> and rapid CA. In Nova Scotia, Canada, the CA recommendation is 0–0.5°C in 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> and in The Netherlands it is 1.0°C in 1.0–1.2 kPa O<sub>2</sub> + 4.5 kPa CO<sub>2</sub> (Kupferman, 2003). Defra AHDB (2017) recommended that UK 'Jonagold' should be stored at 1.5–2.0°C in 1.2 kPa O<sub>2</sub> + 4.0 kPa CO<sub>2</sub>. Gasser and Siegrist (2011) recommended that Swiss 'Jonagold' should be stored in CA at 2°C in either 2 kPa O<sub>2</sub> + 4 kPa CO<sub>2</sub> or 1 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. In France, the 'Jonagold' recommendation is to store at 0.5–1.0°C with a choice of two CA recommendations with increasing storage durations: (i) CA at 2–3 kPa O<sub>2</sub> + 3–4 kPa CO<sub>2</sub>; or (ii) ULO at 1.5–1.8 kPa O<sub>2</sub> + 1.5–2.0 kPa CO<sub>2</sub> (Mathieu-Hurtiger *et al.*, 2010). In Australia, Little *et al.* (2000) recommended

storage at +0.5°C in 1.8 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>. In Argentina, CA-stored 'Jonagold' are held at 0.8°C in 1.5 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> (Benítez, 2001). In Ontario, Canada, the recommendation is either (i) conventional CA with 0°C in 2.0 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>, or (ii) 0–0.5°C in 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> (DeEll, 2012).

'Jonagold' apples stored in DCA-CF stored for up to 9 months were firmer and did not accumulate fermentative volatiles, compared with fruit stored in CA with 1.5 kPa O<sub>2</sub> (DeLong *et al.*, 2004).

Kitteman *et al.* (2015a) tested the possibility of storing 'Jonagold' for 7 months using (i) ULO at a high storage temperature, 5°C, 1 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> + 1-MCP, compared with (ii) ULO at 1°C, 1 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>, or (iii) DCA at 1°C,  $\cong$  0.7 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub>. In addition, 'reverse samples' of 1-MCP-treated apples were stored at 1°C (non 1-MCP-treated ULO room) and apples without 1-MCP treatment were stored at 5°C (1-MCP-treated room). Firmness had the highest values in 1-MCP-treated fruit, without any difference between the 1°C (reverse) and 5°C treatments. Rot incidence was highest in storage condition 1 (ULO + 1-MCP at 5°C) but weight loss was lowest in both condition 1 (ULO + 1-MCP at 5°C) and 3 (DCA at 1°C), compared with condition 2 (ULO at 1°C). Sensory panels rated taste as being the same in all three conditions but purchase preference was poorest in storage condition 2 (ULO at 1 °C). Sensory texture was rated highest in condition 1 (ULO + 1-MCP at 5°C), followed by condition 3 (DCA at 1 °C) and lowest in condition 2 (ULO at 1°C).

### Jonathan

'Jonathan', a parent of 'Jonagold', is considered to be slightly less tolerant of low storage temperature, and CO<sub>2</sub>, compared with 'Jonagold'. In Michigan, USA, the CA recommendation is 0°C in 1.5 kPa O<sub>2</sub> + <3.0 kPa CO<sub>2</sub> (Kupferman, 2003). Gasser and Siegrist (2011) recommended that Swiss 'Jonathan' should be stored at 3–4°C but

provided no CA recommendation. In Australia, Little *et al.* (2000) recommended storage at +0.5°C in 1.8 kPa O<sub>2</sub> + 1.0 kPa CO<sub>2</sub>. They also said that a CA store of 'Jonathan' fruit should never be sealed unless CA generation can start immediately and to ensure that CO<sub>2</sub> does not exceed 1.0 kPa initially. There have been no studies examining storage of 'Jonathan' under DCA conditions. Older general recommendations have been summarized by Herregods (1993, personal communication) and Meheriuk (1993) from a variety of countries as follows:

	Temperature °C	kPa CO <sub>2</sub>	kPa O <sub>2</sub>
Australia (South)	0	3	3
Australia (South)	0	1	3
Australia (Victoria)	2	1	1.5
China	2	2–3	7–10
Germany (Westphalia)	3–4	3	1–2
Israel	–0.5	5	1–1.5
Slovenia	3	3	3
Slovenia	3	3	1.5
Switzerland	3	3–4	2–3
USA (Michigan)	0	3	1.5

### McIntosh

'McIntosh', discovered in Canada in 1811, is probably the second oldest cultivar, after 'Reinette du Canada', in the list of top 20 cultivars grown in the world. It is mostly cultivated in colder apple regions of Canada, USA and parts of Europe. 'McIntosh' is known to be sensitive to low storage temperatures and not very tolerant to O<sub>2</sub> below 1.0 kPa. 'Marshall McIntosh' is especially sensitive to low O<sub>2</sub> and may develop low-O<sub>2</sub> injury. The CA recommendation is 3–3.5°C in 1.5–2.5 kPa O<sub>2</sub> + 1.5–4.5 kPa CO<sub>2</sub> (non-'Marshall McIntosh' clones) and 2.5 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> ('Marshall McIntosh') (Kupferman, 2003). In Ontario, Canada, the CA recommendation is 3°C in 2.5 kPa O<sub>2</sub> + gradual

increase in CO<sub>2</sub> (2.5 kPa for first 6 weeks and then gradually increase to 4.5 kPa CO<sub>2</sub>) (DeEll, 2012). Since 'McIntosh' is more sensitive to CO<sub>2</sub> if treated with 1-MCP, the first 6 weeks is 0.5 kPa CO<sub>2</sub> and it can be kept at 2.5 kPa for the first 6 weeks if both 1-MCP and DPA (reduces CO<sub>2</sub> damage) are applied (DeEll, 2012). Gasser and Siegrist (2011) recommended Swiss 'McIntosh' be stored at 3–4°C but provided no CA recommendation. 'McIntosh' stores well in DCA-CF (Prange *et al.*, 2003b; DeLong *et al.*, 2004). Older general recommendations are summarized by Herregods (1993, personal communication) and Meheriuk (1993) from a variety of countries as follows:

	Temperature °C	kPa CO <sub>2</sub>	kPa O <sub>2</sub>
Canada (British Columbia)	3	4.5	2.5
Canada (British Columbia)	3	5	2.5
Canada (British Columbia)	1.7	5	2.5
Canada (Nova Scotia)	3	4.5	2.5
Canada (Nova Scotia)	3	1.5	1.5
Canada (Ontario)	3	5	2.5–3
Canada (Ontario)	3	1	1.5
Canada (Quebec)	3	5	2.5
Germany (Westphalia)	3–4	3–5	2
USA (Massachusetts)	1–2	5	3
USA (Michigan)	3	3	1.5
USA (New York)	2	3–5	2–2.5
USA (New York)	2	3–5	4
USA (Pennsylvania)	1.1–1.7	0–4	3–4

### Melrose

The parents of 'Melrose' are 'Jonathan' and 'Delicious' and, like its parents, it is low-temperature and somewhat CO<sub>2</sub> tolerant. Hansen (1977) recommended 0–1°C in 6 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub>. Meheriuk (1993) recommended 0–3°C in 2–5 kPa CO<sub>2</sub> + 1.2–3 kPa O<sub>2</sub> for 5–7 months. At 1°C and 85–90% RH, weight losses and decay were negligible

and there were no physiological disorders in storage in 7 kPa CO<sub>2</sub> with 7 kPa O<sub>2</sub>, while bitter pit was observed in fruit stored in air (Jankovic and Drobnjak, 1994). Jankovic and Drobnjak (1992) showed that storage at 1°C and 85–90% RH in 7 kPa O<sub>2</sub> + < 1 kPa CO<sub>2</sub> for 168 days reduced respiration, delayed ripening and produced firmer, juicier fruit judged to have better taste at the end of storage than those stored in air. CA also reduced the loss of total sugars, total acids, ascorbic acid and starch during storage compared with storage in air. There have been no studies examining storage of ‘Melrose’ under DCA conditions. Older general recommendations have been summarized by Herregods (1993, personal communication) and Meheriuk (1993) from a variety of countries as follows:

	Temperature °C	kPa CO <sub>2</sub>	kPa O <sub>2</sub>
Belgium	2	2	2–2.2
France	0–3	3–5	2–3
Germany (Westphalia)	2–3	3	1–2
Slovenia	1	3	3
Slovenia	1	3	1.2

### Mutsu (‘Crispin’)

The parents of ‘Mutsu’ are ‘Golden Delicious’ and ‘Indo’, which are also the parents of ‘Orin’. ‘Mutsu’, like ‘Golden Delicious’, is low-temperature and CO<sub>2</sub> tolerant. Fidler (1970) recommended 1°C in 8 kPa CO<sub>2</sub> with data supplied only for stores not fitted with a scrubber. Hansen (1977) recommended 0–1°C in 6 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> and Meheriuk (1993) recommended 0–2°C in 1–3 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> for 6–8 months. Dilley *et al.* (1989) recommended 0°C or 3°C in 1.5 kPa O<sub>2</sub> + up to 3 kPa CO<sub>2</sub>. In Ontario, Canada, the recommendation is 0°C with 2.5 kPa O<sub>2</sub> and 2.5 kPa CO<sub>2</sub> (DeEll, 2012). There have been no studies examining storage of ‘Mutsu’ under DCA conditions. Older general recommendation have been summarized by Herregods (1993, personal communication) and Meheriuk (1993) from a variety of countries as follows:

	Temperature °C	kPa CO <sub>2</sub>	kPa O <sub>2</sub>
Australia (Victoria)	1	1	1.5
Australia (Victoria)	0	3	3
Denmark	0–2	3–5	3
Germany (Westphalia)	1–2	3–5	1–2
Japan	0	1	2
Slovenia	1	3	1
USA (Michigan)	0	3	1.5
USA (Pennsylvania)	–0.5 to +0.5	0–2.5	1.3–1.5

### Nicoter (Kanzi®)

‘Nicoter’ is a relatively new cultivar, appearing commercially in 2006. Its parents are ‘Gala’ and ‘Braeburn’ which are also the parents of ‘Scifresh’ (‘Jazz’). Like ‘Braeburn’, ‘Nicoter’ does not seem to be tolerant of low storage temperature and is CO<sub>2</sub> intolerant. Defra AHDB (2017) recommended that UK ‘Kanzi®’ could be stored until late March at 3.0–3.5°C in 2.5 kPa O<sub>2</sub> + < 0.7 kPa CO<sub>2</sub> and sealing (initiation of CA) should be delayed for 21 days. Kitemann *et al.* (2015a) concluded that for ‘Kanzi’ apples in the Lake Constance region of Germany, storage at 3°C in 1.0 kPa O<sub>2</sub> + < 1.0 kPa CO<sub>2</sub> and a CA start immediately after room filling seemed to be suitable to reduce the occurrence of internal browning and to maintain fruit quality during long-term storage. ‘Nicoter’ has been stored experimentally in DCA-CF for many years, beginning in 2007 (Zanella and Stürz, 2015), and is stored commercially in DCA-CF in the South Tyrol, Italy (Prange *et al.*, 2013).

### Orin

The parents of ‘Orin’ are ‘Golden Delicious’ and ‘Indo’, which are also the parents of ‘Mutsu’. In the absence of publicly available information, an acceptable CA recommendation for ‘Orin’ is likely to be very similar to ‘Mutsu’. There have been no studies examining storage of ‘Orin’ under DCA conditions.

### Pinova ('Corail', 'Sonata', Piñata®)

In France, the 'Pinova' recommendation is to store at 0.5–1.0°C with a choice of two CA recommendations with increasing storage durations: (i) CA at 2–3 kPa O<sub>2</sub> + 2–2.5 kPa CO<sub>2</sub>; or (ii) ULO at 1.5–1.8 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> (Mathieu-Hurtiger *et al.*, 2010). Gasser and Siegrist (2011) recommended 2°C in 2 kPa O<sub>2</sub> + 4 kPa CO<sub>2</sub> or in 1 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. Kiteman *et al.* (2015a) tested the possibility of storing 'Pinova' for 7 months using (i) ULO at a high storage temperature, 5°C, 1 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> + 1-MCP, compared with (ii) ULO at 1°C, 1 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>, or (iii) DCA at 1°C,  $\cong$  0.7 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub>. In addition, 'reverse samples' of 1-MCP-treated apples were stored at 1°C (non 1-MCP-treated ULO room) and apples without 1-MCP treatment were stored at 5°C (1-MCP-treated room). The firmness did not differ amongst the various conditions. Rot incidence was lowest in storage condition 1 (ULO at a high storage temperature, 5°C, + 1-MCP) and weight loss was highest in storage condition 2 (ULO at 1°C). Sensory panels rated taste and purchase preference as being the same in all three conditions, but sensory texture was poorest in storage condition 2 (ULO at 1°C), compared with the other two conditions. 'Pinova' has been stored experimentally in DCA-CF (Zanella and Stürz, 2015) and is stored commercially in DCA-CF in the South Tyrol, Italy (Prange *et al.*, 2013).

### Reinette du Canada

'Reinette du Canada', which is known from the early 19th century, is probably the oldest cultivar listed in the top 20 apple cultivars grown worldwide. Gasser and Siegrist (2009) recommended 4°C in 2–3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. Guerra *et al.* (2010) found that 'Reinette du Canada' could be stored for 30 weeks at 1.5°C in 3.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, compared with only 10 weeks in refrigerated air. There have been no studies examining storage of 'Reinette du Canada' under DCA conditions.

### Rome Beauty ('Rome', 'Red Rome')

'Rome Beauty' comes from a seedling discovered in 1817 in Ohio, USA. It is considered to be CO<sub>2</sub> intolerant (Kupferman, 2003). Thus, CO<sub>2</sub> should remain well below the oxygen level at all times and temperatures should be held slightly elevated during CA storage (> 0°C), especially in cooler growing regions. He reported that the CA recommendation for Michigan, USA, was 0°C in 1.5 kPa O<sub>2</sub> + < 3 kPa CO<sub>2</sub>, and in Italy 1–2°C in 2.0 kPa O<sub>2</sub> + 2.0 kPa CO<sub>2</sub>. ASHRAE (1968) recommended –1.1°C to 0°C in 2–3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. Singh *et al.* (1972) used 1.5 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> for 90 days and found that the amino and organic acid contents were considerably increased and the respiration rate was reduced both during and after storage compared with those stored in air. Sugars were not affected. There have been no studies examining storage of 'Rome Beauty' under DCA conditions.

### Scifresh (Jazz®)

'Scifresh' is a relatively new cultivar, appearing commercially in 2004. Its parents are 'Gala' and 'Braeburn', which are also the parents of 'Nicoter' ('Kanzi'). Defra AHDB (2017) recommended, based on experience in New Zealand and France, that Jazz® apples in the UK could be stored until April at 0.5–0.8°C in 2 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> (if manually controlled) or 1.5–1.8 kPa O<sub>2</sub> + 1–1.5 kPa CO<sub>2</sub> (if automatically controlled). In France, Mathieu-Hurtiger *et al.* (2014) made the following comments on 'Scifresh' ('Jazz®'). Storage temperatures should start with 3 weeks at 3°C before dropping the temperature to 1°C. The fruit is sensitive to soft scald during the temperature drop. Their only CA recommendation was ULO (1.5–1.8 kPa O<sub>2</sub> + < 1.5 kPa CO<sub>2</sub>), since it stores very well without XLO. Gasser and Siegrist (2011) recommended 3–3.5°C in either (i) CA at 2.0 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> or (ii) ULO at 1.0 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>. Siegrist and Cotter (2012) conducted further trials that showed that a stepwise decrease in

temperatures from 3.5°C to 1.5°C increased soft scald damage. Cooling the apples at 3.5°C and then storing them at 3°C under CA avoided soft scald without any negative effects on fruit quality. Both CA (2.0 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>) and ULO (1.0 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>) gave good results for 8 months of storage, with ULO being recommended for longer storage. In general, this cultivar offers outstanding high tolerance to storage diseases. ‘Scifresh’ has been stored experimentally in DCA-CF (Zanella and Stürz, 2015).

### Tsugaru

Hong *et al.* (1997) showed that at 0°C fruit firmness retention was better in 1 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> storage than in air. Also CA reduced internal browning, cork spot and decay, mainly caused by *Botrytis cinerea* and *Penicillium expansum*. There have been no studies examining storage of ‘Tsugaru’ under DCA conditions.

### Apricot (*Prunus armeniaca*)

Storage recommendations include: 2–3 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> (SeaLand, 1991) and 2–3 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> at 0–5°C (Kader, 1989; Bishop, 1996) but Kader stated that CA storage was not used commercially. It was reported that CA storage can lead to an increase in internal browning in some cultivars (Hardenburg *et al.*, 1990). Fruits of two cultivars were stored at 0–0.5°C in four different CA with 0, 2.5, 5 or 7.5 kPa CO<sub>2</sub> + 5.0 kPa O<sub>2</sub>. Both the cultivars ICAPI 17 COL, but particularly ICAPI 30 COL, proved suitable for CA storage for up to 3 weeks. Only in the cultivar ICAPI 30 COL did development of some ripening parameters seem to be related to CO<sub>2</sub> concentration (Andrich and Fiorentini, 1986). Folchi *et al.* (1995) stored fruit of the cultivar ‘Reale d’Imola’ at 0°C or 6°C, either in air or in 0.3 kPa O<sub>2</sub> for up to 37 days. Fruit stored in 0.3 kPa O<sub>2</sub> had significantly increased ethanol and acetaldehyde content at 6°C, whereas the methanol

content was not significantly affected by low O<sub>2</sub>. Low O<sub>2</sub> reduced TSS when fruit was stored at 6°C and increased fruit firmness at both storage temperatures. Physiological disorders, such as flesh discoloration and browning, appeared after 5 days of CA storage at 6°C and after 12 days at 0°C. Folchi *et al.* (1994) stored the cultivar ‘Reale d’Imola’ in < 0.1 kPa CO<sub>2</sub> + 0.3 kPa O<sub>2</sub> and showed an increase in fruit aldehyde and ethanol content with increased storage time. Truter *et al.* (1994) showed that at –0.5°C, storage in 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> resulted in a loss of reduced mass compared with those stored in air. Ali Koyuncu *et al.* (2009) stored apricots at 0°C and 90 ± 5% RH and found that those stored in CA retained their quality better than those in MA packages or in air.

Fruit of the cultivars ‘Rouge de Roussillon’ and ‘Canino’ were picked half-ripe or mature and then subjected to pretreatment with high levels of CO<sub>2</sub> (10–30%) for 24, 48 or 72 h before storage in air and were compared with fruits stored directly in air or in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. Underripe fruit exposed to 20 kPa CO<sub>2</sub> for 24 or 48 h, or CA storage, remained firm for longer than those stored in air. CO<sub>2</sub> pretreatment also appeared to reduce the incidence of brown rot caused by *Monilinia* sp. TA decreased continuously during cold storage and after 24 days was the same for CO<sub>2</sub> pretreated and air-stored fruit, but remained high for the CA-stored fruit. Respiration rate of apricots stored in air showed a typical climacteric peak, which was delayed by the CO<sub>2</sub> pretreatments and was completely inhibited by CA storage. The best CO<sub>2</sub> pretreatment was exposure to 20 kPa CO<sub>2</sub> for 24–48 h. Higher concentrations or a longer duration of exposure resulted in fermentation. The benefits of CO<sub>2</sub> pretreatments were most noticeable at the beginning of cold storage but gradually declined, and, once the temperature was increased, they disappeared within about 24 h (Chambroy *et al.*, 1991). For storage of apricots for canning, Van der Merwe (1996) recommended –0.5°C with 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> for up to 2 weeks, while Hardenburg *et al.* (1990) recommended 2.5–3 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> at 0°C for the cultivar ‘Blenheim (Royal)’

which retained their flavour better than air-stored fruit when they were subsequently canned.

### Apricot, Japanese (*Prunus mume*)

Also called mume. Mature-green fruits of the cultivar 'Ohshuku' were stored in 2–3 kPa O<sub>2</sub> + 3, 8, 13 or 18 kPa CO<sub>2</sub> at 20°C and 100% RH. Surface yellowing and flesh softening were delayed at the higher CO<sub>2</sub> levels. Pitting injury occurred in air-stored fruits, but no injury was observed in CA-stored fruits for up to 23 days in 3–13 kPa CO<sub>2</sub>. The results indicated that CA-stored fruits had a shelf-life of 15, 15, 19 and 12 days using 3, 8, 13 and 18 kPa CO<sub>2</sub>, respectively (Kaji *et al.*, 1991). Mature-green fruits of the cultivars 'Gojiro', 'Nankou', 'Hakuoukoume' and 'Shirakaga' were stored for 7 days at 25°C in various CA combinations. Fermentation occurred at O<sub>2</sub> concentrations of up to 2 kPa at 25°C. Fruit held at O<sub>2</sub> concentrations of up to 1 kPa developed browning injury and produced ethyl alcohol at a high rate. The percentage of fruit with water core injury was high in conditions of low O<sub>2</sub> and zero CO<sub>2</sub>. Ethylene production rates were suppressed at high CO<sub>2</sub> concentrations or at O<sub>2</sub> levels of < 5 kPa. They found that storing fruit in approximately 10 kPa CO<sub>2</sub> + 3–4 kPa O<sub>2</sub> could help to retain optimum quality at 25°C, but that such CA storage conditions delayed the respiratory climacteric slightly and increase ethyl alcohol production after the third day of storage (Koyakumar *et al.*, 1995). The cultivar 'Gojiro' was stored for 3 days at 25°C in several different CA conditions. Compared with fruit in ambient air, O<sub>2</sub> uptake and ethylene production decreased when fruit was exposed to 19.8 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>; they decreased even more in 5 or 2 kPa O<sub>2</sub>. However, at 2 kPa O<sub>2</sub> and 0 CO<sub>2</sub>, the percentage of physiologically injured fruits increased considerably in or after storage. They suggested that removing ethylene and maintaining about 8 kPa CO<sub>2</sub> + at least 2 kPa O<sub>2</sub> are important to retain fruit quality (Koyakumar *et al.*, 1994).

### Artichoke, Globe (*Cynara scolymus*)

Typical storage conditions were given as 0–5°C in 2–3 kPa CO<sub>2</sub> + 2–3% O<sub>2</sub> (Bishop, 1996), 0.5–1.5°C in 0–2.5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for 20–30 days (Monzini and Gorini, 1974), 3–5 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> (SeaLand, 1991), 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> in cold storage (unspecified) for 1 month, which reduced browning (Pantastico, 1975), and 0–5°C in 2–3 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub>, but CA storage only had a slight effect on storage life (Saltveit, 1989; Kader, 1992). Kader (1985) had previously recommended 0–5°C in 3–5 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> but indicated that it was not used commercially. Poma Treccarri and Anoni (1969) recommended 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for up to 1 month. They showed that it reduced browning discoloration of the bracts. Miccolis and Saltveit (1988) described storage of large, mature artichokes in humidified air at 7°C for 1 week, 2.5°C for 2 weeks, or 0°C for 3 weeks. All storage conditions resulted in significant quality loss, and there was little or no beneficial effect of 1, 2.5 or 5 kPa O<sub>2</sub> or 2.5 or 5 kPa CO<sub>2</sub> in storage at 0°C. Bracts and receptacles blackened in all atmospheres after 4 weeks of storage. The degree of blackening was insignificant in air, but became severe as the O<sub>2</sub> level decreased over the range from 5% to 1% and CO<sub>2</sub> increased over the range of 2.5–5%. CA storage of large artichokes was therefore not recommended. The cultivar 'Violeta' was stored for 15–28 days at 1°C and 90–95% RH in 1–6 kPa O<sub>2</sub> + 2–8 kPa CO<sub>2</sub>. The best results, for physical, chemical and organoleptic properties, and a post-storage shelf-life of 3–4 days, were obtained with storage for 28 days in 2 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> (Artes-Calero *et al.*, 1981). Gil *et al.* (2003) showed that storage in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> prolonged storage life of 'Blanca de Tudela' for long periods but storage in 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> caused off-odours and CO<sub>2</sub> injuries in the inner bracts and receptacles. They found that respiration rates decreased down to 15–25 ml CO<sub>2</sub>/kg/h in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> and 10–20 ml CO<sub>2</sub>/kg/h in 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> storage, while it was 30–40 ml CO<sub>2</sub>/kg/h in air. Storage trials were carried out by Mencarelli (1987a) at 1 ± 0.5°C and 90–95% RH in a range of



5 and 10 kPa O<sub>2</sub> + 2–6 kPa CO<sub>2</sub> for up to 45 days. Those stored in 5 and 10 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> remained turgid with good organoleptic quality. The addition of 2 kPa CO<sub>2</sub>, however, controlled the development of superficial mould, but 4 kPa CO<sub>2</sub> had adverse effects.

### Artichoke, Jerusalem (*Helianthus tuberosus*)

Depending on cultivar, tubers can be stored for up to 12 months at 0–2°C and 90–95% RH; therefore there is little justification for using CA storage (Steinbauer, 1932). Tubers freeze at –2.2°C but exposure to –5°C causes little damage. Tubers are not sensitive to ethylene (Kays, 1997). They are a major source of inulin and storage of tubers in 22.5 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> significantly retarded the rate of inulin degradation, apparently through an effect on enzyme activity (Denny *et al.*, 1944).

### Asian Pear, Nashi (*Pyrus pyrifolia*, *P. ussuriensis* var. *sinensis*)

Zagory *et al.* (1989) reviewed work on several cultivars and found that storage experiments at 2°C in 1%, 2% or 3 kPa O<sub>2</sub> of the cultivars ‘Early Gold’ and ‘Shinko’ showed no clear benefit compared with storage in air. ‘Early Gold’ fruit appeared to be of reasonably good quality after up to 6 months of storage at 2°C. ‘Shinko’ became prone to internal browning after the third month, perhaps due to CO<sub>2</sub> injury. Low O<sub>2</sub> atmospheres reduced yellow colour development, compared with air. They also reported that the cultivar ‘20th Century’ was shown to be sensitive to CO<sub>2</sub> concentrations above 1% when exposed for longer than 4 months or to 5 kPa CO<sub>2</sub> for more than 1 month. Kader (1989, 2003) recommended 0–5°C in 0–1 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> but claimed that it had limited commercial use. In other work, Meheriuk (1993) gave provisional recommendations of 2°C with 1–5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for about 3–5 months and said that a longer storage period may result in internal

browning. Meheriuk also stated that some cultivars are subject to CO<sub>2</sub> injury but indicated that this is when they are stored in concentrations of over 4%.

### Asparagus (*Asparagus officinalis*)

Storage at 0°C in 12 kPa CO<sub>2</sub> or at 5°C or just above in 7 kPa CO<sub>2</sub> retarded decay and toughening (Hardenburg *et al.*, 1990). Lill and Corrigan (1996) showed that during storage at 20°C, spears stored in 5–10 kPa O<sub>2</sub> + 5–15 kPa CO<sub>2</sub> had an increased postharvest life from 2.6 days in air to about 4.5 days. Other recommendations include: 0°C and 95% RH in 9 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Lawton, 1996); 5–10 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> (SeaLand, 1991); 5°C with a maximum of 10 kPa CO<sub>2</sub> and a minimum of 10 kPa O<sub>2</sub> (Fellows, 1988); 1°C in 10 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> (Monzini and Gorini, 1974); 0–5°C in 5–10 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>, which had a good effect but was of limited commercial use (Kader, 1985, 1992); 1–5°C with 10–14 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>, which had only a slight effect (Saltveit, 1989); and 0–3°C in 10–14 kPa CO<sub>2</sub> (Bishop, 1996). Lipton (1968) showed that levels of 10 kPa CO<sub>2</sub> could be injurious, causing pitting, at storage temperatures of 6.1°C, but at 1.7°C no CO<sub>2</sub> injury was detected. However, spears were injured at both temperatures in atmospheres containing 15 kPa CO<sub>2</sub>. Lipton also showed that storage in 10 kPa CO<sub>2</sub> reduced rots due to *Phytophthora*, but there were no effects in atmospheres containing 5 kPa CO<sub>2</sub>. Other work has also shown that fungal development during storage could be prevented by flushing with 30 kPa CO<sub>2</sub> for 24 h or by maintaining a CO<sub>2</sub> concentration of 5–10% (Andre *et al.*, 1980a). Hardenburg *et al.* (1990) indicated that brief (unspecified) exposure to 20 kPa CO<sub>2</sub> can reduce soft rot at the butt end of spears. McKenzie *et al.* (2004) found that storage in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> maintained the activity of several sugar-metabolizing enzymes at, or above, harvest levels compared with storage in air, where activities fell. CA storage also eliminated the increase in acid invertase activity found during storage in air. There were significant differences in the pool

sizes of sugars in the different compartments (vacuole, cytoplasm, free space). CA storage also resulted in a lower concentration of glucose in the free space, but it appeared to engage the ethanolic fermentation pathway, though only transiently. CA storage may have permitted more controlled use of whichever vacuole sugar the tissue normally drew upon (fructose) and delayed an increase in plasma membrane permeability. Exposing asparagus to total N<sub>2</sub> for 8 h at room temperature resulted in reduced respiration rate and slowing the decrease in headspace O<sub>2</sub> in LDPE MA packaging. In the total N<sub>2</sub> MA packaging, spears lost < 12% fresh weight after 8 days at 10°C, showed less increase in the accumulation of fibre and lignin and reduced loss of chlorophyll, sugars and AA (Techavuthiporn and Boonyariththongchai, 2016).

### Aubergine, Eggplant (*Solanum melongena*)

Storage in 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> was recommended by SeaLand (1991). High CO<sub>2</sub> in the storage atmosphere can cause surface scald browning, pitting and excessive decay and these symptoms are similar to those caused by chilling injury (Wardlaw, 1938). CO<sub>2</sub> concentrations of 5%, 8% or 12% during storage all resulted in CO<sub>2</sub> injury, characterized by external browning without tissue softening (Mencarelli *et al.*, 1989). Viraktamath *et al.* (1963, quoted by Pantastico, 1975) claimed that atmospheres containing CO<sub>2</sub> levels of 7% or higher were injurious.

### Avocado (*Persea americana*)

Fellows (1988) gave a general recommendation of a maximum of 5 kPa CO<sub>2</sub> and a minimum of 3 kPa O<sub>2</sub>. In South Africa, Van der Merwe (1996) recommended 5.5°C with 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for up to 4 weeks. In Australia, CSIRO (1998) recommended 2–5 kPa O<sub>2</sub> + 3–10 kPa CO<sub>2</sub> at 12°C. In other work, 3–10 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> was recommended for both Californian and tropical avocados (SeaLand, 1991). A general

recommendation by Lawton (1996) was 5°C and 85% RH in 9 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. Typical storage conditions were given as 10–13°C in 3–10 kPa CO<sub>2</sub> + 2–5% O<sub>2</sub> by Bishop (1996). A general comment was that storage at 5–13°C in 3–10 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> had a good effect and was of some commercial use (Kader, 1985, 1992, 1993) and that the benefits of reduced O<sub>2</sub> and increased CO<sub>2</sub> were good. He reported that skin browning can occur and off-flavour can develop if the CO<sub>2</sub> level is too high, and internal flesh browning and off-flavour can occur if O<sub>2</sub> levels are too low. In experiments described by Meir *et al.* (1993) at 5°C, increasing CO<sub>2</sub> levels over the range of 0.5%, 1%, 3% or 8% and O<sub>2</sub> levels of 3% or 21% slowed fruit softening, inhibited peel colour change and reduced the incidence of chilling injury. The most effective CO<sub>2</sub> concentration was 8% with either 3 or 21 kPa O<sub>2</sub>. They found that fruit could be stored for 9 weeks in 3 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> and after CA storage the fruit ripened normally and underwent typical peel colour changes. Some injury to the fruit peel was observed, probably attributable to low O<sub>2</sub> concentrations, with peel damage seen in 10% of fruit held in 3 kPa O<sub>2</sub> with either 3 or 8 kPa CO<sub>2</sub>, but it was reported to be too slight to affect their marketability.

Continuous exposure to 20 kPa CO<sub>2</sub> for 7 days at the beginning of storage or for 1 day a week throughout the storage period at either 2°C or 5°C maintained fruit texture and delayed ripening compared with fruit stored throughout in air (Saucedo-Veloz *et al.*, 1991). Storage in total N<sub>2</sub> or total CO<sub>2</sub>, however, caused irreversible injury to the fruit (Stahl and Cain, 1940). CA storage was reported to reduce chilling injury symptoms. There were fewer chilling injury symptoms in the cultivars 'Booth 8', 'Lula' and 'Taylor' in CA storage than in air (Haard and Salunkhe, 1975; Pantastico, 1975). Spalding and Reeder (1975) also showed that in storage at 0 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or 10 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> fruit had less chilling injury and less anthracnose (*Colletotrichum gloeosporioides*) during storage at 7°C than fruit stored in air. Specific recommendations have been made on some cultivars as follows.

### Anaheim

Bleinroth *et al.* (1977) recommended 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> at 7°C, which gave a storage life of 38 days compared with only 12 days in air at the same temperature.

### Fuerte

Both Overholser (1928) and Wardlaw (1938) showed that 'Fuerte' stored at 45°F (7.2°C) had a 2-month storage life in 3 or 4–5 kPa CO<sub>2</sub> + 3 or 4–5 kPa O<sub>2</sub>. This was more than a month longer than they could be stored at the same temperature in air. Biale (1950) reported similar results. Pantastico (1975) showed that storage at 5–6.7°C had double or triple the storage life in 3–5 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> than in air. Bleinroth *et al.* (1977) recommended 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> at 7°C, which gave a storage life of 38 days compared with only 12 days in air at the same temperature. Pre-storage treatment with 3 kPa O<sub>2</sub> + 97% N<sub>2</sub> for 24 h at 17°C significantly reduced chilling injury symptoms during subsequent storage at 2°C for 3 weeks. Fruit softening was also delayed by this treatment. The treated fruit had a lower respiration rate and ethylene production not only during storage at 2°C but also when removed to 17°C (Pesis *et al.*, 1994).

### Gwen

Lizana *et al.* (1993) described storage at 6°C and 90% RH in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>, or 0.03 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> (control) for 35 days followed by 5 days at 6°C in air and 5 days at 18°C to simulate shelf-life. After 35 days all the fruits in CA storage retained their firmness and were in excellent condition, regardless of treatment, while the control fruits were very soft.

### Hass

Yearsley *et al.* (2003) showed that storage at 5°C in 1 kPa O<sub>2</sub> resulted in stress that showed itself in the biosynthesis of ethylene, but

levels rapidly returned to trace levels when fruit was returned to non-stressed atmospheres. No ethanol was detected in fruit after storage for up to 96 h in CO<sub>2</sub> levels of up to 20%. Lizana and Figuero (1997) compared several CA storage conditions and found that 6°C with 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> was optimum. Under these conditions the fruit could be stored for at least 35 days and then kept at the same temperature in air for a further 15 days and was still in good condition. Corrales-Garcia (1997) found that fruit stored at 2°C or 5°C for 30 days had higher chilling injury stored in air than fruit in CA storage in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 15 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. Storage in 15 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> was especially effective in reducing chilling injury. Intermittent exposure to 20 kPa CO<sub>2</sub> increased storage life at 12°C and reduced chilling injury during storage at 4°C compared with those stored in air at the same temperatures (Marcellin and Chaves, 1983). Burdon *et al.* (2008) concluded that fruit quality was better after CA storage than after storage in air, and that DCA storage was better than 'standard' CA storage. The effect of DCA storage was to reduce the incidence of rots when ripe. Inclusion of CO<sub>2</sub> at 5% in CA retarded fruit ripening but stimulated rot expression and they concluded that it should not be used for CA storage of New Zealand-grown 'Hass'. DCA storage prolonged fruit storage life but when removed from storage the fruits ripened more rapidly and more uniformly with fewer rots than those from 'conventional' CA storage (Burdon *et al.*, 2009). They recommended setting the DCA O<sub>2</sub> level as soon as possible after harvest. Fruits were stored for 6 weeks at 5°C in different CA conditions and then ripened in air at 20°C by Burdon *et al.* (2008). Those that had been stored in < 3 kPa O<sub>2</sub> + 0.5 kPa CO<sub>2</sub> ripened in 4.6 days, compared with 7.2 days for those that had been stored in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> and 4.8 days for those that had been stored in air.

### Lula

Stahl and Cain (1940) recommended 3 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>. Storage at 10°C in 3–5 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> delayed fruit softening and in 9 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> maintained acceptable

eating quality and appearance for 60 days (Hardenburg *et al.*, 1990). Earlier work had also shown that 'Lula' could be stored for 60 days at 10°C in 9 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> or for 40 days at 7.2°C in 10 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> (Pantastico, 1975). Spalding and Reeder (1972) showed that 'Lula' could be stored for 8 weeks at 4–7°C and 98–100% RH in 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>.

### Pinkerton

'Pinkerton' is very susceptible to grey pulp. Both CA and MA storage delayed, but did not prevent, this postharvest disorder (Kruger and Truter, 2003).

### Banana (*Musa*)

There are very wide variations in recommendations for optimum CA conditions and 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> at 14°C is used in some commercial shipments. In other cases, higher levels of CO<sub>2</sub> and lower levels of O<sub>2</sub> have shown beneficial effects, but these levels have also been reported to have detrimental effects. Recommended storage conditions for green (pre-climacteric) bananas include the following: 10–13.5 kPa O<sub>2</sub> for 'Latundan' (*Musa* AAB) (Castillo *et al.*, 1967 quoted by Pantastico, 1975); 15°C in 2 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> (Smock *et al.*, 1967); 15°C and 5–8 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for 'Bungulan' (*Musa* AAA) in the Philippines (Calara, 1969; Pantastico, 1975); 5% or lower levels of CO<sub>2</sub> in combination with 2 kPa O<sub>2</sub> (Woodruff, 1969); 20°C in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 182 days for 'Williams' in Australia and the fruit still ripened normally when removed to air (McGlasson and Wills, 1972); 15°C in 2 kPa O<sub>2</sub> + 6–8 kPa CO<sub>2</sub> for 'Dwarf Cavendish' (Pantastico, 1975); 20°C in 1.5–2.5 kPa O<sub>2</sub> + about 7–10 kPa CO<sub>2</sub> (IIR, 1978); 20°C in 1.5–2.5 kPa O<sub>2</sub> + 7–10 kPa CO<sub>2</sub> for 'Williams' (Sandy Trout Food Preservation Laboratory, 1978); 5 kPa CO<sub>2</sub> + 4 kPa O<sub>2</sub> (Hardenburg *et al.*, 1990); O<sub>2</sub> as low as 2% at 15°C if the CO<sub>2</sub> level is around 8% for 'Lakatan' (*Musa* AA) in South-east Asia (Abdullah

and Pantastico, 1990); 2–5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (SeaLand, 1991); 12–16°C in 2–5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (Kader, 1993); 12–16°C in 2–5 kPa CO<sub>2</sub> + 2–5% O<sub>2</sub> (Bishop, 1996); 11.5°C in 7 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> in South Africa (Van der Merwe, 1996); 14–16°C and 95% RH in 2–5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (Lawton, 1996); 14°C in 4 or 6 kPa O<sub>2</sub> + 4 or 6 kPa CO<sub>2</sub> for 'Robusta' from the Windward Islands (Ahmad *et al.*, 2001); 12°C in 3 kPa O<sub>2</sub> + 9 kPa CO<sub>2</sub> or 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 'Prata' (*Musa* AAB) in Brazil for 30 days (Botrel *et al.*, 2004) and 12.5 ± 0.5°C and 98 ± 1.0% RH in 2 kPa O<sub>2</sub> + 4 kPa CO<sub>2</sub> or 3 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> for 'Prata Ana' (*Musa* AAB) (Santos *et al.*, 2006c). They also showed that 'Prata Ana' at colour stage 2 (mature-green) stored in 2 kPa O<sub>2</sub> + 4 kPa CO<sub>2</sub> and 3 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> did not develop to colour stage 7 even after 40 days, whereas those in air reached colour stage 7 in 24 days and those in 4 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> in 32 days.

CO<sub>2</sub> above certain concentrations can be toxic to bananas. Symptoms of CO<sub>2</sub> toxicity are softening of green fruit and an undesirable texture and flavour. Parsons *et al.* (1964) observed that 'Cavendish' (*Musa* AAA) fruit stored in < 1 kPa O<sub>2</sub> failed to ripen normally and developed off-flavour and a dull yellow-to-brown skin with a 'flaky' grey pulp. Also Wilson (1976) showed that fermentation takes place when bananas are ripened at 15.5°C in an atmosphere containing 1 kPa O<sub>2</sub>.

IIR (1978) reported that in experiments with 25 CA storage combinations of 0.5, 1.5, 4.5, 13.5 and 21 kPa O<sub>2</sub> combined with 0, 1, 2, 4 and 8 kPa CO<sub>2</sub>, all at 20°C, the optimum conditions appeared to be 1.5–2.5 kPa O<sub>2</sub> + possibly 7–10 kPa CO<sub>2</sub>. Under these conditions there was an extension of the green life of the fruit of about six times compared with fruit stored in air. They also reported the extension in the postharvest life of bananas at 20°C in CA storage at 0.5 kPa O<sub>2</sub> rather than at 1.5–2.5 kPa O<sub>2</sub>. Abdul Rahman *et al.* (1997) indicated that fermentation occurred in 'Berangan' (*Musa* AAA) at 12°C in 0.5 kPa O<sub>2</sub> but not in 2 kPa O<sub>2</sub>. They also showed that after storage for 6 weeks in both 2 kPa O<sub>2</sub> and 5 kPa O<sub>2</sub> fruits had a

lower respiration rate at 25°C than those that had been stored for the same period in air.

Wills *et al.* (1990) stored 'Cavendish' in a total N<sub>2</sub> atmosphere at 20°C for 3 days soon after harvest. These fruit took about 27 days to ripen in subsequent storage in air, compared with non-treated fruit which ripened in about 19 days. Klieber *et al.* (2002) found that storage in total N<sub>2</sub> at 22°C did not extend storage life compared with storage in air, but resulted in brown discoloration. Parsons *et al.* (1964) found that bananas could be stored satisfactorily for several days at 15.6°C in an atmosphere of 99% N<sub>2</sub> + 1 kPa O<sub>2</sub>. However, they stated that this treatment should only be applied to high-quality fruit without serious skin damage and with short periods of treatment, otherwise fruit failed to develop a full yellow colour when subsequently ripened, even though the flesh softened normally.

After treatment with sulfur dioxide at 2 or 8 µg/kg for 12 h, Williams *et al.* (2003) found that at 15°C bananas could be stored for 4 weeks in air and for 6 weeks in CA. CA conditions before the initiation of ripening had beneficial effects in delaying ripening, with no detrimental effects on subsequent ripening or eating quality of the fruit when ripened in air. Reduced O<sub>2</sub> appeared to be mainly responsible for delaying ripening (Ahmad and Thompson, 2006). The delaying effect of low O<sub>2</sub> on ripening was greater than that of high levels of CO<sub>2</sub>. CA storage produced firmer bananas and extended their shelf-life. 'Robusta' stored at 4 or 6 kPa O<sub>2</sub> + 4 or 6 kPa CO<sub>2</sub> had an extended storage life by 12–16 days beyond that of fruit stored in air and was of good eating quality when ripened (Ahmad *et al.*, 2001).

CA has been shown to reduce postharvest disease. Sarananda and Wilson Wijeratnam (1997) found that fruits stored at 14°C in 1–5 kPa O<sub>2</sub> had lower levels of crown rot than those stored in air. This fungistatic effect continued even during subsequent ripening in air at 25°C. They also found that CO<sub>2</sub> levels of 5% and 10% actually increased rotting levels during storage at 14°C. In contrast, Botrel *et al.* (2004) showed that storage in 3 kPa O<sub>2</sub> + 9 kPa CO<sub>2</sub> improved firmness

and reduced the incidence of anthracnose (*Colletotrichum musae*) better than storage in air or the other CA treatments tested.

CA storage can also affect the shelf-life of fruit after it has begun to ripen. Bananas that had been initiated to ripen by exposure to exogenous ethylene then immediately stored in 1 kPa O<sub>2</sub> at 14°C remained firm and green for 28 days but ripened almost immediately when transferred to air at 21°C (Liu, 1976a, b). Ahmad and Thompson (2006) showed that the marketable life of 'Giant Cavendish' could be extended 2.3–3.8 times, depending on the combination of O<sub>2</sub> + CO<sub>2</sub> used, compared with storage in air. However, they found that there were detrimental effects on fruit quality when 2 kPa O<sub>2</sub> was used and overall 4 kPa O<sub>2</sub> was most effective in extending storage life. CO<sub>2</sub>, in the range tested (4–8%), appeared to have no positive or negative effects on marketable shelf-life or fruit quality. Klieber *et al.* (2003) also found that exposure of 'Williams' to O<sub>2</sub> below 1% at 22°C after ripening initiation induced serious skin injury that increased in severity with prolonged exposure from 6 h to 24 h and also did not extend shelf-life. Fruit of the cultivar 'Sucrier' that had been initiated to ripen with ethylene and then placed in PE bags (0.03 mm) at 20°C showed inhibited ripening and a fermentation flavour (Romphopha *et al.*, 2004). Storage of 'Williams', which had been initiated to ripen, in total N<sub>2</sub> at 22°C had a similar shelf-life to storage in air. However, areas of brown discoloration appeared on bananas placed in N<sub>2</sub> storage (Klieber *et al.*, 2002).

### Basil (*Ocimum basilicum*)

In Thailand, Penchaiya *et al.* (2003) stored sweet basil at 13°C under different CA conditions and found that atmospheres containing 5 or 10 kPa CO<sub>2</sub> were detrimental compared with 0 kPa CO<sub>2</sub> and the optimum CA conditions of the ones they tested were 1.5 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>. However, Sirivatanapa (2006) found that in storage in Thai ambient conditions their shelf-life was 3–5 days, while CA storage at 0–5°C with

1–5% O<sub>2</sub> and 5–15% CO<sub>2</sub> extended their shelf life up to 25 days.

### Bayberry, Chinese (*Myrica rubra*)

Qi *et al.* (2003) found that the optimum conditions for keeping the fruit fresh was 2 ± 1°C and ‘the lower the O<sub>2</sub> concentration, the better the inhibition effects against microorganisms’. CA storage reduced the development of soft and mushy fruit by 67% after 22 days storage at 5°C compared with storage in air. This effect was attributed to a reduction in ethylene production.

### Bean, Runner (*Phaseolus vulgaris*)

Also called green beans, French beans, kidney beans and snap beans. Guo *et al.* (2008) showed that the rate of respiration (peak rate of 109.2 mg CO<sub>2</sub>/kg/h), weight loss, soluble solid and surface colour changes were lower during 12 days of storage at 0°C compared with 8°C or 25°C. Kader (1985) reported that storage at 0–5°C in 5–10 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> had a fair effect on storage but was of limited commercial use. Saltveit (1989) recommended 5–10°C in 4–7 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub>, which was said to have only a slight effect, but, for those destined for processing, 5–10°C in 20–30 kPa CO<sub>2</sub> + 8–10 kPa O<sub>2</sub> was recommended. Storage in 5–10 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> retarded yellowing; also discoloration of the cut end of the beans could be prevented by exposure to 20–30 kPa CO<sub>2</sub> for 24 h (Hardenburg *et al.*, 1990). The same conditions were recommended by SeaLand (1991). High CO<sub>2</sub> levels could result in development of off-flavour, but storage at 45°F (7.2°C) in 5–10 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> retarded yellowing (Anandaswamy and Iyengar, 1961). Snap beans may be held for up to 3 weeks in 8–18 kPa CO<sub>2</sub>, depending on the temperature. CO<sub>2</sub> concentrations of 20% or greater always resulted in severe injury, as loss of tissue integrity followed by decay. At 1°C, 8 kPa CO<sub>2</sub> was the maximum level tolerated, while 18 kPa CO<sub>2</sub> caused injury at 4°C, but not at 8°C. The cultivars

‘Strike’ and ‘Opus’ snap beans were stored for up to 21 days at 1°C, 4°C or 8°C in 2 kPa O<sub>2</sub> + up to 40 kPa CO<sub>2</sub>, then transferred to air at 20°C for up to 4 days (Costa *et al.*, 1994). Sanchez-Mata *et al.* (2003) found that at 8°C, 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> was best in extending shelf-life and preserved the nutritive value compared with storage in air, 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, or 1 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>.

### Beet (*Beta vulgaris*)

Beets can be stored for many months using refrigerated storage at 2°C, but some varieties have a very short marketable life. For example, in Savoy beet, there was a decrease in the β-carotene, AA and chlorophyll contents which was faster in summer, when the shelf-life was 4 days, than in winter, when it was 6 days (Negi and Roy, 2000). SeaLand (1991) reported that unspecified CA storage conditions had only a slight to no effect on beet storage. Shipway (1968) showed that atmospheres containing over 5 kPa CO<sub>2</sub> could damage red beet. Monzini and Gorini (1974) recommended 0°C with 3 kPa CO<sub>2</sub> and 10 kPa O<sub>2</sub> for 1 month.

### Blackberry (*Rubus* spp.)

Berries retained their flavour well for 2 days when they were cooled rapidly and stored in an atmosphere containing up to 40 kPa CO<sub>2</sub> (Hulme, 1971). Kader (1989) recommended 0–5°C in 10–15 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> for optimum storage. Agar *et al.* (1994b) showed that the cultivar ‘Thornfree’ can be stored at 0–2°C in 20–30 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for 6 days and had up to 3 days subsequent shelf-life in air at 20°C. Perkins-Veazie and Collins (2002) found that storage of the cultivars ‘Navaho’ and ‘Arapaho’ at 2°C and 95% RH resulted in reduced decay at 15 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> compared with those stored in air. However, there was some decrease in anthocyanins and some off-flavours were detected after 14 days for the CA-stored fruit. Storage in 20–40 kPa CO<sub>2</sub> could be used to maintain the quality of

machine-harvested blackberries for processing during short-term storage at 20°C (Hardenburg *et al.*, 1990). There was a moderate reduction in vitamin C in blackberries during storage in high CO<sub>2</sub> concentrations (10–30 kPa CO<sub>2</sub>), but reducing O<sub>2</sub> level had little effect (Agar *et al.*, 1997).

### Blackcurrant (*Ribes nigrum*)

Stoll (1972) recommended 2–4°C in 40–50 kPa CO<sub>2</sub> + 5–6 kPa O<sub>2</sub>. After storage at 18.3°C with 40 kPa CO<sub>2</sub> for 5 days, the fruit had a subsequent shelf-life of 2 days with 2–3% of the fruit being unmarketable due to rotting (Wilkinson, 1972). Skrzyński (1990) described experiments where fruit of the cultivar ‘Roodknop’ was held at 6–8°C for 24 h and then transferred to 2°C for storage for 4 weeks in one of the following: 20 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>; 20 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for 14 days, then 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>; 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>; 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>; or in air. The best retention of AA was in both the treatments with 20 kPa CO<sub>2</sub>. In years with favourable weather preceding the harvest, storage in 20 kPa CO<sub>2</sub> completely controlled the occurrence of moulds caused mainly by *Botrytis*, *Mucor* and *Rhizopus* species. Agar *et al.* (1991) stored the cultivar ‘Rosenthal’ at 1°C in 10, 20 or 30 kPa CO<sub>2</sub>, all with 2 kPa O<sub>2</sub>, or a high CO<sub>2</sub> environment of 10, 20 or 30 kPa CO<sub>2</sub>, all with > 15 kPa O<sub>2</sub>. The optimum CO<sub>2</sub> concentration was found to be 20%. Ethanol accumulation was higher in CA storage than in a high CO<sub>2</sub> environment. Fruit could be stored for 3–4 weeks under CA storage or high CO<sub>2</sub>, compared with 1 week for fruit stored at 1°C in air. For juice manufacture, Smith (1957) recommended storage at 2°C in 50 kPa CO<sub>2</sub> for 7 days, followed by a further 3 weeks at 2°C in 25 kPa CO<sub>2</sub>. With longer storage periods there was an accumulation of alcohol and acetaldehyde but the juice quality was not affected.

CA storage can affect the nutrient content of blackcurrants. Harb *et al.* (2008a) found that storage of ‘Titania’ in air for up to 6 weeks did not significantly affect total terpene volatiles, which were similar to those

in freshly harvested berries. They found that decreasing O<sub>2</sub> and increasing CO<sub>2</sub> levels retarded the capacity to synthesize terpenes for 3 weeks, but during storage for an additional 3 weeks it led to a partial recovery. However, storage in 18 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> resulted, in most cases, in a lower biosynthesis of volatile constituents compared with storage in air for 6 weeks. Non-terpene compounds, mainly esters and alcohols, were also increased in fruit during storage in air compared with those in CA storage where there was an initial reduction, but they later recovered. There was a moderate reduction in vitamin C during storage in high CO<sub>2</sub> concentrations (10–30 kPa CO<sub>2</sub>) in blackcurrants but reducing O<sub>2</sub> level had little effect (Agar *et al.*, 1997).

### Blueberry, Bilberry, Whortleberry (*Vaccinium corymbosum*, *V. myrtillus*)

Kader (1989) recommended 0–5°C in 15–20 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> and Ellis (1995) recommended 0.5°C and 90–95% RH in 10 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> for ‘medium term’ storage. Storage for 7–14 days at 2°C in 15 kPa CO<sub>2</sub> delayed decay by 3 days after they had been returned to ambient temperatures, compared with storage in air at the same temperature (Ceponis and Cappellini, 1983). They also showed that storage in 2 kPa O<sub>2</sub> had no added effect over the CO<sub>2</sub> treatment. The optimum atmosphere for the storage of the cultivar ‘Burlington’ was 0°C with 15 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> in which fruit maintained good quality for over 6 weeks (Forney *et al.*, 2003). After 6 weeks of storage at 0°C in 15 kPa O<sub>2</sub> + 25 kPa CO<sub>2</sub>, the concentration of ethanol was about 18 times higher and ethyl acetate was about 25 times higher than when stored in 15 kPa O<sub>2</sub> + 0 or 15 kPa CO<sub>2</sub> (Forney *et al.*, 2003). Harb and Streif (2006) successfully stored the cultivar ‘Bluecrop’ for up to 6 weeks in 12 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>, which retained fruit firmness, had low decay and the absence of off-flavour. ‘Duke’ fruits could be kept in air at 0–1°C in acceptable condition for up to 3 weeks. However, for storage for up to 6 weeks, up to 12 kPa CO<sub>2</sub> + 18–21 kPa O<sub>2</sub> was found to give the

best results. Storage in 6–12 kPa CO<sub>2</sub> maintained their firmness, while those stored in > 12 kPa CO<sub>2</sub> rapidly softened at both 2 kPa O<sub>2</sub> and 18 kPa O<sub>2</sub> and had poorer flavour, firmness and acidity content. Also with increasing CO<sub>2</sub> levels the respiration rate increased in all CO<sub>2</sub> levels to higher respiration rates than in fruit stored in air. Chiabrando and Peano (2006) found that storage at 1°C in 3 kPa O<sub>2</sub> + 11 kPa CO<sub>2</sub> for 60 days maintained fruit firmness, total soluble solids and TA better than storage in air at 3°C and 90–95% RH. DeEll (2002) reported that storage in 15–20 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> reduced the respiration rate and softening, but exposure to < 2 kPa O<sub>2</sub> and/or > 25 kPa CO<sub>2</sub> could cause off-flavours and brown discoloration, depending on the cultivar, duration of exposure and temperature. Zheng *et al.* (2003) showed that the antioxidant levels in ‘Duke’ increased in 60–100 kPa O<sub>2</sub> as compared with 40 kPa O<sub>2</sub> or air during 35 days of storage. O<sub>2</sub> levels of between 60% and 100% also resulted in an increase in total phenolics and total anthocyanins.

Schotsmans *et al.* (2007) found that fungal development was reduced by CA storage. After 28 days at 1.5°C in 2.5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub>, the cultivars ‘Centurion’ and ‘Maru’ had only half as much blemished fruit compared with those stored in air. Incidence of fungal diseases, mainly caused by *Botrytis cinerea*, could be efficiently controlled by CO<sub>2</sub> levels over 6% (Harb and Streif, 2004b). DeEll (2002) reported that 15–20 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> reduced the growth of *B. cinerea* and other decay-causing organisms during transport and storage. Zheng *et al.* (2003) showed that fruit stored in ≥ 60 kPa O<sub>2</sub> had significantly less decay. Song *et al.* (2003) showed that the percentage marketability of the cultivar ‘Highbush’ was 4–7% greater than in the controls after treatment with 200 ppb O<sub>3</sub> for 2 or 4 days in combination with 10 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub>. Rodriguez and Zoffoli (2016) stored ‘Highbush’ blueberries at 0°C in atmospheres of 5/1 kPa, 7/15 kPa or 15/5 kPa CO<sub>2</sub>/O<sub>2</sub> for up to 40 days. Partial pressures of CO<sub>2</sub> higher than 8 kPa or O<sub>2</sub> partial pressures lower than 2 kPa resulted in increased fruit softening,

but this varied between the cultivars they tested. They also found that CO<sub>2</sub> partial pressure higher than 8 kPa or O<sub>2</sub> partial pressure lower than 2 kPa induced fruit softening in the cultivar ‘Brigitta’. Paniagua *et al.* (2014) compared storage of blueberries at 0°C and 4°C in air, 10 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> or 10 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub>. Atmospheres with 10 kPa CO<sub>2</sub> reduced decay incidence, particularly when combined with 2.5 kPa O<sub>2</sub>, though the low O<sub>2</sub> fruit tended to soften more quickly.

After 28 days, Schotsmans *et al.* (2007) showed that storage at 1.5°C in 2.5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> resulted in only half as much blemished fruit compared with storage in air, and fungal development was minimized in one of the two cultivars tested by the CA storage. Cantín *et al.* (2012) tested various levels of CO<sub>2</sub> combined with 3 kPa O<sub>2</sub> on disease control of blueberries. They found that 24 kPa CO<sub>2</sub> increased fruit softening and resulted in off-flavours. They also combined CA storage with SO<sub>2</sub> fumigation and found that SO<sub>2</sub> fumigation followed by storage in 3 kPa O<sub>2</sub> combined with either 6 or 12 kPa CO<sub>2</sub> reduced decay, extended market life and maintained nutritional value.

### Breadfruit (*Artocarpus altilis*, *A. communis*)

The optimum conditions for CA storage were given by Sankat and Maharaj (2007) as 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> at 16°C.

### Broccoli, Sprouting (*Brassica oleracea* var. *italica*)

A considerable amount of research has been published on CA and MA storage. Klieber and Wills (1991) reported that at 0°C and 100% RH broccoli could be stored for about 8 weeks and the storage life could be further extended in < 1 kPa O<sub>2</sub>, but storage in 6 kPa CO<sub>2</sub> resulted in injury after 4–5 weeks. Other recommendations include: 5–10 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (SeaLand, 1991; Jacobsson *et al.*, 2004b); 10 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> (McDonald,



1985; Deschene *et al.*, 1991); 0°C in 2–3 kPa O<sub>2</sub> + 4–6 kPa CO<sub>2</sub> (Ballantyne *et al.*, 1988); 0–5°C in 5–10 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (Kader, 1985, 1992; Saltveit, 1989), the latter claiming that it had a ‘high level of effect’; 0–5°C in 5–10 kPa CO<sub>2</sub> + 1–2% O<sub>2</sub> (Bishop, 1996); a maximum of 15 kPa CO<sub>2</sub> and a minimum of 1 kPa O<sub>2</sub> (Fellows, 1988); 1.8–10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for 3–4 weeks (Gorini, 1988); 0°C or 5°C in 0.5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> or 10°C in 1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> for ‘Marathon’ florets (Izumi *et al.*, 1996a); 15°C in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> (Saijo, 1990); and 3 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> + 95% N<sub>2</sub> at 2°C (Yang *et al.*, 2004). The respiration rate of the cultivar ‘Emperor’ decreased as the O<sub>2</sub> concentration of the atmosphere decreased even down to 0% over a 24 h period at 10°C or 20°C (Praeger and Weichmann, 2001). Kubo *et al.* (1989a, b) reported that 60 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> reduced the respiration rate. However, Makhlof *et al.* (1989b) found that after 6 weeks at 1°C in 10 kPa CO<sub>2</sub> or more, the rate of respiration increased at the same time as the development of undesirable odours and physiological injury.

Generally CA storage reduced the rate of loss of chlorophyll and other phytochemicals. The cultivar ‘Stolto’ was stored for 6 weeks at 1°C in 0 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub>, 6 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> or 15 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub>. Chlorophyll retention was better in CA than in air, mainly due to increased CO<sub>2</sub> concentration (Makhlof *et al.*, 1989a). Yang *et al.* (2004) also showed that losses of chlorophyll and AA were reduced during CA compared with storage in air. The chlorophyll degradation was effectively delayed by storage at 0–1°C in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> or in atmospheres with < 1.5 kPa O<sub>2</sub>. Total carotenoid content remained almost constant during 8 weeks of storage in air or 0–5 kPa CO<sub>2</sub> + 0.8–3.0 kPa O<sub>2</sub>, but increased in storage in 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> (Yang and Henze, 1988). Storage in 5 and 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> were, by visual assessment, the most effective in reducing chlorophyll loss. Most CA combinations had no effect on flavour, but the samples stored at 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> had an undesirable flavour after 8 weeks. CO<sub>2</sub> and O<sub>2</sub> levels had little effect on broccoli firmness (Yang and Henze, 1987).

Colour and texture were not significantly influenced by CA compared with storage at the same temperature in air (Berrang *et al.*, 1990). Storage for 13 weeks in 15 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> resulted in a modest retention of chlorophyll and a 30% reduction of trim loss as compared with air. Storage of the cultivar ‘Green Valiant’ in 8 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>, 6 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or 8 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> reduced chlorophyll loss by 13%, 9% and 32%, respectively, and reduced trim loss by 40%, 45% and 41%, respectively, and the respiration rate as compared with air (McDonald, 1985). When freshly cut heads of the cultivars ‘Commander’ and ‘Green Duke’ were stored in air at 23°C or 10°C, the florets rapidly senesced. Chlorophyll levels declined by 80–90% within 4 days at 23°C and within 10 days at 10°C. Storage at 5°C or 10°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at approximately 80% RH strongly inhibited loss of chlorophyll (Deschene *et al.*, 1991). The effect of storage of the cultivars ‘Marathon’, ‘Montop’ and ‘Lord’ in either 2 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> or 0.5–1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> on antioxidant activity was similar to that for storage in air, but antioxidant activity and the vitamin C content increased in stored broccoli compared with fresh broccoli florets (Wold *et al.*, 2006). Li *et al.* (2016c) compared five CA conditions on storage of broccoli at 10°C. The combinations were: 70 kPa O<sub>2</sub> + 30 kPa CO<sub>2</sub>; 60 kPa O<sub>2</sub> + 40 kPa CO<sub>2</sub>; 50 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub>; 40 kPa O<sub>2</sub> + 60 kPa CO<sub>2</sub>; 30 kPa O<sub>2</sub> + 70 kPa CO<sub>2</sub> at 10°C. They found that the 50 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> combination reduced the metabolism of the broccoli, particularly the respiration rate, and delayed senescence, though ATP levels were relatively higher. Xu *et al.* (2006) investigated the effects of CA storage on changes in the phytochemical content. Glucoraphanin content and induction of quinone reductase (QR) activity (both have been shown to have beneficial effects on human health) in broccoli was affected by temperature and CA conditions. They compared storage in CA with air and showed that there was an increase in the glucoraphanin content and the QR activity over the first 5 days of storage at 5°C, while CA conditions with reduced O<sub>2</sub> concentrations (1 kPa O<sub>2</sub>, 1 kPa O<sub>2</sub> + 10 kPa

CO<sub>2</sub>) led to a steady decrease of glucoraphanin content and QR activity during 20 days of storage at 5°C. The highest content of glucoraphanin and QR activity was found in broccoli stored under 21 kPa O<sub>2</sub>+10 kPa CO<sub>2</sub> at 5°C, which maintained their visual quality, glucoraphanin content and QR activity for 20 days. Eason *et al.* (2007) reported that storage of broccoli for 96 h after harvest at 20°C in 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> resulted in the loss of less water and sugars, had lower protease activity and no significant loss of colour compared with storage in air. They examined differential gene expression in broccoli tissues in response to postharvest CA storage and found a number of novel genes without previously assigned functions with up-regulated and down-regulated expression.

CA storage generally reduced postharvest losses due to disease but results have been mixed. Among the atmospheres tested by Makhlouf *et al.* (1989b), 6 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> was the best for storage for 3 weeks since it delayed the development of soft rot and mould and there was no physiological injury. Compared with storage in air, broccoli stored in atmospheres containing 5–10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> had a small reduction in decay, while storage in 0.8 and 1.5 kPa O<sub>2</sub> resulted in more rapid decay (Yang and Henze, 1987). Storage in 11 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> at 4°C significantly reduced the growth of microorganisms and extended the length of time they were subjectively considered acceptable for consumption (Berrang *et al.*, 1990). The effects of CA storage on the survival and growth of *Aeromonas hydrophila* was examined. Two lots of each were inoculated with *A. hydrophila* 1653 or K144. A third lot served as a non-inoculated control. Following inoculation they were stored at 4°C or 15°C in a CA storage system previously shown to extend their storage life, or in ambient air. Without exception, CA storage lengthened the time that they were considered to be acceptable for consumption. However, CA storage did not significantly affect populations of *A. hydrophila* or *Listeria monocytogenes* (Berrang *et al.*, 1989).

CA has been evaluated on minimally processed broccoli florets. 'Marathon' florets

were held for 4 days at 10°C in containers with an air flow (20.5 kPa O<sub>2</sub> and < 0.5 kPa CO<sub>2</sub>), a restricted air flow (down to 17.2 kPa O<sub>2</sub> and up to 3.7 kPa CO<sub>2</sub>), no flow (down to 1.3 kPa O<sub>2</sub> and up to 30 kPa CO<sub>2</sub>) and N<sub>2</sub> flow (< 0.01 kPa O<sub>2</sub> and < 0.25 kPa CO<sub>2</sub>). Sensory analysis of cooked broccoli indicated a preference for the freshly harvested and air-stored broccoli, with samples stored in no-flow conditions showing the opposite results (Hansen *et al.*, 1993). 'Green Valiant' heads and florets were stored at 4°C. Minimally processing broccoli heads into florets increased the rate of respiration throughout storage at 4°C in air, in response to wounding stress. Ethylene production was also stimulated after 10 days. Atmospheres for optimal preservation of the florets were evaluated using continuous streams of the following defined atmospheres: 0 kPa CO<sub>2</sub> + 20% O<sub>2</sub> (air control); 6 kPa CO<sub>2</sub> + 1% O<sub>2</sub>; 6 kPa CO<sub>2</sub> + 2% O<sub>2</sub>; 6 kPa CO<sub>2</sub> + 3% O<sub>2</sub>; 3 kPa CO<sub>2</sub> + 2% O<sub>2</sub>; and 9 kPa CO<sub>2</sub> + 2% O<sub>2</sub>. The atmosphere consisting of 6 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> resulted in extended storage of broccoli florets from 5 weeks in air to 7 weeks. Prolonged chlorophyll retention and reduced development of mould and offensive odours and better water retention were especially noticeable when the florets were returned from CA at 4°C to air at 20°C. It was concluded that minimal processing had little influence on optimal storage atmosphere, suggesting that recommendations for intact produce are useful guidelines for MA packaging of minimally processed vegetables (Bastrash *et al.*, 1993). Cefola *et al.* (2016) reported that CO<sub>2</sub> levels of 10–15% could be damaging during storage of fresh-cut broccoli and the optimum conditions were 5% O<sub>2</sub> and ≤ 5% CO<sub>2</sub>.

### Brussels Sprouts (*Brassica oleracea* var. *gemmifera*)

High-quality sprouts could be maintained for 10 weeks in storage at 1.5–2.0°C in air, which could be extended to 12 weeks in 5 kPa CO<sub>2</sub> (Peters and Seidel, 1987). Beneficial effects have been reported with storage in atmospheres containing 2.5, 5 and 7 kPa

CO<sub>2</sub> (Pantastico, 1975). SeaLand (1991) recommended 5–7 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub>. Typical storage conditions were given as 0–5°C in 5–7 kPa CO<sub>2</sub> + 1–2% O<sub>2</sub> by Bishop (1996). Storage in 5–7.5 kPa CO<sub>2</sub> + 2.5–5 kPa O<sub>2</sub> helped to maintain quality at 5°C or 10°C, but internal discoloration could occur at 0°C with O<sub>2</sub> levels below 1% (Hardenburg *et al.*, 1990). Kader (1985) and Saltveit (1989) recommended 0–5°C + 5–7 kPa CO<sub>2</sub> and 1–2 kPa O<sub>2</sub> which had a good effect on storage but was of no commercial use. Kader (1989) subsequently recommended 0–5°C in 5–10 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> which he reported to have excellent potential benefit, but limited commercial use. Other recommendations include 0–1°C in 6 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub> or 6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for up to 4 months (Damen, 1985) and 7 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> for 80 days (Niedzielski, 1984). ‘Lunette’, ‘Rampart’ and ‘Valiant’ in storage in 0.5, 1, 2 or 4 kPa O<sub>2</sub> had lower respiration rates relative to those in air, but rates were similar among the four low O<sub>2</sub> levels. Ethylene production was low at 2.5°C and 5°C in all atmospheres, but at 7.5°C it was 20–170% higher in air than in low O<sub>2</sub>. Ethylene production virtually stopped during exposure to 1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> or 20 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, but increased markedly when removed to air storage. Since low O<sub>2</sub> levels retarded yellowing and 10 kPa CO<sub>2</sub> retarded decay development, the combination of low O<sub>2</sub> with high CO<sub>2</sub> effectively extended the storage life of sprouts at 5°C and 7.5°C. The beneficial effect of CA storage was still evident after return of the samples to air. The sprouts retained good appearance for 4 weeks at 2.5°C, whether stored in CA or in air. Storage in 0.5 kPa O<sub>2</sub> occasionally induced a reddish tan discoloration of the heart leaves and frequently an extremely bitter flavour in the non-green portion of the sprouts (Lipton and Mackey, 1987). The cultivar ‘Rampart’ was stored either still on the stems or loose at 2–3°C and < 75% RH, or 1°C and < 95% RH on the stem in air or in 6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. Loose sprouts became severely discoloured at the stalk end after only 1 week in air and after 2 weeks in

CA. Those on the stem remained fresh in CA for 9 weeks at 2–3°C and for 16 weeks at 1°C (Pelleboer, 1983).

### Butter Bean, Lima Bean (*Phaseolus lunatus*)

Storage of fresh beans in CO<sub>2</sub> concentrations of 25–35% inhibited fungal and bacterial growth without adversely affecting their quality (Brooks and McCulloch, 1938).

### Cabbage (*Brassica oleracea* var. *capitata*)

Cabbage is perhaps the most common vegetable to be stored commercially in CA storage conditions. Stoll (1972) recommended 0°C in 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for red and savoy cabbages and 0–3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for white cabbage. Danish cultivars were successfully stored for 5 months at 0°C in 2.5–5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Isenberg and Sayles, 1969). Hardenburg *et al.* (1990) showed that the optimum conditions for CA storage were 2.5–5 kPa CO<sub>2</sub> + 2.5–5 kPa O<sub>2</sub> at 0°C, while SeaLand (1991) recommended 5–7 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for green, red or savoy. Kader (1985) recommended 0–5°C in 5–7 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> or 0–5°C in 3–6 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> (Kader, 1992), which was claimed to have had a good effect and was of some commercial use for long-term storage of certain cultivars. Saltveit (1989) recommended 0–5°C in 3–6 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub>, which had a high level of effect. Typical storage conditions were given as 0°C in 5 kPa CO<sub>2</sub> + 3% O<sub>2</sub> by Bishop (1996) for white cabbage. After storage of several white cabbage cultivars at 0–1°C for 39 weeks, the percentage recovery of marketable cabbage after trimming was 92% in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> and only 70% for those that had been stored in air (Geeson, 1984). Cabbage stored for 159 days in air had a 39% total mass loss and in 3–4 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> for 265 days the total mass loss was only 17%. Cabbages in CA storage showed better retention of green colour,

fresh appearance and texture than those in air (Garipey *et al.*, 1985). The cultivars 'Lennox' and 'Bartolo' were stored in air, 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 2.5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. Disease incidence was lower with both the CA storage conditions and there were no trimming losses. CA also helped to retain green colour in 'Lennox' (Prange and Lidster, 1991). Storage at 0.5–1.5°C and 60–75% RH in 3 kPa O<sub>2</sub> + 4.5–5 kPa CO<sub>2</sub> lengthened the period for which cabbage could be stored by at least 2 months and improved quality compared with storage at 2–7°C in 60–75% RH in air (Zanon and Schragl, 1988).

Storage of the cultivar 'Tip Top' at 5°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> gave the slowest rate of deterioration and at 2.5°C a CO<sub>2</sub> concentration > 2.5% was detrimental. The cultivar 'Treasure Island' had a longer storage life than 'Tip Top' in all CO<sub>2</sub>/O<sub>2</sub> combinations. At 2.5°C in 5 kPa CO<sub>2</sub> + 20.5 kPa O<sub>2</sub>, 97% of 'Treasure Island' was saleable after 120 days (Schouten, 1985). Garipey *et al.* (1984) showed that after storage at 3.5–5 kPa CO<sub>2</sub> + 1.5–3 kPa O<sub>2</sub> for 198 days there was a total mass loss of 14%, compared with 40% for storage in air, and better retention of colour, fresher appearance and firmer texture. The cultivar 'Winter Green' was stored at 1.3°C in 5–6 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> + 92% N<sub>2</sub> and traces of other gases for 32 weeks compared with storage in air at 0.3°C. The average trimming losses were < 10% for the CA-stored cabbage while those in air exceeded 30%; and CA-stored cabbages retained their colour, flavour and texture better (Raghavan *et al.*, 1984). Huang *et al.* (2002) stored the cultivar 'Chu-chiu' at 0–1°C in 2–6 kPa O<sub>2</sub> (mostly at 3%) + 3–5 kPa CO<sub>2</sub> (mostly at 5%) and found that the storage life was doubled to 4 months compared with air storage. The factors that terminated storage were high weight loss and trim loss, inferior colour and freshness, loss of flavour and rooting at the cut end. Two cultivars of savoy cabbage, 'Owasa' and 'Wirosa', were stored at 0–1°C in 4 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> + 93% N<sub>2</sub>, which retained their quality and slowed down the degradation of vitamin C and chlorophyll pigments compared with air storage (Krala *et al.*, 2007).

The need for fresh air ventilation at regular intervals to maintain ethylene concentrations at low levels was emphasized by Meinel *et al.* (1988).

AA concentration increased during storage at 0°C in air. This increase was reduced in an atmosphere of 1 kPa O<sub>2</sub> which was shown also to delay the yellowing of the outer laminae and maintained higher chlorophyll content (Wang and Ji, 1989). Berard (1985) found that storage at 1°C in 2.5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> considerably delayed de-greening, and eliminated abscission and loss of dormancy during the first 122 days of storage, compared with those stored in air.

There is strong evidence that CA storage can reduce some diseases. Storage at –0.5–0°C in 5–8 kPa CO<sub>2</sub> prevented the spread of *Botrytis cinerea* and total storage losses were lower in CA storage than in air (Nuske and Muller, 1984). Pendergrass and Isenberg (1974) also reported less disease, also mainly caused by *B. cinerea*, and better head colour was observed with storage in 5 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> + 92.5% N<sub>2</sub> compared with storage in air at 1°C and 75%, 85% or 100% RH. Berard (1985) described experiments in which 25 cultivars were placed at 1°C and 92% RH. She found that those in 2.5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> usually had reduced or zero grey speck disease and reduced incidence and severity of vein streaking compared with those stored in air for up to 213 days, but not in every case. Black midrib and necrotic spot were both absent at harvest but in comparison with storage in air those stored in 2.5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> had increased incidence of black midrib and it also favoured the development of inner head symptoms on susceptible cultivars. In CA the incidence of necrotic spot in the core of the heads of cultivar 'Quick Green Storage' was increased, which was particularly evident in a season when senescence of cabbage was most rapid. Even though both disorders were initiated in the parenchyma cells, black midrib and necrotic spot had a distinct histological evolution and affected different cultivars under similar conditions of growth and storage (Berard *et al.*, 1986).

### Cactus Pear, Prickly Pear, Tuna, *Opuntia* (*Opuntia ficus indica*, *O. robusta*)

Testoni and Eccher Zerbini (1993) recommended 5°C in 2 or 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> which reduced the incidence of chilling injury and rot development. Cantwell (1995) cited the beneficial effects of lining boxes of fruit with PE film, especially with paper or other absorbent material to absorb condensation.

Besides the fruit, the flattened stems of the prickly pear called *cladodes nopal* and *nopalitos* are eaten in Mexico. Guevara *et al.* (2003) tested the effects of passive and semi-active MA packaging on the postharvest life and quality of *cladodes* stored at 5°C. In semi-active MA packaging, they injected elevated partial pressures of CO<sub>2</sub> (20, 40 or 80 kPa) in the packages immediately after sealing. Passive MA packaging (where no CO<sub>2</sub> was added) had an atmosphere of up to 8.9 kPa O<sub>2</sub> and 7 kPa CO<sub>2</sub> after 35 days of storage at 5°C. Semi-active atmospheres with initial CO<sub>2</sub> pressures of 40 or 80 kPa increased the losses in texture, weight, chlorophyll content, dietary fibre content and colour. Passive MA packaging and semi-active MA packaging with 20 kPa CO<sub>2</sub> significantly decreased the losses in the above-mentioned parameters and also decreased the microbial counts (total aerobic mesophiles, mould and yeasts), but slightly increased the total anaerobic mesophiles counts. The microorganisms identified were *Pseudomonas*, *Leuconostoc*, *Micrococcus*, *Bacillus*, *Ruminococcus*, *Absidia*, *Cladosporium*, *Penicillium* and *Pichia*. Therefore, fresh prickly pear cactus stems could be stored for up to 32 days in MA packaging with ≤ 20 kPa CO<sub>2</sub> without significant losses in quality or any significant increase in microbial counts.

### Capsicum, Sweet Pepper, Bell Pepper (*Capsicum annum* var. *grossum*)

Recommendations include: 8–12°C in 0 kPa CO<sub>2</sub> + 2 or 3–5 kPa O<sub>2</sub>, which had a slight to fair effect but was of limited commercial use (Kader, 1985, 1992; Saltveit,

1989); 0–3 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (SeaLand, 1991); and 8°C and > 97% RH in 2 kPa CO<sub>2</sub> + 4 kPa O<sub>2</sub> (Otma, 1989). Storage with O<sub>2</sub> levels of 3–5% was shown to retard respiration, but high CO<sub>2</sub> could reduce loss of green colour and also result in calyx discoloration (Hardenburg *et al.*, 1990). Storage life of 'California Wonder' at 8.9°C could be extended from 22 days in air to 38 days in 2–8 kPa CO<sub>2</sub> + 4–8 kPa O<sub>2</sub> (Pantastico, 1975). Storage of the cultivar 'Jupiter' for 5 days at 20°C in 1.5 kPa O<sub>2</sub> + 98.5% N<sub>2</sub> resulted in post-storage respiratory rate suppression for about 55 h after transfer to air (Rahman *et al.*, 1995). 'California Wonder' stored in low O<sub>2</sub> had lower internal ethylene contents than those in air and storage in 1 kPa O<sub>2</sub> resulted in significantly lower internal CO<sub>2</sub> than storage in 3, 5, 7 or 21 kPa O<sub>2</sub>. Colour retention was greater in storage in low O<sub>2</sub> atmospheres than in air (Luo and Mikitel, 1996). Rahman *et al.* (1993b) found that there was a residual effect on respiration rate of storage for 24 h at 20°C in 1.5, 5 or 10 kPa O<sub>2</sub>. The residual effect lasted for 24 h when they were transferred to air, with 1.5 kPa O<sub>2</sub> having the most effect. Extending the storage period in 1.5 kPa O<sub>2</sub> to 72 h extended the residual effect from 24 h to 48 h.

Polderdijk *et al.* (1993) studied the interaction of CA and humidity. 'Mazurka' fruits were stored for 15 days at 8°C in atmospheres containing 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or air at 85%, 90%, 95% or 100%. Fruits were then stored for 7 days in air at 20°C and 70% RH and the incidence of an unspecified decay during this period increased relative to the humidity increased during storage. Storage in 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> reduced the incidence of post-storage decay compared with storage in air. Luo and Mikitel (1996) also reported that CA could reduce decay. After 2 weeks of storage at 10°C, 'California Wonder' had 33% decay in those stored in air but only 9% of those stored in 1 kPa O<sub>2</sub> and this reduction in decay continued throughout the 4 weeks of storage. Storage in 3 and 5 kPa O<sub>2</sub> atmospheres slightly reduced decay for a short time, while the incidence of decay in fruits stored in 7 kPa O<sub>2</sub> was not significantly different to those stored in air. Conesa *et al.* (2007)

evaluated several atmospheres on the storage of fresh-cut peppers and found that an atmosphere containing 50–80 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> prevented fermentation and inhibited growth of spoilage microorganisms. Those exposed to 0, 0.5, 1, 3 or 9 kPa O<sub>2</sub> (all with 0 kPa CO<sub>2</sub>), and to 0 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub>, had lower respiration rates than those in the range 20–100 kPa O<sub>2</sub> in both 0 or 20 kPa CO<sub>2</sub>. High O<sub>2</sub> had little effect on respiration rate at 14°C and no effect at 2°C and 7°C in stimulating both CO<sub>2</sub> production and O<sub>2</sub> consumption compared with normal air, but 20 kPa CO<sub>2</sub> in the range 20–100 kPa O<sub>2</sub> increased the respiration rate, probably because physiological injury occurred at 14°C.

### **Carambola, Star Fruit (*Averrhoa carambola*)**

Storage at 7°C and 85–95% RH with either 2.2 kPa O<sub>2</sub> + 8.2 kPa CO<sub>2</sub> or 4.2 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> resulted in low losses of about 1.2% during 1 month and fruit retained a bright yellow colour with good retention of firmness, TSS and acidity compared with fruit stored in air (Renel and Thompson, 1994). Fresh-cut carambola were dipped in AA, citric acid or Ca-EDTA and then stored in various concentrations of O<sub>2</sub> from 0.4 to 20.3 kPa. They found that the combination of 1% AA and storage in 0.4 kPa O<sub>2</sub> reduced browning and 'loss of visual quality' for up to 12 days, which was 3 days longer than those stored in 0.4 kPa O<sub>2</sub> without AA treatment (Teixeira *et al.*, 2008).

### **Carrot (*Daucus carota* subsp. *sativus*)**

CA storage was not recommended by Hardenburg *et al.* (1990), since carrots stored in 5–10 kPa CO<sub>2</sub> + 2.5–6 kPa O<sub>2</sub> had more mould growth and rotting than those stored in air. Increased decay was also reported by Pantastico (1975) in atmospheres of 6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> compared with storage in air. Ethylene or high levels of CO<sub>2</sub> in the storage atmosphere could give the roots a bitter flavour (Fidler, 1963). However, storage in 1–2 kPa O<sub>2</sub> at 2°C for 6 months was previously

reported to have been successful (Platenius *et al.*, 1934) and Fellows (1988) recommended CA storage in a maximum of 4 kPa CO<sub>2</sub> and a minimum of 3 kPa O<sub>2</sub>. Storage at atmospheres containing 1, 2.5, 5 or 10 kPa O<sub>2</sub> inhibited both sprouting and rooting during storage at 0°C, but again was reported to result in increased mould infection. Atmospheres containing 21 or 40 kPa O<sub>2</sub> reduced mould infection, but increased sprouting and rooting (Abdel-Rahman and Isenberg, 1974). Ayhan *et al.* (2008) found that storing minimally processed carrots in 80 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> retained their quality better than in 5% O<sub>2</sub> + 0 kPa CO<sub>2</sub>. Izumi *et al.* (1996b) stored carrot slices, sticks and shreds in air or in 0.5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> at 0°C, 5°C and 10°C. CA storage was beneficial in reducing decay, weight loss and white discoloration on shreds and microbial growth on sticks. White discoloration and microbial growth occurred only at 0°C and 5°C.

### **Cassava, Tapioca, Manioc, Yuca (*Manihot esculenta*)**

These can deteriorate within a day or so of harvesting, due to a physiological disorder called vascular streaking (Thompson and Arango, 1977). The cultivar 'Valencia' was harvested from 12-month-old plants, coated with paraffin wax and stored at 25°C for 3 days in either 54–56% RH or 95–98% RH, with O<sub>2</sub> levels of either 21% (air) or 1% by Aracena *et al.* (1993). They found that storage in 54–56% RH in air resulted in 46% vascular streaking, while storage in 1 kPa O<sub>2</sub> had reduced vascular streaking down to 15%. However, at high humidity, irrespective of the O<sub>2</sub> level, vascular streaking incidence was reduced to only 1.4%. They concluded that the occurrence of vascular streaking was primarily related to water stress in the tissue, while O<sub>2</sub> was secondarily involved.

### **Cauliflower (*Brassica oleracea* var. *botrytis*)**

Respiration rate was decreased in 3 kPa O<sub>2</sub> compared to storage in air (Romo Parada

*et al.*, 1989). Recommended storage conditions include: 0°C with 10 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> for 5 weeks (Wardlaw, 1938); 0°C in 10 kPa CO<sub>2</sub> + 11 kPa O<sub>2</sub>, also for 5 weeks (Smith, 1952); 0°C in 0–3 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> (Stoll, 1972); 5–6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> (Tataru and Dobreanu, 1978); 0°C in 5–10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for 50–70 days (Monzini and Gorini, 1974); 2–5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (SeaLand, 1991); a maximum of 5 kPa CO<sub>2</sub> and a minimum of 2 kPa O<sub>2</sub> (Fellows, 1988); 0–5°C in 2–5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> which had a fair effect but was of no commercial use (Kader, 1985, 1992); and storage at 0°C in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> which reduced weight loss and discoloration but did not affect free sugar or glucosinolate profiles compared with those stored in air at the same temperature for up to 56 days (Hodges *et al.*, 2006). Adamicki (1989) described successful storage of autumn cauliflowers at 1°C in 2.5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for 71–75 days or in the same atmospheres but at 5°C for 45 days. Romo Parada *et al.* (1989) showed that curds stored at 1°C and 100% RH in 3 kPa O<sub>2</sub> + either 2.5 or 5 kPa CO<sub>2</sub> were still acceptable after 7 weeks of storage, while 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> caused softening, yellowing and increased leakage. Curds of the cultivar ‘Primura’ were stored successfully for 6–7 days at 0–1°C and 90–95% RH in circulating air and for 20–25 days in 4–5 kPa CO<sub>2</sub> + 16–27 kPa O<sub>2</sub> (Saray, 1988). Work in Holland showed very little benefit of CA storage, but storage at 0–1°C and > 95% RH with 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> gave a better external appearance but had no effect on curd quality (Mertens and Tranggono, 1989). Both summer and autumn crops stored at 1°C or 5°C in 2.5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> had better leaf colour, curd colour, firmness and market value than in air. Curds of the autumn crop stored in 2 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> had a superior composition to those stored in other CA at both storage temperatures (Adamicki and Elkner, 1985). Kaynaş *et al.* (1994) successfully stored the cultivar ‘Iglo’ in 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for a maximum of 6 weeks with a shelf-life of 3 days at 20°C, which was double their storage life in air.

In other work, CA did not extend storage life and CO<sub>2</sub> levels of 5% or more or 2 kPa O<sub>2</sub> or less injured the curds (Hardenburg *et al.*,

1990). After storage at either 4.4°C or 10°C with levels of 5 kPa CO<sub>2</sub> in the storage atmosphere, some injury was evident after the curds were cooked (Ryall and Lipton, 1972). Mertens and Tranggono (1989) concluded that, in storage at 0–1°C and < 95% RH for 4 or 6 weeks with subsequent shelf-life studies at 10°C and 85% RH, storage in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> had a very small effect, if any, on the respiration rate of the curds. Work on the cultivar ‘Pale Leaf 75’ by Tomkins and Sutherland (1989) with curds stored at 1°C for up to 47 days in air, or in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or 0 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, showed that curds stored in 2 kPa O<sub>2</sub> alone suffered severe, irreversible injury and were discarded after 27 days of storage. After 47 days storage in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> plus the 4-day marketing period, curd quality was acceptable owing to the reduction in incidence of soft rot and black spotting noted in air-stored curds. In air, curds were unsaleable after only 27 days of storage plus the 4-day marketing period. Storage at 4°C in 18 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> for 21 days had no significant effect on the growth of microorganisms compared with storage in air at the same temperature (Berrang *et al.*, 1990). Storage for 8 days at 13°C in air or 15 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> + 64% N<sub>2</sub> accelerated the deterioration of microsomal membranes during storage and caused an early loss in lipid phosphate (Voisine *et al.*, 1993). Menniti and Casalini (2000) stored cauliflowers at 0°C and found that concentrations of 10, 15 or 20 kPa CO<sub>2</sub> in the atmosphere delayed leaf yellowing and rot caused by *Alternaria brassicicola*, but caused injury inside the stem and developed off-flavours and odours after cooking.

### Celeriac, Turnip-rooted Celery (*Apium graveolens* var. *rapaceum*)

CA storage was not recommended, because 5–7 kPa CO<sub>2</sub> + low O<sub>2</sub> increased decay during 5 months of storage (Hardenburg *et al.*, 1990). Pelleboer (1984) also reported that storage in air at 0–1°C gave better results than CA storage. SeaLand (1991) reported that CA storage had a slight to no effect and Saltveit (1989) recommended 0–5°C in 2–3 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub>, which had only a slight effect.

### Celery (*Apium graveolens* var. *dulce*)

Recommended storage conditions include: 2–5 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> (SeaLand, 1991); 0°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, which reduced decay and loss of green colour (Hardenburg *et al.*, 1990); 0–5°C in 0 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> or 0–5°C in 0–5 kPa CO<sub>2</sub> + 1–4 kPa O<sub>2</sub>, which had a fair effect (Kader, 1985, 1992); 0–5°C in 3–5 kPa CO<sub>2</sub> + 1–4 kPa O<sub>2</sub>, which had only a slight effect (Saltveit, 1989); and 3–4 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub>, which retained better texture and crispness than storage in air (Garipey *et al.*, 1985). Storage at 0°C, 4°C or 10°C for 7 days in 25 kPa CO<sub>2</sub> resulted in browning at the base of the petioles, reduced flavour and a tendency for petioles to break away more easily (Wardlaw, 1938). Storage in 1–4 kPa O<sub>2</sub> was shown to preserve the green colour only slightly and although 2.5 kPa CO<sub>2</sub> may be injurious, levels of 9% during 1 month's storage caused no damage (Pantastico, 1975). Reyes (1989) reviewed work on CA storage and concluded that at 0–3°C in 1–4 kPa CO<sub>2</sub> + 1–17.7 kPa O<sub>2</sub>, storage could be prolonged for 7 weeks; and he specifically referred to his recent work which showed that at 0–1°C in 2.5–7.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> market quality could be maintained for 11 weeks. Total weight loss of < 10% over a 10-week period was reported by storing celery in 1 kPa O<sub>2</sub> + 2 or 4 kPa CO<sub>2</sub> at 0°C. Significant increases in marketable celery resulted when ethylene was scrubbed from some atmospheres. It was suggested that improved visual colour, appearance and flavour and increased marketable yield justified the use of 4 kPa CO<sub>2</sub> in storage (Smith and Reyes, 1988). The cultivar 'Utah' stored at 0–1°C in 1.5 kPa O<sub>2</sub> had better marketable quality after 11 weeks than when stored in air. Marketable level was improved by using 2.5–7.5 kPa CO<sub>2</sub> in the storage atmosphere, but not by 2–4 kPa CO<sub>2</sub> (Reyes and Smith, 1987).

CA storage was shown to affect disease development. At 1°C, the growth *in vitro* of *Sclerotinia sclerotiorum* on celery extract agar was most suppressed in an atmosphere containing 7.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub>, but only slightly suppressed in 4 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or in 1.5 kPa O<sub>2</sub> compared with storage in

air. Watery soft rot caused by *S. sclerotiorum* was severe on celery stored in air for 2 weeks at 8°C. A comparable level of severity took 10 weeks to develop at 1°C. At 8°C, suppression of this disease was greatest in atmospheres of 7.5–30 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub>, but only slightly reduced in 4–16 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or in 1.5–6 kPa O<sub>2</sub> alone (Reyes, 1988). A combination of 1 or 2 kPa O<sub>2</sub> + 2 or 4 kPa CO<sub>2</sub> prevented black stem development during storage (Smith and Reyes, 1988). Decay was most severe on celery stored in 21 kPa O<sub>2</sub> compared with CA storage; *Botrytis cinerea* and *S. sclerotiorum* were isolated most frequently from decayed celery (Reyes and Smith, 1987). After storage at 1°C, celery grown in Ontario, Canada, developed a black discoloration of the stalks that appeared outwardly in a striped pattern along the vascular strands. In cross-section the vascular strands were discoloured and appeared blackened. A CA storage atmosphere of 3 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> almost completely eliminated the disorder. Ethylene and pre-storage treatment with sodium hypochlorite had little or no influence on the occurrence of the disorder but there was some indication of difference in cultivar susceptibility (Walsh *et al.*, 1985). Gómez and Artés (2004a) stored celery stalks at 4°C for 35 days in atmospheres containing air or 5 kPa O<sub>2</sub> + 5, 15 or 25 kPa CO<sub>2</sub>. All the CA combinations reduced the respiration rates, which were some 70% of that found in air. Growth of leaves was negatively correlated with CO<sub>2</sub> concentration: the higher the CO<sub>2</sub> level, the lower was the leaf growth. CA also decreased the development of pithiness and improved sensory quality, avoiding the cut butt end from browning and keeping the green colour. Furthermore, decay development, which for air affected 10% of stalks, was inhibited. No off-odours or off-flavours were detected in any treatment. However, under 5 kPa O<sub>2</sub> + 25 kPa CO<sub>2</sub> a slight browning of the internal petioles was observed. After 5 weeks under 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub>, no decay developed and stalks showed the best quality. Storing fresh-cut celery, which had been inoculated with *Listeria monocytogenes*, in 95 kPa O<sub>2</sub> + 5 kPa N<sub>2</sub> proved most beneficial compared with storage in



air, MA packaging and the other atmospheres tested in maximizing both safety and quality retention during 1 week of storage at 7°C (González-Buesa *et al.*, 2014).

### Cherimoya (*Annona cherimola*)

Ludders (2002) observed that the main obstacles to the successful marketing of cherimoyas were their rapid perishability and susceptibility to chilling injury. It was suggested that this could be overcome using CA storage. Storage recommendations include: 8°C in 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (Hatton and Spalding, 1990); 10°C, with a range of 8–15°C, in 5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (Kader, 1993); and 8.5°C and 90% RH in 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for 22 days (De la Plaza *et al.*, 1979). Fruit of the cultivar ‘Fino de Jete’ was stored at 9°C in air, 3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>, 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> or 3 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub>. Low O<sub>2</sub> resulted in the greatest reductions in respiration rate, sugars content and acidity, whereas high CO<sub>2</sub> resulted in the greatest reductions in ethylene production and softening rate. CO<sub>2</sub> at 3% and 6% delayed the softening by 5 and 14 days beyond 3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> and air storage, respectively. This allowed sufficient accumulation of sugars and acids to reach an acceptable quality. The results suggest that 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> and 3 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> atmospheres can extend storage life by 2 weeks longer than storage in air (Alique and Oliveira, 1994). The cultivar ‘Concha Lisa’ was harvested in Chile 240 days after pollination. Respiration rates during storage at 10°C in air showed a typical climacteric pattern with a peak after some 15 days. The climacteric was delayed by storage in 15% or 10 kPa O<sub>2</sub> and fruit kept in 5 kPa O<sub>2</sub> did not show a detectable climacteric rise and did not produce ethylene. All fruit ripened normally after being transferred to air storage at 20°C. However, the time needed to reach an edible condition was affected by O<sub>2</sub> level, with 11 days in 5 kPa O<sub>2</sub>, 6 days in 10 kPa O<sub>2</sub> and 3 days in 20 kPa O<sub>2</sub> (Palma *et al.*, 1993). Berger *et al.* (1993) showed that waxed (unspecified) fruit of the cultivar

‘Bronceada’ could be stored at 10°C, 90% RH in 0 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for 3 weeks without visible change. Fruit of ‘Fino de Jete’ was stored for 4 weeks at 10–12°C in chambers supplied with a continuous flow of CO<sub>2</sub>. The CO<sub>2</sub> treatment prolonged the storage life of cherimoyas by at least 3 weeks compared with those stored in air (Martinez-Cayuela *et al.*, 1986). Escribano *et al.* (1997) found that pre-treatment of the cultivar ‘Fino de Jete’ for 3 days at 6°C in 20 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> maintained fruit firmness and colour compared with those not treated. They also reported that there were some interactions between cultivar and CA storage treatment. Not all reports on CA storage have been positive. De la Plaza (1980) and Moreno and De la Plaza (1983) showed that fruit of the cultivars ‘Fino de Jete’ and ‘Campa’ stored in 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> had a higher respiration rate and deteriorated more quickly than fruit stored at the same temperature in air.

### Cherry, Sour (*Prunus cerasus*)

SeaLand (1991) recommended 10–12 kPa CO<sub>2</sub> + 3–10 kPa O<sub>2</sub> and Ionescu *et al.* (1978) recommended 0°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> but for up to 20 days, which resulted in about 7% loss. English Morello cherries were stored at 2°C in 25 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>, 15 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> or air for 20 days by Wang and Vestheim (2002). They found that for the cultivar ‘Crisana (Paddy)’, decay was greatly reduced in 25 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>, which also gave the best colour and retention of TA and TSS.

### Cherry, Sweet (*Prunus avium*)

Storage recommendations include: 20–25 kPa CO<sub>2</sub> or 0.5–2 kPa O<sub>2</sub>, which helped to retain firmness, green stems and bright fruit colour (Hardenburg *et al.*, 1990); 0–5°C and 95% RH in 10–15 kPa CO<sub>2</sub> + 3–10 kPa O<sub>2</sub> (Lawton, 1996); 20–25 kPa CO<sub>2</sub> + 10–20 kPa O<sub>2</sub> (SeaLand, 1991); 0–5°C in 10–15 kPa CO<sub>2</sub>

+ 3–10 kPa O<sub>2</sub> (Bishop, 1996); 0–5°C in 10–12 kPa CO<sub>2</sub> + 3–10 kPa O<sub>2</sub> (Kader, 1985); 0–5°C in 10–15 kPa CO<sub>2</sub> + 3–10 kPa O<sub>2</sub> (Kader, 1989); 0°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for 30 days with 9% losses for cultivars ‘Hedelfingen’ and ‘Germersdorf’ (Ionescu *et al.*, 1978); 1°C + 95% RH in 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> 21 days plus 2 days at 20°C to simulate shelf-life (Luchsinger *et al.*, 2005); 0 ± 0.5°C in 20 or 25 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for the cultivar ‘0900 Ziraat’ for up to 60 days (Akbulduk *et al.*, 2008); and 1°C and 95% RH in 2 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> recommended for ‘Sweetheart’ for up to 6 weeks, which retained the anthocyanin content and the PPO activity was at its lowest (Remón *et al.*, 2003). At 0°C, Wang and Long (2014) found that respiration rate of cherries declined logarithmically from about 10 kPa to about 1 kPa but respiration rate from 21 kPa to about 10 kPa was affected very little. They also showed that storage atmospheres of 1.8–14.4 kPa O<sub>2</sub> + 5.7–12.9 kPa CO<sub>2</sub>, generated by commercial MA packaging, delayed fruit softening. Stow *et al.* (2004) found that there were no consistent effects of the 16 combinations of 0, 5, 10 and 20 kPa CO<sub>2</sub> with 1, 2, 4 and 21 kPa O<sub>2</sub> and, overall, CA storage was not superior to air storage for the cultivars ‘Stella’, ‘German Late’, ‘Colney’, ‘Pointed Black’ and ‘Lapins’. It was concluded that the maximum storage life could be obtained if the fruit had been cooled to 1°C within 36 h of harvest and thereafter maintained at 0°C in air. Rotting was the major cause of loss during storage at 0°C and especially during subsequent shelf-life at 10°C or 20°C. Shellie *et al.* (2001) had previously shown similar results in experiments with the cultivar ‘Bing’ stored for 14 days at 1°C in 6 kPa O<sub>2</sub> + 17 kPa CO<sub>2</sub> + 82% N<sub>2</sub>. They had similar market quality as cherries stored in air. In spite of these findings, there is considerable evidence that both CA and MA storage can be beneficial.

For the cultivars ‘Napoleon’, ‘Stella’ and ‘Karabodur’, Eris *et al.* (1994) found that the optimum conditions were at 0°C and 90–95% RH in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. The conditions they tested were 0 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, 20 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> and 0 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. Chen *et al.* (1981) compared

a range of CA storage treatments on the cultivar ‘Bing’ at –1.1°C for 35 days and found that 0.03 kPa CO<sub>2</sub> + 0.5–2 kPa O<sub>2</sub> maintained the greenness of the stems, brighter fruit colour and higher acidity than other treatments. Storage in 10 kPa CO<sub>2</sub> had similar effects, with the exception of maintaining the stem greenness. Folchi *et al.* (1994) stored the cultivar ‘Nero 1’ in < 0.1 kPa CO<sub>2</sub> + 0.3 kPa O<sub>2</sub> and showed an increase in aldehyde and ethanol content of fruit with increased storage time. In a comparison between atmospheres of 4 and 20 kPa O<sub>2</sub> + 5 or 12 kPa CO<sub>2</sub>, the best results for up to 12 days for the cultivar ‘Burlat’ were 12 kPa CO<sub>2</sub>, independently of O<sub>2</sub> concentration. In these conditions there was a higher acidity level, lower anthocyanin content and lower levels of peroxidase and polyphenoloxidase activities (Remón *et al.*, 2004). The cultivars ‘Star’, ‘Kordia’ and ‘Regina’ were stored at 1°C with 90–93% RH for up to 3 weeks in 10 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>, 15 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> or 20 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> followed by storage in air for 1 week at 20°C with 60% RH. CA storage decreased the respiration rate, compared with air, and improved stem colour retention and condition, but there was little difference between CA and air storage on sugar and acid levels (Gasser *et al.*, 2004). ‘Lapins’ cherries stored in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> were firmer and had higher vitamin C and TA than those in MA packaging or CA with higher O<sub>2</sub> levels, but TSS contents were not significantly affected by CA (Tian *et al.*, 2004). Storage at 1°C with 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> inhibited the enzymatic activities of PPO and peroxidase, reduced MDA content, effectively prevented flesh browning, decreased fruit decay and extended storage life more than storage in air, MA or 70 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>. Storage in 70 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> was more effective at inhibiting ethanol production in flesh and reducing decay than other treatments, but showed increased fruit browning after 40 days of storage. ‘Sweetheart’ cherries were stored for 6 weeks at 1°C and 95% RH with 2 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> while maintaining an excellent quality throughout their storage and shelf-life of 3 days at 20°C. Under these conditions they had the

highest acceptability and appearance score, anthocyanin content remained unchanged and PPO activity was at its lowest level (Remón *et al.*, 2003).

CA has been shown to reduce disease and Haard and Salunkhe (1975) reported that storage with CO<sub>2</sub> levels of up to 30% reduced decay. Tian *et al.* (2004) also reported that storage of the cultivar ‘Lapins’ at 1°C in 70 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> was effective in reducing decay, but stimulated fruit browning after 40 days. Storage in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> and 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub>, with N<sub>2</sub> the balance, inhibited grey mould (*Botrytis cinerea*) development after 18 days at 1°C plus 6 days at ambient (Wermund and Lazar, 2003). Serradilla *et al.* (2013) compared storage of cultivar ‘Ambrunés’ in air, 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> and 8 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>. They found that both 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> and 8 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> effectively controlled the growth of mesophilic aerobic bacteria, psychrotrophs, *Pseudomonas* spp., yeasts and the fungi *Aureo basidium* spp., *Penicillium* spp., *Leuconostoc* spp., and *Rahnella* spp.

CA has been used for controlling insects for phytosanitary purposes. The treatment times required to completely kill specific insects by O<sub>2</sub> levels at or below 1 kPa suggest that low O<sub>2</sub> atmospheres are potentially useful as postharvest quarantine treatments for some fruit. Fruit of the cultivar ‘Bing’ was treated with 0.25 kPa or 0.02 kPa O<sub>2</sub> (balance N<sub>2</sub>) at 0°C, 5°C or 10°C to study the effects of these insecticidal low-O<sub>2</sub> atmospheres on fruit postharvest physiology and quality attributes. Development of alcoholic off-flavour, associated with ethanol accumulation, was the most common and important detrimental effect that limited fruit tolerance to low O<sub>2</sub> (Ke and Kader, 1992b).

### Chestnut, Chinese (*Castanea mollissima*)

Respiration rate was reduced to a steady low level at 0–5°C but was not further reduced by decreasing O<sub>2</sub> or increasing CO<sub>2</sub> in the atmosphere (Wang *et al.*, 2000b). The decay rate when exposed to 40 kPa CO<sub>2</sub> for

20 days was only 1% after subsequent cold storage for 120 days, but an off-flavour was detected when treated for more than 20 days. Also there were no adverse effects of any CO<sub>2</sub> concentrations tested if the treatment duration was not more than 10 days. The presence of an off-flavour was irreversible when the CO<sub>2</sub> concentration was > 50 kPa and treatment duration was longer than 20 days (Liang *et al.*, 2004). Storage of the cultivar ‘Dahongpao’ at 0°C in 0–5 kPa O<sub>2</sub> for up to 120 days showed that off-flavours were produced by exposure to concentrations below 2 kPa O<sub>2</sub> for 20 days or more. Storage in 0–2 kPa O<sub>2</sub> could stimulate the decomposition of starch, while in 3–5 kPa O<sub>2</sub> the decomposition rate could be slowed. Those exposed to 1–3 kPa O<sub>2</sub> had higher total sugars compared with other treatments, but the mean decay rate in 0–2 kPa O<sub>2</sub> was 10% and with those stored in air it was 7%. The decay level in 3–4 kPa O<sub>2</sub> atmosphere for 20 days or 5 kPa O<sub>2</sub> for 25 days was only 1% at the end of 120 days of cold storage. In general, 3 kPa O<sub>2</sub> for 20 days was the best (Wang *et al.*, 2004a).

### Chestnut, Sweet (*Castanea sativa*)

Also called Spanish chestnut, Portuguese chestnut and European chestnut. The cultivars ‘Catot’ and ‘Platella’ were stored at 1°C in 20 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> and their freshness, taste and flavour were maintained and after 105 days they looked as fresh and bright as those freshly harvested (Mignani and Vercesi, 2003). This treatment controlled fungal infections except for those caused by *Aspergillus niger*, but storage in 2.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> was less effective. Rouves and Prunet (2002) found that the best storage conditions of those that they tested for the cultivars ‘Marigoule’ and ‘Bouche de Betizac’ were –1°C in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. There was ‘no water loss’, mould development was much slower and taste remained unaltered but some germination occurred in ‘Marigoule’. ‘Comballe’ and ‘Marron de Goujonac’ did not store well in any of the conditions tested.

### Chicory, Endive, Belgian Endive, Escarole, Radicchio, Witloof Chicory (*Cichorium* spp.)

The cultivated forms of *C. intybus* var. *foliosum* are grown for their leaves, which are used in salads, or *C. intybus* var. *sativum* for their roots. The latter is used mainly as a coffee substitute and only *C. Intybus* var. *foliosum* and *C. endivia* are dealt with here. Storage of witloof at 0°C in 4–5 kPa CO<sub>2</sub> + 3–4 kPa O<sub>2</sub> delayed greening of the tips in light and delayed opening of the heads (Hardenburg *et al.*, 1990). Saltveit (1989) also recommended 4–5 kPa CO<sub>2</sub> and 3–4 kPa O<sub>2</sub> for witloof, but at 0–5°C, and it was reported to have had only a slight effect. Storage at 5°C in 10 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> prevented red discoloration, leaf edge discoloration and other negative quality aspects, but in storage at 1°C there was increased red discoloration (Vanstreels *et al.*, 2002). In a comparison of 5, 10, 15 and 20 kPa CO<sub>2</sub> in storage at 0°C, Bertolini *et al.* (2003) found that 10 kPa CO<sub>2</sub> was the most effective in suppressing *Botrytis cinerea* in red chicory. Later Bertolini *et al.* (2005) stored radicchio ‘Rosso di Chioggia’ at 0°C for up to 150 days. They artificially inoculated some heads with *B. cinerea* and found that lesions caused by *B. cinerea* decreased with increasing concentrations of CO<sub>2</sub> over the range of 5–20 kPa for up to 60 days. Subsequently only 10 and 15 kPa CO<sub>2</sub> were effective, while after 120 days all the concentrations had low efficacy. In naturally infected heads, 5 and 10 kPa CO<sub>2</sub> was effective in preventing *B. cinerea* even after 150 days storage and it also reduced the fungus spreading to adjacent heads. Heads stored in 15 kPa CO<sub>2</sub> for 150 days showed phytotoxic effects and increased their vulnerability to rots. Monzini and Gorini (1974) recommended 0°C in 1–5 kPa CO<sub>2</sub> + 1–5 kPa O<sub>2</sub> for 45–50 days. Van de Velde and Hendrickx (2001) investigated storage of cut Belgian endive in atmospheres ranging from 2 to 18 kPa CO<sub>2</sub> + 2–18 kPa O<sub>2</sub> and found that the optimum concentration was 10 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> + 80 kPa N<sub>2</sub>. Wardlaw (1938) reported that storage in 25 kPa CO<sub>2</sub> could cause the central leaves to turn brown.

### Chillies (*Capsicum annuum*, *C. frutescens*)

Also called chilli peppers, hot peppers, cherry peppers. Storage at 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> was recommended by SeaLand (1991). Kader (1985, 1992) recommended 8–12°C in 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> which he observed had a fair effect but was of no commercial use, but 10–15 kPa CO<sub>2</sub> was beneficial at 5–8°C. Saltveit (1989) recommended 12°C in 0–5 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for the fresh market, which had only a slight effect, and 5–10°C in 10–20 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> was recommended for processing. Storage of green chillies for 6 weeks at 10°C and 80–90% RH in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> resulted in less decay and wrinkling and they were firmer with higher TSS and vitamin C than those stored in air or other CA tested (Ullah Malik *et al.*, 2009).

### Chinese Cabbage (*Brassica pekinensis*)

The recommended storage conditions in air were 0–1°C, with a maximum storage period of 6 weeks (Mertens, 1985). Hardenburg *et al.* (1990) recommended 0°C in 1 kPa O<sub>2</sub>. Wang and Kramer (1989) also showed that storage in 1 kPa O<sub>2</sub> extended storage life and reduced the decline in AA, chlorophyll and sugars in the outer leaf laminae. Saltveit (1989) recommended 0–5°C in 0–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub>, which had only a slight effect. Apeland (1985) stored the cultivars ‘Tip Top’ and ‘Treasure Island’ at 2.5°C or 5.0°C in 0.5, 2.5 or 5.0 kPa CO<sub>2</sub> + 1–20.5 kPa O<sub>2</sub>. With ‘Tip Top’ the storage was best at 5°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. The results with ‘Treasure Island’ were inconclusive, because the control heads kept very well but with an average of only 68% saleable after 120 days. Pelleboer and Schouten (1984) reported that after 4 months of storage of the cultivars ‘Chiko’ and ‘WR 60’ at 2–3°C in 0.5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or less, the average percentage of healthy heads was 72%, and after 5 months 60% or more were healthy. After subsequent storage at 15°C and 85% RH to simulate shelf-life, the corresponding percentages were 58% and 43%. Following storage

in 6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 6 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub> there was a rapid fall in quality, confirming that 6 kPa CO<sub>2</sub> could be harmful.

Hermansen and Hoftun (2005) stored the cultivar 'Nerva', inoculated with *Phytophthora brassicae*, at 1.5°C in air, 0.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or 3.0 kPa CO<sub>2</sub> + 3.0 kPa O<sub>2</sub> for 94–97 days. The infection caused by *P. brassicae* was significantly higher in both the CA treatments than in air, but *in vitro* studies gave the opposite results. Brown midribs, spots or streaks of brown tissue of the leaves and midribs and pepper spots are common physiological disorders. Chilling injury (brown midribs) found in 'Parkin' was highest in heads stored in air. CA storage reduced this disorder and the lowest percentage of brown midribs was found in 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. No chilling injury was found. Adamicki (2003), working with the F<sub>1</sub> hybrid cultivars 'Asten', 'Bilko', 'Gold Rush', 'Maxim', 'Morillo', 'Parkin' and 'RS 6064', also reported that low concentrations of O<sub>2</sub> and CO<sub>2</sub> greatly reduced the incidence of physiological disorders, including brown ribs. However, storage at 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> resulted in lower percentage of marketable heads and higher percentage of damaged leaves and heads compared with storage in air for 100–130 days at 0°C, 2°C or 5°C and 95–98% RH. CO<sub>2</sub> concentrations > 5 kPa resulted in greater losses of weight and trim, caused mainly by leaf rotting (Adamicki and Gajewski, 1999). They also found that 1.5–3.0 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> was the best atmosphere for long-term storage. For most of the F<sub>1</sub> cultivars tested, a storage temperature 2°C was better than 0°C.

### Citrus Hybrids (*Citrus* spp.)

Chase (1969) stored 'Temple' oranges (*C. sinensis* × *C. reticulata*) and 'Orlando' tangelos (*C. sinensis* × *C. paradisi*) at 1°C in atmospheres containing 5 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>. They found little benefit from CA storage compared with storage in air for 'Orlando' but when 'Temple' oranges were stored for 5 weeks in CA followed by 1 week at 21°C in air their flavour was superior to those stored in air throughout.

### Cranberry (*Vaccinium* spp.)

There are two species: the American cranberry (*Vaccinium macrocarpon*) and the European cranberry (*V. oxycoccus*). Kader (1989) recommended 2–5°C in 0–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub>, but Hardenburg *et al.* (1990) reported that CA storage was not successful in increasing storage life. In contrast, Gunes *et al.* (2002) found that in storage of the cultivars 'Pilgrim' and 'Stevens' at 3°C in atmospheres of 21 kPa O<sub>2</sub> + 15 or 30 kPa CO<sub>2</sub> there was decreased bruising, physiological breakdown and decay of berries compared with those stored in air. Respiration rate and weight loss were also decreased, but fruit softening increased compared with storage in air. There was also an increase in acetaldehyde, ethanol and ethyl acetate during storage in 21 kPa O<sub>2</sub> + 15 or 30 kPa CO<sub>2</sub> but the levels varied with cultivar and storage atmosphere, with the highest in 2 and 70 kPa O<sub>2</sub> and in 100 kPa N<sub>2</sub>. Overall, 30 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> appeared to be optimum. No sensory analysis was included to confirm whether accumulations of fermentation products at this atmosphere were acceptable for consumers. However, they also found that the storage atmosphere did not affect the content of total phenolics or flavonoids, but the total antioxidant activity of the fruit increased overall by about 45% in fruit stored in air. This increase did not occur during storage in 30 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>.

### Cucumber (*Cucumis sativus*)

Over a 24 h period at 10°C or 20°C, the respiration rate of the cultivar 'Tyria' decreased as the O<sub>2</sub> concentration decreased down to 0.5 kPa, but at 0 kPa O<sub>2</sub>, the respiration rate increased because of fermentation (Praeger and Weichmann, 2001). CA recommendations include: 14°C with 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Stoll, 1991); 5–7 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (SeaLand, 1992); a maximum of 10 kPa CO<sub>2</sub> and a minimum of 3 kPa O<sub>2</sub> (Fellows, 1988); 8.3°C in 3–5 kPa O<sub>2</sub>, which gave a slight extension in storage life (Pantastico, 1975); 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Ryall and Lipton, 1972);

8–12°C in 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>, which had a fair effect but was of no commercial use (Kader, 1985, 1992); 12°C in 0 kPa CO<sub>2</sub> + 1–4 kPa O<sub>2</sub> for the fresh market and 4°C in 3–5 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for pickling (Saltveit, 1989); and 12.5°C in 5% O<sub>2</sub> + 5 kPa CO<sub>2</sub> (Schales, 1985). Mercer and Smittle (1992) stored cucumbers of the cultivar ‘Gemini II’ in 0, 5 or 10 kPa CO<sub>2</sub> + 5 or 20 kPa O<sub>2</sub> at 5°C or 6°C for 2, 4 or 6 days, or at 5°C for 5 days, or at 3°C for 10 days then 2–4 days at 25°C. High CO<sub>2</sub> and low O<sub>2</sub> delayed the onset of chilling injury symptoms, but did not prevent their development. Chilling injury symptoms increased with longer exposure to chilling temperatures and were associated with solubilization of cell wall polysaccharides. Storage in 5 kPa O<sub>2</sub> was shown to retard yellowing (Lutz and Hardenburg, 1968). Pre-treatment with O<sub>2</sub> has been shown to be beneficial. In storage at 5°C, pre-treatment with 100 kPa O<sub>2</sub> for 48 h lowered the respiration rate compared with storage in either 5 kPa O<sub>2</sub> or air. It also delayed the appearance of fungal decay by 2 days, delayed the onset and reduced severity of chilling injury, halved weight loss and delayed the appearance of shrivelling by 4 days compared with control (Sri-laong *et al.*, 2005). Reyes (1989) showed that the virulence of mucor rot (*Mucor mucedo*) and grey mould (*Botrytis cinerea*) were suppressed in 7.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub>.

### Durian (*Durio zibethinus*)

Durian is a climacteric fruit. It was reported by Pan (2008) that in South China it could be stored in air at ambient temperatures for 3 weeks. Tongdee *et al.* (1990) showed that storage of 85% mature ‘Mon Tong’ at 22°C in atmospheres with 5 or 7.5 kPa O<sub>2</sub> inhibited ripening, but fruit ripened normally when returned to air. They also found that in 2 kPa O<sub>2</sub> ripening was inhibited and the fruit failed to ripen when removed and stored in air. CO<sub>2</sub> levels of up to 20 kPa in air did not affect the speed of ripening or the quality of the ripe fruit. Fruit of the cultivars ‘Chanee’, ‘Kan Yao’ and ‘Mon Tong’ was harvested at three maturity stages and

stored at 22°C. The respiration rate and ethylene production at harvest and the peak climacteric respiratory value were higher in fruit harvested at a more advanced stage. Storage in 10 kPa O<sub>2</sub> resulted in a significant reduction in respiration rate and ethylene production, but the onset of ripening and ripe fruit quality were not affected. Ripening was inhibited in fruit stored in 5–7.5 kPa O<sub>2</sub> but recovered when fresh air was subsequently introduced. O<sub>2</sub> at this level did not affect ripening in fruit harvested at an advanced stage of maturity. Fruit stored in 2 kPa O<sub>2</sub> failed to resume ripening when removed to air. Fruit stored in 10 or 20 kPa CO<sub>2</sub> was either not affected or showed a slight reduction in ethylene production. Thus high levels of CO<sub>2</sub> alone did not influence the onset of ripening or other quality attributes of ripe fruit. Atmospheres of 5, 10, 15 or 20 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> had a greater effect on the condition of the aril than either high CO<sub>2</sub> or low O<sub>2</sub> alone. The aril remained hard in the less mature fruit stored in 10 kPa O<sub>2</sub> + 15 or 20 kPa CO<sub>2</sub>. Kader (1993) recommended 10.5°C in 5–20 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>.

### Feijoa (*Feijoa sellowiana*)

Aziz Al-Harthy *et al.* (2009) reported that fruits of the cultivar ‘Opal Star’ stored at 4°C in 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>, 2 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> or 0 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> for up to 10 weeks retained their green colour while those stored in air went yellow. CA also had some effects on reducing the rate of softening compared with storage in air, with 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> and 5 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> giving the best results.

### Fig (*Ficus carica*)

Storage in high CO<sub>2</sub> atmospheres reduced mould growth without affecting the flavour of the fruit (Wardlaw, 1938). Hardenburg *et al.* (1990) also showed that storage with enriched CO<sub>2</sub> was a useful supplement to refrigeration. Storage at 0–5°C in 15 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> was recommended by Kader

(1985), as was 0–5°C + 15–20 kPa CO<sub>2</sub> and 5–10 kPa O<sub>2</sub> by Kader (1989, 1992). SeaLand (1991) also recommended 15 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. Tsantili *et al.* (2003) stored the cultivar ‘Mavra Markopoulou’ at –1°C in either air or 2 kPa O<sub>2</sub> + 98 kPa N<sub>2</sub> for 29 days. Those stored in air became soft during storage for longer than 8 days, but those in CA remained in much better condition. Colelli *et al.* (1991) showed that good quality of ‘Mission’ figs was maintained for up to 28 days when they were kept at 0°C, 2.2°C or 5°C in atmospheres enriched with 15 or 20 kPa CO<sub>2</sub>. The benefits of exposure to high CO<sub>2</sub> levels were a reduction of the incidence of decay and the maintenance of a bright external appearance. Ethylene production was lower and softening was slower in figs stored at high CO<sub>2</sub> concentrations compared with those kept in air. Ethanol content of the fruit stored in 15 or 20 kPa CO<sub>2</sub> increased slightly during the first 3 weeks and moderately during the fourth week, while acetaldehyde concentration increased during the first week and then decreased. It was concluded that postharvest life can be extended by 2–3 weeks at 0–5°C in atmospheres enriched with 15–20 kPa CO<sub>2</sub> but off-flavours could be a problem by the fourth week of storage.

### Garlic (*Allium sativum*)

Storage in 0 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> was recommended (SeaLand, 1991) and Monzini and Gorini (1974) recommended 3°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for 6 months. Cantwell *et al.* (2003) showed that storage of the cultivars ‘California Late’ and ‘California Early’ at 0–1°C in CO<sub>2</sub> atmospheres of 5 kPa, 10 kPa, 15 kPa and 20 kPa reduced sprout growth, decay and discoloration but CO<sub>2</sub> concentrations over 15 kPa could lead to injury after 4–6 months.

For fresh peeled garlic storage at 5°C and 10°C with either 5–15 kPa CO<sub>2</sub> or 1–3 kPa O<sub>2</sub> was effective in retarding discoloration and decay for 3 weeks (Cantwell *et al.*, 2003). They also showed that, when garlic shoots were stored at 0°C in ten combinations of 3–6.5 kPa O<sub>2</sub> + 0–12 kPa CO<sub>2</sub>, the optimum atmosphere was 3 kPa O<sub>2</sub> + 8 kPa

CO<sub>2</sub>, which maintained levels of chlorophyll, reducing and total sugars and freshness for 235 days. There was some spoilage and rotting, but rotting decreased with decreasing O<sub>2</sub> concentration, while high CO<sub>2</sub> reduced mould growth (Zhou *et al.*, 1992a). Zhou *et al.* (1992b) and Zhang and Zhang (2005) used CA storage of sprouted garlic in chambers flushed with N<sub>2</sub> from a carbon molecular sieve to reduce the O<sub>2</sub> content to 1–5 kPa. CO<sub>2</sub> was allowed to increase to 2–7% by product respiration. After 240–270 days, the quality of sprouted garlic remained high. After fumigating, the shoots were treated with a fungicide and packed in plastic bags with a silicon window and stored at 0–1°C with 95% RH. These treatments reduced the respiration rate and the loss of chlorophyll and inhibited cellulose production. After 280 days, 96% had good freshness, greenness and crispness.

### Gooseberry (*Ribes uva-crispa*, *R. grossularia*)

Robinson *et al.* (1975) showed that with storage in low O<sub>2</sub> the respiration rate was reduced at all temperatures tested (Table 12.1). Prange (not dated) stated that they could be stored at 0–1°C for 3 weeks but storage duration could be extended to 6–8 weeks, using 1°C in 10–15 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub>. Harb and Streif (2004a) stored the cultivar ‘Achilles’ at 1°C in air, 6 kPa CO<sub>2</sub> + 18 kPa O<sub>2</sub>, 6 kPa CO<sub>2</sub> + 18 kPa O<sub>2</sub>, 12 kPa CO<sub>2</sub> + 18 kPa O<sub>2</sub>, 18 kPa CO<sub>2</sub> + 18 kPa O<sub>2</sub>, 12 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 18 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or 24 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. Storage in air led to a reduction in firmness, darkening of fruit colour, mealy texture and lower acidity level compared with CA storage. Storage in 2 kPa O<sub>2</sub> + 18 kPa CO<sub>2</sub> resulted in off-flavours, but in 12 kPa CO<sub>2</sub> +

**Table 12.1.** Effects of temperature and reduced O<sub>2</sub> level on the respiration rate (CO<sub>2</sub> production in mg/kg/h) of Leveller gooseberries (Robinson *et al.*, 1975).

	0°C	5°C	10°C	15°C	20°C
Air	10	13	23	40	58
3 kPa O <sub>2</sub>	7		16		26

2 kPa O<sub>2</sub> no off-flavour was detected. They recommended 12–15 kPa CO<sub>2</sub> + 18 kPa O<sub>2</sub> for up to 7 weeks. Certain cultivars of gooseberries have been held for as long as 3 months at 0°C in air (McKay and Van Eck, 2006), but for longer storage they suggested harvesting at the green-mature stage and placing them at –0.5°C to –0.9°C with 93% RH in 2.5–3.0 kPa O<sub>2</sub> + 20–25 kPa CO<sub>2</sub>. They also found that gooseberries were sensitive to ethylene.

### Grape (*Vitis vinifera*)

Storage recommendations were 1–3 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (SeaLand, 1991). Magomedov (1987) showed that different cultivars required different CA storage conditions. The cultivars ‘Agadai’ and ‘Dol’chatyi’ stored best in 3 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> whereas for ‘Muskat Derbentskii’ 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 3 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> were more suitable. ‘Muskat Derbentskii’, ‘Dol’chatyi’ and ‘Agadai’ had a storage life of up to 7, 6 and 5 months, respectively, in CA storage. The best results for storage of the cultivar ‘Moldova’ were in either 8 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> or 10 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub>. In these conditions, 89%, 80% and 75% of first grade grapes were obtained after 5, 6½ and 7½ months of storage, respectively. In another trial with ‘Muscat of Hamburg’ and ‘Italia’, grapes stored in air or in CA for 3–7 months were assessed in relation to profitability. The best results were again obtained in storage of 8 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> for ‘Muscat of Hamburg’ and of 5 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> for ‘Italia’ (Khitron and Lyublinskaya, 1991). Turbin and Voloshin (1984) showed that for storage for less than 5 months, 8 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> was most suitable for ‘Muskat Gamburgskii’ (‘Hamburg Muscat’), 5–8 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for ‘Italia’ and 8 kPa CO<sub>2</sub> + 5–8 kPa O<sub>2</sub> for ‘Galan’. ‘Waltham Cross’ and ‘Barlinka’ were stored in CA at –0.5°C for 4 weeks. The percentage of loose berries in ‘Waltham Cross’ was highest (> 5%) in 21 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> than in lower O<sub>2</sub> levels (Laszlo, 1985). The cultivar ‘Agiorgitiko’ was stored at 23–27°C for 10 days either in 100 kPa CO<sub>2</sub> or in air. Fruits from both treatments were held at 0°C for 20 h before

analysis (Dourtoglou *et al.*, 1994). ‘Kyoho’ stored well in 4 kPa O<sub>2</sub> + 30 kPa CO<sub>2</sub> but an alcoholic flavour and browning occurred after 45 days. Storage in 4 kPa O<sub>2</sub> + 9 kPa CO<sub>2</sub> or 80 kPa O<sub>2</sub> + 20 kPa N<sub>2</sub> retained good quality during 60 days of storage, without off-flavours (Deng *et al.*, 2006). Crisosto *et al.* (2003a) studied 5, 10, 15, 20 and 25 kPa CO<sub>2</sub> factorially combined with 3, 6 and 12 kPa O<sub>2</sub> on ‘Redglobe’. Optimum conditions for late-harvested fruit (19% TSS) were 10 kPa CO<sub>2</sub> + 3, 6 or 12 kPa O<sub>2</sub> for up to 12 weeks of storage and for early-harvested fruit (16.5% TSS) it was 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> for up to 4 weeks. Pedicel browning was accelerated in grapes exposed to 10 kPa CO<sub>2</sub> for early-harvested and above 15% for late-harvested fruit. However, the same authors (Crisosto *et al.*, 2003b) found that the combination of 15 kPa CO<sub>2</sub> with 3, 6 or 12 kPa O<sub>2</sub> was optimum for late-harvested ‘Thompson Seedless’ for up to 12 weeks and CA should not be used for commercially early-harvested grapes.

Sahin *et al.* (2015) compared storage of ‘Red Globe’ grapes at 0°C with 95% RH for 90 days in 21 kPa O<sub>2</sub> + 0.03 kPa CO<sub>2</sub>, 2 kPa O<sub>2</sub> + 0.03 kPa CO<sub>2</sub>, 2 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>, 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 1 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>. Those stored in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> were the firmest and had the highest TSS content and lowest decay, while those that had been stored in 1 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> had the lowest rachis browning.

CA has been shown to reduce disease development and may be able to be developed as an alternative to sulfur dioxide fumigation. Kader (1989, 1992) recommended 0–5°C in 1–3 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> but reported that CA storage was incompatible with sulfur dioxide fumigation. However, Berry and Aked (1997) working on ‘Thompson Seedless’ stored at 0–1°C for up to 12 weeks showed that atmospheres containing 15–25 kPa CO<sub>2</sub> inhibited infection with *Botrytis cinerea* by between 95% and 100% without detrimentally affecting flavour. Mitcham *et al.* (1997) showed that high CO<sub>2</sub> levels in the storage atmosphere could be used for controlling insect pests of grapes. In trials where four cultivars were exposed to 0°C in 45 kPa CO<sub>2</sub> + 11.5 kPa O<sub>2</sub>, complete



control of *Platynota stultana*, *Tetranychus pacificus* and *Frankliniella occidentalis* was achieved without injury to the grapes. 'Flame Seedless' and 'Crimson Seedless' grapes were inoculated with *B. cinerea* and then exposed to 40 kPa CO<sub>2</sub> at 0°C for 48 h and stored in either in air or 12 kPa O<sub>2</sub> + 12 kPa CO<sub>2</sub>. This combined treatment (CO<sub>2</sub> + CA) reduced the grey mould incidence compared with storage in air from 22% to 0.6% after 4 weeks and from 100% to 7.4% after 7 weeks of storage, compared with fruit not treated and stored in air at 0°C (Teles *et al.*, 2014).

### Grapefruit, Pummelo (*Citrus × paradisi*)

Typical storage conditions were given as 10–15°C in 5–10 kPa CO<sub>2</sub> + 3–10% O<sub>2</sub> by Bishop (1996) although CA storage was not considered beneficial. Erkan and Pekmezci (2000) found that 1 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at 10°C to be optimum for 'Star Ruby' grown in Turkey. They were stored in these conditions for 125 days without losing much quality. SeaLand (1991) recommended 5–10 kPa CO<sub>2</sub> + 3–10 kPa O<sub>2</sub> for grapefruit from California, Arizona, Florida, Texas and Mexico. Storage at 10–15°C in 5–10 kPa CO<sub>2</sub> + 3–10 kPa O<sub>2</sub> had a fair effect on storage, but Kader (1985, 1992) reported that CA storage was not used commercially. Storage experiments have shown that there was some indication that fruit stored at 4.5°C in 10 kPa CO<sub>2</sub> for 3 weeks had less pitting (a symptom of chilling injury) than those stored in air. Also pre-treatment with 20–40 kPa CO<sub>2</sub> for 3 or 7 days at 21°C reduced physiological disorders on fruit stored at 4.5°C for up to 12 weeks (Hardenburg *et al.*, 1990). Hatton *et al.* (1975) and Hatton and Cubbedge (1982) also showed that exposing grapefruit before storage to 40 kPa CO<sub>2</sub> at 21°C reduced chilling injury symptoms during subsequent storage.

### Guava (*Psidium guajava*)

Short-term exposure to 10, 20 or 30 kPa CO<sub>2</sub> had no effect on respiration rate, but ethylene

biosynthesis was reduced by all three levels of CO<sub>2</sub> (Pal and Buescher, 1993). Teixeira *et al.* (2009) recommended storage of the cultivar 'Pedro Sato' at 12.2° in 5 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> for up to 28 days, but they found that storage in 15 or 20 kPa CO<sub>2</sub> resulted in CO<sub>2</sub> injury. Mature-green guavas were exposed to air or 5 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> for 24 h before storage in air at either 4°C or 10°C for 2 weeks. They were then transferred to 20–23°C for 3 days to simulate shelf-life. The colour of the CA-treated fruit developed more slowly than in those kept in air throughout and they were considered to be of better quality after storage and shelf-life and showed no chilling injury symptoms even after storage at 4°C for 3 weeks, while chilling injury occurred on those stored in air throughout (Bautista and Silva, 1997). Freshly harvested mature-green 'Lucknow-49' fruit was stored at 8°C with 85–90% RH in 5 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub> or 10 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. CA-stored fruits could be kept in an unripe condition for 1 month, while those stored in air showed severe chilling injury symptoms, high weight loss and spoilage softening and colour change (Pal *et al.*, 2007). Subsequently Singh and Pal (2008b) investigated 2.5, 5, 8 and 10 kPa O<sub>2</sub> factorially combined with 2.5, 5 and 10 kPa CO<sub>2</sub> (balance N<sub>2</sub>) at 8°C and 85–90% RH. They successfully stored the cultivars 'Lucknow-49', 'Allahabad Safeda' and 'Apple Colour' for 30 days in 5 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 8 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. The fruits were then transferred to ambient conditions of 25–28°C and 60–70% RH and they all ripened successfully. Chilling injury and decay incidence were reduced during ripening of fruit that had been stored in CA compared with those that had been stored in air. However, larger amounts of ethanol and acetaldehyde accumulated in fruit held in atmospheres containing 2.5 kPa O<sub>2</sub>. Teixeira *et al.* (2016) stored the cultivar 'Pedro Santo' at 12.2°C in 5 kPa O<sub>2</sub> combined with 1, 5, 10, 15 or 20 kPa CO<sub>2</sub>. CO<sub>2</sub> injury occurred in fruit stored in both 15 and 20 kPa CO<sub>2</sub>. They therefore recommended 5 kPa O<sub>2</sub> + 1 or 5 kPa CO<sub>2</sub> as giving the optimum conditions over 28 days storage.

### Horseradish (*Armoracia rusticana*)

CA storage was reported to have had only a slight to no effect (SeaLand, 1991) and was not recommended by Saltveit (1989). Weichmann (1981) studied different levels of CO<sub>2</sub> of up to 7.5 kPa during 6 months of storage at 0–1°C and found no detrimental effects of CO<sub>2</sub> but no advantages either over storage in air. He did, however, find that those stored in 7.5 kPa CO<sub>2</sub> had a higher respiration rate and higher total sugar content than those stored in air. Gui *et al.* (2006) reported that exposure to high levels of CO<sub>2</sub> could result in some reduction in enzyme activity but activity returned to what it was when they were removed and stored in air.

### Jujube (*Ziziphus jujuba*)

Han *et al.* (2006) found that the best CA storage condition for the cultivar ‘Dong’ was 1.5 ± 0.5°C and 95% RH in 5–6 kPa O<sub>2</sub> + 0–0.5 kPa CO<sub>2</sub> for 90 days.

### Kiwifruit (*Actinidia chinensis*)

Also called Chinese gooseberry and yang tao. Storage recommendations include: 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (SeaLand, 1991); 0°C and 90% RH in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (Lawton, 1996); 0–5°C in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (Kader, 1985); 0°C in 3–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (Kader, 1989, 1992); 0.5–0.8°C and 95% RH with either 1 kPa O<sub>2</sub> + 0.8 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 4–5 kPa CO<sub>2</sub> for 6 months (Brigati and Caccioni, 1995); and 0–5°C in 5–10 kPa CO<sub>2</sub> and 1–2 kPa O<sub>2</sub> (Bishop, 1996). Intermittent storage at 0°C of 1 week in air and 1 week in air with 10 kPa or 30 kPa CO<sub>2</sub> was shown to reduce fruit softening (Nicolas *et al.*, 1989). Storage in CA was shown to increase storage life by 30–40%, with optimum conditions of 3–5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, but levels of CO<sub>2</sub> above 10 kPa were found to be toxic to the fruit (Brigati and Caccioni, 1995). Tulin Oz and Eris (2009) found that storage of ‘Hayward’ at 0°C and 85–90% RH in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> retained their quality for 5 months.

They recommended harvesting when the TSS content was 5.5–6.5%. Yildirim and Pekmezci (2009) recommended 0°C and 90–95% RH in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 4 months, which suppressed ethylene production and delayed softening. However, they found that there was a slight decrease in vitamin C content in both CA storage and air storage. Storage at 0°C and 90–95% RH in 3 kPa CO<sub>2</sub> + 1–1.5 kPa O<sub>2</sub> maintained fruit firmness during long-term storage, but 1 kPa CO<sub>2</sub> + 0.5 kPa O<sub>2</sub> gave the best storage conditions while maintaining an acceptably low level of the incidence of rotting (Brigati *et al.*, 1989). Özer *et al.* (1999) found that ‘Hayward’ could be stored for 6 months at 0 ± 0.5°C and 90–95% RH, especially in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. However, subsequent shelf-life at 20 ± 3°C and 60 ± 5% RH should be limited to 15 days. Steffens *et al.* (2007b) investigated factorial combinations of 0.5, 1.0 and 1.5 kPa O<sub>2</sub> combined with 8, 12 and 16 kPa CO<sub>2</sub> on ‘Bruno’. Sensory analysis showed that fruit stored well in 1.0 + 8 kPa CO<sub>2</sub> and 1.5 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> without developing off-flavours (Steffens *et al.*, 2007b). ‘Hayward’ was stored over two seasons in 0–21 kPa O<sub>2</sub> + 0–5 kPa CO<sub>2</sub> at 0–10°C. CO<sub>2</sub> delayed softening but lowering O<sub>2</sub> to near 0 kPa did not inhibit softening completely at 0°C. At temperatures higher than 3°C the additional effect of CA storage was limited (Hertog *et al.*, 2004). Testoni and Eccher Zerbini (1993) showed that storage in CO<sub>2</sub> concentrations higher than 5 kPa resulted in irregular softening in the core.

Tonini and Tura (1997) showed that storage at –0.5°C in combination with 4.8 kPa CO<sub>2</sub> + 1.8 kPa O<sub>2</sub> reduced rots (*Botrytis cinerea* and *Phialophora* spp.) and softening compared with storage in air. The combination of either 1 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> or 1.5 CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> was even more effective in controlling rots caused by *Phialophora* spp. In contrast, in central and northern Italy, Tonini *et al.* (1999) found that CA storage at –0.8°C for 120–140 days favoured the spread of *B. cinerea*. They found that postponing the reduction of O<sub>2</sub> and the increase in CO<sub>2</sub> for 30–50 days avoided increasing *Botrytis* rots without any adverse effects on

fruit firmness. Storage at 0.5–1°C with 92–95% RH in 4½–5 kPa CO<sub>2</sub> + 2–2½ kPa O<sub>2</sub> and ethylene at 0.03 ppm or less delayed flesh softening but strongly increased the incidence of Botrytis stem-end rot (Tonini *et al.*, 1989). Brigati and Caccioni (1995) also showed that the high level of CO<sub>2</sub> could lead to an increase in the incidence of *B. cinerea*. Li *et al.* (2017a) confirmed the benefit of CA storage in maintaining fruit firmness and that 5 kPa CO<sub>2</sub> was better than 2 kPa CO<sub>2</sub>. The effect of 5 kPa CO<sub>2</sub> was maintained for a couple of weeks after transfer from CA to air storage. Whilst the softening of all tissue zones was slowed by CA, the outer pericarp appeared to be slowed to a greater extent than the inner pericarp. A consequence of the differential effect of CA on the two pericarp tissue zones was that after CA storage the inner pericarp had become relatively softer than the outer pericarp when compared with air-stored fruit at the same fruit firmness value.

Kiwifruit is very susceptible to ethylene even at very low concentrations and Brigati and Caccioni (1995) recommended that ethylene should be maintained at < 0.03 ppm in CA stores. Storage in 2–5 kPa O<sub>2</sub> + 0–4 kPa CO<sub>2</sub> reduced ethylene production (Wang *et al.*, 1994) and Tulin Oz and Eris (2009) found that the main effect of 2 kPa O<sub>2</sub> in storage was to reduce the rate of softening and of 5 kPa CO<sub>2</sub> to reduce ethylene production. Antunes and Sfakiotakis (2002) found that after 60, 120 or 180 days of storage at 0°C there was an initiation of ethylene production when fruits were transferred to 20°C, with no lag period from those stored in air or 2% O<sub>2</sub> + 5% CO<sub>2</sub>. However, those removed from storage in 0.7 kPa O<sub>2</sub> + 0.7 kPa CO<sub>2</sub> or 1% O<sub>2</sub> + 1% CO<sub>2</sub> showed reduced capacity to produce ethylene, mainly due to low ACC oxidase activity rather than reduced ACC production or ACC synthase activity. Zhang (2002) successfully stored kiwifruit at 0 ± 0.5°C in 2–3 kPa O<sub>2</sub> + 4–5 kPa CO<sub>2</sub> with a potassium permanganate absorber for over 180 days. After storage the fruit looked fresh with good colour, smell and taste. The percentage of firm fruit was > 92%, and the retention of AA was > 80% compared with levels at the beginning of storage.

Storage of two species of hardy kiwifruit, ‘Ananasnaya’ (*A. arguta*) and ‘Bingo’ (*A. purpurea* × *A. arguta*), at 1°C and 85% RH in air or 1.5 kPa O<sub>2</sub> + 1.5 kPa CO<sub>2</sub> showed that storage in air was adequate for 4 weeks, but for 8 weeks CA was required (Latocha *et al.*, 2014).

### Kohlrabi (*Brassica oleracea* var. *gongylodes*)

Escalona *et al.* (2007) found that storage in air at 0°C at high humidity resulted in yellowing of the stalks that later fell off, which affected their appearance and marketability. Whole or fresh-cut kohlrabi could be stored for 28 days at 5°C and 95% RH in 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> followed by 3 days at 15°C and 60–70% RH in air and still retain good commercial quality without detrimentally affecting their stalks. However, Seal and (1991) had previously reported that CA storage had only a slight to no effect.

### Lanzones, Langsat (*Lansium domesticum*)

Pantastico (1975) recommended 14.4°C in 0 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> which gave a 16-day postharvest life compared with only 9 days at the same temperature in air. He also indicated that the skin of the fruit turned brown during retailing and if they were sealed in PE film bags the browning was aggravated, probably due to CO<sub>2</sub> accumulation.

### Leek (*Allium ampeloprasum* var. *porrum*)

Optimum storage conditions were reported to be: 0°C in 5–10 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> for up to 4–5 months (Kurki, 1979); 0°C in 15–25 kPa CO<sub>2</sub> for 4½ months (Monzini and Gorini, 1974); 3–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (SeaLand, 1991); 0–5°C in 3–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub>, which had a good effect but was claimed to be of no commercial use (Kader, 1985, 1992); and 0–5°C in 5–10 kPa CO<sub>2</sub> + 1–6 kPa O<sub>2</sub>, which had only a slight effect (Saltveit, 1989). Goffins and Herregods (1989)

showed that with storage at 0°C and 94–96 kPa RH in 2 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> + 5 kPa carbon monoxide, leeks could be stored for up to 8 weeks compared with 4 weeks at the same temperature in air. Under those conditions the total losses were 19% while those stored in the same conditions but without carbon monoxide had 28% losses and those in air had 37% losses. Lutz and Hardenburg (1968) reported that CO<sub>2</sub> levels of 15–20 kPa caused injury.

### Lemon (*Citrus limon*)

Storage recommendations include: 5–10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (SeaLand, 1991); 10–15°C in 0–5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Kader, 1985); 10–15°C in 0–10 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> (Kader, 1989); and 10–15°C with 0–10 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> (Bishop, 1996). The rate of colour change could be delayed with high CO<sub>2</sub> and low O<sub>2</sub> in the storage atmosphere but 10 kPa CO<sub>2</sub> could impair the flavour (Pantastico, 1975). Wild *et al.* (1977) reported that lemons may be stored in good condition for 6 months in 10 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> combined with the continuous removal of ethylene.

### Lettuce (*Lactuca sativa*)

Storage recommendations include: 2 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for up to 1 month (Hardenburg *et al.*, 1990); 0 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (SeaLand, 1991); 0°C with 98% RH in a maximum of 1 kPa CO<sub>2</sub> and a minimum of 2 kPa O<sub>2</sub> (Fellows, 1988); 1.5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> (Lawton, 1996); 0–5°C in 0 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> (Bishop, 1996); and 3–5 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub> for 3 weeks (Tataru and Dobreanu, 1978). Storage in 1.5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> inhibited butt discoloration and pink rib at 0°C; the effect did not persist during 5 days of subsequent storage at 10°C (Hardenburg *et al.*, 1990) but CO<sub>2</sub> above 2.5 kPa or O<sub>2</sub> levels below 1 kPa could injure lettuce. Brown stain on the mid ribs of leaves can be caused by levels of CO<sub>2</sub> of 2 kPa or higher, especially if combined with low O<sub>2</sub> (Haard and Salunkhe, 1975). Adamicki (1989) described successful storage of lettuce at 1°C in 3 kPa CO<sub>2</sub> + 1 kPa

O<sub>2</sub> for 21 days with less loss in AA than those stored in air. Kader (1985) recommended 0–5°C in 0 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> or 0–5°C in 0 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> (Kader, 1992), which had a good effect and was of some commercial use when carbon monoxide was added at the 2–3 kPa level. Saltveit (1989) recommended 0–5°C in 0 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> for leaf, head and cut and shredded lettuce; the effect on the former was moderate and on the latter it had a high level of effect. Storage in 0 kPa CO<sub>2</sub> with 1–8 kPa O<sub>2</sub> gave an extension in storage life and hypobaric storage increased storage life from 14 days in conventional cold stores to 40–50 days (Haard and Salunkhe, 1975).

### Lime (*Citrus aurantiifolia*)

CA storage recommendations were 0–10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> by SeaLand (1991). Pantastico (1975) recommended storage in 7 kPa O<sub>2</sub>, which reduced the symptoms of chilling injury; however, he showed that CA storage of ‘Tahiti’ limes increased rind scald decay and reduced juice content. Storage at 10–15°C in 0–10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> was recommended by Kader (1986) or 10–15°C in 0–10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Kader, 1989, 1992). These storage conditions were shown to increase the postharvest life of limes but were said to be not used commercially (Kader, 1985). CA storage was not considered beneficial but typical storage conditions were given as 10–15°C in 0–10 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> by Bishop (1996). Fruits were stored in air or in CA containing 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> and 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> by Sritananan *et al.* (2006). Those stored in air had a higher ethylene production and respiration rate than those in CA. All the CA storage conditions reduced the loss of chlorophyll and change in peel colour compared with fruits stored in air.

### Litchi, Lychee (*Litchi chinensis*)

Kader (1993) recommended 3–5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> at 7°C, with a range of 5–12°C, and

reported that the benefits of reduced O<sub>2</sub> were good and those of increased CO<sub>2</sub> were moderate. Vilasachandran *et al.* (1997) stored the cultivar 'Mauritius' at 5°C in air or in 5, 10 or 15 kPa CO<sub>2</sub> + 3 or 4 kPa O<sub>2</sub>. After 22 days all fruits were removed to air at 20°C for 1 day. Those stored in 15 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> were lighter in colour and retained TSS better than the other treatments, but had the highest levels of off-flavours. Those in all the CA storage treatments had negligible levels of black spot and stem-end rot compared with the controls. On the basis of the above, they recommended 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 4 kPa O<sub>2</sub>. Mahajan and Goswami (2004) stored the cultivar 'Bombay' at 2°C and 92–95% RH for 56 days in 3.5 kPa O<sub>2</sub> + 3.5 kPa CO<sub>2</sub>. Fruits retained their red colour, while those stored in air had begun to turn brown. Loss of acidity and AA content and the smallest increase of firmness and pericarp puncture strength of fruits were for those stored in CA compared with those stored in air. The sensory evaluation of aril colour and taste showed that the fruits held in CA were rated 'good' throughout the 56 days of storage. Ali *et al.* (2016) found that the cultivar 'Gola' stored for 35 days at 5 ± 1°C turned completely brown within 28 days in air, but in an atmosphere of 1 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> browning was delayed, antioxidant activity maintained and the fruit had better organoleptic properties.

### Mamey (*Mammea americana*)

Manzano-Mendez and Dris (2001) showed that storage of the cultivar 'Amarillo' at 15 ± 2°C in 1 kPa CO<sub>2</sub> + 5.6 kPa O<sub>2</sub> for 2 weeks retarded maturation.

### Mandarin, Satsuma (*Citrus spp.*)

These include the satsuma mandarin, *C. unshiu*, and common mandarin, *C. reticulata*. CA had only a slight or no effect on storage (SeaLand, 1991). At 25°C in 60 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> + 20 kPa N<sub>2</sub> the respiration rate was not affected, but in 80 kPa CO<sub>2</sub> + 20 kPa air and

90 kPa CO<sub>2</sub> + 10 kPa air, respiration rates were significantly reduced (Kubo *et al.*, 1989b). Ogaki *et al.* (1973) carried out experiments over a number of years and showed that the most suitable atmosphere for storing satsumas was 1 kPa CO<sub>2</sub> + 6–9 kPa O<sub>2</sub>. A humidity of 85–90% RH produced the best-quality fruit and resulted in only 3% weight loss. Pre-storage treatment at 7–8°C and 80–85% RH was also recommended. In a two-season trial, satsumas were stored for 3 months at 3°C and 92% RH in 2.8–6.5 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> + 93.5–97.2 kPa N<sub>2</sub> or in air. Weight losses were 1.2–1.5% in CA and 6.5–6.7% in air; and fruit colour, flavour, aroma and consistency were also better in CA. AA content was 5.9% higher in the flesh and 10.3% in the peel of CA-stored fruits compared with the controls (Dubodel and Tikhomirova, 1985). Satsumas were stored at 2–3°C and 90% RH in 3–6 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> or in air. The total sugar content of fruit flesh decreased in both CA- and air-stored fruit by 10.7% and 13.0%, respectively, and losses in the peel were even greater. Reducing sugars decreased during storage, especially in the flesh of CA-stored fruits and in the peel of control fruits (Dubodel *et al.*, 1984). Yang and Lee (2003) compared storage in air with 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> and 10.5 kPa CO<sub>2</sub> + 3.9 kPa O<sub>2</sub> and found that organic acids in fruit in all three CA conditions were higher than those in air until 60 days, but there were no differences between treatments after 120 days of storage. They found that 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> maintained the best quality among the CA conditions after 120 days and gave the best retention of firmness. Yang (2001), using the same CA conditions, reported that the TSS of fruit in cold storage increased until 2 months and thereafter sharply decreased, whereas TSS of CA-stored fruit increased slowly throughout the storage period of 120 days. It was concluded that sensory evaluation clearly showed that CA could extend the marketing period of satsumas and retain good flavour. Storage of *C. unshiu* in China in containers with a D45 M2 1 silicone window of 20–25 cm<sup>2</sup>/kg of fruit gave what they described as the optimum CO<sub>2</sub> concentration of < 3 kPa together with O<sub>2</sub> < 10 kPa (Hong *et al.*, 1983).

### Mango (*Mangifera indica*)

CA storage recommendations include: 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Pantastico, 1975; Sealand, 1991); 10°C with 90% RH in 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Lawton, 1996); 10–15°C in 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (Bishop, 1996); 10–15°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Kader, 1986); 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (Kader, 1989, 1992); and 13°C, with a range of 10–15°C, in 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>, or 5–7 kPa O<sub>2</sub> for south-east Asian varieties (Kader, 1993). Fuchs *et al.* (1978) described an experiment where storage at 14°C in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> kept the mangoes green and firm for 3 weeks. Upon removal they attained full colour in 5 days at 25°C, but 9% had rots. An additional week in storage resulted in 40% developing rots during ripening. Bleinroth *et al.* (1977) reported that fermentation occurred with alcohol production during storage at 8°C and > 10 kPa CO<sub>2</sub> for 3 weeks. After 38 days of storage of the cultivar 'Delta R2E2' at 13 ± 1°C harvested at the mature-green stage, ethanol, acetaldehyde and esters were significantly higher in fruit in 1.5 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub>, 1.5 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> than those in air. Storage in 3 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> appeared promising and resulted in no significant fermentation (Lalel and Singh, 2006). Kim *et al.* (2007) stored mature-green fruit at 10°C for 2 weeks in 3 kPa O<sub>2</sub> + 97 kPa N<sub>2</sub>, 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> + 87 kPa N<sub>2</sub> or in air and then ripened them at 25°C in air. They found that CA delayed TA and colour changes and the overall decline in polyphenolic concentration compared with fruit stored in air. Paull and Chen (2004) recommended 3–5 kPa O<sub>2</sub> and 5–10 kPa CO<sub>2</sub>, at 7–9°C and 90% RH. For Florida varieties the optimum conditions were reported to be 3–5 kPa O<sub>2</sub> and 5–10 kPa CO<sub>2</sub> (Bender *et al.*, 2000).

Detrimental effects of CA storage have also been reported. Storage in 1 kPa O<sub>2</sub> resulted in the production of off-flavour and skin discoloration but storage at 12°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> was possible for 20 days (Hatton and Reeder, 1966). Deol and Bhullar (1972) mentioned that there was increased decay in mangoes stored in either CA storage or MA packaging compared with those

stored non-wrapped in air. This is in contrast with the work of Wardlaw (1938), Thompson (1971) and Kane and Marcellin (1979), who all showed that either CA storage or MA packaging reduced postharvest decay of mangoes. However, this difference may be related to anthracnose (*Colletotrichum gloeosporioides*), since the fungus infects the fruit before harvest and begins to show symptoms as the fruit ripens. Since CA storage and MA packaging delay ripening, they would also be expected to delay the development of anthracnose symptoms. CA has been tested on ripe fruit. In a comparison of various conditions ranging from 1.6 to 20.7 kPa O<sub>2</sub> + 0.2 to 10.2 kPa CO<sub>2</sub> with the balance being N<sub>2</sub> in a flow-through system at 5°C, 15°C or 25°C. The optimum combination was found to be around 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for the suppression of the respiration rate of ripe 'Irwin' (Nakamura *et al.*, 2003).

CA storage conditions have been recommended for specific varieties as follows.

### Alphonso

Niranjana *et al.* (2009) showed that fruits stored in air at 8°C showed chilling injury symptoms. However, treatment with a fungicide or hot water treatment of 55°C for 5 min followed by storage at 8°C in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> resulted in fruit with no chilling injury symptoms. CA-stored fruit also retained their antioxidant levels and fresh, hard, green appearance and they ripened normally when subsequently removed to ambient conditions of 24–29°C with 60–70% RH. Sudhakar Rao and Gopalakrishna Rao (2009) stored mature-green fruit at 13°C in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, or 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. CA that contained 5 kPa O<sub>2</sub> significantly reduced the respiration and ethylene peaks, but 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> resulted in abnormal respiration and ethylene production. Storage in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> extended storage life by 4 or 5 weeks. The fruit then ripened to good quality in ambient conditions of 25–32°C. These fruits had a bright yellow skin colour, high TSS and

total carotenoid and sugar content, and were firm and acceptable organoleptically.

### Amelie

Storage at 11–12°C in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 4 weeks was reported to reduce storage rots and give the best eating quality compared with storage in air (Medlicott and Jeger, 1987).

### Banganapalli

Sudhakar Rao and Gopalakrishna Rao (2009) stored mature-green fruit at 13°C in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>, or 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. CA storage in 5 kPa O<sub>2</sub> significantly reduced the respiration and ethylene peaks during ripening, but CA containing 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> resulted in abnormal respiration and ethylene production. In 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> storage life was extended by 5 weeks, followed by an additional week to become fully ripe in ambient conditions of 25–32°C. Fruits stored in optimum 5 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> ripened normally to a bright yellow skin colour, with high TSS, total carotenoid and sugar contents, and were firm with an acceptable organoleptic quality.

### Carabao

Storage in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for 35–40 days was recommended by Mendoza (1978).

### Carlotta

Storage recommendations include 10–11°C in 2 kPa O<sub>2</sub> + 1 or 5 kPa CO<sub>2</sub> for 6 weeks (Medlicott and Jeger, 1987) and 8°C in 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> for 35 days (Bleinroth *et al.*, 1977).

### Delta R2E2

Bleinroth *et al.* (1977) reported that fermentation of fruit occurred with alcohol

production during storage at 8°C and > 10 kPa CO<sub>2</sub> for 3 weeks. After 38 days of storage at 13 ± 1°C harvested at the mature-green stage, ethanol, acetaldehyde and esters were significantly higher in fruit in 1.5 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub>, 1.5 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> than those in air.

### Jasmin

Storage in 10–11°C in 2 kPa O<sub>2</sub> + 1 or 5 kPa CO<sub>2</sub> for 6 weeks (Medlicott and Jeger, 1987) and 8°C in 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> for 35 days were recommended (Bleinroth *et al.*, 1977).

### Julie

Storage at 11–12°C in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 4 weeks was said to reduce storage rots and give the best eating quality compared with those stored in air (Medlicott and Jeger, 1987).

### Haden

Storage at 10–11°C in 2 kPa O<sub>2</sub> + 1 or 5 kPa CO<sub>2</sub> for 6 weeks (Medlicott and Jeger, 1987) and 8°C in 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> for 30 days (Bleinroth *et al.*, 1977) was recommended. Sive and Resnizky (1985) found that ripening was delayed at 13–14°C but lower temperatures resulted in chilling injury. They reported that in CA fruit could be kept in good condition for 6–8 weeks at 13–14°C.

### Keitt

Storage at 13°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> was recommended, after which fruit ripened normally in air (Hatton and Reeder, 1966). Sive and Resnizky (1985) found that ripening was delayed at 13–14°C but lower temperatures resulted in chilling injury. They reported that in CA they could be kept in good conditions for 8–10 weeks at 13–14°C. ‘Keitt’ mangoes were stored at 7°C in 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> + 84 kPa N<sub>2</sub>, 8 kPa CO<sub>2</sub> + 6 kPa

O<sub>2</sub> + 86 kPa N<sub>2</sub>, 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> + 90 kPa N<sub>2</sub> or air at 7°C, 10°C, 13°C or 21–24°C. Of those that were tested the optimum conditions were 7°C under 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> for up to 6 weeks. However, it was claimed that those stored at 13°C in air for up to 21 days were better and showed no chilling injury effect although there was no mention that chilling injury occurred at 7°C (Hailu, 2016). Yahia and Tiznado Hernández (1993) stored Keitt for 0–5 days at 20°C in a continuous flow of 0.2–0.3 kPa O<sub>2</sub> (balance N<sub>2</sub>). Fruits were evaluated during storage and again after being held in air at 20°C for 5 days. There was no fruit injury or reduction in organoleptic quality due to the low O<sub>2</sub> atmosphere and fruits ripened normally. These results indicate that applying low O<sub>2</sub> atmospheres postharvest can be used to control insects in mangoes without adversely affecting fruit quality.

### Kensington

Within the concentrations studied by McLauchlan *et al.* (1994), the optimum atmosphere appeared to be around 4 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> but they suggested that further research was required below 2 kPa O<sub>2</sub> and above 10 kPa CO<sub>2</sub>, as well as at lower storage temperatures to control softening. They also reported that colour development was linearly retarded by decreasing O<sub>2</sub> from 10 kPa to 2 kPa and increasing CO<sub>2</sub> from 0 kPa to 4 kPa; there was no further effect from 4–10 kPa CO<sub>2</sub>. Fruit from storage in low O<sub>2</sub> atmospheres also had high TA after storage, but after 5 days in ambient conditions acid levels fell and they continued to develop typical external colour. In trials by O'Hare and Prasad (1993) over two seasons, over-ripeness was more common after 5 weeks at 13°C (and 10°C in the first season) and was associated with a decline in TA. They found that chilling injury occurred after 1 week at 5°C, but storage in 5 or 10 kPa CO<sub>2</sub> alleviated chilling injury symptoms. Storage in 5 kPa O<sub>2</sub> had no significant effect on chilling injury. Increased pulp ethanol concentration was associated with CO<sub>2</sub> injury in 10 kPa CO<sub>2</sub>, but 3 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> at 13°C

proved to be beneficial for extending the storage life of 'Kensington Pride' by up to 6 weeks, with good fruit quality and maintaining a high concentration of the major volatile compounds responsible for the aroma of ripe mangoes (Singh and Zaharah, 2015). Fresh-cut slices from ripe 'Kensington' mango were prepared aseptically and stored at 3°C. Storage in 2.5 kPa O<sub>2</sub> was effective at controlling tissue darkening and the development of a 'glassy' appearance, while increasing CO<sub>2</sub> levels to 5–40 kPa had little positive effect on shelf-life, appearing to promote tissue softening. A combination of low oxygen and calcium allowed 'Kensington' slices to be held for at least 15 days at 3°C. Green-mature 'Kensington Pride' mangoes were stored at 13°C in 90 l chambers containing 2 kPa O<sub>2</sub> + 3 kPa, 6 kPa and 9 kPa CO<sub>2</sub> or air in one experiment and O<sub>2</sub> at 1 kPa, 2 kPa and 3 % factorily combined with CO<sub>2</sub> at 6 kPa and 8 % for 21 and 35 days in both experiments. The fruits were allowed to ripen at 21±1°C to eating soft stage. Increased concentrations of CO<sub>2</sub> increased the concentration of total fatty acids as well as palmitic acid, palmitoleic acid, stearic acid and linoleic acid and caused the ratio of palmitic to palmitoleic acid to be more than 1 and reduced the production of monoterpenes, sesquiterpenes and aromatics in the pulp of ripe fruit. An increased concentration of CO<sub>2</sub> also increased the production of total esters as well as major individual esters, especially ethyl butanoate. Storage in 2 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> or 3 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> gave the best results in extending the postharvest life while maintaining a high concentration of the major volatile compounds responsible for the aroma of ripe mangoes (Lalel and Singh, 2004).

### Kent

Antonio Lizana and Ochagavia (1997) reported that storage at 12°C in 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> increased postharvest life to 29 days, which was 8 days longer than when stored in air. Storage of mature unripe fruit for 21 days at 12°C showed that CO<sub>2</sub> levels of 50 kPa and 70 kPa resulted in high levels of ethanol production rates and symptoms of



CO<sub>2</sub> injury, while storage in 3 kPa O<sub>2</sub> appeared to have little effect on ethanol production (Bender *et al.*, 1994).

### Maya

Sive and Resnizky (1985) found that at 13–14°C ripening was delayed and lower temperatures caused chilling injury, but in CA at 13–14°C the fruit could be kept in good conditions for 6–8 weeks.

### Rad

Storage at 13°C in combinations of 4, 6 and 8 kPa CO<sub>2</sub> + 4, 6 and 8 kPa O<sub>2</sub> was studied. Ripening was delayed by CA storage for 2 weeks compared with storage in air, with 4 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> giving the highest acceptability by taste panellists (Noomhorm and Tiasuwan, 1988).

### San Quirino

Storage at 10–11°C in 2 kPa O<sub>2</sub> + 1 or 5 kPa CO<sub>2</sub> for 6 weeks (Medlicott and Jeger, 1987) or at 8°C in 10 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> for 35 days was recommended (Bleinroth *et al.*, 1977).

### Tommy Atkins

Antonio Lizana and Ochagavia (1997) reported that in storage at 12°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> there was a postharvest life of 31 days. Sive and Resnizky (1985) found that at 13–14°C ripening was delayed and lower temperatures caused chilling injury but in CA the fruit could be kept in good conditions for 6–8 weeks. Storage of mature unripe fruit for 21 days at 12°C showed that CO<sub>2</sub> levels of 50 kPa and 70 kPa resulted in high levels of ethanol production rates and symptoms of CO<sub>2</sub> injury, while the 3 kPa O<sub>2</sub> concentration seemed to have little effect on ethanol production (Bender *et al.*, 1994).

CA storage can be used to control pest infestation. In storage trials to control fruit fly in mango fruits, it was found that fruits exposed to 50 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for 5 days or 70–80 kPa CO<sub>2</sub> + < 0.1 kPa O<sub>2</sub> for 4 days did not suffer adverse effects when they were subsequently ripened in air (Yahia *et al.*, 1989). CA in combination with other treatments has been shown to reduce disease. Immersing fruit in hot water (52°C for 5 min) plus benomyl (Benlate 50 WP 1 g/l) provided good control of stem-end rot on mangoes following inoculation with either *Dothiorella dominicana* or *Lasiodiplodia [Botryodiplodia] theobromae* during storage for 14 days at 25–30°C. In the same storage conditions prochloraz (as Sportak 45EC), DPXH6573 (40EC), RH3866, (25EC) and calcium chloride did not control stem-end rot (*D. dominicana*). During long-term storage at 13°C in 5 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> for 26 days followed by air for 11 days at 20°C, benomyl at 52°C for 5 min followed by prochloraz at 25°C for 30 s provided effective control of stem-end rot and anthracnose. The addition of guar gum to hot benomyl improved control of stem-end rot in the combination treatment. Benomyl at 52°C for 5 min alone was ineffective. Other diseases, notably Alternaria rot (*A. alternata*) and dendritic spot (*D. dominicana*), emerged as problems during CA storage. *A. alternata* and dendritic spot were also controlled by benomyl at 52°C for 5 min followed by prochloraz. *Penicillium expansum*, *Botrytis cinerea*, *Stemphylium vesicarium* and *Mucor circinelloides* were reported as postharvest pathogens of ‘Kensington Pride’ mango (Johnson *et al.*, 1990a, b).

### Mangosteen (*Garcinia mangosteen*)

Pakkasarn *et al.* (2003b) found that storage for 28 days at 13°C in 10 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> was effective in reducing ethylene production and calyx chlorophyll loss, delaying peel colour and firmness changes and decreasing weight loss and respiration rate. Fruit in air could be stored for only 16 days. Pakkasarn *et al.* (2003a) also found that fruit in 2 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> for 24 days developed

a fermented flavour. Storage at 13°C in 2–6 kPa O<sub>2</sub> + 10–15 kPa CO<sub>2</sub> had reduced weight loss and retarded peel colour development, softening and subsequent hardening, calyx chlorophyll loss, respiration rate and ethylene production compared with those stored in air.

### Medlar (*Mespilus germanica*)

Selcuk and Erkan (2015) stored medlars in air, MA packages and CA of 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> at 0°C for 60 days. Total phenolics concentrations, total flavonoids concentrations, antioxidant activity, AA retention and total condensed tannins retention were the highest in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>.

### Melon (*Cucumis melo*)

Few positive effects of CA storage have been shown. Kader (1985, 1992) recommended 3–7°C in 10–15 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for cantaloupes, which had a good effect on storage but was said to have limited commercial use. For honeydew, Kader (1985, 1992) recommended 10–12°C in 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>, which had a fair effect but was of no commercial use. Saltveit (1989) recommended 5–10°C in 10–20 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>, which had only a slight effect. Storage in 10–15 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> was recommended for cantaloupe and 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for honeydew and casaba (SeaLand, 1991). Typical storage conditions were given as 2–7°C in 10–20 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> by Bishop (1996) for cantaloupes. Christakou *et al.* (2005) stored galia in 30 kPa CO<sub>2</sub> + 70 kPa N<sub>2</sub> at 10°C, which helped to reduce the loss in quality. However, vitamin C decreased during storage for those stored in CA and those stored in air. Perez Zungia *et al.* (1983) compared storage of the cultivar ‘Tendral’ in 0 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> with storage in air and found that the CA-stored fruit had better overall quality. ‘Durango’ hybrid fruit were stored for 20 days at 10 or 15°C and supplied daily with a 5 min flow of air containing 5.1 kPa CO<sub>2</sub>. CO<sub>2</sub> treatment

resulted in a reduced TSS, firmness and acidity and increased the rate of colour change (Rodriguez and Manzano, 2000). CA had limited effects on disease. Mencarelli *et al.* (2003) carried out storage experiments on organically grown winter melons at 10°C and 95% RH for 63 days plus 5 days in air at 20°C. They concluded that CA storage in 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>, 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> had few beneficial effects compared with storage in air. Storage in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> resulted in some control of decay and the fruit stored at 2 kPa O<sub>2</sub> tasted better after storage and were sweeter and firmer than those stored in air. Martinez-Javega *et al.* (1983) stored the cultivar ‘Tendral’ in 12 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> at temperatures within the range of 2–17°C and found no effects on levels of decay or chilling injury compared with fruits stored at the same temperatures in air.

### Melon, Bitter (*Momordica charantia*)

Fruits of bitter melon were harvested at ‘horticultural maturity’ and stored for 2 weeks in various conditions. Fruit quality was similar after storage at 15°C in 21, 5 or 2.5 kPa O<sub>2</sub> + 0, 2.5, 5 or 10 kPa CO<sub>2</sub> or in air. Fruits stored for 3 weeks in 2.5 kPa O<sub>2</sub> + 2.5 or 5 kPa CO<sub>2</sub> showed greater retention of green colour and had less decay (unspecified) and splitting than those stored in air (Zong *et al.*, 1995). Bitter melons are reported to be subject to chilling injury by Dong *et al.* (2005). They immersed fruit in hot water at 42°C for 5 min, then packed them in commercial PE film bags and stored them at 4°C for 16 days with no indication of chilling injury.

### Mushroom (*Agaricus bisporus*)

Recommended optimum storage conditions include: 5–10 kPa CO<sub>2</sub> (Lutz and Hardenburg, 1968; Ryall and Lipton, 1979); 10–15 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> (SeaLand, 1991); and 0°C in 8 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> (Zheng and Xi, 1994). Kader (1985, 1992) and Saltveit (1989)

recommended 0–5°C in 10–15 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub>, which had a fair or moderate effect but was of limited commercial use. Marecek and Machackova (2003) found that storage in 1 or 2 kPa O<sub>2</sub> + 0 or 4 kPa CO<sub>2</sub> was best in keeping the fruit's quality and firmness. Storage with CO<sub>2</sub> levels of 10–20 kPa inhibited mould growth and retarded cap and stalk development, O<sub>2</sub> levels of < 1 kPa were injurious (Pantastico, 1975). Anonymous (2003) confirmed that high CO<sub>2</sub> prevented head opening at 12°C, but also found that atmosphere with high CO<sub>2</sub> concentration resulted in more cap browning and that O<sub>2</sub> concentration did not have any effect on colour change. Zheng and Xi (1994) found that colour deterioration was inhibited by storage at 0°C and also that low O<sub>2</sub> and 10 kPa CO<sub>2</sub> inhibited cap opening and internal browning but caused a 'yellowing' of the cap surface. Tomkins (1966) reported that 10 kPa CO<sub>2</sub> could delay deterioration but levels of over 10 kPa could cause a 'pinkish' discoloration. Button mushrooms were stored in MA packages with 95–100 kPa CO<sub>2</sub> and then ventilated after 0, 12, 24 or 48 h by puncturing the film at four corners. CO<sub>2</sub> exposure for 12 h reduced browning and increased antioxidant activity, which in turn maintained flavour, quality and consumer acceptance. In addition, the MDA content was significantly lower, while catalase and POD activities were significantly increased by high CO<sub>2</sub> treatment (Lin *et al.*, 2017).

### Mushroom, Cardoncello (*Pleurotus eryngii*)

Amodio *et al.* (2003) stored slices at 0°C either in air or in 3 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> for 24 days and found that those in CA had a better appearance than those in air. None of the treatments showed visible presence of microbial growth, though there was a slight increase in mesophilic and psychrophilic bacteria and yeasts for slices kept in air, while there was no increase, or a slight decrease, for those in CA. Lovino *et al.* (2004) also used 3 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> at 0°C but found a combination of CA plus microperforated plastic film to be the optimum. After 3 weeks and 48 h shelf-life, MA packaging in CA

storage showed higher marketable quality than the CA alone.

### Mushroom, Oyster (*Pleurotus ostreatus*)

CA storage was shown to have little effect on increasing storage life at either 1°C or 3.5°C, but at 8°C with a combination of 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 20 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> or 30 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> the oyster mushrooms had a reduced respiration rate and retained their quality for longer than those stored in air (Bohling and Hansen, 1989). Henze (1989) recommended storage at 1°C and 94% RH in 30 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for about 10 days and Pantastico (1975) recommended 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for 21 days.

### Mushroom, Shiitake (*Lentinus edodes*)

Storage at an unspecified temperature with 40 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> extended storage life four times longer than when stored in air (Minamide, 1981, quoted by Bautista, 1990). Storage at 0°C in 5 kPa, 10 kPa, 15 kPa or 20 kPa CO<sub>2</sub> + 1 kPa, 5 kPa and 10 kPa O<sub>2</sub> plus an air control was studied by Pujantoro *et al.* (1993). They found that 1 or 5 kPa O<sub>2</sub> levels resulted in poor storage. Respiration rates were suppressed by decreasing O<sub>2</sub> and increasing CO<sub>2</sub> concentrations at 5°C, 15°C, 20°C and 30°C. The RQ breakpoints were mainly controlled by O<sub>2</sub> concentration and were little affected by CO<sub>2</sub> concentration, suggesting that better keeping quality could be obtained from low O<sub>2</sub> atmosphere conditions above the RQ breakpoint at low temperature (Hu *et al.*, 2003).

### Natsudaidai (*Citrus natsudaidai*)

Storage in 60 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub> + 20 kPa N<sub>2</sub> had little or no effect on fruit quality (Kubo *et al.*, 1989a), but Kajiura and Iwata (1972) recommended storage at 4°C in 7 kPa O<sub>2</sub> + 93 kPa N<sub>2</sub>. They showed that fruit could be stored for 2 months at 3°C or 4°C or for 1 month at 20°C. At 3°C and 4°C, TSS and TA

were little affected by the O<sub>2</sub> concentration, but the fall in AA was less marked at low O<sub>2</sub>. However, O<sub>2</sub> levels below 3 kPa resulted in a fermented flavour but the eating quality was similar in fruit stored at all O<sub>2</sub> concentrations above 5 kPa. At 20°C there was more stem-end rot but *Penicillium* sp. development was reduced by low O<sub>2</sub>. Storage at 3°C and 4°C and below 1.5 kPa O<sub>2</sub> resulted in the albedo becoming watery and yellow and the juice sacs turning from orange to yellow, while respiration rate fell. Kajiura (1972) showed that fruit in 4°C had less button browning in 5 kPa CO<sub>2</sub> and above. TSS was not affected by CO<sub>2</sub> concentration, but TA was reduced in 25 kPa CO<sub>2</sub>. At 20°C, button browning was retarded above 3.5–5.0 kPa CO<sub>2</sub> but there were no differences in TSS content and TA between different CO<sub>2</sub> levels. An abnormal flavour and a sweet taste developed in storage above 13 and 5 kPa CO<sub>2</sub>, respectively. Above 13 kPa CO<sub>2</sub> the peel turned red. In subsequent work, Kajiura (1973) stored fruit at 4°C and either 98–100% RH or 85–95% RH in air mixed with 0, 5, 10 or 20 kPa CO<sub>2</sub> for 50 days and found that high CO<sub>2</sub> retarded button browning at both humidities and the fruit developed granulation and loss of acidity at 85–95% RH. In 98–100% RH combined with high CO<sub>2</sub> there was increased water content of the peel and the ethanol content of the juice but there was abnormal flavour. At 85–95% RH no injury occurred and CO<sub>2</sub> was beneficial, its optimum level being much higher. In another trial, fruit was stored in 0 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub>. In 0 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> there was little effect on button browning or acidity but granulation was retarded and an abnormal flavour and ethanol accumulation occurred. In 5 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub> the decrease in acidity was retarded and there was abnormal flavour development at high humidity but at low humidity it provided the best quality stored fruit.

### Nectarine (*Prunus persica* var. *nectarina*)

Storage recommendations are similar to peach, including 5 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> for 6 weeks (Hardenburg *et al.*, 1990). Storage at 0.5°C and 90–95% RH in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> was recommended for 14–28 days

(SeaLand, 1991). Other storage recommendations were given as 0–5°C in 5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (Kader, 1985) or 0–5°C in 3–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> which had a good effect, but CA storage had limited commercial use. The storage life at –0.5°C in air was about 7 days, but fruit stored at the same temperature in atmospheres containing 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> could be stored for 5–7 weeks (Van der Merwe, 1996). Folchi *et al.* (1994) stored the cultivar ‘Independent’ in < 0.1 kPa CO<sub>2</sub> + 0.3 kPa O<sub>2</sub> and the fruit showed an increase in aldehyde and ethanol content with increased storage time. Lurie (1992) reported that the cultivars ‘Fantasia’, ‘Flavourtop’ and ‘Flamekist’ stored well for up to 6 weeks at 0°C in an atmosphere containing 10 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, and internal breakdown and reddening were almost completely absent. Although this CA prevented internal breakdown and reddening, after extended storage fruit did not develop the increased TSS content or extractable juice during post-storage ripening that occurred in non-stored fruit. Therefore, while preventing storage disorders, CA did not reduce the loss of ripening ability that occurred during storage. Later, Lurie *et al.* (2002) suggested that a certain level of ethylene production was essential for normal ripening of nectarines after cold storage. They linked this with prolonged storage where they did not ripen and developed a dry, woolly texture. They found that the disorder can be alleviated by storing the fruit in the presence of exogenous ethylene.

Anderson (1982) stored the cultivar ‘Regal Grand’ at 0°C in 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for up to 20 weeks with intermittent warming to 18–20°C every 2 days, which almost entirely eliminated the internal browning associated with storage in air at 0°C. Conditioning the fruit prior to storage at 20°C in 5 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> for 2 days until they reached a firmness of 5.5 kg was necessary.

### Okra (*Hibiscus esculentus*, *Abelmoschus esculentus*)

Storage at 7.2°C in 5–10 kPa CO<sub>2</sub> kept okra in good condition (Pantastico, 1975) but 0 kPa

CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> was recommended by SeaLand (1991). Kader (1985) also recommended 8–10°C in 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> which had a fair effect but was of no commercial use, though CO<sub>2</sub> at 5–10% was beneficial at 5–8°C. Saltveit (1989) recommended 7–12°C in 4–10 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> but it had only a slight effect. Ogata *et al.* (1975) stored okra at 1°C in air or 3 kPa O<sub>2</sub> + 3, 10 or 20 kPa CO<sub>2</sub> or at 12°C in air or 3 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub>. At 1°C there were no effects of any of the CA treatments. At 12°C the CA storage resulted in lower AA retention but it improved the keeping quality. At levels of 10–12 kPa CO<sub>2</sub> off-flavour may be produced (Anandaswamy, 1963).

### Olive (*Olea europaea*)

Storage of fresh olives at 8–10°C in 5–10 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> had a fair effect, but was not being used commercially (Kader, 1985). Kader (1989, 1992) recommended 5–10°C in 0–1 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub>. In other work Kader (1989) found that green olives of the cultivar ‘Manzanillo’ were damaged when exposed to CO<sub>2</sub> levels of 5 kPa and above, but storage was extended in 2 kPa O<sub>2</sub>. In air they found that in storage life was 8 weeks at 5°C, 6 weeks at 7.5°C and 4 weeks at 10°C, but in 2 kPa O<sub>2</sub> the storage life was extended to 12 weeks at 5°C and 9 weeks at 7.5°C. Garcia *et al.* (1993) stored the cultivar ‘Picual’ at 5°C in air, 3 kPa CO<sub>2</sub> + 20 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 1 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. They found that 5 kPa O<sub>2</sub> gave the best results in terms of retention of skin colour, firmness and acidity, but had higher incidence of postharvest losses compared with those stored in air. They found no added advantage of increased CO<sub>2</sub> levels. Agar *et al.* (1997) compared storage of the cultivar ‘Manzanillo’ in air or 2 kPa O<sub>2</sub> + 98 kPa N<sub>2</sub> at 0°C, 2.2°C and 5°C. They found that fruit firmness was not affected by the CA storage compared with air storage, but decay was reduced in the low O<sub>2</sub> compared with air at all three temperatures. They concluded that the olives could be stored for a maximum of 4 weeks at 2.2–5°C in 2 kPa O<sub>2</sub>. Ramin and Modares (2009) found that at 7°C increasing the level of CO<sub>2</sub> in storage over the range of

1.6–6.4 kPa resulted in improved retention of firmness and colour and lowest respiration rate, with 2 kPa O<sub>2</sub> + 6.4 kPa CO<sub>2</sub> giving the best results.

### Onion (*Allium cepa* var. *cepa*)

CA stores with a total capacity of 9300 t have recently been constructed in Japan for onions (Yamashita *et al.*, 2009). Robinson *et al.* (1975) reported that low O<sub>2</sub> levels in storage reduced the respiration rate of bulbs at all temperatures studied (Table 12.2). CA storage conditions gave variable success and were not generally recommended, but 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> was shown to reduce sprouting and root growth (Hardenburg *et al.*, 1990). SeaLand (1991) recommended 0 kPa CO<sub>2</sub> with 1–2 kPa O<sub>2</sub>. Typical storage conditions were given as 0°C and 65–75% RH with 5 kPa CO<sub>2</sub> and 3 kPa O<sub>2</sub> by Bishop (1996). Monzini and Gorini (1974) recommended 4°C in 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for 6 months. Fellows (1988) recommended a maximum of 10 kPa CO<sub>2</sub> and a minimum of 1 kPa O<sub>2</sub>. Kader (1985) recommended a temperature range of 0–5°C and 75% RH in 1–2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>. In later work Kader (1989) recommended 0–5°C in 0–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub>. Saltveit (1989) recommended 0–5°C in 0–5 kPa CO<sub>2</sub> and 0–1 kPa O<sub>2</sub>, which was claimed to have had only a slight effect. He also found that CA storage gave better results when onions were stored early in the season just after curing.

Smittle (1988) found that more than 99% of the bulbs of the cultivar ‘Granex’ were marketable after 7 months of storage at 1°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, though the weight loss was 9%. Bulbs stored at 1°C in 5 kPa

**Table 12.2.** Effects of temperature and reduced O<sub>2</sub> level on the respiration rate (CO<sub>2</sub> production in mg/kg/h) of Bedfordshire Champion onions; adapted from Robinson *et al.* (1975).

	Temperature °C				
	0	5	10	15	20
Atmosphere					
Air	3	5	7	7	8
3 kPa O <sub>2</sub>	2		4		4

CO<sub>2</sub> + 3 kPa O<sub>2</sub> kept well when removed from storage, while bulbs from 10 kPa CO<sub>2</sub> + 3 and 5 kPa O<sub>2</sub> became unmarketable at a rate of about 15% per week due to internal breakdown during the first month out of store. Bulb quality, as measured by low pungency and high sugar, decreased slowly when onions were stored at 27°C or at 1°C or 5°C at 70–85% RH in CA. Quality decreased rapidly when the cultivar ‘Granex’ was stored in air at 1°C or 5°C. Adamicki (1989) also described successful storage of onions at 1°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. Adamicki and Kepka (1974) found that there were no changes in the colour, flavour or chemical composition of onions after 2 months of storage even in 15 kPa CO<sub>2</sub>. Chawan and Pflug (1968) examined storage of the cultivars ‘Dowing Yellow Globe’ and ‘Abundance’ in several combinations of 5 and 10 kPa CO<sub>2</sub> + 1, 3 and 5 kPa O<sub>2</sub> at 1.1°C, 4.4°C and 10°C. The best combination was 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at 4.4°C and the next best was 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> at 4.4°C. Internal spoilage of bulbs was observed in 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at 1.1°C but none at 4.4°C. Adamicki and Kepka (1974) also observed a high level of internal decay in bulbs stored in 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> at 1°C, but none at 5°C in the same atmospheric composition. The number of internally decayed bulbs increased with length of storage. Their optimum results for long-term storage were in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at either 1°C or 5°C. Stoll (1974) stated that there were indications that the early trimmed lots did not store as well in CA conditions as in conventionally refrigerated rooms, and he reported better storage life at 0°C rather than at 2°C or 4°C when the same 8 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> atmosphere was used. Sitton *et al.* (1997) also found that storage in high CO<sub>2</sub> caused injury but neck rot (*Botrytis* sp.) was reduced in CO<sub>2</sub> levels of greater than 8.9 kPa. They found that bulbs stored in 0.5–0.7 kPa O<sub>2</sub> were firmer, of better quality and had less neck rot than those stored in O<sub>2</sub> levels of < 0.7 kPa.

Waelti *et al.* (1992) converted a refrigerated highway trailer to CA storage and they found that the cultivar ‘Walla Sweet’ stored well for 84 days in 2 kPa O<sub>2</sub> but subsequently the shelf-life was only 1 week. Yamashita

*et al.* (2009) tested 1°C and 80% RH in 1 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> for storage up to 196 days and found that sprouting and root growth were inhibited in CA storage and 98.2% were considered marketable, compared with 69.2% for those stored at –0.5°C and 80% RH in air after 196 days. The onion cultivar ‘Hysam’ was stored in air or 2 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> or 2 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> at 0.5 ± 0.5°C and 80 ± 3% RH for 9 weeks. Both the CA levels resulted in reduced pungency and flavour. This was by reducing both flavour precursors and enzyme activity (Uddin and MacTavish, 2003).

Storage of calçots (floral stems of the second-year onion *Allium cepa* L.) for 60 days at 1°C in 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> resulted in higher consumer preference and higher antioxidant activity (DPPH radical dot assay), while those stored in air had higher antioxidant activity (FRAP assay), total phenolic content and total flavonoids (Zudaire *et al.*, 2017).

### Physiological disorders

CO<sub>2</sub> injury or internal spoilage of onion bulbs is a physiological disorder that can be induced by elevating the CO<sub>2</sub> concentration around the bulbs. This effect was aggravated by low temperature (< 5°C). Chawan and Pflug (1968) observed internal spoilage of bulbs in storage at 1.1°C in < 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, but there was no spoilage in the same CA at 4.4°C. It was stated that ‘internal spoilage of the bulbs was due to an adverse combined effect of temperature and gas concentration’. Also Adamicki and Kepka (1974) reported that Böttcher had stated that for the cultivars ‘Ogata’ and ‘Inoue’ there was very strong internal spoilage at concentrations of 10 kPa CO<sub>2</sub> and above. They also found that after storage at 1°C (but not at 5°C) for more than 220 days there were 23–56% of internally spoiled bulbs when CO<sub>2</sub> concentrations were 10% or higher. Adamicki *et al.* (1977) found that internal breakdown was observed in 68.6% of the bulbs when they were stored at 1°C in sealed PE bags with a CO<sub>2</sub> concentration higher than 10 kPa; also they found similar disorder in bulbs stored at 5°C in sealed PE bags, and

this was probably due to the very high concentration of CO<sub>2</sub> (28.6 kPa). Adamicki and Kepka (1974) indicated that internal spoilage of onions was due to the combined effects of high CO<sub>2</sub> concentrations, low temperature and a relatively long period of storage. Smittle (1989) found that the cultivar 'Granex' stored at 5°C in 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for 6 months had internal breakdown of tissue due to CO<sub>2</sub> toxicity.

### Sprouting

CA storage is being increasingly used for onions since it can replace MH, which is used for sprout suppression. Previously Isenberg (1979) had concluded that CA storage was an alternative to the use of sprout suppression in onions, but stressed the need for further testing of the optimum condition of O<sub>2</sub> and CO<sub>2</sub> required for individual varieties. Adamicki and Kepka (1974) also quoted work where onions stored in an atmosphere with 5–15 kPa CO<sub>2</sub> at room temperature showed a decrease in the percentage of bulbs that sprouted. They also reported that bulbs of cultivar 'Wolska' stored for 163 or 226 days in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> and then transferred to 20°C sprouted about 10 days later than those transferred from air storage. There is evidence that CA storage can have residual effects after the bulbs have been removed from store. Bulbs of cultivar 'Wolska' stored for 163 or 226 days in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> and then transferred to 20°C sprouted about 10 days later than those transferred from air storage (Adamicki and Kepka, 1974). Yamashita *et al.* (2009) reported that storage at 1°C and 80% RH under 1 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> for up to 196 days inhibited sprouting and root growth. In 10 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at 4.4°C, Chawan and Pflug (1968) found that bulbs showed no sprouting after 34 weeks, while bulbs stored in air had 10% sprouting for 'Dowing Yellow Globe' and 15% for 'Abundance'. In a detailed study of 15 different cultivars and various CA combinations, Gadalla (1997) found that all the cultivars stored in 10 kPa CO<sub>2</sub> for 6 months or more had internal spoilage (but after 3 months) with more spoilage when 10 kPa

CO<sub>2</sub> was combined with 3 kPa or 5 kPa O<sub>2</sub>. All CA combinations, 1, 3 or 5 kPa O<sub>2</sub> + 0 or 5 kPa CO<sub>2</sub>, reduced sprouting but those combinations that included 5 kPa CO<sub>2</sub> were the most effective. Also the residual effects of CA storage were still effective after 2 weeks at 20°C. No external root growth was detected when bulbs were stored in 1 kPa O<sub>2</sub> compared with 100% for the bulbs in air. There was also a general trend to increased rooting with increased O<sub>2</sub> over the range of 1–5 kPa. Bulbs stored in 5 kPa CO<sub>2</sub> had less rooting than those stored in 0 kPa CO<sub>2</sub>. Both these latter effects varied between cultivars. Most of the cultivars stored in 1 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> and some cultivars stored in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> were considered marketable after 9 months of storage. He also found that CA storage was most effective when applied directly after curing, but a delay of 1 month was almost as good. It was therefore concluded that, with a suitable cultivar and early application of CA, it is technically possible to store onions at 0°C for 9 months without chemical sprout suppressants. Sitton *et al.* (1997) found bulbs stored in 0.5–0.7 kPa O<sub>2</sub> in cold storage sprouted more quickly when removed to 20°C, compared with those stored in O<sub>2</sub> levels of > 0.7 kPa.

### Orange (*Citrus sinensis*)

CA storage was not considered beneficial but typical storage conditions were given as 5–10°C in 0–5 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> by Bishop (1996). Storage at 5–10°C in 5 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> had a fair effect on storage but was not being used commercially (Kader, 1985). Kader (1989, 1992) recommended 5–10°C in 0–5 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub>. Recommended storage of the cultivar 'Valencia' grown in Florida was 1°C in 0 or 5 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub> for 12 weeks followed by 1 week in air at 21°C. In these conditions the oranges retained better flavour and had less skin pitting than those stored in air, but CO<sub>2</sub> levels of 2.5–5 kPa, especially when combined with 5 or 10 kPa O<sub>2</sub>, adversely affected flavour retention (Hardenburg *et al.*, 1990). SeaLand (1991) recommended 5 kPa

**Table 12.3.** The effects of CA storage on the quality of oranges; modified from Chase (1969).

O <sub>2</sub> kPa	CO <sub>2</sub> kPa	Rotting (%) <sup>a</sup>	Rind breakdown (%) <sup>a</sup>	Flavour score <sup>b</sup>
	Air	97	0	66
15	0	58	32	97
15	2½	28	13	95
15	5	62	0	93
10	0	67	49	74
10	2½	60	3	69
10	5	81	3	52
5	0	47	9	38
5	2½	36	2	20
5	5	79	0	5

<sup>a</sup>After 20 weeks storage at 1°C plus 1 week at 21°C.

<sup>b</sup>After 12 weeks storage at 1°C plus 1 week at 21°C.

CO<sub>2</sub> + 10 kPa O<sub>2</sub>. Smoot (1969) showed that for 'Valencia' those stored in the combinations of 0 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub> and 0 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> had less decay than those stored in air. However, ASHRAE (1968) showed that CA storage could have deleterious effects on fruit quality, particularly through increased rind injury and decay or on fruit flavour. Chase (1969) also showed that CA storage could affect fruit flavour, rind breakdown and rotting (Table 12.3).

### Pak Choi (*Brassica campestris* L. subsp. *chinensis* var. *communis*)

Harbaum-Piayda *et al.* (2010) stored pak choi at 2°C and 99% RH in either air or 1.5–2.5 kPa O<sub>2</sub> + 5–6 kPa CO<sub>2</sub>. The level of flavonoids increased more in CA than in air, but hydroxycinnamic acids were unaffected.

### Papaya (*Carica papaya*)

At 18°C, storage in 10 kPa CO<sub>2</sub> reduced decay (Hardenburg *et al.*, 1990). Storage recommendations include: 10–15°C in 5–8 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (Bishop, 1996); 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for surface transport (SeaLand, 1991); 10–15°C in 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> had a fair effect but was said to be 'not

being used commercially' (Kader, 1985); 10–15°C in 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (Kader, 1989, 1992); and 12°C or 10–15°C in 5–8 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (Kader, 1993). Sankat and Maharaj (1989) found that fruit of the cultivar 'Known You Number 1' stored at 16°C were still in an acceptable condition after 17 days of storage in 5 kPa CO<sub>2</sub> + 1.5–2 kPa O<sub>2</sub> compared with only 13 days in air. The cultivar 'Tainung Number 1' stored in the same conditions as 'Known You Number 1' had a maximum life of 29 days in CA storage and was considered unacceptable after 17 days stored in air. In Hawaii, storage in 2 kPa O<sub>2</sub> + 98 kPa N<sub>2</sub> was shown to extend the storage life of fruit at 10°C compared with storage in air at the same temperature (Akamine and Goo, 1969). Akamine (1969) showed that storage of 'Solo' in 10 kPa CO<sub>2</sub> for 6 days at 18°C reduced decay compared with fruit stored in air, but fruit stored in air after storage in 10 kPa CO<sub>2</sub> decayed rapidly. Akamine and Goo (1969) showed that storage at 13°C in 1 kPa O<sub>2</sub> for 6 days or 1.5 kPa O<sub>2</sub> for 12 days ripened about 1 day later than those stored in air. Chen and Paull (1986) also showed that storage in 1.5–5 kPa O<sub>2</sub> delayed ripening of 'Kapaho Solo' compared with those stored in air and that there were no further benefits on storage by increasing the CO<sub>2</sub> level to either 2 kPa or 10 kPa. Hatton and Reeder (1968) showed that after storage in 13°C for 3 weeks in 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> followed by ripening at 21°C, fruit were in a fair condition with little or no decay and were of good flavour. Cenci *et al.* (1997) showed that the cultivar 'Solo' stored at 10°C in 8 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> could be kept for 36 days and still have an adequate shelf-life of 5 days at 25°C. Fonseca *et al.* (2006) compared storage in 3 kPa O<sub>2</sub> + 3 or 6 kPa CO<sub>2</sub> with storage in air and found that the postharvest diseases of anthracnose, chocolate spot, stem-end rot and black spot caused more losses on both 'Sunrise Solo' and 'Golden' stored in air than in a CA with 6 kPa CO<sub>2</sub>. However, they also found that the advantages of the CA were not effective in fruit harvested in the Brazilian summer. A cultivar was developed at the Malaysian Agriculture Research and Development Institute called 'Paola'. It was shown to have



a shelf-life of an extra 2 weeks compared with 'Eksotika'. After 30 days in CA the fruit was shown to retain its quality and could be transported by sea-freight over long distances (Kwok, 2010). Also in Malaysia, Broughton *et al.* (1977) showed that scrubbing ethylene from the store containing the cultivar 'Solo Sunrise' had no effects on storage life but they recommended that 20°C with 5 kPa CO<sub>2</sub> with ethylene removal was the optimum storage condition for 7–14 days. They also found that chilling injury could occur at 15°C. In later work it was shown that ethylene application can accelerate ripening by 25–50% (Wills, 1990).

Techavuthiporn *et al.* (2009) found that shredded unripe papaya stored at 4°C in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> retained their quality for 5 days. Shredded unripe papaya is the major ingredient of 'Som-Tam' in Thailand, which is complicated to prepare and deteriorates quickly.

Storage at 20°C in 0 kPa CO<sub>2</sub> + < 0.4 kPa O<sub>2</sub> as a method of fruit fly control in papaya resulted in increased incidence of decay, abnormal ripening and the development of off-flavour after 5 days of exposure (Yahia *et al.*, 1989).

### Passionfruit (*Passiflora edulis*)

Yonemoto *et al.* (2004) stored hybrid passionfruit (*P. edulis* forma *edulis* × *P. edulis* forma *flavicarpa*) in 20.9, 23.0 or 29.0 kPa O<sub>2</sub> at either 20°C or 15°C. The acidity of fruit kept at 20°C decreased from 2.9% to 2.4%, 2.1% and 2.0% after 10 days of storage at 20.9, 23.0 and 29.0 kPa O<sub>2</sub>, respectively. There were no differences in acidity of fruit kept at 15°C after 5 days of storage, but acidity was significantly lower in fruit in 23.0 or 29.0 kPa O<sub>2</sub> after 10 days compared with 20.9 kPa.

### Peach (*Prunus persica*)

CA storage recommendations include: 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (SeaLand, 1991); 0–5°C in 3–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (Bishop, 1996);

0–5°C in 5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub>, which had a good effect but was claimed to be of limited commercial use (Kader, 1985); 0–5°C in 3–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> for both freestone and clingstone peaches (Kader, 1989, 1992); 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, which maintained quality and retarded internal breakdown for about twice as long as those stored in air (Hardenburg *et al.*, 1990); and –0.5°C in 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> for up to 4 weeks for canning (Van der Merwe, 1996). The storage life at –0.5°C in air was about 7 days. It was found that conditioning the fruits at 20°C in 5 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> for 2 days prior to storage until they reach a firmness of 5.5 kg (as measured with a Magness and Taylor penetrometer) was necessary. Wade (1981) showed that storage of the cultivar 'J.H. Hale' at 1°C resulted in chilling injury symptoms (flesh discoloration and soft texture), but storage at the same temperature in atmospheres containing 20 kPa CO<sub>2</sub> resulted in only moderate levels even after 42 days. Bogdan *et al.* (1978) showed that the cultivars 'Elbarta' and 'Flacara' could be stored at 0°C for 3–4 weeks in air or about 6 weeks in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. Truter *et al.* (1994) showed that the cultivars 'Oom Sarel', 'Prof. Neethling' and 'Kakamas' could be successfully stored at –0.5°C in either 1.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for 6 weeks with a mass loss of only 1%. This compared with mass losses of 20.7% and 12.6% for 'Oom Sarel' or 'Prof. Neethling', respectively, stored in air at the same temperature. CA storage, however, did not affect decay incidence but the fruit was of an acceptable quality even after 6 weeks of storage and was also considered to be still highly suitable for canning. Brecht *et al.* (1982) showed that there was a varietal CA storage interaction. They stored five cultivars at –1.1°C in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> and found that 'Loadel' and 'Carolyn' could be successfully stored for up to 4 weeks while 'Andross', 'Halford' and 'Klamt' could be stored for only comparatively short periods. Lurie *et al.* (2002) suggested that a certain level of ethylene production is essential for normal ripening after cold storage and linked this with prolonged storage where the fruit did not ripen and developed a dry, woolly texture.

CA storage plus intermittent warming to 18°C in air for 2 days every 3–4 weeks has shown promising results (Hardenburg *et al.*, 1990). Anderson (1982) stored the cultivar ‘Rio Oso Gem’ at 0°C in 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for up to 20 weeks with intermittent warming to 18–20°C every 2 days, which almost entirely eliminated the internal browning associated with storage in air at O<sub>2</sub>.

### Pear (*Pyrus communis*)

CA storage recommendations include: a maximum of 5 kPa CO<sub>2</sub> and a minimum of 2 kPa O<sub>2</sub> for the cultivar ‘Bartlett’ (Fellows, 1988); 0.5°C in 0.5–1.0 kPa CO<sub>2</sub> + 2–2.2 kPa O<sub>2</sub> (Koelet, 1992); 0–1 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> (SeaLand, 1991); 0°C and 93% RH in 0.5 kPa CO<sub>2</sub> + 1.5 kPa O<sub>2</sub> (Lawton, 1996); 0–5°C

in 0–1 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> (Kader, 1985); 0–5°C in 0–3 kPa CO<sub>2</sub> + 1–3 kPa O<sub>2</sub> (Kader, 1992); -1 to -0.5°C in <1 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for both (Conference) and (Concorde) (Johnson, 1994), -1°C to -0.5°C in < 1 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for ‘Conference’ (Sharples and Stow, 1986); 0.8–1 kPa CO<sub>2</sub> + 2–2.5 kPa O<sub>2</sub> or 0.1 kPa CO<sub>2</sub> or less + 1–1.5 kPa O<sub>2</sub> for ‘Anjou’ (Hardenburg *et al.*, 1990); 0.5–1°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for ‘Conference’ and ‘Doyenne du Comice’ and 0.5–1°C in 6 kPa CO<sub>2</sub> + about 15 kPa O<sub>2</sub> for ‘William’s Bon Chretien’ produced in England (Fidler and Mann, 1972); and 0°C in 2 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for ‘Conference’, ‘Doyenne du Comice’ and ‘William’s Bon Chretien’ produced in Switzerland (Stoll, 1972). Exposure of ‘Anjou’ to 12 kPa CO<sub>2</sub> for 2 weeks immediately after harvest had beneficial effects on the retention of ripening capacity (Hardenburg *et al.*, 1990).

**Table 12.4.** Recommended CA storage conditions for selected pear cultivars by country and region (Herregods, 1993).

	°C	kPa CO <sub>2</sub>	kPa O <sub>2</sub>
<b>Cultivar Comice</b>			
Belgium	0 to -0.5	< 0.8	2 to 2.2
France	0	5	5
Germany (Westphalia)	0	3	2
Italy	-1 to 0.5	3–5	3–4
Spain	-0.5	3	3
Switzerland	0	5	3
USA (Oregon)	-1	0.1	0.5
<b>Cultivar Conference</b>			
Belgium	-0.5	< 0.8	2–2.2
Denmark	-0.5 to 0	0.5	2–3
England	-1 to 5	< 1	2
Germany (Saxony)	1	1.1–1.3	1.3–1.5
Germany (Westphalia)	-1 to 0	1.5–2	1.5
Holland	-1 to -0.5	Much < 1	3
Italy	-1 to -0.5	2–4	2–3
Italy	-1 to -1.5	1–1.5	6–7
Slovenia	0	3	3
Spain	-1	2	3.5
Switzerland	0	2	2
<b>Cultivar Packham’s Triumph</b>			
Australia (South)	-1	3	2
Australia (Victoria)	0	0.5	1
Australia (Victoria)	0	4.5	2.5
Germany (Westphalia)	-1 to 0	3	2
New Zealand	-0.5	2	2
Slovenia	0	3	3
Switzerland	0	2	2

Richardson and Meheriuk (1989) reported that the cultivars 'Alexander Lucas' and 'Gellerts Butterbirne' were very sensitive to CO<sub>2</sub> and advised a maximum storage period of only 6 months and 4–5 months, respectively. Herregods (1993, personal communication) and Meheriuk (1993) compared recommendations for CA storage for selected cultivars from different countries (Table 12.4). Van der Merwe (1996) recommended specific conditions for storage of fruit produced in South Africa (Table 12.5). Recommended storage conditions for various pear cultivars were given by Richardson and Meheriuk (1989) (Table 12.6). Kupferman (2001b) also summarized storage conditions for pears (Table 12.7).

**Table 12.5.** Recommended CA storage conditions for pears at  $-0.5^{\circ}\text{C}$  (Van der Merwe, 1996).

Cultivar	kPa O <sub>2</sub>	kPa CO <sub>2</sub>	Duration
Bon Chretien	1	0	4 months
Buerré Bosc	1.5	1.5	4 months
Forelle	1.5	0–1.5	7 months
Packham's Triumph	1.5	1.5	9 months

Chen and Varga (1997) showed that storage in 0.5 kPa O<sub>2</sub> + < 0.1 kPa CO<sub>2</sub> resulted in a high incidence of scald and black speck. After 6½ months of storage at  $-0.5^{\circ}\text{C}$  the average percentage of 'Beurre d'Anjou' with symptoms of scald ranged from zero in fruit stored in 1.0 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> or 1.5 kPa O<sub>2</sub> + 0.5 kPa CO<sub>2</sub> to 2% in fruit stored in 2.5 kPa O<sub>2</sub> + 0.8 kPa CO<sub>2</sub> and 100% in fruit stored in air. For the cultivar 'Bartlett', Zlatić *et al.* (2016) showed that storage under 0.8 kPa O<sub>2</sub> + < 0.5 kPa CO<sub>2</sub> at  $-1^{\circ}\text{C}$  or  $+1^{\circ}\text{C}$  maintained firmness and suppressed synthesis of aroma volatiles and synthesis of esters, including two character-impacting compounds: methyl and ethyl (2*E*,4*Z*)-deca-2,4-dienoate. Synthesis of hexyl acetate was suppressed under ULO storage regardless of temperature, while ethyl acetate synthesis was suppressed only by ULO at  $-1^{\circ}\text{C}$ . The levels of most aroma volatiles were recovered after the following 10 days of shelf-life, though with significantly lower recovery for methyl and ethyl (2*E*,4*Z*)-deca-2,4-dienoate in fruit under ULO storage. Although synthesis of aroma volatiles was most suppressed under ULO at  $-1^{\circ}\text{C}$ , butyl and hexyl

**Table 12.6.** Recommended CA storage conditions for pears (Richardson and Meheriuk, 1989).

Cultivar	Temperature $^{\circ}\text{C}$	kPa O <sub>2</sub>	kPa CO <sub>2</sub>
Abate Fetel	-1 to 0	3–4	4–5
Alexander Lucas	-1 to 0	3	1
Anjou	-1 to 0	1–2½	0.03–2
Bartlett	-1 to 0	1–3	0–3
Buerré Bosc	-1 to 0	1–3	0.03–4
Buerré Hardy	-1 to 0	2–3	0–5
Clapps Favorite	0	2	0–1
Conference	-1 to 0.5	1½–3	0–3
Decana Comizio	0	3	4–5
Decana Inverno	0	3	5
Diels Butterbirne	-1 to 0	2	2
Doyenne du Comice	-1 to 1	2–3	0.8–5
Gellerts Butterbirne	-1 to 0	2	1
Kaiser Alexander	0 to 1	3	3–4
Kosui	-0.5 to 5	1–2	–
Nijiseiki	-1 to 1	3	≤ 1
Packham's Triumph	-1 to 1	1–3	0–5
Passe Crassane	-1 to 1	1.5–3	2–5
Tsu Li	-0.5 to 1	1–2	–
20th Century	-1 to 1	3	≤ 1
Williams Bon Chretien	-1 to 0	1–3	0–3
Ya Li	-0.5 to 0.5	4–5	Up to 5

**Table 12.7.** Optimum CA conditions for selected pear cultivars; modified from Kupferman (2001b).

Cultivar	Country	°C	kPa O <sub>2</sub>	kPa CO <sub>2</sub>	Storage life in months
Anjou	Washington State, USA	-0.5 to 0	1.5	0.3	9
Beurre Bosc	South Africa	-0.5	1.5	1.5	4
Conference	Netherlands	-1	2.5	0.7	7.5
Doyenne du Comice	Netherlands	-0.5	2.5	0.7	5
Doyenne du Comice	South Africa	-0.5	1.5	1.5	6
Doyenne du Comice	New Zealand	-0.5	2	< 1	3
Forelle	South Africa	-0.5	1.5	0	7
Josephine	South Africa	-0.5	1.5	1	8
Packham's Triumph	New Zealand	-0.5	2	< 1	5
Packham's Triumph	South Africa	-0.5	1.5	2.5	9
Rosemarie	South Africa	-0.5	1.5	1	5
Williams Bon Chretien	South Africa	0 to -0.5	1	0	4
Williams Bon Chretien	South Africa	0 to -0.5	1	0	4
Williams Bon Chretien	Washington State, USA	-0.5 to 1	1.5	0.5	4

acetate levels recovered better in fruit under ULO storage at  $-1^{\circ}\text{C}$  than at  $+1^{\circ}\text{C}$ . Acetaldehyde and nonanal were the principal aldehydes present, with levels that were higher in early-harvested fruit and in fruit stored at  $+1^{\circ}\text{C}$ . The cultivars ‘Swiss Bartlett’ and ‘Cold Snap’, with or without 1-MCP treatment, were stored at  $0^{\circ}\text{C}$  in air or CA (18 kPa  $\text{O}_2$  + 2 kPa  $\text{CO}_2$  or 2.5 kPa  $\text{O}_2$  + 2 kPa  $\text{CO}_2$ ) for at least 167 days. There was high incidence of senescent scald and internal breakdown in pears that had not been treated with 1-MCP during storage in air, but symptoms were reduced by both CA and 1-MCP, resulting in minimal to negligible incidence in 1-MCP + 2.5 kPa  $\text{O}_2$ . 1-MCP and CA reduced the rates of ethylene production and delayed softening and yellowing in ‘Swiss Bartlett’, but had negligible to slight effects with ‘Cold Snap’. In both cultivars,  $\gamma$ -aminobutyrate accumulated, though 1-MCP and CA slightly reduced the levels in ‘Cold Snap’ and 1-MCP increased levels in ‘Swiss Bartlett’ fruit. Ascorbate levels were rapidly depleted in ‘Cold Snap’, regardless of treatment; these levels were better maintained by 1-MCP in ‘Swiss Bartlett’ in all storage atmospheres. In both cultivars, glutathione concentrations and redox status fluctuated during storage, though these levels were generally higher in 1-MCP-treated fruit (Lum *et al.*, 2017).

### Pea, Garden Pea, Mangetout, Snow Pea, Sugar Pea (*Pisum sativum*)

Pantastico (1975) reported that in storage at  $0^{\circ}\text{C}$  peas in their pods could be kept in good condition for 7–10 days, while in 5–7 kPa  $\text{CO}_2$  + 5–10 kPa  $\text{O}_2$  they could be kept for 20 days. At  $0^{\circ}\text{C}$  their quality was better after 20 days in 5–7 kPa  $\text{CO}_2$  than in air (Tomkins, 1957). Storage in 5–7 kPa  $\text{CO}_2$  at  $0^{\circ}\text{C}$  maintained eating quality for 20 days (Hardenburg *et al.*, 1990). Fellows (1988) recommended a maximum of 7 kPa  $\text{CO}_2$  and a minimum of 5 kPa  $\text{O}_2$ . However, Suslow and Cantwell (1998) reported that there was no benefit in CA storage compared with storage in air at the same temperature. Monzini and Gorini (1974) recommended  $0^{\circ}\text{C}$  in 5 kPa  $\text{CO}_2$  + 21 kPa  $\text{O}_2$  for 20 days. Saltveit (1989) recommended  $0$ – $10^{\circ}\text{C}$  in 2–3

kPa  $\text{CO}_2$  + 2–3 kPa  $\text{O}_2$  for sugar peas, which had only a slight effect. Pariasca *et al.* (2001) showed that storage at  $5^{\circ}\text{C}$  in 5–10 kPa  $\text{O}_2$  + 5 kPa  $\text{CO}_2$  were the best conditions among those that they tested, since changes in organic acid, free amino acid and sugar contents and pod sensory attributes were slight. The appearance of pods stored in CA was much better than for those stored in air. They also found that 2.5 kPa  $\text{O}_2$  + 5 kPa  $\text{CO}_2$  and 5 kPa  $\text{O}_2$  + 10 kPa  $\text{CO}_2$  had a detrimental effect on quality, since the peas developed slight off-flavours, but this effect was partially alleviated after ventilation.

### Pepino (*Solanum muricatum*)

Storing mature fruit at  $5^{\circ}\text{C}$  in 5 kPa  $\text{O}_2$  + 15 kPa  $\text{CO}_2$  resulted in retention of colour and firmness compared with fruit stored in air (Prono-Widayat *et al.*, 2003). They also recommended a storage life of 3 weeks for ripe fruit stored in 5 kPa  $\text{O}_2$  + 20 kPa  $\text{CO}_2$ , which retained their quality.

### Persimmon (*Diospyros kaki*)

Also called sharon fruit and kaki. General storage recommendations include:  $0$ – $5^{\circ}\text{C}$  in 5–8 kPa  $\text{CO}_2$  and 3–5 kPa  $\text{O}_2$  but reported to have only a fair effect on storage and not commercially useful (Kader, 1985, 1992); 5–8 kPa  $\text{CO}_2$  + 3–5 kPa  $\text{O}_2$  (SeaLand, 1991); and  $0$ – $5^{\circ}\text{C}$  in 5–8 kPa  $\text{CO}_2$  + 3–5 kPa  $\text{O}_2$  (Bishop, 1996). Liamnimitr *et al.* (2018) determined the lower  $\text{O}_2$  limit of aerobic respiration as 0.65 kPa, based on the respiratory quotient breakpoint. Next, the parameters of a Michaelis–Menten type respiration model were estimated, and they were used to predict the  $\text{O}_2$  consumption rate at 1.5 kPa, which was set as the target  $\text{O}_2$  partial pressure inside the package and is considered safe to avoid fermentation. The results revealed that the  $\text{O}_2$  concentration at a steady state in the designed bulk MAP could be established at 1.3–1.6 kPa, which successfully corresponded with the target value. Furthermore, losses in flesh firmness and colour, and the external damage of fruits were reduced in the bulk MAP when compared to storage in the other two package

treatments. Overall, the bulk MAP designed could be used for ca. 4 months storage of 'Fuyu' persimmon fruit. The incidence of skin browning was reduced by storing fruit in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> with an ethylene absorber, compared with storage in air (Lee *et al.*, 1993). After 2 months of storage of 'Kyoto' at -0.5°C followed by 5 days at 20°C, optimum flesh firmness was maintained in 0.5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. The highest rot incidence was observed in fruit stored at 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>. Levels of CO<sub>2</sub> of 10–15 kPa increased skin blackening (Brackmann *et al.*, 2004). Storage of 'Rojo Brillante' in 97 kPa N<sub>2</sub> + air for 30 days at 15°C, retained 'commercial' firmness and removed astringency (Arnal *et al.*, 2008). Fruit from trees of the cultivar 'Triumph' that were sprayed with GA at 50 mg/l 2 weeks prior to harvest was 10 times less sensitive to ethylene than fruit from non-treated trees when subsequently stored at 10°C in air or 1.5–2.0 kPa CO<sub>2</sub> + 3.0–3.5 kPa O<sub>2</sub> (Ben Arie *et al.*, 1989). CA recommendations for the cultivar 'Fuyu' include: 1°C and 90–100% RH in 8 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (Hulme, 1971); 0°C in 5–8 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> for 5–6 months (Kitagawa and Glucina, 1984); and 0–2°C in 9 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for 5 months (Chung and Son, 1994). Neuwald *et al.* (2008) found that the best CA condition for 'Fuyu' was 0.5–1 kPa O<sub>2</sub> + 5–15 kPa CO<sub>2</sub> at -0.5°C for 3 months with a shelf-life at 20°C of 3 days. Fruit stored in 0.5–1 kPa O<sub>2</sub> showed less rot than fruit stored in > 10 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> or in air. Also skin browning was reduced when 0.5–1 kPa O<sub>2</sub> was combined with CO<sub>2</sub> up to 10 kPa, but 0.5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> increased skin browning. Martins *et al.* (2004) studied the effect of delaying implementing CA for up to 3 days on subsequent storage of 'Fuyu'. Storage conditions were 0 ± 0.5°C and 95 ± 3% RH throughout, and CA was 8 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>. They concluded that conservation and quality of the fruit were not significantly affected by the different periods of refrigerated storage in air prior to storage in CA. Lee *et al.* (2003b) identified four different types of browning disorders classified by symptoms and the surface zone affected: (i) typical top flesh browning called 'chocolate symptom' on the top; (ii) pitted specks scattered on the surface; (iii) flesh blotch

browning usually on the equatorial to bottom portion; and (iv) pitted blotch browning on the equatorial zone. They suggested that top flesh browning was primarily induced by low O<sub>2</sub> (not by high CO<sub>2</sub>) and that the critical level of O<sub>2</sub> was in a very narrow range. For external development, however, highly elevated CO<sub>2</sub> seemed to be a complementary factor. Speculative primary inducing factors for other disorders were: low O<sub>2</sub> for flesh blotch browning, and high CO<sub>2</sub> for pitted specks and pitted blotch browning.

### Pineapple (*Ananas comosus*)

Typical storage conditions were given as 8–13°C in 5–10 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> by Bishop (1996) but CA storage was not considered beneficial. Dull *et al.* (1967) also considered that CA storage had no obvious effect on fruit quality or the maintenance of fruit appearance. However, Akamine and Goo (1971) found that the storage life of the cultivar 'Smooth Cayenne' was significantly extended in 2 kPa O<sub>2</sub> at 7.2°C and Yahia (1998) tentatively recommended 2–5% O<sub>2</sub> + 5–10 kPa CO<sub>2</sub>. Kader (1989) recommended 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> at 10–15°C. Storage at 10–15°C in 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> had a fair effect but was not being used commercially (Kader, 1985). Kader (1989, 1992) recommended 8–13°C in 5–10 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub>. Kader (1993) also recommended 10°C, with a range of 8–13°C, in 5–10 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub>. SeaLand (1991) recommended 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. Paull and Rohrbach (1985) found that storage at 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or 3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> did not suppress internal browning (sometimes referred to as black heart) in the cultivar 'Smooth Cayenne' stored at 8°C. If fruit was exposed to 3 kPa O<sub>2</sub> in the first week of storage at 22°C followed by 8°C, it was shown that the symptom could be effectively reduced. Dull *et al.* (1967) described an experiment where fruit was harvested at Stage 4 maturity (< 12% shell colour) and stored in atmospheres containing 21, 10, 5 or 2.5 kPa O<sub>2</sub> with the balance N<sub>2</sub>. As the O<sub>2</sub> concentration decreased, so did the respiration rate. Where 5 or 10 kPa CO<sub>2</sub> was added to the atmosphere, the only noticeable effect was a slight further reduction

**Table 12.8.** Effects of CA storage on the frequency and level of internal browning score (where 0 = none and 5 = maximum) on ‘Smooth Cayenne’ pineapples stored at 8°C for 3 weeks and then 5 days at 20°C (Haruenkit and Thompson, 1996).

Composition (kPa)			Internal browning score
O <sub>2</sub>	CO <sub>2</sub>	N <sub>2</sub>	
1.3	0	98.7	0.8
2.2	0	97.8	0.8
5.4	0	94.6	3.8
1.4	11.2	87.4	0.3
2.3	11.2	86.5	0.3
20.8	0	79.2	4.1
LSD ( $p = 0.05$ )			1.4

in respiration rate. For fruit harvested at full maturity and stored at 8°C in 2 kPa O<sub>2</sub> + 0, 2 or 10 kPa CO<sub>2</sub> or in air, all the fruits were still in good condition after 3 weeks. However, in subsequent storage at 22°C, to simulate marketing, some fruits in all the treatments developed internal translucency within 3 days and fruits stored in air and 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> developed internal browning (black heart) (Haruenkit and Thompson, 1993). CA storage was shown to reduce but not eliminate internal browning (Table 12.8). For fresh-cut pineapple, Chonhenchob *et al.* (2007) found that an atmosphere of 6 kPa O<sub>2</sub> and 14 kPa CO<sub>2</sub> increased shelf-life by up to 7 days at 10°C.

### Plantain (*Musa*)

SeaLand (1991) recommended 2–5 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub>. Agoreyo *et al.* (2007) found that CA storage increases the postharvest life when compared with fruit stored in air and that ethylene production in CA was not detected. Maintaining high humidity around the fruit can help to keep it in the pre-climacteric stage so that where fruits were stored in moist coir dust or individual fingers were sealed in PE film they could remain green and pre-climacteric for over 20 days in Jamaican ambient conditions of about 28°C (Thompson *et al.*, 1972b). Such fruit ripened quickly and normally when removed from the plastic film. Since packing in moist coir proved as effective as MA packaging, it could

be that the effect of MA on plantains is due entirely or at least partially to humidity rather than changes in O<sub>2</sub> and CO<sub>2</sub> inside the bags (Thompson *et al.*, 1974d).

### Plum (*Prunus domestica*)

CA recommendations include: 0–5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (SeaLand, 1991); 0–5°C in 0–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> (Bishop, 1996); 0–5°C in 0–5 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> reported to have a good effect but was not being used commercially (Kader, 1985); 0 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> for 12 weeks for the cultivar ‘Soldam’ without any quality changes (Chung and Son, 1994); and 1°C in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for up to 3 weeks for ‘Opal’ and ‘Victoria’, picked when not fully ripe (Roelofs and Breugem, 1994). Van der Merwe (1996) recommended –0.5°C in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> for fruit ripened to a firmness of 5.5 kg before storage for up to 7–8 weeks, depending on cultivar. This recommendation was shown to vary between cultivars and for ‘Santa Rosa’ and ‘Songold’ it was recommended that they should be partially ripened to a firmness of approximately 4.5 kg, storage at 0.5°C in 4 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> for 7 or 14 days, which was sufficient to keep fruit in an excellent condition for an additional 4 weeks in air at 7.5°C. Internal breakdown was almost eliminated by this treatment and the fruit was of excellent eating quality after ripening (Truter and Combrink, 1992). Fruit of the cultivar ‘Angeleno’ was kept in air, 0.25% or 0.02 kPa O<sub>2</sub> at 0°C, 5°C or 10°C. Exposures to the low O<sub>2</sub> atmospheres inhibited ripening, including reduction in ethylene production rate, retardation of skin colour changes and flesh softening, but maintained TA and increased resistance to CO<sub>2</sub> diffusion. The most important detrimental effect of the low O<sub>2</sub> treatments was the development of an alcoholic off-flavour (Ke *et al.*, 1991a). Naichenko and Romanshchak (1984) recommended storage at –2°C for the cultivar ‘Vengerka Obyknovennaya’. At that temperature fruits were kept in good condition for only 19 days in air but they could be stored for up to 125 days in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> + 92 kPa N<sub>2</sub>. ‘Victoria’ plums at

different stages of ripeness were stored in 3 or 21 kPa O<sub>2</sub> and 0, 4 or 7 kPa CO<sub>2</sub> at 0.3°C or 0.5°C for up to 4 weeks. Some fruits were harvested unripe, others when not yet fully ripe but of good colour. In 3 kPa O<sub>2</sub>, fruit colour development was delayed, fruits remained firm and after 4 weeks they tasted better than those stored in 21 kPa O<sub>2</sub>. In high CO<sub>2</sub>, fruit colour development was also delayed and fruit flavour was not affected. Average wastage after 2, 3 and 4 weeks of storage was 9%, 20% and 32%, respectively (Roelofs, 1993a). With the cultivar 'Opal', the best storage conditions were 0°C in 1 or 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> and with the cultivar 'Monsieur Hatif' they were 0°C and 1 or 2 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub>. Both cultivars could be stored for up to 3 weeks without excessive spoilage due to rots and fruit cracking (De Wild and Roelofs, 1992). Folchi *et al.* (1994) stored the cultivar 'President' in < 0.1 kPa CO<sub>2</sub> + 0.3 kPa O<sub>2</sub> but observed an increase in aldehyde and ethanol content with increased storage time. Truter and Combrink (1997) stored the cultivars 'Laetitia', 'Casselmann' and 'Songold' at 0°C for 8 weeks in various CA storage conditions followed by 7 days in air at 10°C. The best treatments of those investigated were 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> but, for 'Songold', for a maximum of 7 weeks. Storage in 1 kPa O<sub>2</sub> at 0.5°C with intermittent warming was shown to have beneficial effects (Hardenburg *et al.*, 1990). In storage at 0.5°C physiological injury to the cultivar 'Monarch' was some 25% less in 2.5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> than in air (ASHRAE, 1968). Tonini *et al.* (1993) showed that there was a reduction in the incidence of internal breakdown in the cultivar 'Stanley' in atmospheres containing 20 kPa CO<sub>2</sub> compared with those stored in air, both at 0°C. Higher concentrations proved phytotoxic. For the cultivar 'Angelo', phytotoxicity was observed during storage at CO<sub>2</sub> concentrations higher than 2.5%. At 1°C in 12 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> the cultivar 'Buhler Fruhwetsche' had a good appearance, taste and firmness after 4 weeks of storage (Streif, 1989). They detected no injury in 'Buhler Fruhwetsche' during storage at CO<sub>2</sub> concentrations below 16 kPa.

Tahir and Olsson (2009) found that CA storage suppressed pathogenic decay but there was a difference in the response of different cultivars to CA storage. They recommended

0.5°C and 90% RH in 1 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> for 'Opal', 'Vallor' and 'Victoria' and 1 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> for 'Jubileum' and 'Vision'.

Good control of plum fruit moth *Cydia funebrana* and the fungus *Monilinia* spp. in fruit harvested in dry weather and at an optimum maturity stage could be achieved by storage for 1 month in air at 0°C and for 6 weeks in 2 kPa O<sub>2</sub> + 2.5 kPa CO<sub>2</sub>. Atmospheres with 1–1.5 kPa O<sub>2</sub> could limit browning of the flesh, maintain a crisp and juicy texture and slow down rot development (Westercamp, 1995).

### Pomegranate (*Punica granatum*)

Fruit of the cultivar 'Hicaz' was stored at 6°C, 8°C or 10°C and 85–90% RH in 1 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, 6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or in air. Those stored in air at 6°C were acceptable after 5 months of storage but those stored in 6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> retained good quality for 6 months. Fruit stored in air at 8°C or 10°C had a storage life of 50 days, whereas fruit stored at these temperatures in CA had a storage life of 130 days (Kupper *et al.*, 1994). CA storage has been shown to affect the development of a postharvest superficial browning disorder called 'husk scald'. The most effective control in the cultivar 'Wonderful' was by storing late-harvested fruit in 2 kPa O<sub>2</sub> at 2°C. However, this treatment resulted in accumulation of ethanol, which resulted in off-flavours, but ethanol and off-flavour dissipated when the fruit was transferred to air at 20°C (Ben Arie and Or, 1986). Kader *et al.* (1984) reported that pomegranate fruit did not respond to pre-storage ethylene treatment.

### Potato (*Solanum tuberosum*)

The amount of O<sub>2</sub> and CO<sub>2</sub> in the atmosphere of a potato store can affect the sprouting, rotting, physiological disorders, respiration rate, sugar content and processing quality of tubers (Table 12.9). Fellows (1988) recommended a maximum of 10 kPa CO<sub>2</sub> and a minimum of 10 kPa O<sub>2</sub>. Black heart developed in tubers held in 1 and 0.5 kPa O<sub>2</sub> at 15°C or 20°C (Lipton, 1967).



**Table 12.9.** Sugars (g/100 g dry weight) in tubers of three potato cultivars stored for 25 weeks under different CA at 5°C and 10°C and reconditioned for two weeks at 20°C (modified from Khanbari and Thompson, 1996).

Cultivar	Gas content		Sucrose	Reducing sugars	Non-reducing sugars	Sucrose	Reducing sugars	Non-reducing sugars
	kPa CO <sub>2</sub>	kPa O <sub>2</sub>						
Record	9.4	3.6	0.757	0.216	0.973	0.910	0.490	1.400
Record	6.4	3.6	0.761	0.348	1.109	1.385	1.138	2.523
Record	3.6	3.6	0.622	0.534	1.156	1.600	0.749	2.349
Record	0.4	3.6	0.789	0.510	1.299	0.652	0.523	1.175
Record	0.5	21.0	0.323	0.730	1.053	0.998	0.634	1.632
	<i>Mean</i>		0.650	0.488	1.138	1.109	0.707	1.816
Saturna	9.4	3.6	0.897	0.324	1.221	0.685	0.233	0.918
Saturna	6.4	3.6	0.327	0.612	0.993	0.643	0.240	0.883
Saturna	3.6	3.6	0.440	0.382	0.822	0.725	0.358	1.083
Saturna	0.4	3.6	0.291	0.220	0.511	0.803	0.117	0.920
Saturna	0.5	21.0	0.216	0.615	0.831	0.789	0.405	1.194
	<i>Mean</i>		0.434	0.473	0.907	0.729	0.271	1.000
Hermes	9.4	3.6	0.371	0.480	0.851	0.256	0.219	0.475
Hermes	6.4	3.6	0.215	0.332	0.547	1.364	0.472	1.836
Hermes	3.6	3.6	0.494	0.735	1.229	0.882	0.267	1.149
Hermes	0.4	3.6	0.287	0.428	0.715	0.585	0.303	0.888
Hermes	0.5	21.0	0.695	0.932	1.627	0.617	0.510	1.127
	<i>Mean</i>		0.412	0.682	1.094	0.741	0.354	1.095

### Mechanical properties

Olsen *et al.* (2003) reported that it appeared that tubers prior to storage had quantitatively stronger tissue compared with tubers after storage and that storage at 7°C in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> altered tissue mechanical properties as well as carbohydrate content and had physiochemical characteristics of tubers stored at 3°C in air.

### Flavour

Lipton (1967) found that 'White Rose' tubers stored at 0.5 kPa O<sub>2</sub> had a sour off-odour when raw, and this off-odour persisted when cooked. Aeration for 8 days following 0.5 kPa O<sub>2</sub> storage diminished this off-odour and off-flavour. The taste was influenced only slightly at 1 kPa O<sub>2</sub>, and no off-odour or off-flavour was found in tubers stored in 5 kPa O<sub>2</sub>.

### Sugars and processing

If potatoes are to be processed into crisps or chips, it is important that reducing sugars

and amino acids are at a level to permit the Maillard reaction during frying. This gives them an attractive golden colour and their characteristic flavour. If levels of reducing sugars are too high, they are too dark after frying and unacceptable to the processing industry (Schallenger *et al.*, 1959; Roe *et al.*, 1990). Potatoes stored at low temperatures (around 4°C) have higher levels of sugars than those stored at higher temperatures of 7–10°C (Khanbari and Thompson, 1994), which can result in the production of dark-coloured crisps. At low storage temperatures, potato tubers accumulate reducing sugars, but the effects of CO<sub>2</sub> can influence their sugar levels (Table 12.10).

With levels of 6 kPa CO<sub>2</sub> + 2–15 kPa O<sub>2</sub> in storage for 178 days at 8°C and 10°C, fry colour was very dark (Reust *et al.*, 1984). Daniels-Lake (2012) reported that, although darkening of fry colour was related to ethylene, this effect interacted with CO<sub>2</sub>. Mazza and Siemans (1990), studying storage in 1–3.2 kPa CO<sub>2</sub>, found that darkening of crisps occurred after there was a rise in CO<sub>2</sub> levels in stores. At levels of 8–12 kPa CO<sub>2</sub> the fry colour was darker than for tubers stored in air (Schmitz, 1991). At O<sub>2</sub> levels of 2 kPa

**Table 12.10.** Effects of CO<sub>2</sub> levels in storage on reducing sugar content in potatoes.

CO <sub>2</sub> %	Effect	Reference
4	Higher reducing sugars than in air	Workman and Twomey, 1969
5	Prevents accumulation of reducing sugars but increased accumulation of sucrose	Denny and Thornton, 1941
6	Accumulation of sugars, especially sucrose	Reust <i>et al.</i> , 1984
5–20	Initial reduction in reducing sugars but later increase to a higher level than air	Burton, 1989

sugar levels were reduced or there was no accumulation at low temperature, but 5 kPa O<sub>2</sub> was much less effective (Workman and Twomey, 1969). The cultivar ' Bintje ' were stored at 6°C in atmospheres containing 0, 3, 6 or 9 kPa CO<sub>2</sub> and 21, 18, 15 or 12 kPa O<sub>2</sub>. During the early phase, the unfavourable effect of high CO<sub>2</sub> on fry colour increased to a maximum at 3 months. In the later phase, fry colour deteriorated in tubers stored in zero CO<sub>2</sub>, and CO<sub>2</sub> enrichment had less effect (Schouten, 1994). Khanbari and Thompson (1994) stored 'Record' tubers at 4°C in 0.7–1.8 kPa CO<sub>2</sub> + 2.1–3.9 kPa O<sub>2</sub> and found that it resulted in a significantly lighter crisp colour, low sprout growth and fewer rotted tubers compared with in air. Storing tubers in anaerobic conditions of total N<sub>2</sub> prevented accumulation of sugars at low temperature but it had undesirable side effects on the tubers (Harkett, 1971). Gökmen *et al.* (2007) stored tubers of 'Agria' and 'Russet Burbank' at 9 ± 1°C and found that there was only a limited increase in the concentrations of sugars during 6 month of storage in air, but the concentrations of sugars increased in CA where the O<sub>2</sub> level was sufficiently low to increase the respiration rate.

### Sprouting and disease

Schouten (1992) discussed the effects of CA storage on sprouting of stored potatoes. He found that sprout growth was stimulated in 3 kPa CO<sub>2</sub> at 6°C, whereas some inhibition of growth occurred in 6 kPa CO<sub>2</sub>. Sprout growth was strongly inhibited in 1 kPa O<sub>2</sub> at 6°C, but pathological breakdown may develop at this O<sub>2</sub> level. Internal disorders were found at O<sub>2</sub> levels of < 1 kPa. Stimulation of sprout growth occurred at slightly higher O<sub>2</sub> levels if CA

storage started from the beginning of the season. Stimulation was less if low O<sub>2</sub> conditions were applied from January onwards in Holland. It was concluded that CA storage at 6°C was not an alternative to chemical control of sprout growth for potatoes used for consumption. In the early phase of storage, Schouten (1994) showed that sprouting was stimulated by 3–6 kPa CO<sub>2</sub> and inhibited by 9 kPa CO<sub>2</sub>. During the later phase, all CO<sub>2</sub> concentrations inhibited sprouting. Khanbari and Thompson (1994) showed that high CO<sub>2</sub> with low O<sub>2</sub> combinations during storage completely inhibited sprout growth, but caused the darkest colour when the potatoes were processed into crisps. However, after reconditioning, tubers gave the same level of sprouting and crisps as light as the other CA storage combination. Reconditioning of stored tubers can reduce their sugar content (Table 12.11) and improve their fry colour (Khanbari and Thompson, 2004). Schouten (1994) recommended reconditioning tubers at 15°C for 2–4 weeks and showed that the treatment improved crisp fry colour, but he also found that reconditioning stimulated sprouting. Tubers stored in 0.7–1.8 kPa CO<sub>2</sub> + 2.1–3.9 kPa O<sub>2</sub> showed a significantly higher weight loss and shrinkage after reconditioning, compared with tubers that had previously been stored in air (Khanbari and Thompson, 1994). They also showed that tubers stored in high CO<sub>2</sub>, especially at 10% or 15%, had earlier onset of rotting. In 0.7–1.6 kPa CO<sub>2</sub> + 2–2.4 kPa O<sub>2</sub> there was also an increase in tuber rotting.

### Quince (*Cydonia oblonga*)

Quince is a pome fruit that shows a climacteric respiratory pattern and flesh browning

**Table 12.11.** Sugar content of tubers from three potato cultivars after being stored for 25 weeks in 9.4% CO<sub>2</sub> + 3.7% O<sub>2</sub> and another 20 weeks in air at 5°C and then reconditioned for two weeks at 20°C; modified from Khanbari and Thompson (1996).

Reconditioning	Cultivar	Sugars g/100 g dry weight			Grey level <sup>a</sup>
		Fructose	Glucose	Sucrose	
None	Record	0.778	0.847	0.520	130.3
None	Saturna	0.660	0.685	0.540	136.8
None	Hermes	0.980	1.120	0.564	122.9
Two weeks at 20°C	Record	0.628	0.563	0.897	132.7
Two weeks at 20°C	Saturna	0.381	0.343	0.489	151.8
Two weeks at 20°C	Hermes	1.030	1.127	1.133	123.6

<sup>a</sup>This is for the fry colour of potato crisps where the minimum acceptable grey level = 149 or above.

is the most limiting factor in storage. It can be stored at 0–2°C for up to 6 months in air and for up to 7 months in 2 kPa O<sub>2</sub> + 3 kPa CO<sub>2</sub> at 2°C (Gunes, 2008).

### Radish (*Raphanus sativus*)

Low O<sub>2</sub> storage was shown to prolong post-harvest life (Haard and Salunkhe, 1975). Lipton (1972) found that storage of topped radishes at 5°C or 10°C in 1 kPa O<sub>2</sub> reduced sprouting by about 50% compared with storage in air. Although storage in 0.5 kPa O<sub>2</sub> further reduced sprouting, it resulted in physiological injury. Storage at 0.6°C in 5 kPa O<sub>2</sub> or at 5°C or 10°C in 1–2 kPa O<sub>2</sub> was recommended by Pantastico (1975). Saltveit (1989) recommended 0–5°C in 2–3 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> for topped radishes but indicated that CA storage had only a slight effect.

### Rambutan (*Nephellium lappaceum*)

Kader (1993) recommended 10°C, with a range of 8–15°C, in 7–12 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>. Storage decay could be controlled by dipping the fruit in 100 ppm benomyl fungicide. Boonyariththongchai and Kanlayanarat (2003b) reported that spintern and peel browning were the main causes of loss in quality and they tested atmospheres containing 1, 5, 10 and 20 kPa CO<sub>2</sub> on storage of the cultivar ‘Rong-rien’ at 13°C or 20°C and 90–95% RH. In air the fruits could be stored

for only 6 days at 20°C and 10 days at 13°C. CO<sub>2</sub> toxicity was observed at 20°C, but at 13°C all the CO<sub>2</sub> concentrations used were able to prolong storage life by at least 2–8 days, with 10 kPa CO<sub>2</sub> being optimum, prolonging storage life to 18 days. This was due to reduced rates of browning, weight loss, respiration rate and ethylene production. Ratanachinakorn (2001) found that the storage life at 13°C was 15 days in 2 or 4 kPa O<sub>2</sub>, 10 days in 6 or 21 kPa O<sub>2</sub>, 5 days in 1 kPa O<sub>2</sub> and 10 days in 5 or 10 kPa CO<sub>2</sub>. Fruit was injured in 20 and 40 kPa CO<sub>2</sub>. During storage there was no change on TSS, acidity or vitamin C. At 13°C in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> fruit could be stored for 16 days with spoilage of < 10%; in 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> spoilage was 23%. O’Hare *et al.* (1994) found that storage of cultivar ‘R162’ in 9–12 kPa CO<sub>2</sub> retarded colour loss and extended shelf-life by 4–5 days but storage in 3 kPa O<sub>2</sub> or 5 ppm ethylene did not significantly affect the rate of colour loss.

### Raspberry (*Rubus idaeus*)

There is strong evidence that CA storage with high levels of CO<sub>2</sub> reduced rots. Kader (1989) and Bishop (1996) recommended 0–5°C in 15–20 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> and Hardenburg *et al.* (1990) reported that storage in 20–25 kPa CO<sub>2</sub> retarded the development of rots. Haffner *et al.* (2002) also reported that storage in 10 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> or 10 kPa O<sub>2</sub> + 31 kPa CO<sub>2</sub> suppressed

rotting significantly compared with storage in air. The cultivars 'Glen Clova', 'Glen Moy', 'Glen Prosen' and 'Willamette' stored in 2 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> had reduced fruit rot (*Botrytis cinerea*), delayed ripening, slower colour development, slower breakdown of acids (including AA), slower breakdown of TSS and firmer fruits compared with those stored in air (Callesen and Holm, 1989). The black raspberry cultivar 'Bristol' was stored at 18°C or 0°C in air or 20 kPa CO<sub>2</sub>. Storage in 20 kPa CO<sub>2</sub> greatly improved postharvest quality at both temperatures by reducing grey mould development (Goulart *et al.*, 1992). Goulart *et al.* (1990) showed that when 'Bristol' black raspberry was stored at 5°C in 2.6, 5.4 or 8.3 kPa O<sub>2</sub> + 10.5 or 19.6 kPa CO<sub>2</sub> the mass loss was less in CA after 3 days than when stored in air. When fruits were removed after 3 days and held for 4 days in air at 1°C, deterioration was highest in those that had been stored in 2.6, 5.4 or 8.3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>. Fruits removed after 7 days retained their quality for up to 12 days at 1°C and showed least deterioration, with the 15 kPa CO<sub>2</sub> treatment giving the best results. In another experiment by Goulart *et al.* (1990), fruits of the cultivar 'Heritage' red raspberry were stored for 3 or 5 days at 5°C in 5 kPa O<sub>2</sub> + 95 kPa N<sub>2</sub>, 20 kPa CO<sub>2</sub>, in air or in plastic bags, where the equilibrium atmosphere was 5 kPa O<sub>2</sub> + 18.1 kPa CO<sub>2</sub>. The deterioration was greatest with fruits held in 5 kPa O<sub>2</sub>, followed by those in air, and no off-flavour was detected in raspberries held in a high CO<sub>2</sub> atmosphere. Robbins and Fellman (1993) recommended 0–0.5°C and 90–95% RH using high CO<sub>2</sub> and reduced O<sub>2</sub>. This was effective in reducing the incidence of rots and maintaining fruit quality. There was almost no reduction in vitamin C for raspberries stored in high CO<sub>2</sub> concentrations (10–30 kPa CO<sub>2</sub>), and reducing O<sub>2</sub> level had little effect (Agar *et al.*, 1997). Forney *et al.* (2015) compared storage in air with storage in 12.5 kPa CO<sub>2</sub> + 7.5 kPa O<sub>2</sub>, both at 1°C and 95% RH. They found that loss of fruit firmness was similar during storage in both CA and air but that CA storage strongly suppressed fruit decay of all the cultivars they tested, with the lag period extended to > 45 days, compared with 19–29 days for those

stored in air. They calculated the lag period of decay and physiological breakdown to the number of days until the first sign of deterioration.

### Redcurrant (*Ribes sativum*)

Storage of soft fruit at 20 kPa CO<sub>2</sub> was shown to slow their respiration rate, double their storage life, suppress fungi and reduce decomposition by 50% both at low temperature and during marketing (Hansen, 1986). He also reported that storage at 0°C required only one-third the concentration of CO<sub>2</sub> required at 7°C. Typical storage recommendations include: 0–5°C in 12–20 kPa CO<sub>2</sub> + 2–5 kPa O<sub>2</sub> (Bishop, 1996); 0.5°C and 90–95% RH in 2 kPa O<sub>2</sub> + 18 kPa CO<sub>2</sub> (Ellis, 1995); and 20 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> for more than 20 weeks (Roelofs, 1993b). The cultivars 'Rotet', 'Rondom', 'Rovada', 'Roodneus' and 'Augustus' were stored for 8–25 weeks at 1°C in 0, 10, 20 or 30 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or higher (not controlled, 15–21 kPa) O<sub>2</sub>. The earliest harvested berries were more susceptible to rots, but increased CO<sub>2</sub> concentration reduced rotting, while reduced O<sub>2</sub> levels only reduced fruit rotting at low CO<sub>2</sub> concentrations. Levels above 20 kPa CO<sub>2</sub> resulted in fruit discolouring in some cultivars after 13 weeks of storage and low O<sub>2</sub> concentrations further increased internal breakdown (Roelofs, 1993b). In storage of the cultivars 'Rovada' (late ripening) and 'Stanza' (mid-season ripening) at 0–1°C in air it was shown that they could be kept for 2–3 weeks, while in atmospheres containing 20 kPa CO<sub>2</sub> or more they could be kept for 8–10 weeks (Van Leeuwen and Van de Waart, 1991). The cultivar 'Rotet' was stored at 1°C in 10, 20 or 30 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or 10, 20 or 30 kPa CO<sub>2</sub> + > 15 kPa O<sub>2</sub>. The optimum CO<sub>2</sub> concentration was found to be 20 kPa. At 1°C fruit could be stored for 8–10 weeks in high CO<sub>2</sub> atmospheres, compared with 1 month for those stored in air (Agar *et al.*, 1991). Six cultivars were stored at 1°C in 1, 2 or 21 kPa O<sub>2</sub> + 0 or 10 kPa CO<sub>2</sub>, in all combinations, for 4½ months. Spoilage was

least in 1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> (Roelofs, 1992). The cultivars 'Rotet', 'Rondom', 'Rovada', 'Augustus', 'Roodneus', 'Cassa' and 'Blanka' were stored in combinations of 0, 20 or 25 kPa CO<sub>2</sub> + 2 or 18–21 kPa O<sub>2</sub> for up to 24 weeks. After 24 weeks, fruit stored in 0 kPa CO<sub>2</sub> at 1°C or –0.5°C had about 95% rotting. The average incidence of fungal rots was lowest (about 5%) at –0.5°C in 20 or 25 kPa CO<sub>2</sub> and increased to about 35% at 1°C in 25 kPa CO<sub>2</sub> and to about 50% at 1°C in 20 kPa CO<sub>2</sub>. Fruit quality, however, was adversely affected in 20 kPa CO<sub>2</sub> and high O<sub>2</sub>, but this effect was in contrast to the findings subsequently reported by Roelofs (1994). The cultivar 'Rotet' was stored at 1°C for 10 weeks in air, 10 kPa CO<sub>2</sub> + 19 kPa O<sub>2</sub>, 20 kPa CO<sub>2</sub> + 17 kPa O<sub>2</sub>, 30 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 20 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> or 30 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> by Agar *et al.* (1994c). They found that the chlorophyll content was most stable in high CO<sub>2</sub> and low O<sub>2</sub> and for fruit stored in air the decrease in chlorophyll content coincided with a decrease in fruit quality. There was almost no reduction in AA during storage in 10–30 kPa CO<sub>2</sub>; also reducing O<sub>2</sub> level had little effect (Agar *et al.*, 1997).

### Rocket (*Diplotaxis tenuifolia*)

Rocket leaves were stored at 4°C in air or 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> and air with 10 kPa CO<sub>2</sub> for up to 14 days. After 10 days, the sensory and microbiological quality of samples stored in air was not commercially acceptable but those in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> maintained visual quality and controlled aerobic mesophilic and psychrotropic microorganisms as well as coliforms. The total flavonoid content remained constant during storage or even increased at the end of the shelf-life in CA, but it was degraded in those in air. In addition, the total content of vitamin C was higher in those in CA than in those in air. A decrease in total antioxidant capacity occurred during storage, particularly in samples stored in air (Martínez-Sánchez *et al.*, 2006).

### Roseapple (*Syzygium malaccensis*)

Also called pomerac. Basanta and Sankat (1995) stored pomeracs at 5°C for up to 30 days in 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 4 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 8 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, 8 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, 11 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> or 14 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>. The optimum storage conditions were found to be 11 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> or 14 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub>, where flavour and appearance were maintained for 25 days. Previous work is quoted where storage at 5°C in air resulted in a postharvest life of 20 days.

### Salsify (*Tragopogen porrifolius*)

Storage in 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> at 0°C was recommended by Hardenburg *et al.* (1990).

### Sapodilla (*Manilkara zapota*)

CA storage recommendations were 20°C in 5–10 kPa CO<sub>2</sub> combined with scrubbing of ethylene (Broughton and Wong, 1979), though the optimum temperature for storage in air conditions varied from as low as 0°C (Thompson, 1996). In other work, fruits stored for 28 days at 10°C failed to ripen properly, indicating chilling injury (Abdul-Karim *et al.*, 1993).

### Soursop (*Annona muricata*)

It was shown that fruit in storage is damaged by exposure to low temperatures (Wardlaw, 1938). Snowdon (1990) recommended 10–15°C. No specific data on CA storage could be found, but in a comparison between 16, 22 or 28°C the longest storage period (15 days) occurred in 16°C with fruits packed in high-density PE film bags without an ethylene absorbent. This packaging treatment also resulted in the longest storage duration at the two higher temperatures, but the differences between treatments at those temperatures were much less marked (Guerra *et al.*, 1995).

### Spinach (*Spinaca oleraceae*)

Storage recommendations include: 0–5°C in 10–20 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> which had a fair effect but was of no commercial use (Kader, 1985); 0–5°C in 5–10 kPa CO<sub>2</sub> + 7–10 kPa O<sub>2</sub> which had only a slight effect (Saltveit, 1989); 10–20 kPa CO<sub>2</sub> + 21 kPa O<sub>2</sub> (SeaLand, 1991); and a maximum of 20 kPa CO<sub>2</sub> (Fellows, 1988). Wardlaw (1938) found that storage in high concentrations of CO<sub>2</sub> could damage the tips of leaves but Hardenburg *et al.* (1990) reported that storage in 10–40 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> retarded yellowing.

Young, tender spinach leaves 8–10 cm long and leaves harvested at the traditional length of more than 15 cm were stored at 7.5°C in air or in 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>. CA maintained initial appearance for up to 12 days but carotenoids were in a lower concentration than those that had been stored in air. Chlorophyll levels in both maturities of leaves was highest after storage in 5 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> but leaves showed rapid deterioration in visual quality even more than those stored in air (Martinez-Damian and Trejo, 2002). Storage in air at 2°C prevented a reduction in AA content and L-ascorbate peroxidase activity. AA content, L-ascorbate peroxidase activity and yellowing decreased rapidly in air at 25°C. CA storage at 10°C in 2 kPa O<sub>2</sub> + 3 or 10 kPa CO<sub>2</sub> controlled enzyme activity and contributed to the preservation of AA and L-ascorbate peroxidase activity (Mizukami *et al.*, 2003).

### Spring onion (*Allium cepa*)

Also called scallion and green onion. Optimum recommended storage conditions by Hardenburg *et al.* (1990) were 5 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> at 0°C for 6–8 weeks, while 10–20 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> was recommended by SeaLand (1991). Kader (1985) recommended 0–5°C in 10–20 kPa CO<sub>2</sub> + 1–2 kPa O<sub>2</sub> for green onions, but it had only a fair effect and was of limited commercial use. Saltveit (1989) recommended 0–5°C in 0–5 kPa CO<sub>2</sub> + 2–3 kPa O<sub>2</sub> which also had only a slight effect.

### Squash (*Cucurbita* spp.)

SeaLand (1991) recommended 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> which had a slight to no effect and Hardenburg *et al.* (1990) reported that storage at 5°C in low O<sub>2</sub> was of little or no value. However, Wang and Kramer (1989) found that storage in 1 kPa O<sub>2</sub> reduced chilling injury symptoms at 2.5°C compared with storage in air. Symptoms occurred after 9 days of storage compared with 3 days of storage in air. The *C. pepo* cultivar ‘Romanesco’ was stored for 19 days at 5°C in 21 ± 1 kPa O<sub>2</sub> + 0, 2½, 5 or 10 kPa CO<sub>2</sub> and then held for 4 more days at 13°C in air. High CO<sub>2</sub> levels reduced the respiration rate and the development of chilling injury symptoms at all three harvest maturities (lengths 16 cm, 20 cm and 22 cm). At the end of the 23-day storage period, 82% of the 22 cm fruits held in 10 kPa CO<sub>2</sub> appeared saleable, but they had a slight off-flavour and were soft; 79% of those held in 5 kPa CO<sub>2</sub> were saleable, firm and free from off-flavour; samples from storage in 2.5 kPa CO<sub>2</sub> and in air were unacceptable because of decay and pitting. It is suggested that CO<sub>2</sub> concentrations around 5 kPa may be useful for storing zucchini at about 5°C, a temperature that normally causes chilling injury in zucchini (Mencarelli, 1987b). The extent of surface pitting of the *C. pepo* cultivar ‘Ambassador’ stored at 2.5°C was less in 1 kPa O<sub>2</sub> than when stored in air (Wang and Ji, 1989). At 0°C those stored in 1 kPa O<sub>2</sub> had reduced chilling injury compared with those stored in air (Wang and Kramer, 1989). Mencarelli *et al.* (1983) also showed that storage of zucchini at 2.5°C resulted in chilling injury, but the effect could be delayed by reducing the O<sub>2</sub> to 1–4%. In later work *C. pepo* fruits stored for 2 weeks in 1, 2 or 4 kPa O<sub>2</sub> had reduced respiration rates and ethylene production, particularly at 5°C and 10°C. Both rates increased during subsequent storage in air for 2 days at 10°C, but much more in those that had been held previously at 2.5°C or 5°C than in those from 10°C. About 75% and 55%, respectively, of the burst in respiration rate during storage in air in samples from 2.5°C and 5°C was due to exposure to low temperature; the

remainder was attributed to the effect of low  $O_2$  levels. For ethylene production, the corresponding values were about 95% and 70%. All fruits stored at 5°C for 2 weeks were virtually free of chilling injury, surface mould, decay and off-flavour and almost all the fruits were rated good to excellent in appearance. About three-quarters of the fruits were still in this category after two additional days at 10°C. At 5°C, low  $O_2$  atmospheres had no effect. Storage at 2.5°C resulted in severe chilling injury, which was ameliorated by holding them in 1, 2 or 4 kPa  $O_2$  instead of 8 or 21 kPa  $O_2$ . The best quality maintenance of 'Astra' zucchini was in storage at 6°C in air and 3°C in 5 kPa  $CO_2$  + 3 kPa  $O_2$  or 3 kPa  $CO_2$  + 3 kPa  $O_2$  but storage at 3°C caused chilling injury of young fruits (10–16 cm long) after 2 weeks (Gajewski and Roson, 2001). Subsequently Gajewski (2003) reported that best results were obtained if 'Astra' zucchini was kept at 6°C in either 5 kPa  $CO_2$  + 3 kPa  $O_2$  or 3 kPa  $CO_2$  + 3 kPa  $O_2$ . Zheng *et al.* (2008) stored the zucchini cultivar 'Elite' at 5°C in 21, 100 or 60 kPa  $O_2$ . Squash in 100 kPa  $O_2$  had the lowest respiration rate and those in 60 kPa  $O_2$  had the lowest ethylene production. The transcript levels of antioxidative genes were relatively constant during storage in high  $O_2$ , but the expressions of alternative oxidase,  $O_2$  radical scavenging enzyme activities and total phenol content were slightly increased in 100 and 60 kPa  $O_2$ . They interpreted these changes in relation to high  $O_2$  increasing tolerance to chilling injury.

### Strawberry (*Fragaria* spp.)

CA recommendations include: 15–20 kPa  $CO_2$  + 10 kPa  $O_2$  (SeaLand, 1991); 1°C, 90% RH, 20 kPa  $CO_2$  and 17 kPa  $O_2$  (Lawton, 1996); 0–5°C in 15–20 kPa  $CO_2$  + 5–10 kPa  $O_2$  (Bishop, 1996); a maximum of 20 kPa  $CO_2$  and a minimum of 2 kPa  $O_2$  (Fellows, 1988); 1.5 kPa  $CO_2$  + 1.5 kPa  $O_2$  for 3 weeks compared with only 4 days in air (Van der Merwe, 1996); and 0–5°C in 15–20 kPa  $CO_2$  + 5–10 kPa  $O_2$  (Kader, 1985, 1989). Woodward and Topping (1972) showed that high levels of  $CO_2$  in the storage atmosphere at

1.7°C reduced rotting caused by infection with *Botrytis cinerea*, but levels as high as 20 kPa  $CO_2$  were injurious. However, they showed that there was a temperature interaction with  $CO_2$  levels. In  $CO_2$  concentrations up to 20%, fruit stored at 3°C was in good condition after 10 days, but after 15–20 days there was a distinct loss of flavour, which appeared to be the main limiting factor to storage in the UK. Storage in 10–30 kPa  $CO_2$  or 0.5–2 kPa  $O_2$  slowed respiration rate and reduced disease levels, but 30 kPa  $CO_2$  or < 2 kPa  $O_2$  might cause off-flavours (Hardenburg *et al.*, 1990). Storage in 20 kPa  $CO_2$  by Hansen (1986) slowed the respiration rate of 'Elsanta' more than that of 'Elvira' and doubled storage life of both cultivars compared with those stored in air.

Fruits of the cultivar 'Redcoat' were forced-air cooled and stored in 0, 12, 15 or 18 kPa  $CO_2$  for 0, 2, 7 or 14 days.  $CO_2$  increased firmness but fruits stored for more than 7 days softened rapidly in all  $CO_2$  concentrations and fruits stored for 14 days were as soft as those not exposed to  $CO_2$  at all. The amount of storage decay after 14 days was less in 12, 15 or 18 kPa  $CO_2$  than in 0.5 kPa  $CO_2$ . Organoleptic evaluation identified fruit quality differences between treatments but the results were not consistent (Smith *et al.*, 1993). 'Redcoat' fruit was stored at several temperatures and for various intervals in CA containing atmospheres within the ranges of 0–18 kPa  $CO_2$  + 15–21 kPa  $O_2$ . CA storage indicated that the addition of  $CO_2$  to the storage environment enhanced fruit firmness. Fruits kept in 15 kPa  $CO_2$  for 18 h were 48% firmer than non-treated samples were initially. Response to increasing  $CO_2$  concentration was linear. There was no response to changing  $O_2$  concentration. Maximum retention of firmness was at 0°C. In some instances, there was a moderate enhancement of firmness as time in storage increased.  $CO_2$  reduced the quantity of fruit lost due to rotting. Fruits that were soft and bruised at harvest became drier and firmer in a  $CO_2$ -enriched environment (Smith, 1992). Fruits of various strawberry cultivars were stored by Smith *et al.* (1993) at 0°C for 42 h in 15 kPa  $CO_2$  + 18 kPa  $O_2$  to study the effect on firmness. Compared with initial

samples and control samples stored in air, firmness was increased in 21 of the 25 cultivars evaluated. The CO<sub>2</sub> treatment had no effect on colour, TSS or pH. Fruits of 15 strawberry cultivars were picked at two different stages of maturity, pre-cooled or not, and stored for 5 or 7 days at 0°C or 20°C in 10 or 20 kPa CO<sub>2</sub>. Increased atmospheric CO<sub>2</sub> was associated with excellent control of storage decay caused by *Botrytis* and *Penicillium* species and it also slowed fruit metabolism, thus preserving aroma and quality (Ertan *et al.*, 1990).

Fruits of the cultivar 'Chandler' were kept in air, 0.25 kPa O<sub>2</sub>, 21 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> or 0.25 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> (balance N<sub>2</sub>) at 5°C for 1–7 days to study the effects of CA on volatiles and fermentation enzymes. Acetaldehyde, ethanol, ethyl acetate and ethyl butyrate concentrations were greatly increased, while isopropyl acetate, propyl acetate and butyl acetate concentrations were reduced by the three CA storage atmospheres compared with those of control fruits stored in air. CA storage enhanced pyruvate decarboxylase and alcohol dehydrogenase activities, but slightly decreased alcohol acetyl transferase activity. The results indicate that the enhanced pyruvate decarboxylase and alcohol dehydrogenase activities cause ethanol accumulation, which in turn drives the biosynthesis of ethyl esters; the increased ethanol concentration also competes with other alcohols for carboxyl groups for esterification reactions; the reduced alcohol acetyl transferase activity and limited availability of carboxyl groups due to ethanol competition decrease production of other acetate esters (Ke *et al.*, 1994b). Almenar *et al.* (2006) compared storage in air, 3 kPa CO<sub>2</sub> + 18 kPa O<sub>2</sub>, 6 kPa CO<sub>2</sub> + 15 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 11 kPa O<sub>2</sub>, and 15 kPa CO<sub>2</sub> + 6 kPa O<sub>2</sub> on 'Reina de los Valles' wild strawberry fruit. They found that 10 kPa CO<sub>2</sub> + 11 kPa O<sub>2</sub> at 3°C prolonged the shelf-life by maintaining the quality parameters within acceptable values, through inhibition of *Botrytis cinerea*, without significantly modifying consumer acceptance. Three-quarter-coloured 'Chandler' fruits responded better to storage at 4°C in 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> than fully red fruits, maintaining better

appearance, firmness and colour over 2 weeks of storage, while achieving similar acidity and TSS contents with minimum decay development. CA was more effective than air storage in maintaining initial anthocyanin and TSS contents of three-quarter-coloured fruits. Although three-quarter-coloured fruits darkened and softened in storage at 10°C, the CA-stored fruits remained lighter coloured and as firm as the at-harvest values of fully red fruits. CA maintained better strawberry quality than air storage even at an above-optimum storage temperature of 10°C, but CA was more effective at the lower temperature of 4°C (Nunes *et al.*, 2002). Agar *et al.* (1997) found that AA plus dehydroascorbic acid was reduced during storage in 10–30 kPa CO<sub>2</sub>. Reducing the O<sub>2</sub> concentration in the storage atmosphere in the presence of high CO<sub>2</sub> had little effect on the vitamin C content. After CA storage of strawberries, 16 proteins increased and 13 decreased in abundance, and after air storage 11 proteins increased and 17 decreased while expression profile of 12 proteins was significantly changed by both CA and air (Li *et al.*, 2015).

Sliced strawberries (cultivars 'Pajaro' and 'G 3') were dipped in solutions of citric acid, AA and/or calcium chloride and were stored in air or in CA for 7 days at 2.5°C followed by 1 day in air at 20°C. Whole fruits were also stored in the same conditions. The respiration rate of slices was higher than that of whole fruit at both temperatures. CA slowed the respiration rate and ethylene production of sliced fruits. Firmness of slices was maintained by storage in 12 kPa CO<sub>2</sub> and in 0.5 kPa O<sub>2</sub>, or by dipping in 1% calcium chloride and storing in air or CA storage (Rosen and Kader, 1989).

### Swede (*Brassica napus* var. *napobrassica*)

Also called rutabaga. CA storage was reported to have a slight to no effect on keeping quality (SeaLand, 1991). Coating with paraffin wax was reported to reduce weight loss but if this was too thick the reduced O<sub>2</sub> supply could result in internal breakdown (Lutz and Hardenburg, 1968).



### Sweetcorn, Baby Corn (*Zea mays* var. *saccharata*)

Recommendations include: 0–5°C in 10–20 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> which had a good effect but was of no commercial use (Kader, 1985); 0–5°C in 5–10 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> but it had only a slight effect (Saltveit, 1989); 10–20 kPa CO<sub>2</sub> + 2–4 kPa O<sub>2</sub> (SeaLand, 1991); a maximum of 20 kPa CO<sub>2</sub> (Fellows, 1988); and 80 kPa CO<sub>2</sub> + 20 kPa air at 15°C or 25°C, which reduced respiration rate (Inaba *et al.*, 1989). CA has been shown to preserve sugar content; for example, 5°C in 2 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> for 14 days preserved the highest sugar level and reduced deterioration in visual quality (Riad and Brecht, 2003). Ryall and Lipton (1972) also reported that storage in 5–10 kPa CO<sub>2</sub> retarded sugar loss while 10 kPa caused injury. CA of 2 kPa O<sub>2</sub> + 10 or 20 kPa CO<sub>2</sub> reduced respiration rate and maintained higher sugar levels during 10 days at 1°C of fresh-cut sweetcorn (Riad and Brecht, 2001). Storage in high CO<sub>2</sub> retarded conversion of starch to sugar (Wardlaw, 1938) but cobs may be injured with more than 20 kPa CO<sub>2</sub> or < 2 kPa O<sub>2</sub>, though in an atmosphere of 2 kPa O<sub>2</sub> the sucrose content remained higher than in air (Hardenburg *et al.*, 1990). Brash *et al.* (1992) stored cobs of the cultivar ‘Honey ‘n’ Pearl’ at 0°C in 2.5 kPa O<sub>2</sub> or levels of CO<sub>2</sub> in the range of 5–15% for 3–4 weeks followed by 3 days of shelf-life assessment at 15°C in simulated sea-freight export trials in New Zealand. Atmospheres of 2.5 kPa O<sub>2</sub> + 6–10 kPa CO<sub>2</sub> ensured that the cobs retained sweetness and husk leaves stayed greener for longer during shelf-life, compared with air storage. CA storage for 3 weeks also reduced insect numbers. The use of CA to control insect pests on sweetcorn harvested in New Zealand was studied by Carpenter (1993), who concluded that storage at 0–1°C alone was as effective as treatment with CA.

### Sweet Potato (*Ipomoea batatas*)

SeaLand (1991) indicated that CA storage had a slight or no effect compared with storage in air. Typical storage conditions were given as 0–5°C in 5–10 kPa CO<sub>2</sub> + 2–4% O<sub>2</sub>

by Bishop (1996). Storage in 2–3 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub> kept roots in better condition than those stored in air, but CO<sub>2</sub> levels above 10 kPa or O<sub>2</sub> levels below 7 kPa could give alcoholic or off-flavour to the roots (Pantastico, 1975; Hardenburg *et al.*, 1990). The cultivar ‘Georgia Jet’ was stored at 15.5°C in air at 85% RH or in a 7 kPa O<sub>2</sub> + 93 kPa N<sub>2</sub> atmosphere for 3 months. CA storage had no effect on moisture content, β-carotene or total lipids (Charoenpong and Peng, 1990). In storage for 7 days at 20°C in air, 1 kPa O<sub>2</sub> + 99 kPa N<sub>2</sub> or 100 kPa N<sub>2</sub>, the total soluble solid content increased but weak off-odours were detected in roots stored in 100 kPa N<sub>2</sub>. The intensity of off-odours increased as the concentrations of acetaldehyde and ethanol increased in roots during storage. Ethanol concentrations were higher than those of acetaldehyde, which remained low during storage in 1 kPa O<sub>2</sub> and in air, but increased greatly in roots stored in 100 kPa N<sub>2</sub> (Imahori *et al.*, 2007).

### Sweetsop, Sugar Apple (*Annona squamosa*)

Storage of the cultivar ‘Tsulín’, harvested at the green-mature stage, at 20°C in 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> had reduced ethylene production and delayed ripening compared with those stored in air (Tsay and Wu, 1989). Kader (1993) recommended 15°C, with a range of 12–20°C, in 5–10 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub>.

### Tomato (*Lycopersicon esculentum*)

CA recommendations include: 10–15°C in 3–5 kPa CO<sub>2</sub> + 3–5% O<sub>2</sub> (Bishop, 1996); a maximum of 2 kPa CO<sub>2</sub> and a minimum of 3 kPa O<sub>2</sub> (Fellows, 1988); 12°C in 2.5–5 kPa CO<sub>2</sub> + 2.5 kPa O<sub>2</sub> for 3 months (Monzini and Gorini, 1974); 12.8°C in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> (Parsons and Spalding, 1972); 12–20°C in 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for mature-green fruits and 8–12°C in 0 kPa CO<sub>2</sub> and 3–5 kPa O<sub>2</sub> for partially ripe fruit (Kader, 1985); 12–20°C in 2–3 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> (Saltveit, 1989); 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> at 12°C for fruits harvested at the yellow or ‘tinted stage’

(Wardlaw, 1938); 0 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for mature-green tomatoes or 0–3 kPa CO<sub>2</sub> + 3–5 kPa O<sub>2</sub> for breaker to light pink stages (SeaLand, 1991); 13°C in 9.1 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> for 60 days for the cultivar 'Criterium' harvested at the pink stage of maturity (Batu and Thompson, 1995); and 10°C with 90% RH in 8 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for up to 36 days for the cultivars 'Angela' and 'Kada' harvested at the mature-green stage (Vidigal *et al.*, 1979). In mature green and pink fruits, respiration rates and ethylene production were lower in CA than in air (Lougheed and Lee, 1989). At 12.8°C fruit colour and flavour were maintained in an acceptable condition for 6 weeks in 0 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>, while in other work 1–2 kPa CO<sub>2</sub> + 4–8 kPa O<sub>2</sub> at 12.8°C for breaker or pink fruit was recommended (Pantastico, 1975). Adamicki (1989) described successful storage of mature-green tomatoes at 12–13°C in 0 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub>, 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for 6–10 weeks. Parsons *et al.* (1974) harvested tomatoes at the mature-green stage and stored them at 12.8°C for 6 weeks and found that 0 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> was better than those stored in air. Fruits in air were fully red in 6 weeks, those in 0 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or those in 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> fruits were still pink, with the latter combination having an additional delaying effect on ripening. Dennis *et al.* (1979) stored fruits harvested at the mature green stage at 13°C and 93–95% RH in air, 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> or 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for 6–10 weeks. Two greenhouse-grown cultivars ('Sonato' and 'Sonatine') and three field-grown ('Fortuna', 'Hundredfold' and 'Vico') were used. CA-stored fruit were found to ripen more evenly and to have a better flavour when removed from storage than air-stored fruit, with the 5 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub> treatment giving the best-quality fruit. Mature-green fruits of the cultivars 'Saba' and 'Saul' were stored by Anelli *et al.* (1989) for 14 days at 12°C in a room equipped with an ethylene scrubber, or 40 days at 12°C in 3 kPa O<sub>2</sub> and 3 kPa CO<sub>2</sub>. At the end of storage the fruits had marketable percentages of 90% and 85% for ethylene scrubber and CA storage, respectively. *Phytophthora* decay appeared on some fruits by the end of CA storage. Tataru and Dobreanu (1978) showed that tomatoes kept

best in 3 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>. Greenhouse-grown mature-green and pink tomatoes (cultivar 'Veegan') were stored in air or in 0.5, 3.0 or 5.0 kPa O<sub>2</sub> in N<sub>2</sub> or argon. Colour development was not affected by the background gases N<sub>2</sub> or argon except at 3 kPa O<sub>2</sub>, where ripening was delayed by the argon mixture. All the levels of O<sub>2</sub> suppressed colour development in mature-green fruits compared with those stored in air, with maximum suppression in 0.5 kPa O<sub>2</sub>. Most fruits held in 0.5 kPa O<sub>2</sub> rotted after removal to air before colour development was complete.

Green-mature tomatoes of the cultivar 'Ramy' were stored for 20 days at 4°C or 8°C in 3 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> with or without ethylene removal, or in 1.5 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> with ethylene removal using potassium permanganate. Fruit ripening was delayed most in fruits stored in 1.5 kPa O<sub>2</sub> + 0 kPa CO<sub>2</sub> with ethylene removal (Francile, 1992). The colour of fruit harvested at the pink stage did not change when they were stored in 6.4 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> and 9.1 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> even after 50 days and in some cases after 70 days of storage. The red colour development of the tomatoes exposed to < 6.4 kPa CO<sub>2</sub> increased, whereas red colour decreased with CO<sub>2</sub> levels above 9.1 kPa during storage (Batu, 1995). Storage in atmospheres containing > 4 kPa CO<sub>2</sub> or < 4 kPa O<sub>2</sub> gave uneven ripening (Pantastico, 1975). Li *et al.* (1973) showed that O<sub>2</sub> levels below 1 kPa caused physiological injury during storage, owing presumably to fermentation. The minimum O<sub>2</sub> level for prolonged storage at 12–13°C was about 2–4 kPa O<sub>2</sub>, which delayed the climacteric peak. Salunkhe and Wu (1973a) showed that exposing 'green-wrap' fruits of the cultivar 'DX-54' to low O<sub>2</sub> atmospheres at 12.8°C inhibited ripening and increased storage life. Storage at 10 kPa O<sub>2</sub> + 90 kPa N<sub>2</sub> resulted in a storage life of 62 days, and in 3 kPa O<sub>2</sub> + 97 kPa N<sub>2</sub> it was 76 days. The maximum storage life was 87 days at 1 kPa O<sub>2</sub> and 99 kPa N<sub>2</sub>. Low O<sub>2</sub> atmospheres inhibited fruit chlorophyll and starch degradation and also lycopene, β-carotene and sugar synthesis. Parsons and Spalding (1972) inoculated fruit with soft rot bacteria and held them for 6 days at 12.8°C. Decay lesions were smaller on fruits stored in 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> than on those

stored in air, but CA storage did not control decay. However, although at 12.8°C inoculated tomatoes kept better in CA storage than in air, at 7.2°C or 18.3°C they kept equally well in CA storage and air.

Moura and Finger (2003) harvested 'Agri-set' at breaker maturity (< 10% red colour) and stored them in 2, 3, or 4 kPa O<sub>2</sub> or air for 14 days at 12°C then ripening in air at 20°C. They found that there were no differences in volatile content of the fruit except for *cis*-3-hexenal and 2 + 3-methylbutanol. The other volatiles they measured were acetone, methanol, ethanol, 1-penten-3-one, hexanal, trans-2-hexenal, trans-2-heptenal, MHO, *cis*-3-hexenol, 2-isobutylthiazole, 1-nitro-2-phenylethane, geranylacetone, and β-ionone levels among the treatments. Tomatoes were harvested at the mature-green and pink stages of maturity and then stored for 60 days in air or 3.2 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> or 6.4 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> or 9.1 kPa CO<sub>2</sub> + 5.5 kPa O<sub>2</sub> at 15°C for mature-green and 13°C for pink. Fruits stored in air at 15°C had significantly better flavour and were sweeter and more acceptable than those in any of the CA conditions (Batu, 2003).

### Turnip (*Brassica rapa* var. *rapa*)

CA storage of turnip was reported to have had only a slight to no effect (SeaLand, 1991).

### Turnip-rooted Parsley (*Petroselinum crispum* subsp. *tuberosum*)

Storage in 11 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub> helped to retain the green colour of the leaves during storage (Hardenburg *et al.*, 1990), though

optimum temperature for storage in air is about 0°C (Thompson, 1996).

### Watermelon (*Citrullus lanatus*)

It was stated (SeaLand, 1991) that CA storage had only a slight or no effect. This was supported by Radulovic *et al.* (2007) who said, 'Watermelons are exclusively fresh consumed, and CA storage does not offer any benefits to watermelon quality.' Fonseca *et al.* (2004) found that 'Royal Sweet' had an increased respiration rate in storage at 7–9°C in < 14 kPa O<sub>2</sub> compared with air and they therefore recommended storage at 1–3°C in > 14 kPa O<sub>2</sub> atmospheres, but they found limited applicability of CA or MA packaging. However, Tamas (1992) showed that the storage life of 'Crimson Sweet' was extended from 14–16 days in ambient conditions to 42–49 days by cooling within 48 h of harvest and storage at 7–8°C and 85–90% RH in 2 kPa CO<sub>2</sub> + 7 kPa O<sub>2</sub>. Changes in fruit appearance, flavour and aroma were minimal but improvement in the colour intensity and consistency of the flesh were noticeable.

### Yam (*Dioscorea* spp.)

Little information is available in the literature on CA storage but one report indicated that it had only a slight to no effect (SeaLand, 1991). Fully mature *D. opposita* tubers can be stored for long periods of time at high O<sub>2</sub> tensions at 5°C, which was also shown to reduce browning of tubers (Imakawa, 1967).

# 13

## Transport

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Airfreight is used to transport some high-value fruits, but the transport time is so short that CA conditions are not necessary, though small insulated airfreight containers, designed by Envirotainer Worldwide, were produced that used dry ice for cooling, which would have also increased the CO<sub>2</sub> content in the container. However, large and increasing amounts of fresh fruit and vegetables are transported by sea. Some ships are fitted with refrigerated holds (break bulk) but most is commonly by refrigerated (reefer) containers (Fig. 13.1). Reefer containers were first introduced in the 1930s but it was only in the 1950s that large numbers of reefers were transported on ships. Both ships' holds and reefer containers may be fitted with CA facilities.

International transport of fruit and vegetables has had its critics, especially in terms of its use of energy. Schäfer *et al.* (2014) investigated the carbon footprint of asparagus from four sources to supply the German market. Their product carbon footprint was 1.75 kg CO<sub>2</sub>e/kg (CO<sub>2</sub>e = carbon dioxide equivalent) for the locally grown asparagus, 0.82–1.3 kg CO<sub>2</sub>e/kg from Turkey or Greece by truck and 8 kg CO<sub>2</sub>e/kg airfreighted from Peru. Michaliský and Hooda (2015) compared apples, cherries, strawberries, garlic and peas airfreighted into the UK with those produced locally. They found that those imported from non-European countries contained, on average,

embedded greenhouse gas emissions of 10.16 kg CO<sub>2</sub>e/kg (production, airfreight and distribution within the UK) compared with emissions of 0.5 kg CO<sub>2</sub>e/kg for those produced and supplied locally. Berners-Lee (2010) reported the carbon footprint of one apple produced locally and seasonal as 10 g CO<sub>2</sub>e, and for shipped, cold-stored and inefficiently produced as 150 g CO<sub>2</sub>e. For a banana he gave 80 g CO<sub>2</sub>e imported and for oranges 90 g CO<sub>2</sub>e/kg shipped 2000 miles by boat and 500 g CO<sub>2</sub>e/kg by truck and 5.5 kg CO<sub>2</sub>e/kg by airfreight.

Kruger and Truter (2003) stated: 'During the last years, the use of CA storage has considerably contributed towards improving the quality of South African avocados exported to Europe.' Ludders (2002) suggested that the main obstacles to the successful marketing of cherimoya could be overcome by using CA containers for sea-freight transport. Harman (1988) recommended the use of CA containers for the New Zealand fruit industry and considerable research has been carried out by Lallu *et al.* (1997, 2005) and Burdon *et al.* (2005, 2008, 2009) relevant to CA transport of kiwifruit and avocados. Dohring (1997) claimed that:

...avocados, stone fruit, pears, mangoes, asparagus and tangerine made up over 70% of [CA] container volumes in recent years. Lower value commodities e.g. lettuce, broccoli, bananas and apples, make up a



**Fig. 13.1.** Reefer containers plugged in at the port awaiting loading on to a ship.

greater percentage of the overall global fruit and vegetable trade volumes but cannot absorb the added CA costs in most markets.

Bananas are shipped commercially from tropical and subtropical countries in the refrigerated holds of ships, some of which have CA equipment installed. For example, over 1000 t of bananas are shipped weekly from the West Indies in CA reefer ships. CA reefer containers are also used to a limited extent for bananas. Experimental reports published over several decades have indicated various gas combinations that have been suitable for extending the storage life of bananas. In 1997 there were problems with bananas ripening during transport on reefer ships from the West Indies to the UK. Tests were made using CA reefer containers

but eventually CA reefer ships were used, which solved the problem, but added about 2 US cents/kg to the shipping costs. At the present time the bananas are transported across the Atlantic from Central America and the West Indies to Europe, North Africa and the Middle East on reefer vessels using MA packaging, or otherwise CA vessels and CA containers.

Hansen (1986) reported that in transport of fruit and vegetables the benefits of reduced  $O_2$  were very good and those of increased  $CO_2$  were also good, in that they both delayed ripening and high  $CO_2$  also reduced chilling injury symptoms.  $CO_2$  in cylinders was used for soft fruit during transport in 20 kPa  $CO_2$ , but he reported that dry ice could be used, which is cheaper.

He also reported that storage at 0°C required only one-third of the concentration of CO<sub>2</sub> required at 7°C for soft fruit. DeEll (2002) reported that 15–20 kPa CO<sub>2</sub> + 5–10 kPa O<sub>2</sub> reduced the growth of *Botrytis cinerea* and other decay-causing organisms during transport of blueberry. SeaLand (1991) recommended 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> for surface transport of papaya. For transport of bananas, an atmosphere of 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> was successfully used.

### Sea Freight

The length of the postharvest life of many fruits is the limiting factor to sea freight. Rupavatharam *et al.* (2015) reported that, in New Zealand, feijoa growers harvest fruit at a stage of maturity that gives a storage potential of 4 weeks at 4°C but reliable sea freight from New Zealand to Asian, European and North American markets requires a postharvest life of some 6 weeks. Therefore they suggested the need to investigate the possibility of developing a more immature stage of harvest maturity to prolong the potential postharvest period. Singh and Zaharah (2015) found that mangoes could be stored for only 2–3 weeks at 12–13°C coupled with substantial losses in fruit quality. However, CA storage at 13°C in 3 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> extended the postharvest life of the cultivars ‘Kensington Pride’ and ‘R2E2’ by up to 6 weeks while retaining their good quality. This gave potential for sea-freight export.

### Gas environments in purpose-built CA reefer vessels

Purpose-built CA reefer vessels have been used for transport of banana cargoes for decades. The general configuration of reefer carriers is to have four holds (Holds 1 to 4) in sequence from stem to stern, each of which is also divided vertically into four (or possibly five) decks (Decks A to D or E). In a typical CA vessel, each hold is also segregated into two CA ‘zones’ so that Decks A

and B constitute a single CA gas control zone and Decks C and D/E constitute the second CA zone within the same hold. A vessel with four holds would, therefore, have 16 CA gas control zones. When transporting produce such as bananas, it is standard practice to close-pack palletized fruit on each deck. The air temperatures for each deck and the gas concentrations in each CA gas zone are monitored and controlled independently.

### Cooling

For maximum quality maintenance throughout the marketing period, it is crucial to achieve and maintain the optimum storage conditions as soon as possible. This is called the cold chain. In a study of the cold chain for blueberry exports, Paniagua *et al.* (2014) found that delays in cooling had a small effect on final product weight, whereas variation in storage temperature and atmosphere during simulated transport influenced both firmness and the incidence of rotting. Defraeye *et al.* (2016) investigated warm loading of citrus fruit into reefers for cooling during marine transport as an alternative to forced air pre-cooling as it could simplify logistics and be more economic. They found that it was theoretically possible to cool the produce in less than 5 days in a reefer container, but these cooling rates were not achieved in practice in their trials. They also commented that cooling performance of containers depended on the way in which the fruit was stowed and convectively cooled. They packed citrus fruits in ventilated boxes, pre-cooled them in the reefer container and compared a channelling configuration, which reduced airflow by-pass between pallets, with a horizontal configuration, which forced air horizontally through the pallets. The fruit was cooled within about 3 days to seven-eighths cooling time with standard ambient loading. The channelling configuration exhibited similar cooling behaviour but the fruit lost less moisture, had a longer shelf-life and was of better quality. The horizontal configuration performance was worse in all aspects. In the case of bananas, pre-climacteric fruit is loaded into containers or ships’ holds at

ambient temperatures. The cargoes are then subjected to rapid cooling with the objective to reduce the temperature of the return air circulating through the cargo to within 2.2°C (4°F) of the designated storage temperature (set point) within a few days of loading. The duration of this 'reduction period' provides a measure of the efficiency with which the fruit have been cooled. If the reduction period for any consignment of bananas is protracted, this suggests that the respiration rate of the fruit is super-optimal and, as a consequence, the green life of the fruit will probably be less than anticipated.

### Pest control

Controlling pests and diseases on crops being transported internationally is vital in preventing their introduction into countries where the specific pest or disease may not exist. This has led to legislation governing quarantine and other control measures.

### Methyl bromide

MB, a widely used agent in disinfestation procedures, was scheduled to be withdrawn in many countries in 2005 due to its ozone-depleting properties. It was reported by UNEP (1992) that MB was identified, under the authority of the World Meteorological Organization with the National Oceanic and Atmospheric Administration (NOAA) and National Aeronautics and Space Administration (NASA), as causing significant damage to the earth's protective ozone layer and is scheduled for global phasing out under the Montreal Protocol, an international treaty developed to protect the earth from the detrimental effects of ozone depletion. MB for pre-shipment and quarantine purposes was declared exempt from these restrictions. The concentration of MB required depends on the pests to be controlled as well as temperature. For example, Zoffoli and Latorre (2011) reported that for false red spider mite (*Brevipalpus chilensis*) a dose rate of 24 g/m<sup>3</sup> at greater than 26.5°C was required. Phillips *et al.* (2015) simulated shipping of rock melons

from Australia to New Zealand by sea- and airfreight. No eggs survived MB fumigation, but they took 3–20 days to die, whereas phytosanitary inspections of rock melons occurred within 2–7 days of fumigation, which could lead to misinterpretation of effectiveness. Delays were not influenced by MB concentration, but were significantly lengthened by cooler storage temperatures.

Fumigation with ethyl formate combined with high CO<sub>2</sub> for 2½ h in export cartons in simulated cool down from 15°C to 10°C were found to control long-tailed mealybug (*Pseudococcus longispinus*) and citrus mealybug (*Planococcus citri*) without damaging the fruit. These treatments were reported to satisfy 'generally regarded as safe' criteria and were suitable for use in grapes and grapefruits produced under organic certification (Lima, 2015).

### Irradiation

Follett (2014) reported that the USA, Australia and the International Plant Protection Convention had approved the generic radiation dose of 150 Gy for quarantine treatment of Tephritidae fruit flies. The USA has also approved the generic dose of 400 Gy for other insects, except pupae and adult Lepidoptera. A development was reported by Cetinkaya *et al.* (2016), who showed that DNA Comet Assay (single-cell gel electrophoresis assay) may be a practical quarantine control method for irradiated citrus fruit, since it has been possible to estimate the applied low doses as small as 100 Gy when it is combined with image analysis.

Hot air-assisted microwave disinfestation can be performed at farms or centralized packhouses, since the capital cost would be comparatively lower than vapour heat or ionizing radiation treatments. Gamage *et al.* (2015) investigated hot air (40°C) exposure of apples in combination with a pentagonal microwave system (giving an average temperature of 53.4 ± 1.3°C at the core of the fruit) that gave 100% mortality of the most tolerant stage of Queensland fruit fly (*Bactrocera tryoni*) and Jarvis's fruit fly (*B. jarvisi*) without any adverse impact on fruit quality. The development of small-scale

X-ray machines could provide farmers and packhouses with in-house treatment capability and accelerate adoption of the technology. Also a recent change in import regulations by the USA allows for treatment upon entry, which makes it possible for exporting countries to explore new markets prior to investing in treatment facilities at the origin (Bustos-Griffin *et al.*, 2015).

The effects of irradiation on fruit may be detrimental, depending on the dose, but it can also be beneficial. For example, Kang *et al.* (2012) found that X-ray irradiation at 200 and 400 Gy could reduce the respiration rate of grapes and extended their shelf-life. They found that there were no significant other physical or chemical effects of irradiation and the irradiated fruit had a better appearance than the non-irradiated fruit after 14 days of subsequent storage at  $1.5 \pm 0.5^\circ\text{C}$  and  $75 \pm 5\%$  RH.

### Ozone

A company, Purfresh Transport, developed a reefer container with patented technology that has a slow ozone input throughout the transport period. They claimed that ‘ozone molecules kill moulds, yeasts, and bacteria in the air and on surfaces by up to 99%, as well as to consume and regulate ethylene levels’. It was reported to have been successfully tested on avocados, bananas, berries, citrus fruit, cucumbers, grapes, kiwifruit, mangoes, melons, onions, peppers, pineapples, pome fruit, potatoes, stone fruit, tomatoes and ‘tropicals’ (Bordanaro, 2009). The company carried out a trial on transport of nectarines from California to Taiwan taking 14 days. They reported that the Purfresh container ‘outperformed the CA treatment on net weight loss, fruit pressure, brix content, and microbial counts’.

### Quarantine

In-transit pest control was described by Falik *et al.* (2012). They carried out experiments whose results provided the basis for the future establishment of an export quarantine

treatment protocol for capsicum fruit. This was based on in-transit storage at temperatures of  $1.5^\circ\text{C}$  for 21 days to eliminate Mediterranean fruit fly (*Ceratitis capitata*) eggs and larvae from capsicum fruit while maintaining the commercial quality of the fruit. However, in other work, it was reported that storage of capsicums at  $5^\circ\text{C}$  for 2 weeks resulted in chilling injury (González-Aguilar, 2002). Unahawutti *et al.* (2014) found that infestation of mangosteens with the quarantine pest oriental fruit fly (*Bactrocera dorsalis*) and carambola fruit fly (*B. carambolae*) could only occur if fruit had cracks or mechanical injury.

### Monitoring

The manner in which the quality of a crop is monitored and maintained relative to predetermined criteria during export can vary, but the random sampling of boxes at different points along the supply chain has been used for decades. For example, the Windward Islands banana industry in the 1990s used a comprehensive method (Table 13.1) where boxes were tested as they were being loaded on the ship in the Windward Islands and then another sample was taken for testing on arrival in the UK. The data was sent back to the exporting countries immediately so that any problem could be dealt with before the next weekly shipment. All the boxes had an identification code to enable the grower and packer to be identified.

In the past two decades there has been an impetus to develop technologies that would allow the remote monitoring of not only the location but also the condition of merchandise during transit. One of the first systems developed was that which deployed radio frequency identification technology (RFID). Morris *et al.* (2003) described technologies that had been developed for monitoring refrigerated shipping containers that made it possible for actual storage conditions during transport to be monitored in real time, thus forewarning shippers and carriers of potential problems with a consignment before its arrival at its destination or trans-shipping point. At that time, Morris discussed the possibility of integrating the



**Table 13.1.** Levels of defects allowed for Class 1 fruit as used by the banana export industry in the Windward Islands in the 1990s (0.4 inch<sup>2</sup> = 2.58 cm<sup>2</sup>).

Defect	Tolerance level per cluster
Superficial bruise	Up to 0.4 inch <sup>2</sup>
Latex staining	Up to 0.4 inch <sup>2</sup>
Maturity stain	Up to 0.4 inch <sup>2</sup>
Pest damage	Up to 0.4 inch <sup>2</sup>
Red rust damage	Up to 0.4 inch <sup>2</sup>
Scars	Up to 0.4 inch <sup>2</sup>
Sooty mould	Up to 0.4 inch <sup>2</sup>
Flower thrips	Up to 25%
Speckling/pinspotting	Up to 50%
Crown trimming	0
Damaged pedicel/ neck injury	0
Dirty	0
Finger end rot	0
Fused fingers	0
Misshapen/ malformed fingers	0
Mutilated fingers	0
Overgrade	0
Peel burn	0
Residue	0
Ripe and turning	0
Scruffy	0
Short finger	0
Stale fruit	0
Undeveloped	0
Undergrade	0

potential to monitor the physiological status of a consignment with storage life prediction models. At the time of writing, technological advances now allow for the routine real-time monitoring of O<sub>2</sub> and CO<sub>2</sub> concentration, temperature and relative humidity within CA reefer vessels and containers. In the past few years container operators have begun to develop services that more nearly approach those envisaged by Morris. Systems are now in place whereby the atmospheric conditions within containers can be monitored remotely and relayed to the shippers. As yet the interphase with storage life predictive models has yet to be achieved. This goal will require more investment, research and a more comprehensive understanding of the variability of temperature and gas concentration gradients within the cargoes being transported.

As in the case of CA reefer containers, the gas concentrations within each CA zone on board a ship are usually monitored by sensors positioned in the machine space external to or at the periphery of the stow where the circulating air returns to be re-cooled. In order to monitor air temperature, temperature probes are located within the airstreams entering and leaving the decks. These sensors record supply air temperatures (SAT) (also referred to as delivery air temperature (DAT)) and return air temperatures (RAT). On some vessels each deck within the holds may also be fitted with United States Department of Agriculture temperature probes/sensors (USDA probes) attached to flexible cables. It is common for at least two USDA probes to be present on any one deck. This allows air temperatures to be monitored at various locations to the port and starboard side of each deck. Temperature and humidity data loggers are also used to record environmental conditions within the cargo. Often these are placed adjacent to the merchandise in cartons on the top of or in the upper rows of individual pallets. The information recorded by such data loggers provides a more accurate estimate of the air temperatures and humidity in the immediate vicinity of the produce. In comparison, the gas and temperature sensors located externally to the stows provide a more general impression of the status of the air passing to or from the decks and gas control zones.

### CA Transport Technology

Reefers are increasingly of standard size and capacity. Champion (1986) reviewed the state of CA transport as it existed at that time and defined the difference between CA containers and MA containers. CA containers had some mechanism for measuring the changes in gases and adjusting them to a pre-set level. MA containers had the appropriate mixture injected into the sealed container at the beginning of the journey with no subsequent control, which means that the atmosphere would constantly change during transport due to respiration and leakage. The degree of control over the gases

in CA containers is affected by how gas tight the container is; some early systems had a leakage rate of 5 m<sup>3</sup>/h or more, but with improved technology systems can be below 1 m<sup>3</sup> (Garrett, 1992). Much of the air leakage is through the door and fitting plastic curtains inside the door can reduce the leakage, but they are difficult to fit and maintain in practice. Some reefer containers introduced in 1993 had a single door instead of the double doors, which are easier to make gas tight. According to Garrett (1992), the system used to generate the atmosphere in a container falls into three categories:

1. The gases that are required to control the atmosphere are carried with the container in either a liquid or solid form.
2. Membrane technology is used to generate the gases by separation.
3. The gases are generated in the container and recycled with pressure absorption and swing absorption technology.

The first method involves injecting N<sub>2</sub> into the container to reduce the level of O<sub>2</sub> with often some enhancement of CO<sub>2</sub> (Anonymous, 1987). It was claimed that such a system could carry cooled produce for 21 days compared with an earlier model, using N<sub>2</sub> injection only, which could be used only on journeys not exceeding 1 week. The gases were carried in the compressed liquid form in steel cylinders at the front of the container, with access from the outside. O<sub>2</sub> levels were maintained by injection of N<sub>2</sub> if the leakage into the container was greater than the utilization of O<sub>2</sub> through respiration by the stored crop. If the metabolism of the crop was high, the O<sub>2</sub> could be replenished by ventilation.

In containers that use membrane technology, the CO<sub>2</sub> is generated by the respiration of the crop and N<sub>2</sub> is injected to reduce the O<sub>2</sub> level. The N<sub>2</sub> is produced by passing the air through fine porous tubes, made from polysulphones or polyamides, at a pressure of about 5–6 bar. These will divert most of the O<sub>2</sub> through the tube walls, leaving mainly N<sub>2</sub>, which is injected into the container (Sharples, 1989b). A CA reefer container, which has controls that can give a more precise control over the gaseous

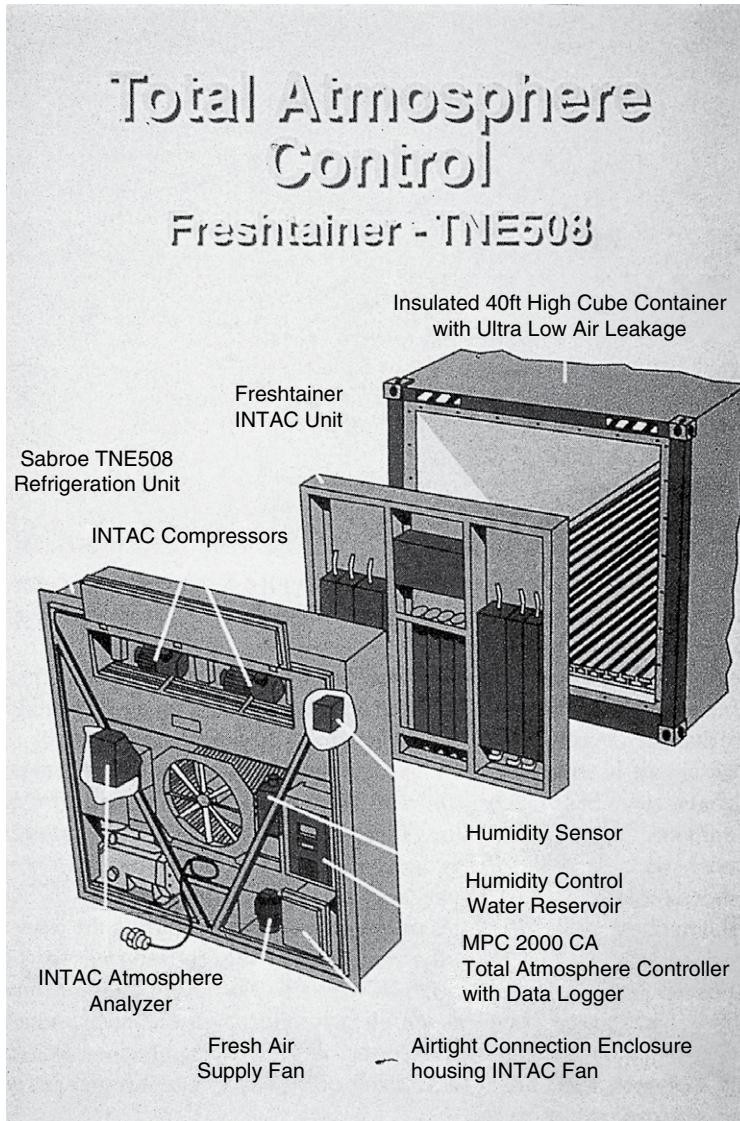
atmosphere, was introduced in 1993. The containers used ventilation to control O<sub>2</sub> levels and a patented molecular sieve to control CO<sub>2</sub> as described in Chapter 8. The levels of gas, temperature and humidity within the container are all controlled by a computer, which is an integral part of the container. It monitors the levels of O<sub>2</sub> from a paramagnetic analyser and the CO<sub>2</sub> from an infra-red gas analyser and adjusts the levels to those that have been pre-set in the computer (Fig. 13.2).

Champion (1986) mentions the ‘Tectrol’ (total environment control) gas sealing specifications that were developed and first used by the Transfresh Corporation of California in 1969. They began with the shipment of stone fruit and avocados from Chile to the USA in 1992. In this system, after loading, the reefer containers were flushed with the desired mixture of gases and then sealed. The reefer unit had N<sub>2</sub> tanks that were used to correct the deviation of O<sub>2</sub>. Lime was included to lower the CO<sub>2</sub> level and magnesium sulfate to reduce ethylene. This system was only satisfactory when O<sub>2</sub> was the only gas to be controlled and the journey was short. The name Tectrol is also used for a system where natural or propane gas is burned with outside air that is introduced into CA stores to lower the O<sub>2</sub> and increase CO<sub>2</sub>.

A system between MA and CA called Fresh-Air Exchange was developed that has mechanical ventilation to provide some control over O<sub>2</sub> and CO<sub>2</sub>. Taeckens (2007) described the version made by Carrier called AutoFresh™:

... when the CO<sub>2</sub> level reached a preset point, the system activates, drawing in outside air to add O<sub>2</sub> and ventilate excess CO<sub>2</sub>. Because Fresh-Air Exchange systems rely on natural atmosphere, the interior of the container being ventilated will generally be approaching a 78 kPa N<sub>2</sub> level, with O<sub>2</sub> and CO<sub>2</sub> in some combination making up the balance.

He thus maintained that the system was useful where the optimal combined CO<sub>2</sub> + O<sub>2</sub> levels were around 21 kPa and gave the example for strawberries at 6 kPa O<sub>2</sub> + 16 kPa CO<sub>2</sub>. A German company, Cargofresh (2009),



**Fig. 13.2.** Diagram of controlled atmosphere reefer container used in 1993.

reported that by connecting ‘an external  $N_2$  supply, an MA container can be “switched” to a fully fledged CA system. In this way, even with a central nitrogen supply, each individual container can be individually regulated.’ They called this system ‘CargoSwitch’ and it was launched in February 2008.

There continue to be challenges in achieving the optimum CA conditions. During

transport there can be considerable variation in temperature and gaseous atmosphere within a hold or container. This is due in part to stacking patterns not being adequate for air circulation. Respiratory heat at ‘normal storage temperatures’ in watts/tonne (W/t) was reported by Cambridge Refrigeration Technology (2001) as: avocado 183–465, asparagus 81–237, mango 133, banana 59–130, cauliflower 44, apple 9–18, grape 3.9–6.8,

orange 18.9 and cherry 17–39. Roger Bancroft (2017, personal communication) stated:

My experience with the out-turn quality from commercial CA transport is that the actual gas concentrations generated in containers is extremely variable so one container's CA is not equivalent to the next. In bulk carriers too although ships' data suggest gas concentrations and air temperatures are held within various parameters, in fact the temperature gradients within the cargo can differ within and between holds, so again the actual storage environment experienced by fruit at different locations can be very different. This will contribute to heterogeneous ripening behaviour [of climacteric fruit] after release from both CA and refrigeration.

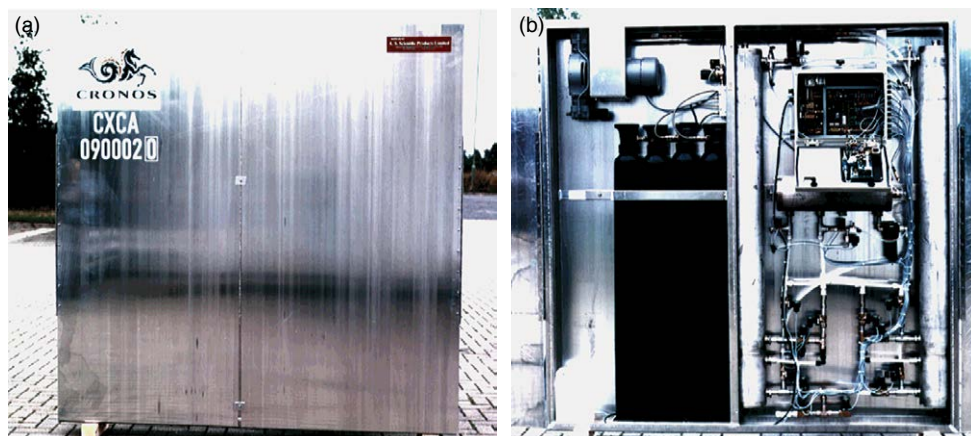
### Cronos CA containers

There are several patented systems available to control the gases in reefer containers. Cronos Containers supply inserts called the 'Cronos CA System', which can convert a standard reefer container to a CA container. The installation may be permanent or temporary and is self-contained, taking some 3 h the first time. If the equipment has already been installed in a container and subsequently dismantled, reinstallation can be achieved in about an hour. The complete units measure 2 m × 2 m × 0.2 m, which means that 50 of them can be fitted into a 40 ft (12 m) dry container for transport to where they will be used. This facilitates management of the system. It also means that they take up little of the cargo space (only 0.8 m<sup>3</sup>) when fitted into the container. The unit operates alongside the container's refrigeration system and is capable of controlling, maintaining and recording the levels of O<sub>2</sub>, CO<sub>2</sub> and humidity to the levels and tolerances pre-set into a programmable controller. Ethylene can also be removed from the container by scrubbing. It was claimed that this level of control was greater than any comparable CA storage system, increasing shipping range and enhancing the quality of fresh fruit, vegetables, flowers, fish, meat, poultry and similar products. The system is easily attached to the container floor and bulkhead and takes power

from the existing reefer equipment, with minimal alteration to the reefer container. The design and manufacture are robust to allow operation in the harsh (marine) environments that will be encountered in typical use. The system fits most modern reefers and is easy to install, set up, use and maintain. A menu-driven programmable controller provides the interface to the operator, who simply has to pre-set the required percentages of each gas to levels appropriate for the produce in transit. The controls are located on the front external wall of the container, and once set up a display will indicate the measured levels of O<sub>2</sub>, CO<sub>2</sub> and relative humidity. The system consists of a rectangular aluminium mainframe on to which the various sub-components are mounted (Fig. 13.3).

The compressor is located at the top left of the mainframe and is driven by an integral electric motor supplied from the control box. Air is extracted from the container and pressurized (up to 4 bar) before passing through the remainder of the system. A pressure relief valve is incorporated in the compressor along with inlet filters. A bleed supply of external air is ducted to the compressor and is taken via the manifold with a filter mounted externally to the container.

The compressed air is cooled prior to passing through the remainder of the system. A series of coils wound around the outside of the air cooler radiate heat back into the container. The compressed air then passes into this component, which removes the pressure pulses produced by the compressor and provides a stable air supply. The water trap passing into the CA storage system then removes moisture. Water from this component is ducted into the water reservoir and used to increase the humidity when required. Two activated alumina drier beds are used in this equipment, each located beneath one of the N<sub>2</sub> and CO<sub>2</sub> beds. Control valves are used to route the air through parts of the system as required by the conditioning process. Mesh filters are fitted to the outlet that vents N<sub>2</sub> and CO<sub>2</sub> back into the container and to the outlet that vents oxygen to the exterior of the container. N<sub>2</sub> and CO<sub>2</sub> beds are located above the dryer



**Fig. 13.3.** Cronos CA reefer container (a) exterior and (b) interior.

sieve beds and contain zeolite for the absorption of nitrogen and  $\text{CO}_2$ .

Ethylene can be removed from the container air if required. A single ethylene absorption bed is used which contains a mixture of activated alumina and Hisea material (a clay mineral-based system patented by BOC). Air from the container is routed to the bed that remains pressurized for several hours and then depressurized via the  $\text{O}_2$  venting lines. The  $\text{O}_2$  flow is then routed through the bed for 20 minutes in order to scrub it. After this the process is repeated.

To increase the relative humidity within the container, an atomized spray of water is injected as required into the main airflow through the reefer. The water supply is taken from a reservoir located at the base of the mainframe which is fed from the reefer defrost system and the water trap. Air from the instrument air buffer is used to form the spray by drawing up the water as required. A further stage of moisture removal is carried out using dryer beds that are charged with activated alumina. From here the air is routed either directly to the nitrogen and  $\text{CO}_2$  beds, or via the ethylene bed, depending on the type of conditioning needed. The reefer process tends to decrease the humidity within the container, with water being discharged from the defrost equipment. When the CA storage system is used, this water is drained into a 5 l

reservoir located within the mainframe and used to increase the humidity if required. Air from the instrument air buffer is used to draw the water into an atomizing injection system located in the main reefer airflow. The control valve is operated for a short period, and once the additional water spray has been mixed in with the air in the container the humidity level is measured and the valve operated again if required.

Four 10 l bottles of  $\text{CO}_2$  are located in the mainframe. To increase the level of  $\text{CO}_2$  within the container, the gas is vented via a regulator.  $\text{CO}_2$  is retained in the molecular sieve along with the nitrogen. Normally this would be returned to the container, but if it is required to remove  $\text{CO}_2$  then the flow of gas (which is mainly  $\text{CO}_2$  at this point) is diverted for the last few seconds to be vented outside the container. This supply of  $\text{CO}_2$  is intended to last for the duration of the longest trip. A check to ensure that the bottles are full is included as one of the pre-trip checks.

$\text{O}_2$  removal is accomplished using a molecular sieve, which is pressurized at up to 3 bar.  $\text{N}_2$  and  $\text{CO}_2$  are retained in the bed. The separated  $\text{O}_2$  is taken from the sieve bed to the manifold at the bottom of the mainframe and then vented outside the container. It is possible to reduce the level of oxygen to around 4 kPa, though this does depend on a maximum air leakage into the container and in certain cases  $\text{O}_2$

levels of 2 kPa or even 1.5 kPa can be maintained (Tim Bach, personal communication, 1997).

To add O<sub>2</sub>, air from outside the container is allowed to enter by opening a control valve for a short period. Once the additional air has been mixed in with the air in the container, the oxygen level is measured and the valve operated again if required.

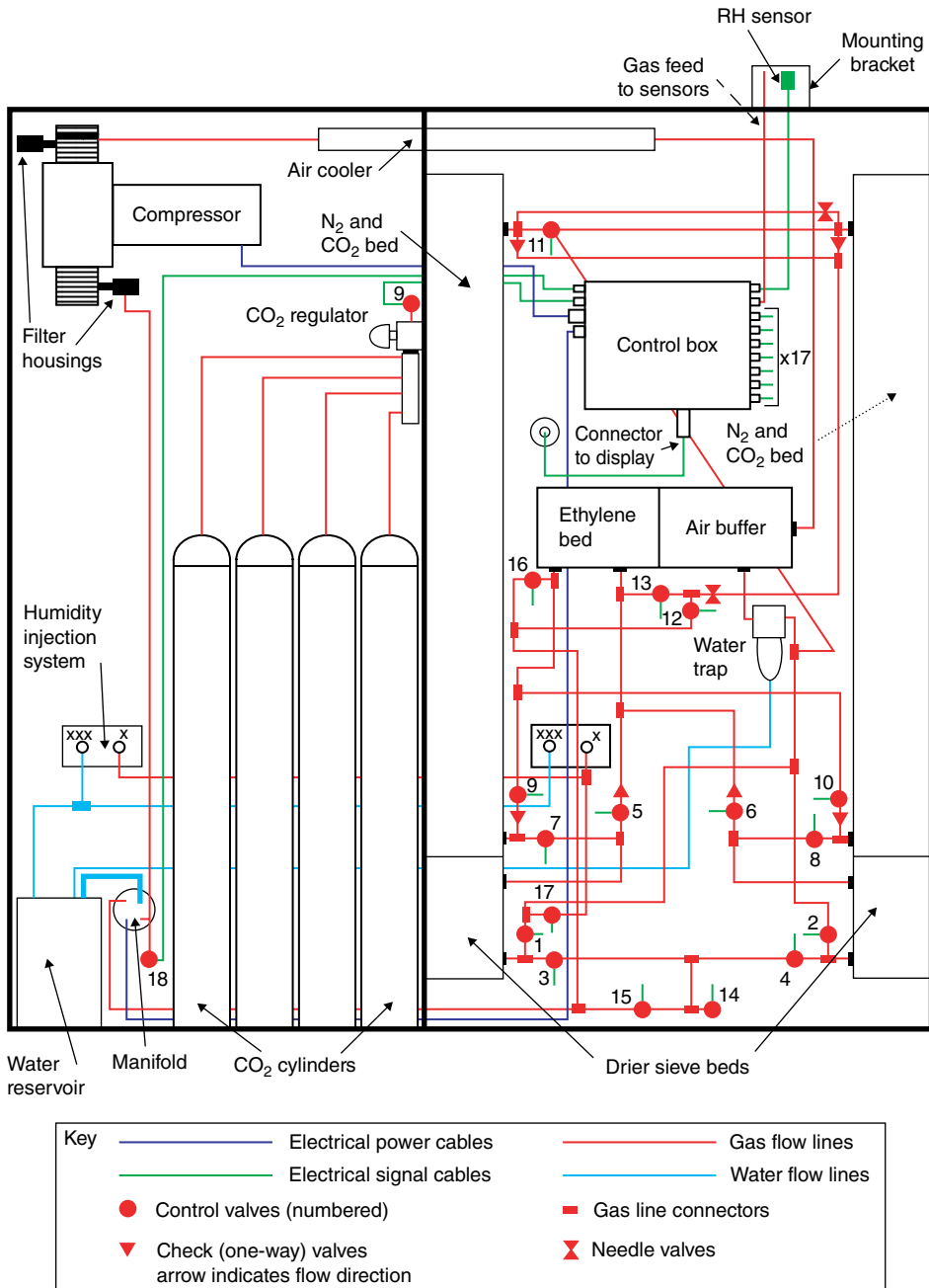
The N<sub>2</sub> and CO<sub>2</sub> retained in the main sieve beds are returned to the alumina dryer beds in order to recharge them and remove moisture from them. Finally the gas is returned to the main container via a valve located at the bottom of the mainframe. If the CO<sub>2</sub> is to be removed, a modified process occurs. The levels of O<sub>2</sub>, CO<sub>2</sub> and humidity are continually monitored and the values passed back to the display. A supply of gas is taken from the general atmosphere within the container via a small pump located within the control box. The gas is passed on to an O<sub>2</sub> transducer and then a CO<sub>2</sub> transducer before being returned to the container. The location of these components is shown in Fig. 13.4. A transducer located in the airflow through the mainframe and connected into the control box measures humidity. The above sequence of operation is carried out according to instructions provided by the display. In particular, the measured values of O<sub>2</sub>, CO<sub>2</sub> and humidity are compared in the display with those preset by the operator and the gas conditioning cycle is adjusted accordingly. Once set up, the process is automatic and no further intervention is required by the operator unless a fault occurs. Further detail on setting up the system is provided in the next section of the manual. In addition to providing the operator interface, the display has a communications socket that allows data to be transferred (e.g. to/from an external computer). The most frequent use of this facility will be the downloading of the data logged during normal operation of the system – the levels of O<sub>2</sub>, CO<sub>2</sub> and relative humidity as periodically recorded. Data is output in a form that can be used in a spreadsheet. It is also possible to use the external computer to modify system settings and carry out other diagnostic operations.

## Hypobaric Containers

Various other methods have been used for modifying the atmosphere during fruit and vegetable transport, including hypobaric containers (see also Chapter 11). A commercial hypobaric container was developed in the mid 1970s and used for fish and meat as well as fruit and vegetables. However, to prevent implosion it was very strongly constructed and therefore heavy and expensive. The Grumman Corporation in the USA constructed a hypobaric container that they called 'Dormovac'. It was operated at 2.2–2.8°C and a pressure of 15 mm Hg and they tested it in commercial situations, but were unable to make it profitable, resulting in losses of some \$50 million (Grumman Park, undated). Alvarez (1980) described experiments where papaya fruits were subjected to sub-atmospheric pressure of 20 mm Hg at 10°C and 90–98% RH for 18–21 days during shipment in a hypobaric container from Hawaii to Los Angeles and New York. Both ripening and disease development were inhibited. Fruits ripened normally after removal from the hypobaric containers, but abnormal softening, unrelated to disease occurred in 4–45% of fruits of one packer. It was found that hypobaric stored fruits had 63% less peduncle infection, 55% less stem-end rot and 45% fewer fruit surface lesions than those stored in a refrigerated container at normal atmospheric pressure. A hypobaric system was described by Burg (2004) called Vacu-Fresh<sup>sm</sup>. It had a very slow removal of air and therefore it was claimed that there was no desiccation problem of the fruit and vegetables transported in these containers (Fig. 13.5).

## MA Packaging

MA packaging has been developed for particular transport situations. For long-distance transport of bananas a system was developed and patented by the United Fruit Company called 'Banovac' (Badran, 1969).



**Fig. 13.4.** Schematic layout of Cronos CA system (courtesy of Tim Bach).

The system uses PE film bags, 0.04 mm thick, in which the fruit is packed (typically 18.14 kg) and a partial vacuum is applied and the bags are sealed (Fig. 13.6).

Typical gas contents that develop in the bags through fruit respiration are about 5 kPa CO<sub>2</sub> and 2 kPa O<sub>2</sub> during transport at 13–14°C (De Ruiter, 1991).



**Fig. 13.5.** Intermodal tank container from Welfit Oddy (Pty) Ltd, Port Elizabeth, South Africa.



**Fig. 13.6.** Banavac MA packaging: (a) box with plastic bag before packaging; (b) packed boxes loaded on to a pallet and stowed in a ship for transport.

## CA Transport Trials and Case Studies

### Some sources of variability in MA and CA shipments

Despite improved understanding of physiology, and advances in handling practices and technology, those involved in the transportation of fresh produce are only too aware that there is always a risk that out-turn quality of a proportion of shipments will

result in a biological and/or economic loss. Such outcomes may be the consequence of a single factor or the interaction of multiple factors. Companies and research establishments have worked on CA transport and the details of several experiments are described in the literature.

One of the recurrent problems that cause failure in CA-stored produce is the inefficient management of the required concentration of  $O_2$  and  $CO_2$ . Whether or not CA



stores function as required, occasionally the physiological status of the stored produce or a sub-component is such that neither refrigeration nor the gas conditions are able to prevent premature ripening, deterioration or senescence. Physiological stress and diseases may also play a role in negating the beneficial influence of MA or CA technologies. The following case studies seek to illustrate these points.

### Mangoes

Maekawa (1990) reported that green-mature 'Irwin' mangoes were transported in a refrigerated truck at 12°C on a car ferry from Kume Island (Japan) to Tokyo Port and then from Tokyo to Tsukuba University by land. The total journey took 4 days. On arrival the fruits were stored in 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> at 8°C, 10°C or 12°C. Fruit quality was retained at all temperatures for 1 month. No chilling injury was observed, even at the lowest temperature. Dos Prazeres and de Bode (2009) carried out shipment trials for 21 days at 10°C and 95% RH in 3 kPa O<sub>2</sub>. They reported that all fruit arrived in good condition with 'no quality losses' and with a shelf-life of up to 10 days. They also did static trials of storage for 43 days and then an actual shipment taking 49 days and found that 'Kent' and 'Tommy Atkins' proved the best cultivars, since they arrived in good condition. All major aroma volatile compounds in ripe fruit decreased as the level of CO<sub>2</sub> was increased over the range of 3, 6 or 9 kPa at 2 kPa O<sub>2</sub> during 21 days storage of the cultivar 'Kensington Pride' at 13°C. Storage in 2 kPa O<sub>2</sub> + 6 kPa CO<sub>2</sub> at 13°C appeared to be promising for extending the shelf-life and maintaining fruit quality, while storage in 2 kPa O<sub>2</sub> + 2 kPa CO<sub>2</sub> appeared to be better at maintaining the aroma compounds of ripe fruit (Lalel *et al.*, 2005). Successful simulated commercial export of mangoes using the 'Transfresh' system of CA container technology was described in Ferrar (1988). Peacock (1988) described a CA transport experiment in Australia where mangoes were harvested from three commercial sites in Queensland when the TSS was 13–15%. Fruits were de-stalked in two

of the three sources and washed, treated with 500 ppm benomyl at 52°C for 5 min, cooled and sprayed with prochloraz (200 ppm), dried and sorted and finally size graded and packed in waxed fibreboard cartons. After packing, fruits were transported by road to a pre-cooling room overnight (11°C). Pulp temperatures were 18–19°C the following morning. After 36 h of transport to Brisbane and overnight holding in a conventional cool room, the fruits were placed in a CA shipping container. At loading, the pulp temperature was 12°C. Fruits were stored in the container at 13°C with 5 kPa O<sub>2</sub> + 1 kPa CO<sub>2</sub> for 18 days. Ripening was significantly delayed in CA storage, though there were problems with CO<sub>2</sub> control, related to the efficiency of the scrubbing system. There was virtually no anthracnose disease but there were very high levels of stem-end rot. On removal from the CA container, fruits immediately began to ooze sap and the loss continued for at least 24 h.

### Kiwifruit

Lallu *et al.* (1997) described experiments on CA transport of kiwifruit in containers where the atmosphere was controlled by either N<sub>2</sub> flushing and Purafil (potassium permanganate absorbed on to a clay mineral) or lime and Purafil and a third with no CA. N<sub>2</sub> flushing + Purafil maintained CO<sub>2</sub> levels of approximately 1 kPa and O<sub>2</sub> at 2–2.5 kPa while with lime + Purafil the CO<sub>2</sub> levels were 3–4 kPa and the O<sub>2</sub> levels increased steadily to 10 kPa. On arrival the ethylene levels in the three containers were less than 0.02 µl/l, indicating that the potassium permanganate was not necessary. However, they concluded that CA containers can result in benefits to fruit quality.

### Blueberries

Storage in high concentrations of CO<sub>2</sub> is effective in delaying deterioration as well as reducing the susceptibility of fruit to fungal rots. The carriage instructions for a sample

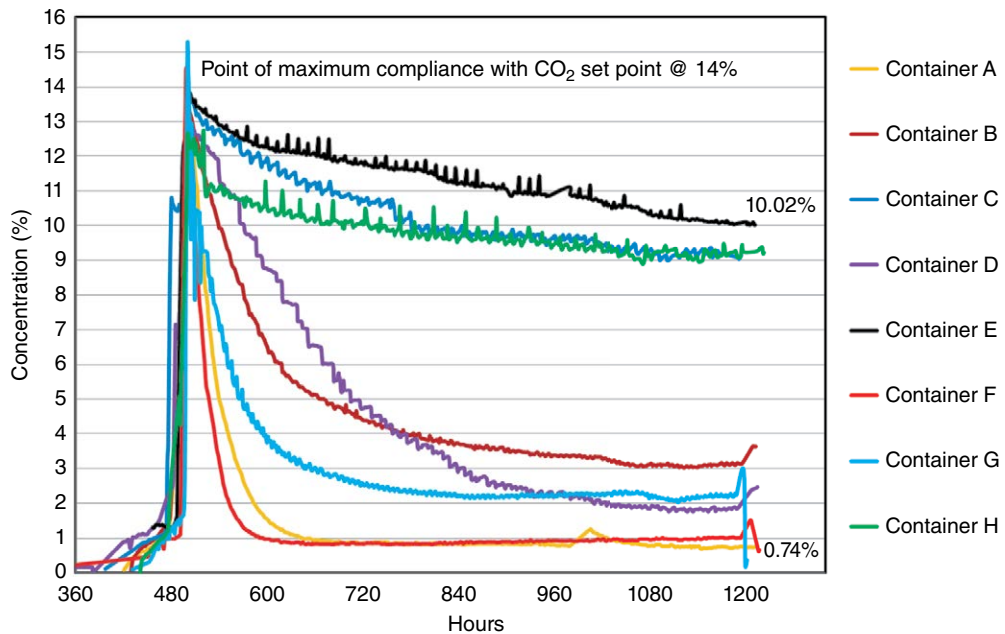
of eight containers of Chilean blueberries, analysed in 2013 (R.D. Bancroft, commercial unpublished data) and considered here, required refrigeration to  $-1^{\circ}\text{C}$  and CA conditions of 8 kPa  $\text{O}_2$  and 14 kPa  $\text{CO}_2$ . The CA units deployed had the capacity to regulate  $\text{O}_2$  and  $\text{CO}_2$  by the introduction of  $\text{N}_2$  and  $\text{CO}_2$ , respectively. Analysis of the temperature logs for all the containers showed good compliance with the temperature set point. In six of the eight containers the concentration of  $\text{O}_2$  reached the set point within 2–3 days of CA initiation. In one of the containers the concentration of  $\text{O}_2$  was eventually reduced to a mean of 3.5 kPa, while in another container the average  $\text{O}_2$  for the voyage was 11 kPa. As regards  $\text{CO}_2$ , the majority of the containers never reached consistent parity with the set point of 14 kPa. Certain  $\text{CO}_2$  maxima were attained and then the concentration of this gas declined over time (Fig. 13.7).

These observations suggest that the gas atmosphere storage environments that became established in these containers were

more variable than might be expected. A further complication was that each pallet stack was covered in a polyethylene cover (sometimes called a poly-foil). The covers were perforated with 12 mm holes at a frequency of 22 perforations/ $\text{m}^2$ . It seems likely, therefore, that the gas concentrations immediately adjacent to the blueberries deviated in some manner from those recorded by the CA units attached to the containers.

Analyses of the out-turn quality of the fruit transported in these containers did show some slight variation, with the proportion of uncompromised fruit ranging from 83% to 90%, but such trends were found to correlate with the variety and source of the fruit rather than the CA conditions.

In this study, although it transpired that the operation instructions cited ‘Controlled Atmosphere’ storage, in reality what was achieved was a ‘primed’ form of MA storage. Here the level of  $\text{CO}_2$  was elevated at the beginning of the transit period and thereafter declined. This is understood to be ‘normal’ practice for the export of blueberry fruits



**Fig. 13.7.** The relative compliance of a series of eight CA containers to a  $\text{CO}_2$  Set Point of 14%. Each container held palletized blueberries transported from Chile to the UK. The period of transit from gas initiation to de-vanning was approximately 32 days.

from Chile and it was advised that this system had worked well over the years. If so, then it is suggested that whatever the gas atmospheres registered in the CA control units, the partial pressures of O<sub>2</sub> and CO<sub>2</sub> within the pallets themselves must be adequate to permit the journey to Europe. From a commercial point of view it would be useful to know more precisely what gas levels really existed within the pallets and whether these could be achieved without aspiring to a 14 kPa CO<sub>2</sub> level at the beginning of the voyage and the maintenance of 8 kPa O<sub>2</sub> throughout.

### Avocados

Eksteen and Truter (1989) stored avocados in 'Freshtainer' 40 ft CA containers controlled by microprocessors. The set conditions were  $7.4 \pm 0.5^\circ\text{C}$  for 8 days followed by  $5.5 \pm 0.5^\circ\text{C}$  for 7 days, with  $2 \pm 0.5$  kPa O<sub>2</sub> and a maximum of  $10 \pm 0.5$  kPa CO<sub>2</sub>. They reported that the containers were capable of very accurate control within the specified conditions. Also the fruit quality was better than that of the controls, which were 2 weeks storage at  $5.5^\circ\text{C}$  in air in insulated containers.

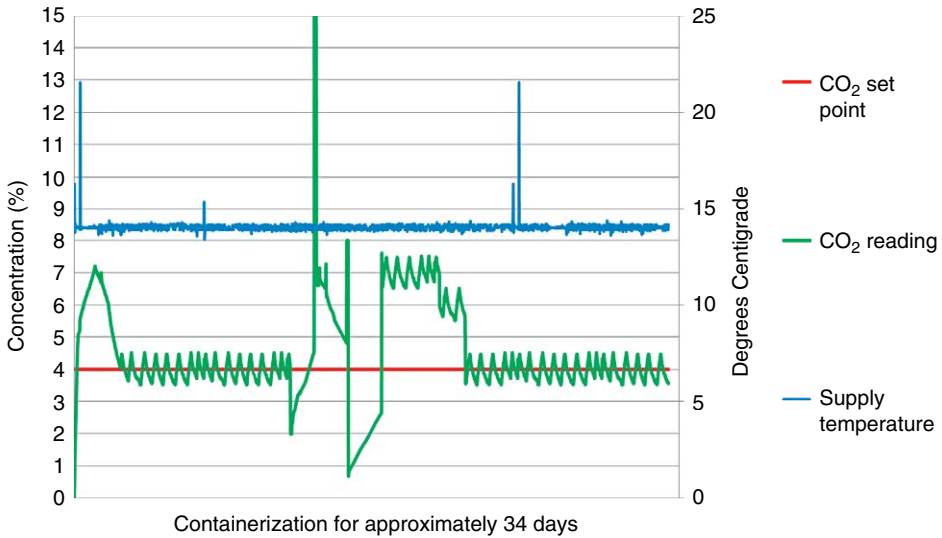
### Bananas

The storage technologies deployed to transport bananas across the Atlantic to the European markets are well established. A small sample of 11 CA containers, investigated in 2016 (R.D. Bancroft, commercial unpublished data), suggested, however, that the degree to which the gas concentrations within individual containers conform to the standard set points can be very variable. The transit period for all containers discussed here was approximately 34 days. This involved a sea voyage of a nominal 9 days followed by road haulage across Europe to the UK. There were two transhipment periods. The set points for temperature, O<sub>2</sub> and CO<sub>2</sub> concentrations were  $14^\circ\text{C}$ , 3 kPa and 4 kPa, respectively. When powered, none of the containers failed to regulate their air temperatures as programmed. The out-turn quality of three

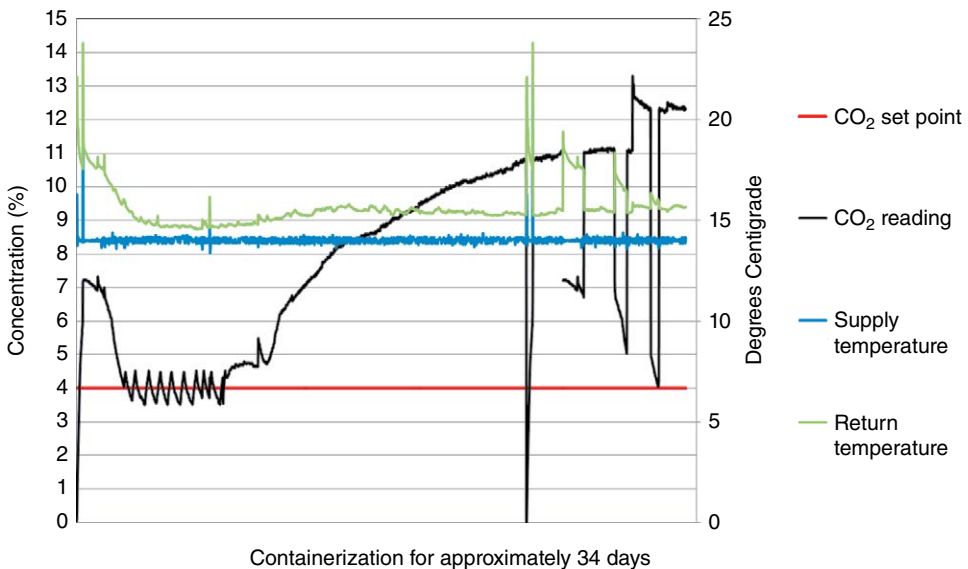
of the containers, however, resulted in claims for loss of produce. Of the other eight CA containers, the control of CO<sub>2</sub> was satisfactory. Four of these containers also regulated O<sub>2</sub> to the set point, but the mean O<sub>2</sub> concentration of the remaining four containers varied from 11 kPa to 14 kPa.

Of the CA containers with poor out-turn results, one exhibited no gas control whatsoever throughout the entire transit period. As a consequence the endogenous levels of CO<sub>2</sub> began to rise during the last 6 days, due to the premature ripening of the banana fruit. The plots of CO<sub>2</sub> associated with the last two failed containers are reproduced in Figs 13.8 and 13.9. In one container (Fig. 13.8) the data indicates a discontinuity in CO<sub>2</sub> control following discharge from the vessel in Europe. This lasted for a period of about 10 days. Although CO<sub>2</sub> regulation was re-imposed during the later period of haulage, this was insufficient to prevent damage to the cargo.

In the second example (Fig. 13.9), despite appropriate cooling and CO<sub>2</sub> control during early transit, the concentration of CO<sub>2</sub> then rose consistently some 16 days after containerization. The data set then showed intermittent reductions in CO<sub>2</sub> that would have been associated with the opening of the container on a daily basis during the last 3 days of storage. Concurrent with the increase in CO<sub>2</sub> gas concentrations found in the air being recycled from within the container is a rise in return air temperature (RAT). This provided further evidence that the respiratory activity within the cargo had changed. The timing of this transition indicates not only that at least some of the bananas had begun to ripen but also that it is very likely that these fruit were already over-mature when they had been originally stowed. Once a sub-component of the cargo had begun to ripen, the process will have become autocatalytic and the remaining fruit will have also become susceptible to premature ripening. Under such conditions the CA control unit would no longer be able to influence the levels of CO<sub>2</sub> within the container. In this example, ultimately the inherent physiology of the fruit negated the potential of the CA unit to conserve the cargo.



**Fig. 13.8.** Example of the progressive failure to control CO<sub>2</sub> concentration in a CA container transporting bananas across the Atlantic to Europe and finally by road haulage to the UK. This was attributed to periods of power loss or management oversight associated with transshipment.



**Fig. 13.9.** Example of the progressive failure to control CO<sub>2</sub> concentration in a CA container transporting bananas across the Atlantic to Europe and finally by road haulage to the UK. The cause was the premature ripening of fruit.

Eksteen and Truter (1989) stored bananas in 'Freshtainer' 40 ft CA containers controlled by microprocessors. The set conditions for bananas were:  $12.7 \pm 0.5^\circ\text{C}$  for 8 days followed by  $13.5 \pm 0.5^\circ\text{C}$  for 11 days,

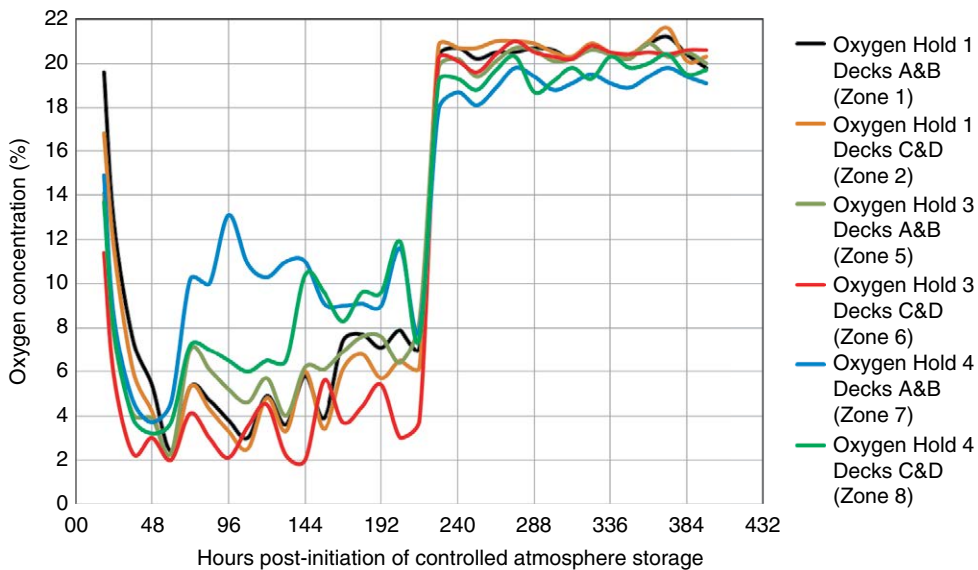
with  $2.0 \pm 0.5$  kPa O<sub>2</sub> and a maximum of  $7 \pm 0.5$  kPa CO<sub>2</sub>. The fruit quality was better than that of the controls, which were 2 weeks of storage at  $13.5 \pm 0.5^\circ\text{C}$  in air in insulated containers.

### Over-mature bananas

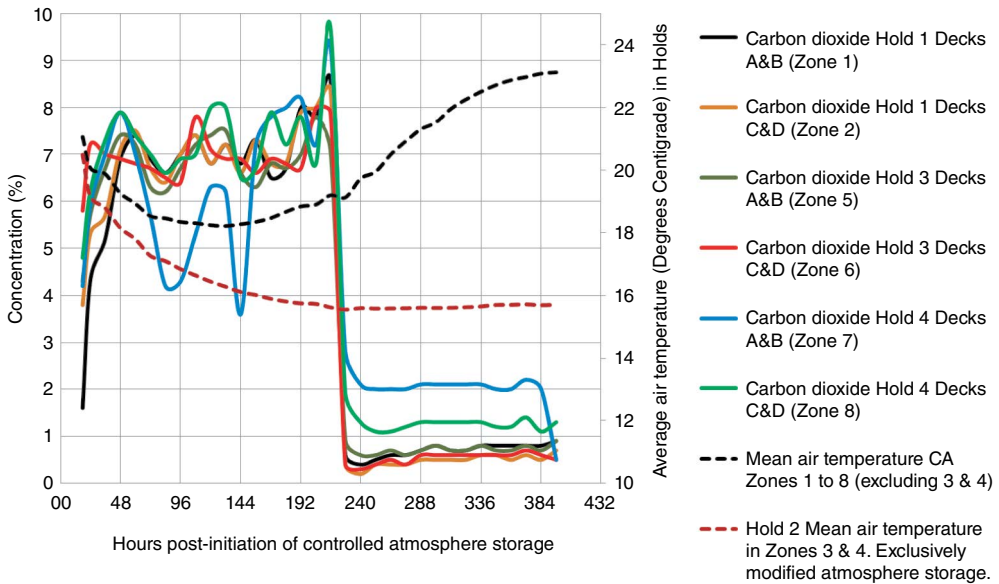
As seen in the previous case studies above, the efficiency with which the air temperature and gas concentrations may be managed within a storage environment does not preclude the loss in quality of a cargo. [Figure 13.10](#) shows the O<sub>2</sub> concentrations established in six CA gas control zones aboard a CA reefer vessel transporting bananas from Central America to Europe in 2008. The set point for O<sub>2</sub> was 3%. Although the majority of the zones did not achieve this level of reduction during the initial phase of the voyage, experience suggests that O<sub>2</sub> concentrations between 3% and 8% would not necessarily have compromised the storage potential of the generality of the cargo. With regards to the management of CO<sub>2</sub>, relative to a set point of 5% CO<sub>2</sub>, following the imposition of CA conditions five of the six CA gas control zones equilibrated to approximately 6.8% CO<sub>2</sub> ([Fig. 13.11](#)). Although not ideal, in themselves such O<sub>2</sub> and CO<sub>2</sub> concentrations should have been sufficient to conserve the cargo during its passage to Europe. However, the elevated levels of CO<sub>2</sub> recovered from the different CA gas control zones indicated

that the fruit was not quiescent and was respiring more than would have been expected. Within a few days it transpired that at least some of the bananas in all the CA holds had begun to ripen prematurely, so much so that the ship's crew terminated CA storage 9 days into the voyage and allowed ambient air into all the CA gas control zones.

On this particular vessel, Hold 2 transported bananas in Banavac packaging. With a temperature set point of 13.3°C, this will have generated an MA environment sufficient to conserve sound fruit for approximately 2 months. The evidence that the fruit committed to the CA holds was over-mature on loading was confirmed by an analysis of the air temperatures recorded by the USDA probes present on all the decks. [Figure 13.11](#) depicts the mean air temperature registered by the USDA probes in the CA gas control zones in Holds 1, 3 and 4. This indicates that, on loading, the mean air temperature adjacent to the fruit was 21°C. Some 5 days later this had been reduced to 18°C before rising progressively from then on to some 23°C at discharge. A consistent reduction in air temperature to within 2.22°C of the set point did not occur in any of these



**Fig. 13.10.** Variation in O<sub>2</sub> concentrations recorded in the holds of a CA reefer vessel transporting bananas in 2008. The premature ripening of the cargo eventually resulted in the termination of active gas storage some 9 days into the voyage.



**Fig. 13.11.** Variation in CO<sub>2</sub> concentrations recorded in the holds of a CA reefer vessel transporting bananas in 2008. The mean air temperature recorded in the active CA holds (Gas Control Zones 1 to 8, excluding 3 and 4) indicated that the cargo had been over-mature on loading. In contrast, bananas held in modified atmosphere packaging in Hold 2 (Gas Control Zones 3 and 4) stored well and were discharged in good condition.

holds. In comparison, the air temperatures in Hold 2 show that the reduction period for the fruit held in MA packaging was 108 h. Thereafter the mean air temperature in Hold 2 was 15.8°C and a respectable 2.5°C above the set point for this cargo. There was no reported loss in quality in those fruit transported in Hold 2. There is no suggestion that the vessel failed to comply with the carriage instructions issued by the shipper and hence the loss in quality of the bananas recovered on discharge was due to their advance maturity on loading.

#### *Mature bananas and latent diseases*

Under normal circumstances, reefer vessels exporting bananas from ports in South or Central America and the Caribbean to discharge ports (disports) around the Mediterranean would expect to discharge their cargoes within 23–30 days. Polyethylene bags, often referred to as ‘polypacks’ (Stover and Simmonds, 1987) and Banavac plastic membranes are used to conserve bananas in MA conditions for many of these trade

routes. Commercially, the green-life expectancy for sound, hard-green bananas in polypacks is some 28 days, while that in partially evacuated Banavac bags can be as long as 50–60 days.

Occasionally the routine transport of bananas to Europe and the Mediterranean is disrupted by mechanical failures of one sort or another. The degree to which the storage potential of the cargo may be foreshortened or jeopardized will often depend on whether the carrier can keep the merchandise adequately refrigerated and ventilated, the extent to which any transit period is prolonged, and the biological condition of the fruit.

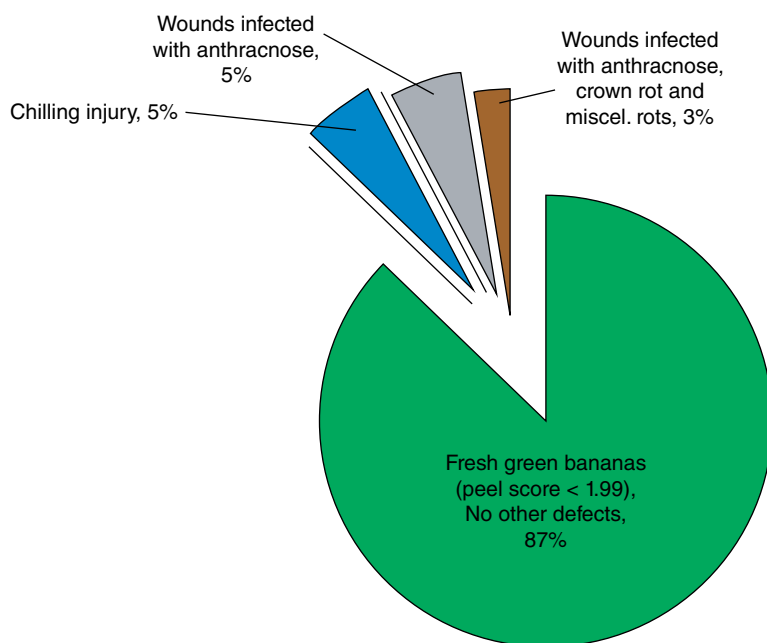
In 2006 a cargo of bananas was being transported from Ecuador to the Middle East on a conventional reefer vessel. The carriage instructions indicated that the supply air temperature (SAT) within the holds was to be 13.3°C. Engine problems prolonged the voyage and discharge occurred 8–9 weeks after the first loading episode. The cargo consisted of several brands of bananas all conserved in Banavac bags. On account of the extended storage period, it was expected

that at least some of the fruit would have ripened naturally in transit; however, analysis of a sample of cartons during discharge indicated that other factors were also responsible for the loss in quality of sub-components of the cargo (R.D. Bancroft, commercial unpublished data). The nature and proportion of defects observed varied according to brand and whether fruit was categorized as non-ripe or ripening. The premature ripening of many of the fruit was considered to have been caused by ethylene generated from damaged fruit and those chronically infected with anthracnose and/or crown rot. On discharge, the most expedient segregation of the cargo was into those cartons that contained non-ripened green bananas and those that had ripened or had begun to turn colour. Differences between brands immediately became apparent (Figs 13.12–13.15).

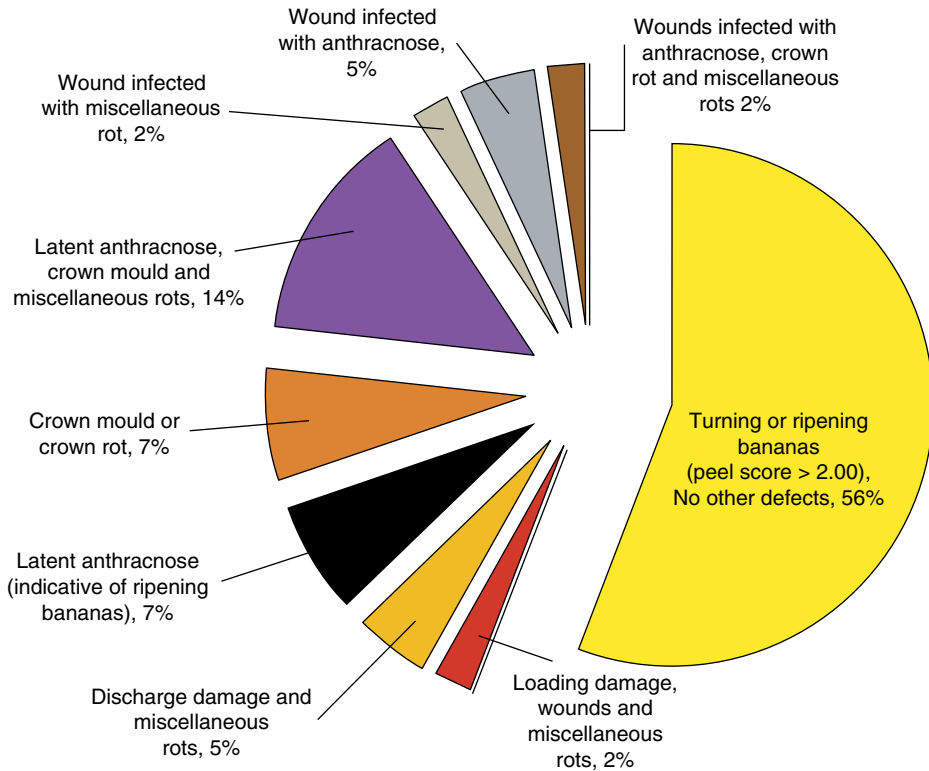
In Brand A, for example, of those fruit that remained green, 87% of the fruit sampled showed no external defects and, relative to a peel coloration scale of 1 (hard-green) to 7 (yellow with flecks of brown), registered a score of < 1.99 (Fig. 13.12). Where damage

was observed in these fruit, 8% was attributable to wounds infected with fungi, principally anthracnose and the disease complex known as crown rot. Another 5% of the fruit showed signs of chilling injury. In contrast in Brand B, 82% of the consignment remained unblemished, with 68% registering a colour score of < 1.99 and the residual 14% having a peel colour in excess of 2 (Fig. 13.14). In Brand B, 15% of the fruit samples supported fungal infections and 3% had chilling injury. When the ripe or ripening fruit were analysed, 56% of the Brand A bananas were unblemished compared with 40% in Brand B (Figs 13.13 and 13.15). Damage caused by rough handling during loading accounted for 2–3% and that engendered during discharge adversely impacted 5% of fruit of Brand A and 10% of fruit of Brand B.

Of particular interest is the proportion of ripe or ripening bananas predated by anthracnose and crown rot. In Brand A, 12% of the samples showed signs of the development of latent anthracnose and 7% were clearly attacked by crown rot. A further 18% showed signs of infection by a mixture of



**Fig. 13.12.** Out-turn quality of non-ripened bananas of Brand A held in Banavac packaging and transported for 8–9 weeks in a conventional reefer vessel.



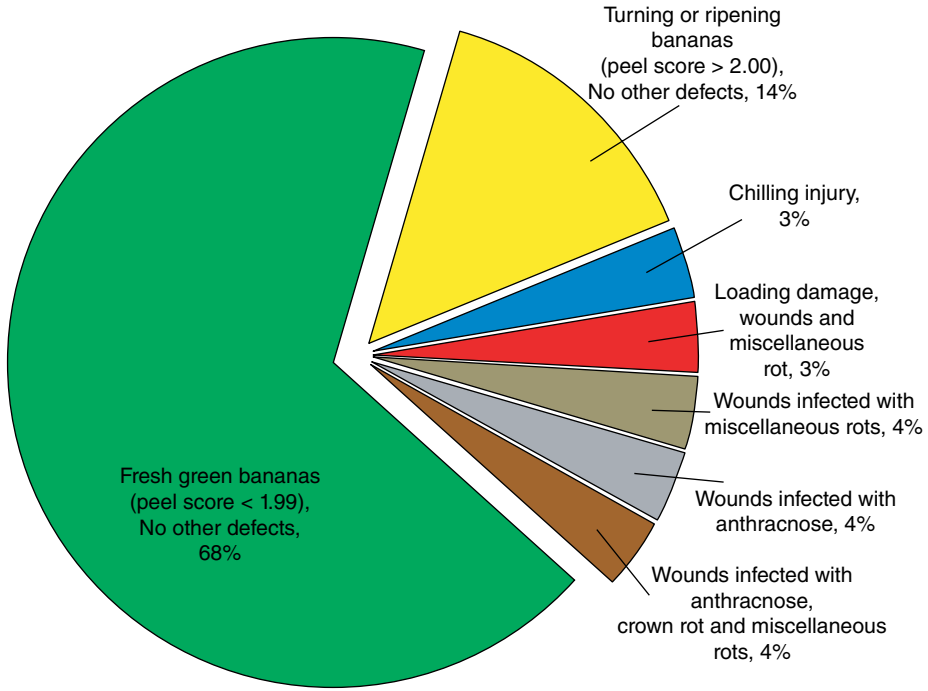
**Fig. 13.13.** Out-turn quality of ripe and ripening bananas of Brand A held in Banavac packaging and transported for 8–9 weeks in a conventional reefer vessel. Premature ripening of many of the fruit was considered to have been caused by ethylene gas generated from damaged fruit and those chronically infected with anthracnose and/or crown rot.

fungi, including anthracnose and crown rot. In Brand B, latent anthracnose occurring on its own was observed in 5% of fruit whereas crown rot was the principal disease in 25% of the bananas sampled. Mixed infections accounted for a further 15%. These observations suggested that latent anthracnose might have been more prevalent in Brand A while crown rot was found more often in Brand B. Both diseases will have arisen from latent infections. That is, although unobserved, these diseases will have been present in the fruit at the time of loading and will have become active during the voyage. The ripening gas ethylene is produced by damaged tissues and both crown rot (Fig. 13.16) and anthracnose pathogens (Fig. 13.17). The observations made at discharge indicate that, despite the retarding influence of the MA conditions created in the Banavac bags, damage to the

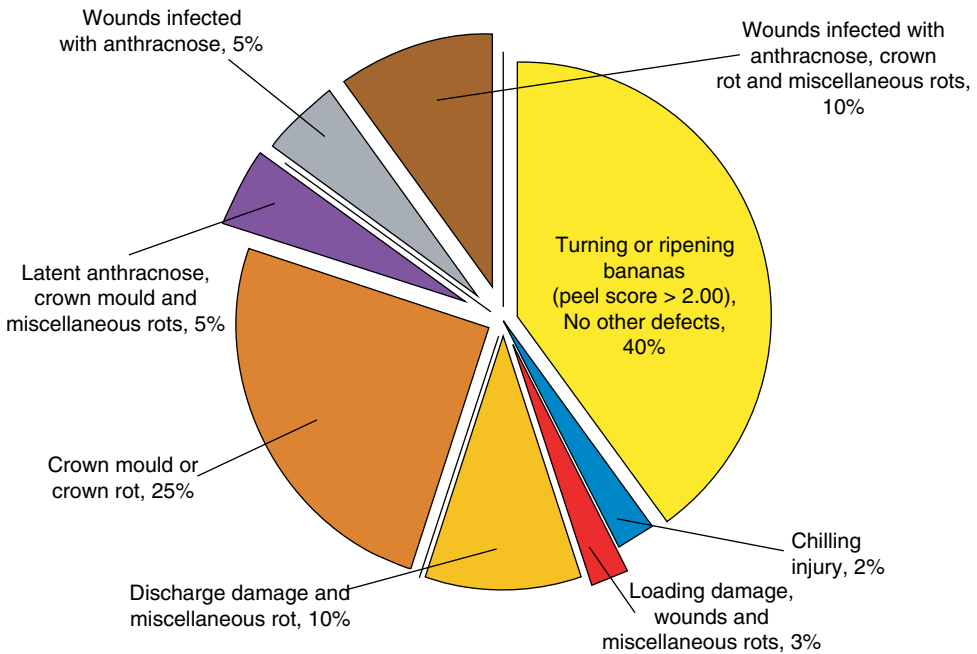
cargo at loading augmented by the activity of anthracnose and crown rot is likely to have made a significant contribution to the premature ripening of the fruit.

Observations at out-turn also suggested that the storage potential of some of the Brand A cargo may have been compromised in Ecuador due to the extended period in which the fruit remained exposed to ambient temperature conditions during loading. This would have diminished the green-life potential of the fruit from the outset. Aware of the potential damage that could be caused by a protracted voyage, the SAT was reduced to 13°C some 38 days from the date of initial loading. In fact, on occasion the SAT was found to have dropped below 13°C. This is believed to have brought about the chilling injury observed in the cargo at discharge. It is unlikely that the merchandise would have





**Fig. 13.14.** Out-turn quality of non-ripened bananas of Brand B held in Banavac packaging and transported for 8–9 weeks in a conventional reefer vessel.



**Fig. 13.15.** Out-turn quality of ripe and ripening bananas of Brand B held in Banavac packaging and transported for 8–9 weeks in a conventional reefer vessel. Premature ripening of many of the fruit was considered to have been caused by ethylene gas generated from damaged fruit and those chronically infected with anthracnose and/or crown rot.



**Fig. 13.16.** Crown rot infecting the crown and pedicels of banana fruit. Ethylene released by the rot and damaged tissues will have triggered ripening in the surrounding fruit.



**Fig. 13.17.** An example of a wound infected with anthracnose generating ethylene, which has begun to trigger ripening in the surrounding hands of bananas.

expressed these defects if the temperature regime had remained unaltered.

It is apparent that the prolonged period of storage to which the banana cargo was exposed did engender premature ripening. However, the extent to which this occurred was also a reflection of the innate potential of the cargo. Of the losses identified at out-turn, sampling suggests that a significant

proportion was as a consequence of inherent infections and damage sustained by the merchandise prior to loading. The provision of MA packaging will have conserved healthy and robust fruit but would not have been able to prevent losses brought about by the advanced physiological age of some of the fruit, mechanical damage or the intervention of disease.

## MA Packaging

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Perhaps the first use of MA packaging in plastic film was for meat in 1979 by the retailer Marks and Spencer in the UK. Since then it has become increasingly used for a wide range of foods, including fresh fruit and vegetables. Bradshaw (2007) found that food packaging added more than 20% to the cost of buying fruit at a leading UK supermarket. A customer who buys fresh fruit worth £12.41 can expect to spend an additional £4.67 if they choose those that are pre-packed rather than loose. The same supermarket stated that they had reduced packaging of fresh fruit and vegetables by 33% since 2000. However, packaging can reduce losses and preserve quality in fruit and vegetables, especially MA packaging.

MA packaging can be defined as an alteration in the composition of gases in and around fresh produce by respiration and transpiration, when such commodities are sealed in plastic films (G.E. Hobson, personal communication). MA packaging, using various plastic films, has great potential in extending the postharvest life of fruit and vegetables, but it has never reached its full potential. It can be clearly demonstrated experimentally that it can have similar effects to CA storage on the postharvest life of crops and the mechanisms of the effects are doubtless the same. Its limited uptake by the industry probably reflects the limited

control of the gases around and within the fruit and vegetables achieved by this technology. This can result in unpredictable effects on postharvest life and quality of the commodity packaged in this way. Some recent concerns about the safety of products in MA packaging may also militate against its use (Oliveira *et al.*, 2015).

Plastic films used in MA packaging for fruit and vegetables are permeable to gases. The degree of permeability depends on the chemical composition and thickness of the film as well as temperature. Other films, called barrier films, are designed to prevent the exchange of gases and are mainly used with non-respiring products. Developments in MA packaging were reviewed by Church (1994) and by Rai and Shashi (2007). These included low permeability films, high permeability films, O<sub>2</sub> scavenging technology, CO<sub>2</sub> scavengers and emitters, ethylene absorbers, ethanol vapour generators, tray-ready and bulk MA packaging systems, easy opening and resealing systems, leak detection, time-temperature indicators, gas indicators, combination treatments, and predictive/mathematical modelling. The concentration of the gases inside the film bag will also vary depending on mass, type, temperature and maturity of fruit or vegetable in the bag and the activity of microorganisms as well as the characteristics and thickness of the film used.

Given an appropriate gaseous atmosphere, MA packaging can be used to extend the postharvest life of fruit and vegetables in the same way as CA storage, but maintaining high humidity also seems to be involved. Plantains were shown to have a considerable extension in their storage life when packed in PE film, but keeping the fruit in a flow through system at high humidity had exactly the same effect. This effect of moisture content is more likely a reduction in stress within a climacteric fruit, which may be caused by a rapid rate of water loss in non-wrapped fruit. This in turn may result in increased ethylene production to internal threshold levels that can initiate ripening (Thompson *et al.*, 1974d). Porat *et al.* (2004) suggested that MA packaging reduces the development of rind disorders in citrus fruit in two ways. The first is by maintaining the fruit in high humidity; the second is maintaining elevated CO<sub>2</sub> and lowered O<sub>2</sub> levels. They found that storage of non-wrapped 'Shamouti' oranges in 95% RH reduced rind disorders to a similar level as those stored in macro-perforated plastic films. Johnson (1994a) reported that MA packaging during distribution and marketing may be at the expense of aroma and flavour.

### Film Types

Since the 1970s, MA packages have become increasingly used for fruit and vegetables and new techniques have been developed, including micro-perforation, anti-fogging layers to improve product visibility and reduce rotting and equilibrium MA packaging. Companies have marketed films that they claimed could greatly increase the storage life of packed fruit and vegetables. These have been marketed with names such as Maxifresh®, Gelpack® and Xtend®.

Different films have differences in relation to O<sub>2</sub> and CO<sub>2</sub> permeability, which is a function of thickness, density, presence of additives and gradient concentration modification (Steffens *et al.*, 2007a). The manufacturing process for films has improved over the years and although the actual

thicknesses of the film quoted may vary slightly, they are currently much more even and reliable. Many types of plastic have been used in the manufacture of films used in MA packaging and these are mainly by-products of the petrochemical industry.

### Cellophane

Cellophane is a thin, transparent sheet made of regenerated cellulose from plant cells and has low gas permeability. It was invented by the Swiss chemist Jacques E. Brandenberger in 1900 and is completely biodegradable but releases carbon disulfide and other by-products of the manufacturing process. In the USA, 25 μ clear PP is sometimes referred to as cellophane. Cellophane is a registered trade mark of Innovia Films Ltd, UK. To make cellulose or cellophane film, cellulose is dissolved in a mixture of sodium hydroxide and carbon disulfide and then recast into sulfuric acid. The cellophane produced is very hydrophilic and therefore moisture sensitive; it has good mechanical properties, desirable strength, is a good water and O<sub>2</sub> barrier but is not heat sealable. Cellophane is often coated with nitrocellulose wax or PVDC (polyvinylidene chloride) to improve barrier properties and in such form it is used for packaging of baked goods, processed meat, cheese and candies. Cellulose acetate is biodegradable; it possesses relatively low gas and moisture barrier properties and has to be plasticized for film production. A plasticizer is a substance added to plastics to make them more pliable.

### CPP

Cast film extrusion is used for laminating thin layers of plastic on top of paper or fibreboard by applying heat and pressure. Cast polypropylene (CPP) is one such film. Coextruded oriented polypropylene (COPP) has a good moisture vapour barrier and the gas barrier properties can be improved by coating it with PVDC. COPP can be perforated mechanically to improve gas exchange.

### EVA

Ethylene-vinyl acetate copolymer has high flexibility in sheet form and has higher permeability to water vapour and gases than LDPE. It is mainly used as a component of the sealant layer in lids and base films. Ethylene-vinyl alcohol copolymer (EVOH) is expensive but moisture sensitive; it has a very high gas barrier and is sandwiched between the main formable and sealant layer to provide protection.

### HDPE

High-density polyethylene is used as one of the layers in the lid material in coextruded forms; it has a higher softening point than LDPE and better gas barrier properties than linear LDPE (LLDPE) but poor clarity. HDPE has 75–90% crystalline structure, with an ordered linear arrangement of the molecules with little branching, and a molecular weight of 90,000 to 175,000 with a typical density range of 0.995–0.970 g/cm<sup>3</sup>. It has greater tensile strength, stiffness and hardness than LDPE. HDPE film has the following specifications: 520–4000 O<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, 3900–10,000 CO<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, at 1 atm for 0.0254 mm thick at 22–25°C at various or unreported relative humidity and 4–10 water vapour g/m<sup>2</sup>/day at 37.8°C and 90% RH (Schlimme and Rooney, 1994).

### High-impact polystyrene

This is an opaque, thermoformable, moderately low gas-barrier material and is used as a component of a laminate or for coextrusion.

### Ionomers

Ionomers are PE-based films. Al-Ati and Hotchkiss (2003) investigated altering PE ionomer films with permselectivities between 4–5 CO<sub>2</sub> and 0.8–1.3 O<sub>2</sub>. The results on fresh-cut apples suggest that packaging

films with CO<sub>2</sub>:O<sub>2</sub> permselectivities lower than those commercially available (< 3) would further optimize O<sub>2</sub> and CO<sub>2</sub> concentration in MA packages, particularly of highly respiring and minimally processed produce. Al-Ati and Hotchkiss (2002) pressed sodium-neutralized ethylene-methacrylic ionomers into films at 120–160°C and found that the heat treatment may improve their potential for applications as selective barriers for MA packaging of fruit and vegetables.

### LDPE

Low-density polyethylene is used for the ubiquitous ‘polythene bag’. LDPE is produced by polymerization of the ethylene monomer which produces a branched-chain polymer with a molecular weight of 14,000–1,400,000 and a density ranging from 0.910 to 0.935 g/cm<sup>3</sup>. LDPE film has the following specifications: 3900–13,000 O<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, 7700–77,000 CO<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, at 1 atm for 0.0254 mm thick at 22–25°C at various or unreported humidities and 6–23.2 water vapour g/m<sup>2</sup>/day at 37.8°C and 90% RH (Schlimme and Rooney, 1994). It is an inert film with low permeability for water vapour but high gas permeability.

### LLDPE

Linear low-density polyethylene is a variation of basic LDPE film but has better impact strength, tear resistance, higher tensile strength and elongation, greater resistance to environmental stress cracking and better puncture resistance. LLDPE combines the properties of LDPE film and HDPE film, giving a more crystalline structure than LDPE film but with a controlled number of branches, which makes it tougher and suitable for heat sealing. It is made from ethylene with butene, hexene or octene, with the latter two co-monomers giving enhanced impact resistance and tear strength. Permeability was given as 7000–9300 O<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, at 1 atm for 0.0254 mm thick at 22–25°C at various or unreported humidities and

16–31 water vapour g/m<sup>2</sup>/day at 37.8°C and 90% RH (Schlimme and Rooney, 1994).

### MDPE

Medium-density polyethylene is used for shrink-film packaging, among other applications. It has a density in the range of 0.926 and 0.940 g/cm<sup>3</sup>, 2600–8293 O<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, 7700–38,750 CO<sub>2</sub> cm<sup>3</sup>/m<sup>2</sup>/day, at 1 atm for 0.0254 mm thick at 22–25°C at various or unreported humidities and 8–15 water vapour g/m<sup>2</sup>/day at 37.8°C and 90% RH (Schlimme and Rooney, 1994).

### OPP

Oriented polypropylene provides high moisture vapour and gas barrier, seven to ten times that of PE. OPP coated with PVDC in low gauge form provides a high barrier to moisture vapour.

### PE

Polyethylene is a thermoplastic polymer consisting of long chains of the monomer ethylene. It is manufactured through polymerization of ethane and is classified mainly on its density and branching. Its mechanical properties depend on the extent and type of branching, the crystal structure and its molecular weight. It was first synthesized by the German chemist Hans von Pechmann in 1898 while heating diazomethane. However, it was not until 1933 that the first industrially practical PE was synthesized by Eric Fawcett, Reginald Gibson and Michael Perrin at the ICI works in the UK. Perrin developed a reproducible high-pressure synthesis for PE that became the basis for industrial LDPE in 1939. Subsequently catalysts have been developed to promote ethylene polymerization at milder temperatures and pressures. By the end of the 1950s both the Phillips catalyst (developed by 1951 by Robert Banks and John Hogan at Phillips Petroleum) and Ziegler

type catalyst (developed by Karl Ziegler in 1953) were being used for HDPE production. A third type of catalyst system called metallocene was discovered in 1976 in Germany by Walter Kaminsky and Hansjörg Sinn. The Ziegler and metallocene catalysts are flexible at copolymerizing ethylene with other olefins and have become the basis for the wide range of polyethylene resins. Many different types of PE are currently available, including ultra-high molecular weight PE, high molecular weight PE, high-density PE and high-density cross-linked PE, but the ones used in MA packaging are: MDPE (medium-density PE), which is used in shrink film; LDPE and LLDPE, which have a high degree of toughness, flexibility and relative transparency; and VLDPE (very low-density PE) used in food packaging and stretch wrap. Polyethylene terephthalate (PET) can be used to manufacture thin oriented films with excellent thermal properties (–70V to +150°C), therefore it is used for pre-cooked meals and ‘boil in the bag’ but not for MA packaging of fresh fruit and vegetables.

### PET or PETE

Polyethylene terephthalate is a saturated thermoplastic polyester resin made by condensing ethylene glycol and terephthalic acid and is commonly used as a synthetic fibre called polyester. In MA packaging this film is either laminated or extrusion-coated with polyethylene. It is used in various forms in MA packaging, such as as a low gauge oriented film of high clarity as lid material and in crystalline or amorphous form as preformed or thermoformed trays.

### PO

Polyolefin is manufactured from olefins, especially ethylene and propylene, into a range of high-shrink multilayer high-speed machineable shrink-films with high tensile strength, high clarity and balanced heat shrinkage in all directions. For example, Cryovac is used for fruit, generally using foam trays.

## PP

Polypropylene is chemically similar to PE and can be extruded or coextruded to provide a sealant layer. It was first polymerized in Spain by Karl Rehn in 1951 and Giulio Natta in 1954.

## PS

Polystyrene is a clear brittle thermoplastic with a high tensile strength but a poor barrier to moisture vapour and gases. It can be blended with styrene-butadiene or styrene polybutadiene to achieve the required properties, but they can affect its clarity.

## PVC

Polyvinyl chloride is the most widely used as the thermo-formable base for MA packages, since it provides good gas barrier and a moderate barrier to moisture vapour, which varies with thickness. The properties of PVC can range in form from soft and flexible to hard and rigid, either of which may be solid or cellular, depending on the type of plasticizer used. It was discovered in 1835 by the French chemist Henri Victor Regnault when it appeared as a white solid inside flasks of vinyl chloride that had been left exposed to sunlight. In the early 20th century Ivan Ostromislensky and Fritz Klatte of the German chemical company Griesheim-Elektron attempted to use PVC in commercial products, but difficulties in processing prevented development. In 1926, Waldo Semon of the B.F. Goodrich Company developed a method to plasticize PVC by blending it with various additives, which made it more flexible and easier to process, resulting in its widespread commercial use. Polyvinylidene chloride (PVDC) is a copolymer of vinylidene chloride and with vinyl chloride is used as a gas barrier coating on polyester and OPP for films used for lids and in film form as a sandwiched barrier layer. It has low permeability to water

vapour and good barrier properties to O<sub>2</sub> and is heat sealable but it is not very strong.

## Xtend®

These films are manufactured from polymers that, it is claimed, give a 'correct balance' between O<sub>2</sub> and CO<sub>2</sub> while excess moisture is allowed to escape. Various films have been developed with a range of permeabilities. Xtend® films prevented water condensation inside the bags (Porat *et al.*, 2004) and are of two types: low water conductance, which give an internal humidity of 93%; and high water conductance, which give an internal humidity of 90% (Lichter *et al.*, 2005). Aharoni *et al.* (2008) used hydrophilic Xtend® films manufactured from various proprietary blends of polyamides with other polymeric and non-polymeric compounds with micro-perforations, which gave humidity levels inside the packs that prevented accumulation of condensed water. They showed that suitable combinations of O<sub>2</sub> and CO<sub>2</sub> were developed in the micro-perforated Xtend®.

## Biodegradable MA Packaging

Discarded plastics, including plastic films used in MA packaging, represent a major source of environmental pollution, which is constantly increasing in spite of publicity and legislation to control their use and disposal. Most plastics are by-products of the petrochemical industry and can take hundreds of years to break down. Plastics made from plants break down relatively quickly and can even have beneficial effects for use in horticulture when composted. The Biodegradable Products Institute in New York, USA, defines biodegradable plastic as 'one in which degradation results from the action of naturally occurring micro-organisms' (<http://www.bpiworld.org/>). The use of biodegradable plastic films in MA packaging is still uncommon, with films manufactured from petrochemicals being used almost exclusively. The advantages of these

petrochemical-based polymers include their high mechanical performance, ease of manufacture and good barrier properties but, overwhelmingly, their low cost.

Several biodegradable films have been developed and tested, including the following.

- PLA is a linear, semi-crystalline, aliphatic, biodegradable polyester that can be produced from lactic acid by the fermentation of renewable sources such as whey, corn, potatoes or molasses. It is used to make many products, including bottles and biodegradable medical devices as well as plastic films.
- Polybutylene succinate (PBS) is a biopolymer from poly-condensation of succinic acid and 1-4 butanediol, which has been used for biopolymer compounds. It is used for mulching crops, bags, bottles and medical devices as well as films
- Mater-Bi® is made from plant material including starches from cereals (particularly maize) and potatoes or from vegetable oils. It is a biodegradable film that has been used for many years as mulches, particularly for growing strawberries.

To test the rate of degradation of bags made from these three polymers in moist soil, Tengrang *et al.* (2015) packed 1 kg of rambutans in PLA, PBS, Mater-Bi® or LDPE bags and stored them at  $13 \pm 2^\circ\text{C}$  and 78% RH. It was found that Mater-Bi® and PLA films took some 100 days to start degrading, while PBS film started to degrade in 16 days and its degradation increased to 31.25% in 161 days. Rambutans that had been stored in the PBS bags had the lowest weight loss and retained their yellowish/red skin colour and fresh green spinterns, which was superior to the other three packages where the fruit had higher weight losses and tended to develop a brown skin. They concluded that packaging made from PBS film could extend the post-harvest life of rambutans for 15 days at  $13 \pm 2^\circ\text{C}$ . Also they found that the PBS bags and the Mater-Bi® bags were easier to produce and use than the PLA bags. The highest levels of both water vapour and  $\text{O}_2$  transmission rates were with Mater-Bi® film followed by PBS and PLA, respectively. Percentages of

elongation of Mater-Bi® in machine direction and transverse were significantly higher than the other two films, whilst the tensile strength of Mater-Bi® was the lowest in both machine direction and transverse, followed by PBS and PLA, respectively.

Kantola and Helén (2001) compared the effects of Mater-Bi® with LDPE film. They stored tomatoes at  $11 \pm 1^\circ\text{C}$  and 75–85% RH for 3 weeks in: perforated Mater-Bi® bags, Biopol®-coated paper board trays overwrapped with perforated Mater-Bi® bags, polylactide-coated paperboard trays overwrapped with perforated Mater-Bi® bags, perforated cellophane bags and perforated LDPE bags as a control. Package type had no significant influence on the tomatoes' sensory quality, but the tomatoes lost more weight in biodegradable packages than in LDPE packages. Giacalone and Chiabrando (2015) packed fresh cherries in plastic baskets, sealed with overwraps of either Mater-Bi® or PP film. They found that the quality of the cherries was similar after storage in both types of package, but the incidence of postharvest diseases was almost always lower in the fruit packed in Mater-Bi®, probably as a consequence of the higher  $\text{CO}_2$  levels within the package. Koidea and Shib (2007) compared the effects of PLA packaging on the microbial and physico-chemical quality of green peppers. They compared the PLA with both perforated and non-perforated LDPE film and stored them for 7 days at  $10^\circ\text{C}$ . The levels of coliform bacteria were increased by less than 1 log CFU/g in the PLA film packaging, 2.3 log CFU/g in the LDPE film non-perforated package and less than 1 log CFU/g in the perforated LDPE film package. The PLA film had higher water vapour permeability than the LDPE. They concluded that MA packaging of green peppers in PLA film can be used to maintain their quality and protect them from microbial and insect contamination. Mistriotis *et al.* (2016) developed a model simulating the gas diffusion through micro-perforations and the permeable membrane for MA packaging using PLA film. They reported that the relatively high water vapour permeability of PLA films compared with OPP films allowed the development of



biodegradable-based packaging for the specific application.

Llana-Ruiz-Cabello *et al.* (2015) tested PLA containing 2%, 5% and 6.5% Proallium® (extract of *Allium* spp.) on MA packaging of minimally processed salads. Antimicrobial activity was observed, mainly in films containing 5% and 6.5% Proallium®. Packing salads in Proallium® film (6.5%) was effective in controlling bacteria for up to 5 days, and up to 7 days in controlling fungi. Mali and Grossmann (2003) made a film from 4.0g of yam starch plus 1.3 or 2.0g glycerol in 100g of 'filmogenic' solution. They tested the film on packaging strawberries stored at 4°C and 85% RH and compared it with PVC film packaging. Both films significantly reduced decay of the fruits compared with control, but PVC reduced softening and weight loss more than the yam starch film. Brandelero *et al.* (2016) prepared biodegradable films from cassava starch, polyvinyl alcohol and sodium alginate and studied the effects of adding 0.5% essential oils from copaiba (*Copaifera reticulata*) or lemongrass (*Cymbopogon citratus*). The 50µm biodegradable film they used had 11.43–8.11 MPa resistance and 11.3–13.22% elongation, with water vapour permeability of 0.5–4.04 × 10<sup>-12</sup> g/s/Pa/m. They compared this film with PVC film on the storage of minimally processed lettuce at 6 ± 2°C. Lettuce stored in biodegradable films for 4 days remained similar to those freshly harvested, while those in PVC film lost quality after only 2 days, though they had up to eight times less mass loss than those stored in biodegradable films. The addition of essential oils to the films had no significant effects on the lettuce. They concluded that these biodegradable films were viable for the storage of minimally processed lettuce.

Peano *et al.* (2014) stored strawberries in two different single-layer films: a biodegradable film (from corn starch) and a 25 µm perforated PP film. These were stored at 2°C for 4 days or 2°C for 2 days followed by room temperature (20°C) for 2 days. The strawberries packed in biodegradable film and stored at 2°C retained their quality and maintained the optimum headspace composition during storage better than those in

perforated PP film. Giuggioli *et al.* (2015) stored raspberries at 1°C for 2 days followed by 18°C for 2 days in the following films: non-commercial biodegradable and compostable film; non-perforated prototypes (Novamont, Italy); non-perforated commercial PP film (Trepack, Italy); and commercial PP macro-perforated film (6 mm holes; Trepack, Italy). After 2 days the fruit colour of those packed in biodegradable film was similar to that of those that had been freshly harvested for both the passive and active atmospheres. When the temperature was increased, it was only the biodegradable film that facilitated storage of fruit for up to 4 days. They had a headspace containing 24.4–25.9 kPa CO<sub>2</sub>. Makino and Hirata (1997) used a film made from biodegradable laminate of a chitosan-cellulose and polycaprolactone for MA packaging for lettuce, broccoli, tomatoes, sweet corn and blueberries. They developed gas permeability coefficients that were designed to be close to the optimal composition needed for MA packaging. They concluded that this laminate was suitable as a packaging material for storage of fresh produce.

Biodegradable zein films are made from a hydrophobic protein produced from maize, but the films may swell and deform in prolonged contact with water, which seems to restrict their use in MA packaging of fruit and vegetables (Yoshino *et al.*, 2002). However, Rakotonirainy *et al.* (2008) reported that broccoli florets packaged in zein films (laminated or coated with tung oil) maintained their original firmness and colour during 6 days of storage. Films have been prepared by extrusion using acetylated and oxidized banana starches with LDPE (Torres *et al.*, 2008).

## Film Permeability

Respiring fresh fruits and vegetables sealed in plastic films will cause the atmosphere to change, in particular O<sub>2</sub> levels to be depleted and CO<sub>2</sub> levels to be increased. The transmission of gases and water vapour through plastic films can vary with the type of material from which they are made, the

use of additives in the plastic, temperature, humidity, the accumulation, concentration and gradient of the gas and the thickness of the material. Generally films are 4–6 times more permeable to CO<sub>2</sub> than to O<sub>2</sub>. However, Barmore (1987) indicated that the relationship between CO<sub>2</sub> and O<sub>2</sub> permeability and that of water vapour is not so simple. Variation in transmission of water vapour can therefore be achieved to some extent, independently of transmission of CO<sub>2</sub> and O<sub>2</sub>, using techniques such as producing multi-layer films by coextrusion or applying adhesives between the layers. Achieving the exact O<sub>2</sub> and CO<sub>2</sub> levels in MA packaging can be difficult. Film permeability to gases is by active diffusion where the gas molecules dissolve in the film matrix and diffuse through in response to the concentration gradient (Kester and Fennema, 1986). Schlimme and Rooney (1994) showed that there was a range of permeabilities that could be obtained from films with basically the same specifications. A formula to describe film permeability was given by Crank (1975) as follows:

$$P = \frac{Jx}{A(p_1 - p_2)}$$

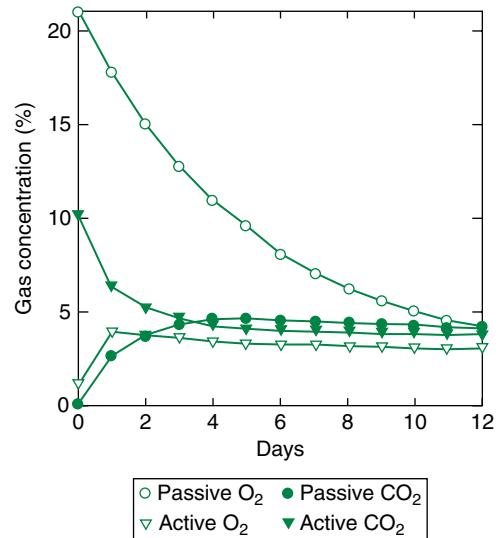
where  $J$  = volumetric rate of gas flow through the film at steady state,  $x$  = thickness of film,  $A$  = area of permeable surface,  $p_1$  = gas partial pressure on side 1 of the film, and  $p_2$  = gas partial pressure on side 2 of the film ( $p_1 > p_2$ ).

Permselectivity is a medical term used to define the preferential permeation of certain ionic species through ion-exchange membranes. Al-Ati and Hotchkiss (2003) applied the term to MA packaging (see 'Ionomers', above).

### Gas Flushing

The levels of CO<sub>2</sub> and O<sub>2</sub> can take some time to change inside the MA pack. In order to speed this process, the pack can be flushed with N<sub>2</sub> to reduce the O<sub>2</sub> rapidly, or the atmosphere can be flushed with an appropriate mixture of CO<sub>2</sub>, O<sub>2</sub> and N<sub>2</sub>. In other cases the pack can be connected to a vacuum

pump to reduce the headspace so that the respiratory gases can change within the pack more quickly. Gas flushing is more important for non-respiring products such as meat or fish, but it can profitably be used with fresh fruit and vegetables. Zagory (1990) showed the effects of flushing fresh chilli peppers stored in plastic film with a mixture containing 10 kPa CO<sub>2</sub> + 1 kPa O<sub>2</sub> compared with no gas flushing (Fig. 14.1). In work described by Aharoni *et al.* (1973), yellowing and decay of leaves of the lettuce cultivar 'Hazera Yellow' were reduced when they were pre-packed in closed PE bags in which the O<sub>2</sub> concentration was reduced by flushing with N<sub>2</sub>. Similar but less effective results were obtained when the lettuce was pre-packed in closed PE bags not flushed with N<sub>2</sub>, or when open bags were placed in PE-lined cartons. Andre *et al.* (1980a) showed that fungal development during storage could be prevented in asparagus spears by packing them in PE bags with silicon elastomer windows and flushing with 30 kPa CO<sub>2</sub> for 24 h or by maintaining a CO<sub>2</sub> concentration of 5–10 kPa.



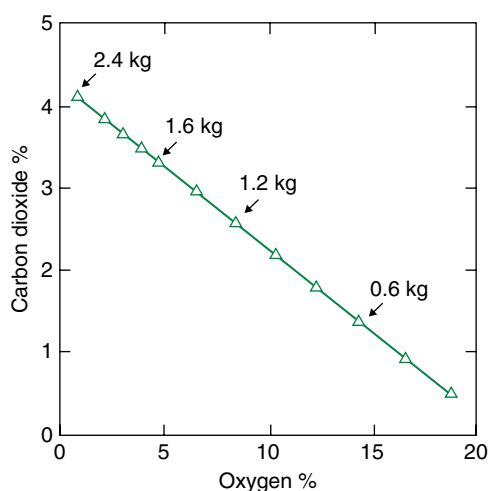
**Fig. 14.1.** A comparison of passive modified atmosphere packaging and active modified atmosphere packaging on the rate of change of CO<sub>2</sub> and O<sub>2</sub> in Anaheim chilli pepper fruit packed in Cryovac SSD-310 film. Reproduced from Zagory (1990) with permission.

## Quantity of Product

The quantity of produce inside the sealed plastic film bag in relation to the size of the bag has been shown to affect the equilibrium gas content (Table 14.1), but the levels of CO<sub>2</sub> and O<sub>2</sub> do not always follow what would be predicted from permeability data and respiration load of the crop. Zagory (1990) also demonstrated the relationship between the weight of produce sealed in 20 cm × 30 cm Cryovac SSD-310 film bags and its O<sub>2</sub> and CO<sub>2</sub> content was linear (Fig. 14.2). However, this varied considerably with a four-fold variation

**Table 14.1.** Effects of the amount of asparagus spears sealed inside plastic bags on the equilibrium CO<sub>2</sub> and O<sub>2</sub> content at 20°C; modified from Lill and Corrigan (1996). Film used was W R Grace RD 106 polyolefin shrinkable multilayer film with anti-fog properties with 23,200 CO<sub>2</sub>, 10,200 O<sub>2</sub> permeability at 20°C in ml/m<sup>3</sup>/atm/day.

Weight (g) of asparagus spears inside the bag	Equilibrium kPa CO <sub>2</sub>	Equilibrium kPa O <sub>2</sub>
100	2.5	9.7
150	3.2	6.2
200	3.5	4.1



**Fig. 14.2.** Equilibrium gas concentration in a modified atmosphere package as a function of weight for Anaheim chili pepper fruit packed in Cryovac SSD-310 film. Reproduced from Zagory (1990) with permission.

in produce weight, which illustrates the importance of varying just one factor. Thompson *et al.* (1972b) also showed that the number of fruit packed in each plastic bag can alter the effect of MA packaging where plantains packed with six fruits in a bag ripened in 14.6 days compared with 18.5 days when fruits were packed individually during storage in Jamaican ambient of 26–34°C.

## Perforation

The appropriate CO<sub>2</sub> and O<sub>2</sub> levels can be achieved for fruit or vegetables with low or medium respiration rates where they are sealed in films such as LDPE, PP, OPP and PVC. However, Rai and Shashi (2007) pointed out that for highly respiring produce such as mushroom, broccoli, asparagus and Brussels sprouts, packaging in these films can result in fermentation. Punching holes in the plastic can maintain a high humidity around the produce, but it may be less effective in delaying fruit ripening because it does not have the same effect on the CO<sub>2</sub> and O<sub>2</sub> content of the atmosphere inside the bag. The holes may be very small and in these cases they are commonly referred to as micro-perforations. Larger holes are sometimes referred to as macro-perforations, but there is no definition as to where one ends and the other begins.

Various studies have been conducted on the effects of perforations in the packing fruit and vegetables at different temperatures on the internal gas content. Perforated bags used for low-respiring fruit and vegetables can have both positive and negative effects. For example, Hardenburg (1955) found that onions in non-perforated bags tended to produce roots, but when the bags were perforated to address this problem the weight loss in the onions increased (Table 14.2). Pears of the cultivars 'Okusankichi' and 'Imamuraaki' had the highest levels of decay if placed in non-sealed bags with 5 pinholes, but bags with ten holes had more decay. Weight loss of fruit after 7 months of storage was 8–9% in non-sealed bags and < 1% in sealed bags. In sealed

**Table 14.2.** Effects of number and size of perforation in 3 lb (1.36 kg) 150 gauge PE film bags of 'Yellow Globe' onions on the relative humidity in the bags, rooting of the bulbs and weight loss after 14 days at 24°C (Hardenburg, 1955).

Number of perforations	Perforation size (mm)	% RH in bag	% bulbs rooted	% weight loss
0	–	98	71	0.5
36	1.6	88	59	0.7
40	3.2	84	40	1.4
8	6.4	–	24	1.8
16	6.4	54	17	2.5
32	6.4	51	4	2.5
Kraft paper with film window	–	54	0	3.4

bags, CO<sub>2</sub> concentration reached 1.9 kPa after 2 months of storage (Son *et al.*, 1983).

### Adjustable Diffusion Leak

A simple method of CA storage was developed for strawberries (Food Investigation Board, 1920). The gaseous atmospheres containing reduced amounts of O<sub>2</sub> and moderate amounts of CO<sub>2</sub> were obtained by keeping the fruit in a closed vessel fitted with an adjustable diffusion leak. Marcellin (1973) described the use of PE bags with silicone rubber panels that allowed a certain amount of gas exchange and have been used for storing vegetables. Good atmosphere control within the bags was obtained for globe artichokes and asparagus at 0°C and green peppers at 12–13°C when the optimum size of the bag and the silicon gas-exchange panel was determined. Rukavishnikov *et al.* (1984) described CA storage in 150–300 µm PE film with windows of membranes selective for O<sub>2</sub> and CO<sub>2</sub> permeability made of polyvinyl trimethyl silane or silicon organic polymers. Apples and pears stored at 1–4°C under the covers had 93% and 94% sound fruit, respectively, after 6–7 months, whereas those stored non-wrapped at similar temperatures resulted in over 50% losses after 5 months. Hong *et al.* (1983) described the storage behaviour of *Citrus unshiu* fruits at 3–8°C in plastic films

with or without a silicone window. The size of the silicone window was 20–25 cm<sup>2</sup>/kg fruit, giving < 3 kPa CO<sub>2</sub> + > 10 kPa O<sub>2</sub>. After 110–115 days, 80.8–81.9% of fruits stored with the silicone window were healthy with good coloration of the peel and excellent flavour, while those without a silicone window had 59.4–76.8% with poor peel coloration and poor flavour. Those stored without wrappings for 90 days had 67% healthy fruit with poor quality and shrivelling of the peel, calyx browning and a high rate of moisture loss. The suitability of the silicone membrane system for CA storage of 'Winter Green' cabbage was studied by Garipey *et al.* (1984) using small experimental chambers. Three different CA starting techniques were then evaluated and there was a control with the cabbages stored in air at the same temperature. The silicone membrane system maintained CA storage conditions of 3.5–5 kPa CO<sub>2</sub> + 1.5–3 kPa O<sub>2</sub>, where 5% and 3%, respectively, were estimated theoretically. After 198 days of storage, total mass loss was 14% in CA storage compared with 40% in air. The three methods used to achieve the CA storage conditions did not have any significant effect on the storability of cabbage. Cabbage stored in CA showed better retention of colour, a fresher appearance and a firmer texture compared with those stored in air. Raghavan *et al.* (1982) used a silicone membrane system and found that carrots could be stored for up to 52 weeks and celery, swedes and cabbage for 16 weeks. Design calculations for selection of the membrane area were presented in the paper. Chinquapin fruit (*Castanea henryi*) packed in PE film bags, each with a silicone rubber window, and stored at 1°C had a longer storage life, a slower decline in vitamin C, sugars and starch, reduced desiccation and a lower occurrence of 'bad' fruits compared with those stored in gunny bags (Pan *et al.*, 2006).

### Absorbents

Chemicals can be used with MA packaging to remove or absorb ethylene, O<sub>2</sub>, CO<sub>2</sub> and water. Such systems are sometimes referred

to as 'active packaging' and have been used for many years. They involve chemicals being placed within the package or they may be incorporated into the packaging material. One such product was marketed as 'Ageless' and used iron reactions to absorb O<sub>2</sub> from the atmosphere (Abe, 1990). AA-based sachets (which also generate CO<sub>2</sub>) and catechol-based sachets are also used as O<sub>2</sub> absorbers. The former are marketed as Ageless G<sup>®</sup> or Toppan C<sup>®</sup>, or Vitalon GMA<sup>®</sup> when combined with iron, and the latter as Tamotsu<sup>®</sup>. Mineral powders are incorporated into some films for fresh produce, particularly in Japan (Table 14.3).

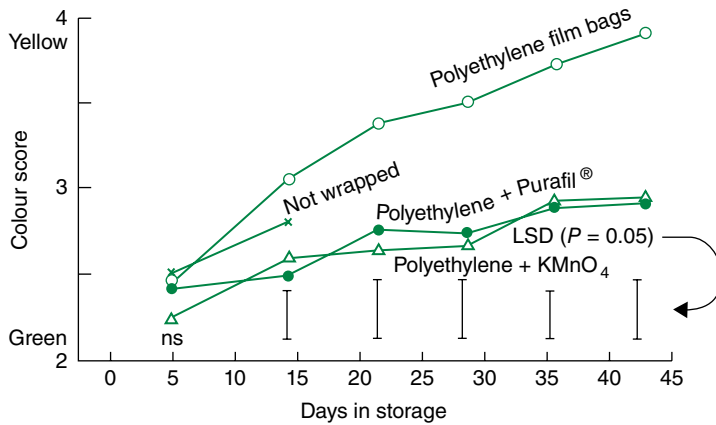
Proprietary products such as Ethysorb<sup>®</sup> and Purafil<sup>®</sup> are available, which are made by impregnating an active alumina carrier (Al<sub>2</sub>O<sub>3</sub>) with a saturated solution of potassium permanganate and then drying it. The carrier is usually formed into small granules; the smaller the granules, the larger is the surface area and therefore the quicker are their absorbing characteristics. Any molecule of ethylene in the package atmosphere that comes into contact with the granule will be oxidized, so the larger surface area is an advantage. The oxidizing reaction is not reversible and the granules change colour from purple to brown, which indicates that they need replacing. Strop (1992) studied the effects of storing broccoli in PE film bags with and without Ethysorb<sup>®</sup>. She found that the ethylene content in the bags after 10 days at 0°C was 0.423 µl/l for those without Ethysorb<sup>®</sup> and 0.198 µl/l for those with Ethysorb<sup>®</sup>. However, Scott *et al.* (1971) showed that the inclusion of potassium permanganate in sealed packages reduced the mean level of ethylene from 395 µl/l to 1.5 µl/l

and reduced brown heart from 68% to 36% in stored pears. Packing limes in sealed PE film bags inside cartons resulted in a weight loss of only 1.3% in 5 days, but all the fruits de-greened more rapidly than those that were packed without PE film, where the weight loss was 13.8% (Thompson *et al.*, 1974b). However, this de-greening effect could be countered by including an ethylene absorbent in the bags (Fig. 14.3). Where potassium permanganate was included in the bags containing bananas, the increase in storage life was 3–4 times compared with non-wrapped fruit and they could be stored for 6 weeks at 20°C or 28°C and 16 weeks at 13°C (Satyan *et al.*, 1992). The rate of ethylene removal from packages using potassium permanganate on an alumina carrier is affected by humidity (Lidster *et al.*, 1985). At the high humidity found in MA packages, the rate of ethylene removal of potassium permanganate was shown to be reduced. Kiwifruit of the cultivar 'Bruno' were stored at -1°C in sealed PE bags of varying thickness, with and without Ethysorb<sup>®</sup>. There was a slower rate of fruit softening and improved keeping quality during 6 months of storage of kiwifruit at -1°C in sealed PE bags 0.04–0.05 mm thick containing an ethylene absorbent. The average composition of the atmosphere in these bags was 3–4 kPa CO<sub>2</sub> + 15–16 kPa O<sub>2</sub> and ethylene at < 0.01 µl/l (Ben Arie and Sonego, 1985).

Moisture scavengers, which can modify package humidity, include silica gel, clays (e.g. montmorillonite), calcium oxide, calcium chloride and modified starch (Day, 2008). A moisture control system was developed to form a liner inside boxes of fruit

**Table 14.3.** Commercially available films in Japan (Abe, 1990).

Trade name	Manufacturer	Compound	Application
FH film	Thermo	Ohya stone/PE	Broccoli
BF film	BF distribution research	Coral sand/PE	Home
Shupack V	Asahi-kasei	Synthetic-zealite	–
Uniace	Idemitsu pet chem	Silicagel mineral	Sweetcorn
Nack fresh	Nippon unicar	Cristobalite	Broccoli and sweetcorn
Zeomic	Shinanen	Ag-zeolite/PE	–



**Fig. 14.3.** Effects of modified atmosphere packaging in polyethylene film and an ethylene absorbent included within the pack on the rate of de-greening of limes stored in tropical ambient conditions of 31–34°C and 29–57% RH (Thompson *et al.*, 1974c).

to control humidity. Antifogging films allow the consumer to see the product clearly through the packaging films, which incorporate humidity absorbers, hydrophilic liners, or micro-perforations in the film (Ozdemir and Floros, 2004). However, moisture scavengers are primarily used in non-respiring foods, not fresh or minimally processed fruit or vegetables.

Modified interactive packaging (MIP) uses an LDPE impregnated with a concentration of minerals. The permeability of the substrate created by porous activated clay particles creates micro-cracks in the film when a container is formed. These micro-cracks allow for an interaction to take place between the produce, O<sub>2</sub>, CO<sub>2</sub>, water, other gases and the film. This allows for better control of the permeability of the film, but in tests the exchange rate did not exceed 3 parts O<sub>2</sub> to 1 part CO<sub>2</sub>. Also high humidity is maintained inside the package with minimal condensation, which is claimed to reduce the risk of spoilage through fungal or bacterial infection. As excess condensation inside the punnet/liner/bag builds up, it was claimed that it was removed via 'osmotic diffusion through the film membrane and vaporised'. The major advantage is that produce moisture is not affected or removed, which would normally result in dehydration and early deterioration (Richard King, FreshTech, personal communication). One

such film is marketed as 'Fresh 'n' Smart'. The film itself has a honeycomb structure with a number of tiny crevices, so that the PE molecular structure is broken down and yet retains its strength, toughness and elasticity. It is recommended that for optimum results, products should be packed in the bags dry and pre-cooled. Once sealed the bags should be kept in well-ventilated cartons or crates to allow air circulation. Fresh 'n' Smart uses resins in the manufacture of its film, which improves both the strength and elasticity. This means that the bags will be less prone to tearing and puncturing, resulting in a very durable and effective film. The resins are of approved food grade.

Adamicki (2001) cooled various vegetables to their optimum storage temperature and placed them either in MIP packages or into the boxes lined with PE film 35 µm thick without minerals. In the MIP packages a gas composition was maintained that resulted in maintenance of the quality and marketable value of the vegetable. The storage periods were 6–9 weeks for broccoli, 4–6 weeks for tomatoes, 4 weeks for sweet peppers and 4–5 weeks for 'Crisphead' lettuce in MIP packages. At 0°C the concentration of CO<sub>2</sub> inside the MIP increased up to 5.6–5.9 kPa in 2 days and then slowly decreased, reaching 3.5–3.7 kPa after 16 days, and stayed around this level to the end of the storage period. The concentration of O<sub>2</sub>

inside the packages decreased from 21 to 3.5 kPa after 11 days and then remained at this level, with small fluctuation, throughout the period of storage. Subsequently Adamicki and Badelek (2006) reported that broccoli stored in MIP had better retention of green colour, firmness, quality and weight loss. Iceberg lettuce stored in MIP packages had less butt discoloration and rotting of leaves and improved marketable quality compared with those in boxes lined with PE film without minerals.

### Ethylene-absorbing films

Various chemicals and formulations, as well as packing methods, have been used as inclusions within MA packaging of fresh fruit and vegetables in order to reduce their deterioration. These chemicals are mainly in the form of sachets that are sealed inside the plastic film bag along with the fruit or vegetables. Less successful are the chemicals that have been incorporated into the plastic film used for packaging. Joyce (1988) described an LDPE film extruded with finely divided crysburite ceramic and marketed by the Odja Shoji Co. Ltd, Japan, as BO Film, which was claimed to have ethylene-absorbing capacity. However, Chairat and Kader (1999) tested commercial products that could be incorporated into PE film and found that the ability of all the products they tested to remove ethylene was far lower than that of potassium permanganate. Zagory (1995) reviewed the subject and reported various materials that have been incorporated into plastic films to remove ethylene, including finely ground coral, a powdered zeolite in viscous oil or grease and oya stone. He also reported that an ethylene-absorbing PE film had been patented by Matsui (1989), which incorporated porous material derived from pumice, zeolite, active carbon, cristobalite or clinoptilolite with a small amount of metal oxide. These porous materials were dispersed and compacted into the PE by heat or pressure without melting the plastic. Plastic films that absorb ethylene are still available; for example, one is marketed by Active Pack International under the name of Bio Fresh®. Fibreboard is available for constructing boxes

for fresh fruit and vegetables and one is marketed as PrimePro® (Chantler Packaging, Mississauga, Ontario). Gelpack® is a patented film made from PE blended with crushed volcanic rock during manufacture creating microscopic cavities in the film.

### Activated carbon

Activated carbon, also called activated charcoal, can hold molecules such as ethylene and CO<sub>2</sub>. When fresh air is passed through activated carbon the molecules are released. Incorporation of 30% activated carbon with 0.3% glucomannan in rice straw paper produced 77% ethylene-scavenging capacity (Sothornvit and Sampoompuang, 2012). This combination was proposed as a possible environmentally friendly scavenger that could be incorporated into MA packages.

### Zeolites

Zeolite are a group of over 200 hydrated alumino-silicate minerals that are made up of interlinked tetrahedra of alumina (AlO<sub>4</sub>) and silica (SiO<sub>4</sub>). Zeolites have an open structure in which other molecules can be trapped and Abeles *et al.* (1992) reported that they can absorb 8% ethylene by weight.

### Palladium

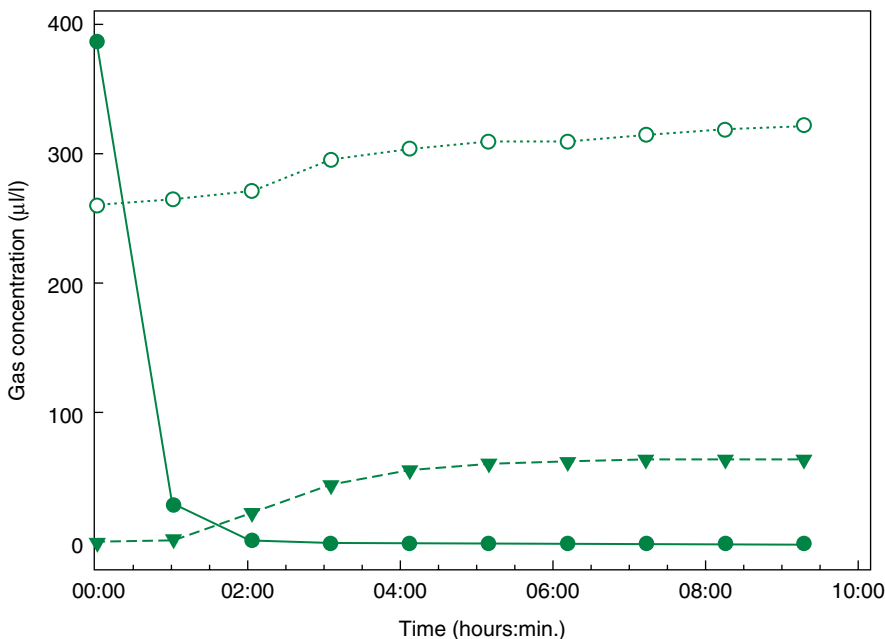
Palladium is not a very reactive metallic element but can absorb up to 900 times its own weight in hydrogen, which passes into the space between atoms. Catalytic converters, used in motor vehicles, may also contain Pd. In MA packaging, palladium chloride and light-activated titanium dioxide are metal catalysts that can be used to accelerate the oxidation reaction of potassium permanganate, thus increasing its absorption capacity by about sixfold (Martinez-Romero and Bailén 2007). Bailén *et al.* (2006) included granular activated carbon impregnated with palladium (ACPd) in MA packaging of tomatoes and compared them with MA packaging alone. ACPd changed the steady-state

atmosphere from 4 kPa O<sub>2</sub> and 10 kPa CO<sub>2</sub> in MA packaging alone to 8 kPa O<sub>2</sub> and 7 kPa CO<sub>2</sub> in ACPd. The addition of ACPd also reduced ethylene accumulation inside packages and reduced the colour change, softening and weight loss and delayed decay. After storage, the tomatoes that had been stored with ACPd had higher taste-panel scores for sweetness, firmness, juiciness, colour, odour and flavour than tomatoes that had been stored in MA packaging alone. Terry *et al.* (2007) described a zeolite impregnated with palladium. At 20°C its ethylene absorption capacity was 4162 µl/g at approximately 100% RH and 45,600 µl/g under low humidity. They compared the ethylene absorption capacity of zeolite impregnated with palladium with potassium permanganate and found it outperformed it by approximately sixfold under high humidity and about 60-fold under low humidity. In a plug flow reactor containing ethylene at 200 µl/l using 0.1 g of zeolite impregnated with palladium, the ethylene level rapidly fell to zero (Fig. 14.4). Sachets containing zeolite impregnated with palladium are marketed by

'It's Fresh!' (Fig. 14.5). Cao *et al.* (2015) found that ethylene removal rate of acidified activated carbon powder impregnated with palladium chloride (10 mg/g) was increased with the addition of copper sulfate (30 mg/g) with a maximum removal of 21.77 ml/g at 25°C and 20.18 ml/g at 5°C. Its ethylene removal capacity could be regenerated to 81.6% by heating at 175°C for 20 min.

### Oxygen scavengers

Sanjeev and Ramesh (2006) reviewed the use of low O<sub>2</sub> in packaged foods. Essentially O<sub>2</sub> scavengers are used in plant foods that have been thermally processed and therefore killed. However, they quote examples of minimally processed potatoes, apples, banana and peaches, where their minimal processing increased the oxidation/reduction potential and could accelerate the formation of such reactive O<sub>2</sub> species as peroxide ions and superoxide anions that could be detrimental. Excessive O<sub>2</sub> can cause the oxidation of phytochemicals, including vitamins,



**Fig. 14.4.** Effect of palladium-promoted material on gaseous composition within sealed jars, initially containing 550 µl ethylene/l in air. CO<sub>2</sub> (O); ethylene (●); ethane (▼). Reproduced from Terry *et al.* (2007) with permission.



pigments, flavour compounds and lipids, as well as facilitating the growth of aerobic microorganisms. Most commercially available  $O_2$  scavengers are used in processed food and moisture scavengers can also be used (Fig. 14.6).  $O_2$  scavengers are reactive, based on iron oxidation to iron oxide, which can reduce the  $O_2$  concentration in the headspace to less than 100 ppm in packages containing non-respiring foods.

## Vapour Generation

### Sulfur dioxide

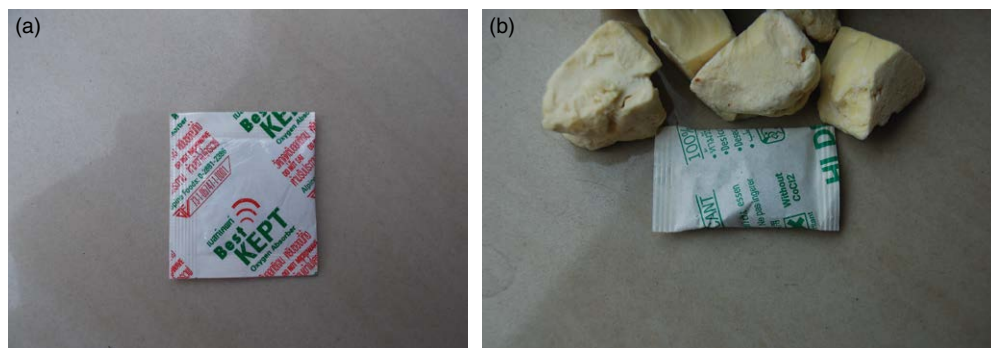
Fumigation of fruit, especially fresh grapes, with  $SO_2$  has been used commercially for



**Fig. 14.5.** 'It's Fresh!' ethylene scavengers used in strawberry retail packs. Reproduced with permission from Simon Lee, It'sFresh!, Cranfield, UK.

many years to control postharvest diseases, mainly grey mould caused by *Botrytis cinerea*.  $SO_2$  has primarily been used as a fumigant before storage or packaging of grapes, but it has also been used on other fruit; for example, in litchis to prevent discoloration (Ragnoi, 1989); and Rodriguez and Zoffoli (2016) found that 100–150  $\mu\text{l/l}$  at 20°C for 30 min controlled decay of blueberries during subsequent storage at 0°C for up to 45 days. Packing  $SO_2$  fumigated fruit into perforated LDPE bags (60  $\mu\text{m}$  thick) was better than packaging the fruit in non-perforated LDPE bags, where the  $CO_2$  steady state partial pressure that controlled decay was close to a level that injured the fruit. In non-perforated LDPE bags a steady-state content of 4.5–6.0 kPa  $CO_2$  + 13–15.0 kPa  $O_2$  developed, depending on cultivar.

Sodium metabisulfite has been used in pads that can be included in MA packaging.  $Na_2S_2O_5$  reacts with water in the atmosphere and releases  $SO_2$ . Grey mould incidence of inoculated grapes was reduced from 91.7% to 1.1% when  $SO_2$  pads were included in the MAP pack (Cayuela *et al.*, 2009). However,  $SO_2$  pads have a limited life and Zutahy *et al.* (2008) reported that the  $SO_2$  level had reduced to below 1  $\mu\text{l/l}$  after 60–80 days at 0°C when  $SO_2$  pads were included in packs of grapes. They described a dual-release  $SO_2$  pad containing a quick-release plus a slow-release phase. In order to extend the life of the pads, they enclosed them in plastic laminate with macro-perforation and found that the optimum laminate had 32 perforations



**Fig. 14.6.** (a) Sachet containing an oxygen scavenger used in processed food and (b) moisture scavenger from a sealed bag of freeze dried durian, both in food processing industry in Thailand.

of 6 mm diameter. This pad was tested in storage at 0°C for up to 116 days, and decay caused by *B. cinerea* and SO<sub>2</sub> damage were greater in the grapes with dual-release pads than the same pads in macro-perforated laminate. Ragnoi (1989), working with litchi, showed that fruits packed in sealed 37.5 µm PE film bags containing 2 kg of fruit and SO<sub>2</sub> pads could be kept in good condition at 2°C for up to 2 weeks, while fruits that were stored without packaging were discoloured and unmarketable.

### Alcohol

Ethanol is in constant and common use for all kinds of cleaning purposes because of its biocidal properties. Chervin *et al.* (2003) reported preliminary results on the optimum ethanol dose for effective disease control in grapes in MA packages. Ethanol vapour was as effective as SO<sub>2</sub> pads in controlling grey mould. Consumer panels detected no significant difference in sensory perception between ethanol-treated and non-treated grapes. Stem browning was higher than with SO<sub>2</sub> treatments in fruit exposed to 5 ml ethanol/kg of fruit compared with a dose rate of 2 ml/kg. Karabulut *et al.* (2004) demonstrated that the addition of ethanol significantly improved the efficacy of a hot water treatment applied to grapes that had been inoculated with *B. cinerea*. A pre-storage dip of grapes in either 33% or 50% ethanol and then stored in MA packaging maintained their quality for up to 7 weeks with 3 days shelf-life at 20°C. The treatment was similar to or better than replacing the ethanol with SO<sub>2</sub> releasing pads (Lichter *et al.*, 2005). However, in *in vitro* studies on *B. cinerea*, ethanol fumigation with 100 µl/l for 3 or 6 min, or 200 µl/l for 6 min actually enhanced its growth (Lagopodi *et al.*, 2009), which may indicate that the fungus can utilize gaseous ethanol, at sub-toxic concentrations, for nutrition. Chu *et al.* (1999) inoculated cherries with spores of *B. cinerea* and then fumigated them with thymol, acetic acid or ethanol at 30 mg/l for 25 min before sealing them in MA packaging. After 10 weeks of storage, both thymol and acetic

acid reduced grey mould from 36% to 0.5% and 6%, respectively, while those exposed to ethanol had similar levels to those not treated. Dong *et al.* (2016) fumigated fresh-cut burdock roots (*Arctium lappa*) with ethanol at 450–600 µl/l for 5 or 12 h and found that ethanol fumigation reduced their respiration rate, lowered the activity of PPO and PAL and reduced total phenolic content, thus prevented browning.

### Hexenal

*Trans-2-hexenal* is released in many fruit and vegetables and is a major volatile compound in ripe tomatoes; it also contributes to the odour of other plants, including the smell of freshly cut green grass. Artes-Hernández *et al.* (2004) stored grapes at 0°C for 41 days in MA packaging sealed in micro-perforated PP film (35 µm thick) followed by 4 days in air also at 0°C and then 3 days at 15°C. They tested placing a filter paper soaked in hexenal (15 or 10 µl) in controlling postharvest diseases. The packs containing hexenal had the highest score for flavour and the lowest level of fungal disease (6%). Fumigation of grapes with 100 or 200 ml of *trans-2-hexenal* per 1.1 l container for 2 weeks at 2°C significantly lowered mould growth but did not affect fruit firmness or TSS content during subsequent storage at 2°C for up to 12 weeks (Archbold *et al.*, 1990). Guo *et al.* (2015) also showed that the incidence of natural or artificial infection by *B. cinerea* in tomatoes was significantly decreased after treatment with *trans-2-hexenal*. They also found that *trans-2-hexenal* increased the mRNA levels of genes encoding ethylene receptors, lipoxygenase and PAL, but did not induce the mRNA accumulation of the pathogenesis-related proteins, such as PR-1a or PR-5. The activities of peroxidases and PAL and the accumulation of phenolic compounds were significantly enhanced after *trans-2-hexenal* treatment. The inclusion of hexenal and *trans-2-hexenal* in MA packaging of apple slices gave a significant extension in their shelf-life and reduced the growth potential of naturally occurring bacteria (Corbo *et al.*, 2000). Exposure of pears to *trans-2-hexenal* vapour for at least

8 h at 20°C before storage at 1°C was required to control *Penicillium expansum* and reduce their patulin content. Treatment did not affect fruit quality and after 6 days at 20°C after storage *trans*-2-hexenal residue in treated fruits was less than the natural content in unripe fruits (Neri *et al.*, 2006).

### Essential oils

Exposure to various essential oils has been successfully tested on insect control in stored grain (Lopez *et al.*, 2008), but also as supplement in MA packaging to control diseases. Fungal spore production, spore germination and germ tube length of *Colletotrichum coccodes*, *B. cinerea*, *C. herbarium* and *Rhizopus stolonifer* was inhibited *in vitro* with lemon-grass oil. Also spraying or dipping fruit in essential oils could control postharvest diseases of several fruits (Tzortzakakis, 2007). In *in vitro* studies of fumigation with the essential oils from *Illicium verum* and *Schizonepetae nuifolia*, both oils gave good control of *B. cinerea* and *C. gloeosporioides*. Shin *et al.* (2014) also found that fumigation with thymol (30 µg/ml) or linalool (120 µg/ml) significantly inhibited mycelial growth and conidia germination of *B. cinerea*. They also found that the occurrence rate of *B. cinerea* and other non-identified fungi was significantly reduced by fumigation of several grape cultivars with thymol. In this study, in addition to the reduced fungal infection, fumigation with thymol had no effect on the sugar content and firmness of the grapes. As was indicated above, Chu *et al.* (1999) found that cherries inoculated with spores of *B. cinerea* and then fumigated with thymol at 30 mg/l developed less grey mould in storage. Also the cherries that had been fumigated with thymol had lower TSS, higher TA and higher stem browning than other treated or non-treated cherries. Plotto *et al.* (2003) found that dipping tomatoes inoculated with *B. cinerea* and *Alternaria arborescens* in emulsions of oils of thyme and oregano at 5000 and 10,000 ppm reduced disease development. However, these treatments did not stop disease development in inoculated tomatoes

and some vapours appeared to be phytotoxic after prolonged exposure. They concluded that essential oils may provide alternatives and supplements to conventional antimicrobial additives in foods. Arras and Usai (2001) in *in vitro* studies found that *Thymus capitatus* essential oil at 250 µl/l was toxic to *Penicillium digitatum*, *P. italicum*, *B. cinerea* and *Alternaria citri*. The fungitoxic activity of *T. capitatus* essential oil at 75, 150, and 250 ppm on healthy oranges, inoculated with *P. digitatum* by spraying and placed in 10 l desiccators, was weak at atmospheric pressure (3–10% inhibition at all three concentrations). In a vacuum (0.5 bar), conidial mortality on the exocarp was 90–97% at all three concentrations. These data proved not to be significantly different ( $p = 0.05$ ) from treatments with thiabendazole at 2000 ppm. Mehra *et al.* (2013) tested biofumigation with essential oils (cinnamon oil, linalool, *p*-cymene, and peppermint leaf oil) and the plant oil-derived biofungicides Sporan (rosemary and wintergreen oils) and Sporotec (rosemary, clove, and thyme oils) on blueberries. The fruits were inoculated with conidial suspensions of *Alternaria alternata*, *B. cinerea*, *Colletotrichum acutatum*, or sterile deionized water as a control before fumigation and stored at 7°C for 1 week. Sporotec volatiles reduced disease incidence in most cases, whereas other treatments had no consistent effect on postharvest decay. Sensory analysis showed negative impacts on sourness, astringency, juiciness, bitterness and blueberry-like flavour. They concluded that the potential for postharvest biofumigation of blueberries under refrigeration appears limited. Valverde *et al.* (2005) showed that the addition of 0.5 µl of eugenol, thymol or menthol inside MA packages containing the cultivar 'Crimson Seedless' reduced weight loss and colour change, delayed rates of pedicel deterioration, retarded TSS:acid ratio development, maintained firmness and reduced decay. Also the total viable counts for mesophilic aerobics, yeasts and moulds were significantly reduced in the grapes packaged with eugenol, thymol or menthol. The authors concluded that the inclusion of one of these essential oils resulted in three additional

weeks of storage compared with MA packaging only. The equilibrium atmosphere in PP bags in storage at 1°C (1.2–1.3 kPa CO<sub>2</sub> + 13–14 kPa O<sub>2</sub>) was reached in 21 days. The inclusion of a sachet impregnated with either 0.5 ml thymol or 0.5 ml menthol inside the bags reduced yeasts, moulds and total aerobic mesophilic colonies and also improved the visual aspect of the rachis. The authors postulated that these essential oils could be an alternative to the use of SO<sub>2</sub> in controlling disease of grapes (Martinez-Romero *et al.*, 2005). Elgayyar *et al.* (2001) evaluated essential oils extracted from anise, angelica, basil, carrot, celery, cardamom, coriander, dill weed, fennel, oregano, parsley and rosemary against *Listeria monocytogenes*, *Staphylococcus aureus*, *Escherichia coli* O:157:H7, *Yersinia enterocolitica*, *Pseudomonas aeruginosa*, *Lactobacillus plantarum*, *Aspergillus niger*, *Geotrichum* and *Rhodotorula*. Oregano essential oil showed the greatest inhibition. Coriander and basil oils were also highly inhibitory to *E. coli* O:157:H7 and to the other bacteria and fungi tested. Anise oil had little inhibitory effect on bacteria but it was highly inhibitory to the fungi. There was no inhibition of either fungi or bacteria with carrot oil. The effects of sachets containing eugenol, carvacrol and *trans*-anethole extracted from herbs were tested in packages of fresh organic wild rocket (*Diplotaxis tenuifolia*), but they had no clear effect on the microbial load (Wiecznyńska *et al.*, 2016).

### Acetic acid

Acetic acid has been well established for its antifungal and antibacterial properties. For example, Sholberg and Gaunce (1995) found that fumigation with acetic acid gave good control of diseases caused by *B. cinerea*, *Penicillium expansum* and *P. italicum*. The levels of acetic acid they used were over the range of 2.0–5.4 mg/l in air at temperatures over the range of 2–20°C on citrus, apples, pears, tomatoes, grapes and kiwifruit. Increasing the humidity from 17% to 98% RH increased the effectiveness of acetic acid fumigation at both 5°C and 20°C. Acetic acid,

at the concentrations used in their trials, had no apparent phytotoxic effects on the fruit. Sholberg *et al.* (2001) artificially or naturally contaminated apples with conidia of *P. expansum* then fumigated them with acetic acid vapour in a 1 m<sup>3</sup> gas-tight chamber at 10°C for times between 1 h and 24 h or dipped in thiabendazole at 450 µl/l. The fruits were then stored at 1°C for 3 months or more. Results showed that fumigation with acetic acid was as effective as thiabendazole in controlling blue mould decay. Perkins *et al.* (2017) found that red bayberries (*Morella rubra*) exposed for 1 h to ultrasonic fog treatment with acetic acid had 18% decay compared with 70% for non-treated fruit after 10 days at 5°C with no adverse effect on their organoleptic properties.

Acetic acid is the main organic compound in vinegar, but the use of vinegar with fresh fruit and vegetables is restricted by its effects on smell and taste. Krusong *et al.* (2015) carried out *in vitro* tests on the effects of baby corn fermented vinegar on cultures of *B. cinerea* and found complete inhibition of their growth. In order to counter the effects on smell and taste, they subsequently either sprayed or exposed strawberries to strawberry flavoured baby corn fermented vinegar vapour. Spraying extended shelf-life at 4°C to 7 days while the latter extended it to 11 days. A trained sensory panel found no significant effect on colour or texture between treated and non-treated, but they detected slight differences in flavour and taste directly after treatment, though not after longer storage. Lagopodi *et al.* (2009) showed that fumigation with acetic acid promoted, inhibited or had no influence on the growth of *B. cinerea in vitro*, with differences dependent on exposure time and dosage. Exposure to acetic acid at 4 µl/l for 6 min gave complete inhibition of the growth of *B. cinerea*.

### Nitric oxide

Wills *et al.* (2000) showed that fumigation of strawberries with NO at 1–4000 µl/l immediately after harvest and then storage at

5°C or 20°C in air containing ethylene at 0.1 µl/l extended postharvest life at both temperatures. The most pronounced effect was obtained with NO in the concentration range 5–10 µl/l, which produced more than 50% extension in shelf-life. Hu *et al.* (2014b) inoculated mangoes with *Colletotricum gloeosporioides* (which causes anthracnose) then fumigated them with the NO donor (sodium nitroprusside, 0.1 mM) at 25°C for 5 min. This fumigation effectively suppressed lesion development. Lichanporn and Techavuthiporn (2013) fumigated longkongs for 3 h with NO vapour, which delayed pericarp browning. However, the fumigated fruit had lower levels of PAL, PPO and peroxidase than the control fruit. Huque *et al.* (2013) treated apple slices with NO (10 µl/l), which showed delayed development of surface browning during storage at 5°C. NO treatment also resulted in a lower level of total phenols, inhibition of PPO activity, reduced ion leakage and reduced rate of respiration, but had no significant effect on ethylene production or levels of lipid peroxide and hydrogen peroxide. The effectiveness of the treatment to inhibit development of surface browning may relate to their ability to minimize the level of phenols activity on the cut surface, possibly in conjunction with a reduced PPO activity. Liu *et al.* (2016b) applied 10 µM methyl jasmonate (MJ) and 1 mM sodium nitroprusside, a NO donor, postharvest to cucumbers and found a significant reduction in chilling injury during storage at 5°C. MJ and sodium nitroprusside reduced H<sub>2</sub>O<sub>2</sub> in cucumber during storage, suggesting that both MJ and NO alleviated chilling injury by inhibiting H<sub>2</sub>O<sub>2</sub> generation. The lower H<sub>2</sub>O<sub>2</sub> induced by MJ and NO was coincident with enhanced catalase activity and gene expression, suggesting that catalase played an important role in control of H<sub>2</sub>O<sub>2</sub> accumulation.

NO acts as an important signal molecule with diverse physiological functions in plants, including promoting cell death during pathogen attacks. Hu *et al.* (2014b) suggested that the resistance of NO-treated mangoes to anthracnose may be attributed to activation of defence responses as well as delay of ripening. NO may also be linked to

cold tolerance. Dong *et al.* (2012) worked on the effects of a saccharide fraction prepared from yeast cell walls on chilling injury of cucumbers during storage at 4°C. They found that the yeast saccharide induced cold tolerance, which was linked with the induction of endogenous NO accumulation. Huque *et al.* (2013) also found that NO-treated apple slices had reduced ion leakage, which is associated with chilling injury.

There appear to be no reports of NO being included in MA packaging, but it seems to have potential and to be worthy of investigation; especially since it occurs naturally. The objective of using it in sachets would simply be to enhance levels already in the plant cells and therefore the effects that are already there.

## Humidity

Thompson *et al.* (1974d) reported that the effect of MA packaging in extending the postharvest life of plantains was at least partly due to the maintenance of a high humidity and not simply to a reduction in O<sub>2</sub> or an increase in CO<sub>2</sub>. The gas permeability of some plastics used for MA packaging is sensitive to environmental humidity. Yahia *et al.* (2005), extrapolating from work on *Opuntia*, conjectured that the humidity outside MA packages has a major effect on their gas exchange rates and hence on the levels of O<sub>2</sub> and CO<sub>2</sub> inside the package. Roberts (1990) showed that gas transmission of polyamides (nylons) can increase by about three times when the humidity is increased from 0% to 100% RH and with ethyl vinyl alcohol copolymers the increase can be as high as 100 times over the same range. Moisture given out by the produce can condense on the inside of the pack. This is especially a problem where there are large fluctuations in temperatures, because the humidity is high within the pack and easily reaches dew-point where the film surface is cooler than the pack air. Anti-fogging chemicals can be added during the manufacture of plastic films. These do not affect the quantity of moisture inside the packs, but

they cause the moisture that has condensed on the inside of the pack to form sheets rather than discrete drops. These can eventually form puddles at the bottom of the pack.

### Temperature

Temperature affects the respiration rate of the produce and the permeability of the film. Tano *et al.* (2007) studied temperature fluctuation between 3°C and 10°C over a 30-day period on packaged broccoli and found that it resulted in extensive browning, softening, weight loss increase, ethanol increase, and infection due to physiological damage and excessive condensation, compared with broccoli stored at constant temperature. At 3°C the atmosphere in the packages started at 3 kPa O<sub>2</sub> + 8 kPa CO<sub>2</sub> but when the temperature was raised to 10°C the CO<sub>2</sub> increased rapidly to a maximum of 15.5 kPa and O<sub>2</sub> decreased to < 1.5 kPa.

### Chilling injury

There is strong evidence in the literature that, under certain conditions, MA packaging can reduce or eliminate the symptoms of chilling injury. Where this effect has been shown includes: carambola (Ali *et al.*, 2004); citrus (Porat *et al.*, 2004); melons (Kang and Park, 2000; Flores *et al.*, 2004); okra (Finger *et al.*, 2008); and papaya (Singh and Rao, 2005). The reason for this effect and its mode of action have not been clearly determined, but Ali *et al.* (2004) concluded that suppression of the enzyme activities in fruit in MA packaging appeared to contribute to increased tolerance to chilling injury. Martinez-Javega *et al.* (1983) reported that the reduction of ethylene production and water loss in MA packaging were necessary in preventing chilling injury symptoms in melons.

### Shrink-wrapping

Besides reducing moisture losses and changing the O<sub>2</sub> and CO<sub>2</sub> levels, shrink-film



**Fig. 14.7.** Shrink-wrapped cucumber in a UK supermarket.

wrapping can also protect the fruit from some damage, e.g. by scuffing during handling and transport, and possibly from some fungal infections. In Europe, cucumbers are commonly marketed in a shrink-wrapped plastic sleeve to help retain their texture and colour (Fig. 14.7). Thompson (1981) found that arracacha roots deteriorated quickly after harvest but could be stored for 7 days if packed in shrink film. Samsouondar *et al.* (2000) found that breadfruits that had been shrink-wrapped with PE film (60 gauge) had an extended storage life. Lemons stored in PVC shrink-wrap retained their quality, with reduced loss of weight and rotting compared with those stored non-wrapped or waxed (Neri *et al.*, 2004).

### Vacuum Packing

Vacuum packaging uses a range of low- or non-permeable films (barrier films) or containers into which the fresh fruit or vegetable is placed and the air is sucked out. There are many different machines used for commercial application, including 'vertical form fill and seal', 'horizontal form and fill seal' and 'single chamber machines'. Banavac is a patented system that uses large PE film bags 0.04 mm thick in which typically 18.14 kg of green bananas are packed, then a vacuum is applied and the bags are sealed (see Fig. 13.6) (Badran, 1969). Nair and Tung (1988) reported that 'Pisang Mas'

bananas stored at 17°C had an extension in their postharvest life of 4–6 weeks when they were kept in evacuated collapsed PE bags by applying a vacuum not exceeding 300 mm Hg. Knee and Aggarwal (2000) tested plastic containers, capable of being evacuated to 50%, on storage of celery, lettuces, grapes, green beans, melons and strawberries at 4°C and broccoli, okras and tomatoes at 8°C. O<sub>2</sub> levels fell below 5% after storage for 3 days for all types of produce in the vacuum containers. There were many negative effects on the produce and overall the vacuum containers showed little advantage over conventional containers.

### Minimal Processing

Minimal processing is also called fresh cut and refers to fruit or vegetables that have been prepared and packaged to a form that can be directly eaten (Fig. 14.8). The damage inflicted in minimal processing of fruit and vegetables can reduce the postharvest life compared with those that remain intact, but MA packaging can be used to compensate for this to some degree. Bastrash *et al.* (1993) found that broccoli heads that had been cut into florets had an increased respiration rate in response to wounding stress throughout storage at 4°C in air and ethylene production was also stimulated after 10 days. They also found that the atmosphere for optimal preservation of broccoli florets was 6 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> at 4°C, which delayed yellowing, reduced development of

mould and offensive odours and resulted in better water retention during 7 weeks of storage. This compared with only 5 weeks in air. These effects were especially noticeable when the florets were returned from CA at 4°C to air at 20°C.

Budu *et al.* (2001) found that the respiration rate of peeled pineapples was very close to that of intact fruits, but their respiration rate decreased in response to slight changes in O<sub>2</sub> and CO<sub>2</sub> concentrations. They concluded that MA packaging should have important commercial applications in reducing the respiration rate and increasing the shelf-life of minimally processed pineapple. For minimally processed pineapple slices Budu *et al.* (2007) found that an MA packaging giving 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> appeared to be the most appropriate. Tancharoensukjit and Chantanawarangoon (2008) found that for fresh-cut pineapples storage in 5 or 10 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> helped to maintain their lightness and visual quality and prolonged their storage life to 9 days. Antonioli *et al.* (2007) sliced pineapples and dipped them in sodium hypochlorite (20 mg/l) for 30 s and then stored them at 5 ± 1°C in a flow-through system with different O<sub>2</sub> and CO<sub>2</sub> concentrations. None of the CA treatments seem to give an advantage, since the slices stored in air had little browning and were free of contamination that would affect the food safety at the end of the storage period. Antonioli *et al.* (2006) did not detect ethylene production during the initial period of 12 h after minimal processing and storage at 5°C. They also found that the initial respiration rate of pineapple slices and chunks was double that observed in the peeled fruits. The respiration rate of slices and chunks was very similar during 14 days of storage. For fresh-cut pineapple Chonhenchob *et al.* (2007) found that an atmosphere of 6 kPa O<sub>2</sub> + 14 kPa CO<sub>2</sub>, which was achieved at equilibrium in the headspace of MA packaging, increased the shelf-life by up to 7 days at 10°C.

Day (1996) also showed that the atmosphere inside the MA packaging can influence the rate of deterioration. In minimally processed fruit and vegetables, 70 and 80 kPa O<sub>2</sub> in MA packaging inhibited undesirable fermentation reactions, delayed browning



Fig. 14.8. Imported minimally processed mangoes in the UK.

caused by damage during processing and inhibited both aerobic and anaerobic microbial growth. Cut-surface browning and flesh softening were inhibited for fruits that had been stored in 100 or 1 kPa O<sub>2</sub> compared with those that had been stored in air. However, the slices from the 100 kPa O<sub>2</sub> and air contained a much lower content of fermentation products associated with off-flavours compared with the slices from apples from 1 kPa O<sub>2</sub>. Lu and Toivonen (2000) sealed 'Spartan' apple slices in 40 µm LDPE film bags having a moderate O<sub>2</sub> transmission rate of about 2.28 cc/m<sup>2</sup>/24 h at 23°C and they remained in good condition for up to 2 weeks at 1°C. Slices from apples that had been previously stored at 1°C in 1 kPa O<sub>2</sub> for up to 19 days developed more cut-surface browning and greater tissue solute leakage and enhanced accumulations of acetaldehyde, ethanol and ethyl acetate compared with slices cut from the fruit that had been stored in 100 kPa O<sub>2</sub> for up to 19 days.

Escalona *et al.* (2006b) concluded that 80 kPa O<sub>2</sub> must be used in MA packaging of fresh-cut butterhead lettuce in combination with 10–20 kPa CO<sub>2</sub> to reduce their respiration rate and avoid fermentation. Day (2003), working on high O<sub>2</sub> storage with iceberg lettuce packed in either 30 µm PVDC or 30 µm OPP and stored at 6–8°C for 7–10 days, found that the CO<sub>2</sub> content inside the PVDC film reached 30–40 kPa, which damaged the lettuce. Inside the OPP film the O<sub>2</sub> level did not exceed 25 kPa and the lettuce retained their good quality. In a study of minimally processed cabbages, Rinaldi *et al.* (2009) found that the CA for 10 days did not extend the storage life at 5°C or 10°C longer than for those stored in air. The CA they used in a flow-through system was combinations of 2–10 kPa O<sub>2</sub> + 3–10 kPa CO<sub>2</sub>.

### Equilibrium Modified Atmosphere Packaging

EMA packaging is a common term for the packaging technology used for minimally processed fruit and vegetables. The internal O<sub>2</sub> concentration of the packages can be predicted for the different steps of the simulated

distribution chain by applying an integrated mathematical model and using highly permeable packaging films. For example, Del-Valle *et al.* (2009) developed a model to determine the EMA packaging for mandarin segments because of their high respiration rate. Mistriotis *et al.* (2016) developed a model simulating the combined gas diffusion through micro-perforations and the permeable polylactic acid (PLA) membrane for EMA packaging systems. By using this model, innovative biodegradable bio-based EMA packaging systems were designed specifically for selected fruits and vegetables using laser micro-perforation and the relatively high water vapour permeability of PLA films compared with oriented polypropylene (OPP) films.

### Modelling

Day (1994) and Ben-Yehoshua *et al.* (1995) reviewed the concept of mathematical modelling of MA packaging for both whole and minimally processed fruit and vegetables. Evelo (1995) showed that the package volume did not affect the equilibrium gas concentrations inside an MA package but did affect the non-steady-state conditions. An MA packaging model was developed using a systems-oriented approach that allowed the selection of a respiration rate model according to the available data, with temperature dependence explicitly incorporated into the model. The model was suitable for assessing optimal MA packaging in realistic distribution chains. O<sub>2</sub> and CO<sub>2</sub> concentrations inside plastic film packages containing fruits of the mango cultivar 'Nam Dok Mai' were modelled by Boon-Long *et al.* (1994). Parameters for polyvinyl chloride (PVC), polyethylene (PE) and polypropylene (PP) films were included. A method of determining respiration rate from the time history of the gas concentrations inside the film package instead of from direct measurements was devised. It was claimed that satisfactory results were obtained in practical experiments carried out to verify the model.

Many other mathematical models have been published (e.g. Cameron *et al.*, 1989;



Lee *et al.*, 1991; Lopez-Briones *et al.*, 1993) which could help to predict the atmosphere around fresh produce sealed in plastic film bags. The latter suggested the following model:

$$\frac{1}{x} = \frac{\hat{A}m}{x_o S} \cdot \frac{1}{K} + \frac{1}{x_o}$$

where  $x$  = O<sub>2</sub> concentration in the pouches (%),  $x_o$  = initial O<sub>2</sub> concentration within the pouches (%),  $\hat{A}$  = proportionality between respiration rate and O<sub>2</sub> concentration including the effect of temperature (ml/g/day/atm),  $K$  = O<sub>2</sub> diffusion coefficient of the film (includes effects of temperature) (ml/m<sup>2</sup>/day/atm),  $S$  = surface area for gas exchange (m<sup>2</sup>), and  $m$  = weight of plant tissue (g).

Lee *et al.* (1991) obtained a set of differential equations representing the mathematical model for an MA system. In this case, the rate of reaction ( $r$ ) is equal to:

$$r = V_m C_1 / [K_M + C_1 (1 + C_2 / K_1)]$$

where  $C_1$  is the substrate concentration (which is O<sub>2</sub> concentration in the case of respiration rate),  $V_m$  and  $K_M$  are parameters of the classical Michaelis–Menten kinetics ( $V_m$  being the maximal rate of enzymatic reaction, and  $K_M$  the Michaelis constant),  $C_2$  is the inhibitor concentration (CO<sub>2</sub> concentration in the case of respiration) and  $K_1$  is the constant of equilibrium between the enzyme–substrate–inhibitor complex and free inhibitor.

Combining this equation with Fick's Law for O<sub>2</sub> and CO<sub>2</sub> permeation, Lee *et al.* (1991) estimated parameters of this model ( $V_m$ ,  $K_M$  and  $K_1$ ) from experimental data and then performed numerical calculation with the equations. CO<sub>2</sub> absorbents or a permeable window can be used to prevent atmospheres developing that could damage the crop. The concentration of the various respiratory gases, O<sub>2</sub>, CO<sub>2</sub> and ethylene, within the crop is governed by various factors, including the gas exchange equation that is a modification of Fick's law:

$$-\delta s / \delta t = (C_{in} - C_{out})DR$$

where  $-\delta s / \delta t$  = rate of gas transport out of the crop,  $C_{in}$  = the concentration of gas

within the crop,  $C_{out}$  = the concentration of gas outside the crop,  $D$  = gaseous diffusion coefficient in air, and  $R$  = a constant specific to that particular crop.

The strawberry cultivars 'Pajaro' and 'Selva' were used to test a model describing gas transport through micro-perforated PP films and fruit respiration involved in MA packaging by Renault *et al.* (1994). They conducted some experiments with empty packs initially filled with either 100 kPa N<sub>2</sub> or 100 kPa O<sub>2</sub>. Simulations agreed very well with experiments only if the cross-sectional area of the micro-perforations was replaced by areas of approximately half the actual areas in order to account for the resistance of air around the perforations. It was also possible to fit the model to gas concentration changes in packs filled with strawberries, although deviations were encountered due to contamination of strawberries by fungi. The model was used to quantify the consequences of the variability of pack properties of the number of micro-perforations per pack and cross-sectional area of these perforations on equilibrium gas concentrations and to define minimum homogeneity requirements for MA packaging. Del Nobile *et al.* (2007) used a simple mathematical model for designing plastic films for minimally processed prickly pear, banana and kiwi-fruit with a laminated PE/aluminium/PET film or a coextruded PO film. Despite the simplicity of the model, they reported that it satisfactorily described and predicted the respiration rate of the fruit during storage at 5°C.

Waghmare *et al.* (2013) confirmed that mathematical modelling can be used to predict respiration rate as a function of time and temperature in MA packaging and tested their model on MA packaging of fresh-cut coriander, cluster beans and beet-root during storage at 10°C, 20°C or 30°C. They concluded that the dependence of respiration rate on temperature and time was well described by Arrhenius and first-order decay models. The effect of impact stress on the expression of genes related to respiratory by-products and ethylene synthesis of the damaged cabbage was studied by Guillard *et al.* (2012). They used a mathematical

model based on interval analysis rather than the probabilistic approach used to predict O<sub>2</sub> and CO<sub>2</sub> evolution in headspaces of MA packaging. They found that their approach needed a minimal amount of unverified assumption concerning uncertainties and also required only a few evaluations of the model. They successfully tested their approach on fresh chicory, mushrooms and blueberries in MA packaging.

### Safety

This subject has been dealt with in detail in a review by Church and Parsons (1995). In a survey of 21 ready-to-eat fruits, 28 whole fresh vegetables, 15 sprout samples and 237 minimally processed salads in Spain, Usall *et al.* (2008) showed that the fresh-cut vegetables had the highest microorganism counts. These were associated with grated carrot, arugula and spinach, and the microorganisms were aerobic mesophilic microorganisms, yeast, moulds, lactic acid bacteria and Enterobacteriaceae. The lowest counts of these microorganisms were generally in fresh-cut endive and lettuce. Counts of psychrotrophic microorganisms were as high as those of mesophilic microorganisms. Microbiological counts for fresh-cut fruit were very low. Sprouts were highly contaminated with mesophilic psychrotrophic microorganisms and Enterobacteriaceae and showed a high incidence of *E. coli* (40% of the samples). Of the samples analysed, four were *Salmonella* positive and two harboured *L. monocytogenes*. None of the samples was positive for *E. coli* O157:H7, pathogenic *Yersinia enterocolitica* or thermo-tolerant *Campylobacter*. There is considerable legislation related to food sold for human consumption. In the UK, for example, the Food Safety Act 1990 states that 'it is an offence to sell or supply food for human consumption if it does not meet food safety requirements'. The use of MA packaging has health and safety implications. One factor that should be taken into account is that the gases in the atmosphere could possibly have a stimulating effect on microorganisms. Farber (1991) stated that

'while MA packaged foods have become increasingly more common in North America, research on the microbiological safety of these foods was still lacking'. The growth of aerobic microorganisms is generally optimum at about 21 kPa O<sub>2</sub> and falls off sharply with reduced O<sub>2</sub> levels, while with anaerobic microorganisms their optimum growth is generally at 0 kPa O<sub>2</sub> and falls as the O<sub>2</sub> level increases (Day, 1996). With many modified atmospheres containing increased levels of CO<sub>2</sub>, the aerobic spoilage organisms, which usually warn consumers of spoilage, are inhibited, while the growth of pathogens may be allowed or even stimulated, which raises safety issues (Farber, 1991). Hotchkiss and Banco (1992) stated that extending the shelf-life of refrigerated foods might increase microbial risks in at least three ways in MA packaged produce:

- Increasing the time in which food remains edible increases the time in which even slow-growing pathogens can develop or produce toxin.
- Retarding the development of competing spoilage organisms.
- Packaging of respiring produce could alter the atmosphere so that pathogen growth is stimulated.

The ability of *E. coli* O157:H7 to grow on raw salad vegetables that have been subjected to processing and storage conditions simulating those routinely used in commercial practice has been demonstrated by Abdul Raouf *et al.* (1993). The influence of MA packaging, storage temperature and time on the survival and growth of *E. coli* O157:H7 inoculated on to shredded lettuce, sliced cucumber and shredded carrot was determined. Packaging in an atmosphere containing 3 kPa O<sub>2</sub> + 97 kPa N<sub>2</sub> had no apparent effect on populations of *E. coli* O157:H7, psychrotrophs or mesophiles. Populations of viable *E. coli* O157:H7 declined on vegetables stored for up to 14 days at 5°C and increased on vegetables stored at 12°C and 21°C. The most rapid increases in populations of *E. coli* O157:H7 occurred on lettuce and cucumbers stored at 21°C. These results suggest that an unknown factor or factors associated with carrots may inhibit the growth of

*E. coli* O157:H7. The reduction in acidity of vegetables was correlated with initial increases in populations of *E. coli* O157:H7 and naturally occurring microflora. Eventual decreases in *E. coli* O157:H7 in samples stored at 21°C were attributed to the toxic effect of accumulated acids. Changes in visual appearance of vegetables were not influenced substantially by growth of *E. coli* O157:H7. 'Golden Delicious', 'Red Delicious', 'McIntosh', 'Macoun' and 'Melrose' apples inoculated with *E. coli* O157:H7 promoted growth of the bacterium in bruised tissue independent of their stage of ripeness. Freshly picked (< 2 days after harvest) 'McIntosh' apples usually had no growth of *E. coli* O157:H7 for 2 days but growth occurred following 6 days of incubation in bruised McIntosh apple tissue. When apples were stored for 1 month at 4°C prior to inoculation with *E. coli* O157:H7, all the five cultivars supported its growth (Dingman, 2000).

Church and Parsons (1995) mentioned that there was a theoretical potential fatal toxigenesis through infections by *Clostridium botulinum* in the depleted O<sub>2</sub> atmospheres in MA packed fresh vegetables. It was claimed that this toxigenesis had not been demonstrated in vegetable products without some sensory indication (Zagory and Kader, 1988, quoted by Church and Parsons, 1995). Roy *et al.* (1995) showed that the optimum in-package O<sub>2</sub> concentration for suppressing cap opening of fresh mushrooms was 6 kPa and that lower O<sub>2</sub> concentrations in storage were not recommended because they could promote growth and toxin production by *C. botulinum*. In a review of hazards related to MA packaging of food, Betts (1996) indicated that vacuum packing of shredded lettuce had been implicated in a botulinum poisoning outbreak.

Jacxsens *et al.* (2002) reported that spoilage microorganisms can proliferate quickly on minimally processed bell peppers and lettuce, with yeasts being the main shelf-life limiting group. They reported that *Listeria monocytogenes* was found to multiply on cucumber slices, survived on minimally processed lettuce and decreased in number on bell peppers due to the combination of acidity and refrigeration. *Aeromonas caviae*

multiplied on both cucumber slices and mixed lettuce but was also inhibited by the acidity of bell peppers. Storage temperature control was found to be of paramount importance in controlling microbial spoilage and safety of equilibrium MA packaged minimally processed vegetables.

Sziro *et al.* (2006) studied microbial safety of strawberry and raspberry fruits after storage in two permeable packaging systems, in combination with an ethylene-absorbing film, calculated to reach 3 kPa O<sub>2</sub> at equilibrium. The first was called high oxygen atmosphere (HOA) where the atmosphere started at 95 kPa O<sub>2</sub> in 5 kPa N<sub>2</sub> and the second was EMA starting from 3 kPa O<sub>2</sub> in 95 kPa N<sub>2</sub>. *E. coli*, *Salmonella* spp. and *L. monocytogenes* were artificially inoculated on to packaged fruits and all three were able to survive in storage at 7°C. Raspberries showed an enhanced inactivation of *Salmonella* during storage in both types of packaging. Growth of *L. monocytogenes* was observed on the calyx of strawberries after 3 days. Generally, increasing the shelf-life of the fruits with EMA and HOA did not give an increased microbial risk.

Another safety issue is the possibility of the films being used in MA packaging being toxic. Schlimme and Rooney (1994) reviewed the possibility of constituents of the polymeric film used in MA packaging migrating to the food that they contain. They showed that it is unlikely to happen, because of their high molecular weight and insolubility in water. All films used can contain some non-polymerized constituents which could be transferred to the food and in the USA the Food and Drugs Administration and also the European Community have regulations related to these 'indirect additives'. The film manufacturer must therefore establish the toxicity and extraction behaviour of the constituents with specified food simulants.

There are other possible dangers in using high concentrations of CO<sub>2</sub> and O<sub>2</sub> in storage. Low O<sub>2</sub> and high CO<sub>2</sub> can have a direct lethal effect on human beings working in those atmospheres. Great care needs to be taken when using MA packaging containing 70 or 80 kPa O<sub>2</sub>, or even higher levels, because of potential explosions.

## Some MA Packaging Recommendations

### Apple

In refrigerated conditions MA packaging was shown to extend the storage life of apples longer than those stored in air but shorter than those stored in CA. For example, in Germany, Hansen (1975) showed that the cultivar 'Jonagold' kept in cold stores at 0.5°C, 2.5°C or 3.5°C retained satisfactory organoleptic properties only until the end of January. However, when they were stored at 3.5°C in PE bags (with an equilibrium CO<sub>2</sub> content of 7 kPa) they retained their quality until April, but in CA storage (6 kPa CO<sub>2</sub> + 3 kPa O<sub>2</sub>) at 2.5°C fruit quality remained satisfactory to the end of May. Under all conditions apples showed no rotting, CO<sub>2</sub> damage or physiological disorders. Hewett and Thompson (1988) also found that storage of the cultivar 'Golden Delicious' at 1°C for 49 days, then 7 days at ambient in PE bags 25 µm thick with 50–70 micro-perforations, made with a 1 mm diameter cold needle, lost less weight, were firmer and had better eating quality than fruit not stored in bags.

Generally, the acceptable time for which apples can remain in MA packages in ambient conditions varies with cultivar, but it is usually 2–4 weeks. An example of this was given by Geeson and Smith (1989), who showed that different apple cultivars required different film types and thicknesses for optimum effect. For 'Bramley's Seedling', 'Egremont Russet' and 'Spartan', they found that after CA storage, apples in 1 kg 30 µm LDPE pillow packs in ambient conditions showed slower flesh softening and skin yellowing and a longer shelf-life than those with no packaging. For 'Cox's Orange Pippin', 20 µm ethylvinyl acetate film was effective in retarding ripening without adversely affecting eating quality. Park *et al.* (2007) found that packaging in 25 µm MA film (film not specified) of the cultivar 'Fuji' for 24 weeks at 0°C was more effective than other film packaging treatments. TA and TSS were higher in MA packaging than control and decay was reduced from 6.8% in control to 2.4% in MA packaging.

### Apricot

Ayhan *et al.* (2009) found that the cultivar 'Kabaasi' stored at 4°C in air could be kept for 7 days without loss of physical and sensory properties, but when they were packed in either bi-oriented polypropylene (BOPP) or cast polypropylene (CPP) films they could be successfully stored for 28 days. Ali Koyuncu *et al.* (2009) stored the cultivar 'Aprikoz' at 0°C and 90 ± 5% RH in 12, 16 or 20 µm stretch film or CA and found that those stored in CA retained their quality better than those in stretch film.

### Arracacia

Scalon *et al.* (2002) found that *Arracacia xanthorrhiza* roots stored without packaging were unfit for marketing after 21 days. However, in ambient conditions in Bogotá (17–20°C and 68–70% RH), Thompson (1981) found that roots deteriorated much more quickly but could be stored for 7 days if wrapped in plastic cling film or plastic shrink film. Scalon *et al.* (2002) also found that roots packaged in PVC film had lower weight losses than those in cellophane film, but those in cellophane film had a better marketing appearance. Ramos *et al.* (2015) found that roots wrapped in PVC film and stored at 5°C or 10°C were commercially viable up to 60 days, while those stored non-wrapped at 5°C had a shelf-life of only 7 days, and at least 25% had chilling injury symptoms.

### Artichoke

Mencarelli (1987a) sealed globe artichokes (immature flower heads) in films of differing permeability or perforated films for 35 days. Perforated heat-sealed films gave poor results, as did low-permeability films, but films sufficiently permeable to allow good gas exchange while maintaining high humidity gave good results, with high turgidity and good flavour and appearance. Storage in crates lined with perforated PE film was

recommended by Lutz and Hardenburg (1968). Five storage trials carried out by Andre *et al.* (1980b) on the cultivar 'Violet de Provence' showed that the storage life in air was 1 week at room temperature and 3–4 weeks at 1°C. This was extended to 2 months by a combination of vacuum cooling, packing in PE bags and storage at 1°C. Gil-Izquierdo *et al.* (2004) found that the cultivar 'Blanca de Tudela' stored at 5°C had an equilibrium atmosphere of 14.4 kPa O<sub>2</sub> + 5.2 kPa CO<sub>2</sub> for those sealed in PVC bags and 7.7 kPa O<sub>2</sub> + 9.8 kPa CO<sub>2</sub> for those in LDPE bags. They showed that loss of water was the major cause of deterioration after 8 days of storage. The vitamin C content of the internal bracts (the edible part) decreased in MA storage, while for those stored without packaging it remained at levels similar to those at harvest. There was also a high phenolic content (896 mg/100 g fresh weight) in the internal bracts and this increased in those in LDPE and PVC but not in those stored without packaging.

### Asparagus

Storage at 2°C in MA packaging retained sensory quality and AA levels (Villanueva *et al.*, 2005). In several trials cooling to 1°C within 6–8 h of harvest, followed by storage at 1°C in PE bags with silicon elastomer windows kept the spears in good condition for 20–35 days (Andre *et al.*, 1980a). Lill and Corrigan (1996) experimented with different MA packs and found that they all significantly extended the shelf-life of the spears by between 83% and 178%, depending on film type, compared with those not wrapped. The least permeable films (W.R. Grace RD 106 and Van Leer Packaging Ltd) gave the longer shelf-life extension; the former film had 2.5–3.5 kPa CO<sub>2</sub> + 4.1–9.7 kPa O<sub>2</sub> and the latter 3.6–4.6 kPa CO<sub>2</sub> + 2.0–4.5 kPa O<sub>2</sub> equilibrium gas content at 20°C.

### Aubergine

MA packaging prolonged the shelf-life and flesh firmness compared with those stored non-wrapped (Arvanitoyannis *et al.*, 2005).

### Avocado

Storage life was extended by 3–8 days at various temperatures by sealing individual fruits in PE film bags (Haard and Salunkhe, 1975). In earlier work it had been shown that fruit of the cultivar 'Fuerte' sealed individually in PE film bags 0.025 mm thick for 23 days at 14–17°C ripened normally on subsequent removal from the bags. The atmosphere inside the bags after 23 days was 8 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> (Aharoni *et al.*, 1968). Thompson *et al.* (1971) showed that sealing various seedling varieties of West Indian avocados in PE film bags greatly reduced fruit softening during storage at various temperatures (Table 14.4).

### Banana

There is strong evidence that MA packaging can extend the pre-climacteric life of banana, which has been shown in several varieties and at different temperatures. Packaging bananas in PE bags alone, gas flushed with 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> or with a partial vacuum (400 mm Hg) resulted in shelf-life extension to 15, 24 and 32 days, respectively, compared with 12 days for those stored non-wrapped at 13 ± 1°C (Chauhan *et al.*, 2006). According to Shorter *et al.* (1987) the storage life was increased five times when bananas were stored in plastic film (where the equilibrium gas content was about 2 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>) with an ethylene scrubber, compared with fruit stored without wraps. Satyan *et al.* (1992) stored 'Williams' in PE tubes 0.1 mm thick at 13°C, 20°C and 28°C and found that the storage life was increased 2–3 times compared with fruit stored without

**Table 14.4.** Effects of storage temperature and packing in 125 gauge PE film bags compared with non-packed on the number of days to softening of avocados (modified from Thompson *et al.*, 1971).

	7°C	13°C	27°C
Not wrapped	32	19	8
Sealed film bags	38	27	11

wraps. Tiangco *et al.* (1987) also observed that green life of the variety 'Saba' (*Musa* BBB) held in MA at ambient temperature had a considerable extension of storage life compared with fruit stored without wraps, especially if combined with refrigeration. In Thailand, Tongdee (1988) found that the green life of 'Kluai Khai' (*Musa* AA) could be maintained for more than 45 days in PE bags at 13°C. This was longer by more than 20 days than fruit stored at 25°C in PE bags. In the Philippines, 'Latundan' (*Musa* AAB) could be stored in PE bags 0.08 mm thick at ambient temperature of 26–30°C for up to 13 days. It was also reported that 'Lakatan' (*Musa* AA) had a lag of 3 days before the beginning of colour change upon removal from the bags (Abdullah and Tirto-soekotjo, 1989; Abdullah and Pantastico, 1990). Marchal and Nolin (1990) found that storage in PE film reduced weight loss and respiration rate, which allowed the bananas to be stored for several weeks. Chamara *et al.* (2000) showed that packaging in LDPE bags 75 µm thick extended the green life of bananas up to 20 days at room temperature of about 25°C and 85% RH. Fruit ripened naturally within 4 days when removed from the packaging. Ali Azizan (1988, quoted by Abdullah and Pantastico, 1990) observed that the TSS and TA in 'Pisang Mas' changed more slowly during storage in MA at ambient temperature than those stored not wrapped.

Bowden (1993) showed that the effect on yellowing of bananas that had been initiated to ripen was related to the thickness of plastic and therefore the gas composition within the film. Film thickness also affected weight loss. For example, at 13°C Wei and Thompson (1993) showed that weight loss of 'Apple' bananas after 4 weeks storage in PE film was 1.5% in 200 gauge, 1.8% in 150 gauge and 2.1% in 100 gauge, while fruit stored without packaging was 12.2 %.

Potassium permanganate-soaked paper in sealed PE bags was the best packaging treatment for 'Harichhaal' (*Musa* AAA). Bavistin + PE bags (with 2% of the surface with perforations) were the most effective treatment in minimizing the microbial spoilage of fruit during storage (Kumar and

Brahmachari, 2005). Clay bricks impregnated with potassium permanganate significantly ( $p = 0.05$ ) lowered in-package ethylene and CO<sub>2</sub>, retained higher O<sub>2</sub> content and resulted in minimum changes in firmness and TSS content (Chamara *et al.*, 2000). Ranasinghe *et al.* (2005) reported that potassium permanganate in combination with silica gel, as a desiccant, and soda-lime, as a CO<sub>2</sub> scrubber, further increased the storage life. They also recommended treatment with emulsions of cinnamon oils combined with MA packaging in 75 µm LDPE as a safe, cost-effective method for extending the storage life of 'Embul' (*Musa* AAB) up to 21 days at 14 ± 1°C + 90% RH and 14 days at 28 ± 2°C without affecting the organoleptic and physico-chemical properties.

MA can also have negative effects. Wills (1990) mentioned that an unsuitable selection in packaging materials can still accelerate the ripening of bananas or enhance CO<sub>2</sub> injury when the ethylene accumulated over a certain period. Wei and Thompson (1993) found that on packaging the cultivar 'Apple' in PE film and storage at 13–14°C symptoms of CO<sub>2</sub> injury were observed when the CO<sub>2</sub> levels were between about 5% and 14%. These levels occurred with some fruit in 150 gauge PE bags and all the fruit in 200 gauge PE bags, but only after 3 weeks, and not at all in 100 gauge bags. Injury was characterized by darkening of the skin and softening of the outer pulp, while the inner pulp (core) remained hard and astringent. Often there would be a distinct irregular ring of dark brown tissue in the outer cross-section of the pulp. In some of the fingers the pulp developed a tough texture.

MA packaging of bananas in PE film bags is commonly used in international transport. With the 'Banovac' system patented by the United Fruit Company (see 'Vacuum packing', above), typical gas contents that develop during transport at 13–14°C in the bags through fruit respiration were about 5 kPa CO<sub>2</sub> + 2 kPa O<sub>2</sub> (De Ruiter, 1991). Besides retaining a desirable high humidity around the fruit, it has been shown that packing in non-perforated bags prolonged the pre-climacteric life of 'Mas' (*Musa* AA)

(Tan and Mohamed, 1990). Plastic bags are used commercially as liners for whole boxes and are mainly intact but some have macro-perforations. Those that were intact were often torn before being loaded into the ripening rooms, in order to ensure that ethylene penetrated to initiate ripening. With the use of forced-air ventilation in modern 'pressure' ripening rooms, this practice was found not to be necessary. Clusters of fruit are also packed individually in small PE bags at source, transported, ripened and sold in the same bags.

In Australia, bananas were transported internally packed in 13.6 kg commercial packs inside PE film bags. Scott *et al.* (1971) reported that in simulated transport experiments the fruit were kept in good condition at ambient temperatures during the 48 h required for the journey. Siriwardana *et al.* (2017) treated 'Embul' bananas (*Musa AAB*) with 1% aluminium sulfate, 1% aluminium sulfate + 0.4% basil oil or distilled water (control), then sealed them in LDPE bags and stored them at 12–14°C. After 14 days the bananas in all three treatments were in good condition and the atmosphere inside the bags was 3.1–3.7 kPa O<sub>2</sub> + 4.2–4.7 kPa CO<sub>2</sub>. Aluminium sulfate + basil oil significantly controlled crown rot disease compared with the other two treatments and had no negative effects on consumer acceptability.

### Basil

Patiño *et al.* (2018) stored basil leaves at 11°C and 85% RH in sealed LDPE bags and in open macro-perforated LDPE bags with and without previous exposition to 1-MCP at 0.3 cm<sup>3</sup>/m<sup>3</sup> for 24 h, which they considered the optimum exposure. The MAP system they used gave a steady-state concentration inside the packaging of about 10.5 kPa O<sub>2</sub> + 4.2 kPa CO<sub>2</sub>. The combined treatment of MAP and 1-MCP increased the storage life of the basil leaves up to 18 days, compared with 9 days for those stored in open bags without 1-MCP exposure, all at 11°C and 85% RH.

### Beans

The quality of 15 cultivars of French beans stored in LDPE at 8–10°C was acceptable for up to 30 days, after which the colour changed to light brown in some of the cultivars. At that time the cultivar 'CH-913' had the highest percentage of beans (92.55%) that were still considered marketable (Attri and Swaroop, 2005).

### Beet

Sugarbeet packed in PVC film maintained good appearance during 22 days storage at an ambient temperature of 15–26°C with a weight loss of 15.8% compared with 55% in the non-wrapped beets (Scalon *et al.*, 2000).

### Black sapote

Black sapote (*Diospyros ebenaster*, *D. digyna*) is a climacteric fruit and the main factor limiting storage was reported to be softening. Fruits wrapped in plastic film had 40–50% longer storage life at 20°C than those not wrapped (Nerd and Mizrahi, 1993).

### Blueberry

Hong *et al.* (2006) reported that packaging blueberries in 0.88 mm (*sic*) PE film significantly reduced the rates of weight loss, rotting and respiration rate during storage at 2°C. It also slowed the reduction in anthocyanin and inhibited enzyme activity compared with those stored non-wrapped. Tectrol MA pallet bags are used in reefer containers and were reported to have proven success in the storage of fresh blueberries for up to 65 days (DeEll, 2002).

### Breadfruit

Storage of breadfruit at 12.5°C sealed in 150 gauge PE bags kept them in good condition

for up to 13 days compared with 80% of the non-wrapped fruit being unmarketable after only 2 days (Thompson *et al.*, 1974a). At 28°C Maharaj and Sankat (1990) found that fruits could be stored for 5 days in sealed 100 gauge PE film bags, while at 12°C or 16°C in sealed 100 gauge PE film bags they could be stored for 14 days. Worrel and Carington (1994) also showed that storing breadfruit sealed in plastic film maintained their green colour and fruit quality. They used 40  $\mu\text{m}$  HDPE film or LDPE film and showed that fruit quality could be maintained for 2 weeks at 13°C. Breadfruits shrink-wrapped in 60 gauge PE film and stored at 16°C had reduced external skin browning and higher chlorophyll levels after 10 days of storage. This treatment also delayed fruit ripening as evidenced by changes in texture, TSS and starch content (Samsouandar *et al.*, 2000).

### Broccoli

Serrano *et al.* (2006) reported that non-wrapped broccoli lost weight and turned yellow and the stems hardened; also there was a rapid decrease in total antioxidant activity, AA and total phenolics, giving a maximum storage period of only 5 days. These changes were delayed in MA packaging, especially in micro-perforated PP film and non-perforated PP film, with total antioxidant activity, AA and total phenolic compounds remaining almost unchanged during the whole storage period of 28 days. Broccoli stored at 5°C in 14 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> or wrapped in PVC film for 3 weeks retained market quality significantly better than when left non-wrapped. Respiration rates of those in CA or in PVC film were reduced by 30–40% compared with the controls (Forney *et al.*, 1989). Cameron (2003) reported that low O<sub>2</sub> could retard yellowing at 5°C but could also produce highly undesirable flavours and odours when O<sub>2</sub> fell below 0.25 kPa. Rai *et al.* (2008) studied various MA packages in storage at 5°C and 75% RH and found that perforated PP film (2 holes, each of 0.3 mm diameter, with a

film area of 0.1 m<sup>2</sup>) could be used to retain chlorophyll and AA levels for 4 days. DeEll *et al.* (2006) reported that, overall, the use of sorbitol, a water absorbent, and sachets of potassium permanganate inside PD-961EZ bags at 0–1°C for 29 days reduced the amount of volatiles that are responsible for off-odours and off-flavours. This was reported to maintain their quality and marketability for longer. Acetaldehyde concentrations were higher in the bags with no sorbitol or potassium permanganate after 29 days, while ethanol was greater in both control bags and those with only potassium permanganate. After storage at 4°C or 10°C in LDPE film packages containing potassium permanganate with an equilibrium gas content of 5 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub>, Jacobsson *et al.* (2004b) found that sensory properties were better than those in other films and similar to freshly harvested broccoli. The optimum equilibrium atmosphere was found to be 1–2 kPa O<sub>2</sub> and 5–10 kPa CO<sub>2</sub>. Jacobsson *et al.* (2004a) measured the gas inside various packages and found that storage in OPP resulted in the highest CO<sub>2</sub> concentration of 6 kPa, while the lowest O<sub>2</sub> concentration of 9 kPa was found in the LDPE package. They also found that broccoli stored in PVC film deteriorated faster than when packaged in the other films and the influence of all the plastic packaging tested was greater at 10°C than at 4°C. Among the packages tested by Cefola *et al.* (2016), the combination of PP/PA in 5 kPa O<sub>2</sub> in 95 kPa N<sub>2</sub> containing CO<sub>2</sub>-absorbing sachets reached an equilibrium condition very close to optimum for broccoli (5 kPa O<sub>2</sub> and  $\leq$  5 kPa CO<sub>2</sub>). Fresh-cut broccoli stored under these conditions for 8 days showed negligible reduction of appearance and odour and were comparable to the freshly harvested samples.

### Brussels sprouts

Storage in PVC film at 0°C for 42 days resulted in reduced browning of cut areas and reduced losses in weight and firmness compared with sprouts that were not wrapped. AA and total flavonoid contents remained



almost constant in the PVC-wrapped sprouts, while radical scavenging activity increased (Vina *et al.*, 2007).

### Cabbage

Peters *et al.* (1986) found that the best plastic covering for the cultivar 'Nagaoka King' stored in wooden boxes was PE sheets (160 cm × 60 cm × 0.02 mm) with 0.4% perforations, wrapped around the top and long sides of the boxes, with the slatted ends of the boxes left open. Omary *et al.* (1993) treated shredded white cabbage with citric acid and sodium erythorbate then inoculated it with *Listeria innocua* and packaged it in lots of 230 g. Four types of plastic film bags were used for packaging, with O<sub>2</sub> transmission rates of 5.6, 1500, 4000 and 6000 cc/m<sup>2</sup>/day and they were stored at 11°C. After 21 days, *L. innocua* populations increased in all packages but the increase was significantly less for cabbage packaged in film with the highest permeability.

### Capsicum

Pala *et al.* (1994) studied storage in sealed LDPE film of various thicknesses at 8°C and 88–92% RH and found that non-wrapped peppers had a shelf-life of 10 days compared with 29 days for those sealed in 70 µm LDPE film (Table 14.5). Also this method of packaging gave good retention of sensory characteristics.

Nyanjage *et al.* (2005) showed that temperature was the major factor in determining the postharvest performance of the cultivar 'California Wonder'. For packaging in open trays, non-perforated or perforated PE bags did not significantly affect colour retention during storage at 4°C, 6.5°C or 17°C, but there was a significantly higher incidence of disease. Perforated PE packaging produced the best overall results.

In other work, the marketable value of peppers stored for 4 weeks at 8°C was highest for those in PE bags with 0.1% perforations and in non-perforated PE bags compared with non-packaged (Kosson and Stepowska, 2005). After 2 weeks of storage in non-perforated PE bags the atmosphere was about 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> and with 0.0001% perforation it was about 18 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. Water condensation on the inner surface of PE bags and on the stored peppers occurred in bags with 0.0001%, 0.001%, 0.01% and 0% perforations but not in those with 0.1% perforations. Banaras *et al.* (2005) found that there were no significant differences in AA and fruit firmness between the different perforated MA packaging of the cultivars 'Keystone', 'NuMex R Naky' and 'Santa Fe Grande'. However, perforated MA packaging was shown to reduce water loss, maintain turgidity of fruits and delay red colour and disease development at both 8°C and 20°C. MA packaging extended postharvest life for another 7 days at 8°C and for 10 days at 20°C compared with non-packaged fruit held at these temperatures. Postharvest water loss and turgidity were similar for fruit stored in packages

**Table 14.5.** Sensory score (1–9) of green capsicums during storage at 8°C and 88–92% RH, where 1–3.9 = non-acceptable, 4–6.9 = acceptable, 7–9 = good; modified from Pala *et al.* (1994).

	Storage time	Sensory score			
		Appearance	Colour	Texture	Flavour
Before storage	0 days	9	9	9	9
Non-wrapped control	15 days	4.3	4.6	2.7	4.2
20 µm LDPE	22 days	5.7	5.5	5.7	5.3
30 µm LDPE	20 days	5.8	6.0	6.0	4.5
50 µm LDPE	20 days	6.0	5.0	6.0	5.0
70 µm LDPE	29 days	7.8	7.8	7.6	7.8
100 µm LDPE	27 days	5.5	5.6	5.7	5.5

with and without 26 holes at 8°C and 20°C. Amjad *et al.* (2009) stored the cultivars 'Wonder King' and 'P-6' in various thicknesses of PE bags at 7°C, 14°C and 21°C. They found that the optimum thickness for 'Wonder King' was 21 µm at 7°C for minimizing weight loss, while 'P-6' fruit stored in 15 µm had the lowest weight loss and their maximum storage life was 20 days. AA levels and total phenolics and carotenoids also varied between cultivars, temperature and thickness of PE. Hughes *et al.* (1981) showed that capsicums sealed in various plastic films had a higher percentage of marketable fruit than those stored in air (see Table 11.2).

### Carambola

Ali *et al.* (2004) harvested the cultivar 'B10' at the mature-green stage. They reported that MA packaging in LDPE film suppressed the incidence of chilling injury in storage at 10°C and also retarded softening and the development of fruit colour. Neves *et al.* (2004) stored the cultivar 'Golden Star' in various thicknesses of LDPE bags at 12 ± 0.5°C and 95 ± 3% RH for 45 days, then in ambient conditions of 22 ± 3°C and 72 ± 5% RH for a further 5 days. They found that fruits in LDPE bags 10µm thick best retained their firmness and TA and had the 'best colour standard', the lowest decay incidence and the highest acceptance scores by the taste panellists, but the lowest TSS.

### Carrot

MA packaging is used commercially to extend the marketable life of immature carrots. Seljasen *et al.* (2003) found that carrots being gently handled and stored in perforated plastic bags at low temperature retained the most favourable taste in the Norwegian distribution chain. MA packaging did not result in an increase in off-flavours or an increase in 6-methoxymellein, which has been associated with off-flavours, or in a reduction in sugars. Storage in PE bags at 0°C

was reported by Kumar *et al.* (1999) to slow the decrease in AA and reduce the levels of TSS and reduce the weight loss to 1% over an 8-day period compared with carrots stored in gunny bags in ambient conditions, which lost 16%.

### Cassava

Thompson and Arango (1977) found that dipping the roots in a fungicide and packing them in PE film bags directly after harvest reduced the physiological disorder vascular streaking during storage for 8 days at 22–24°C compared with roots stored non-wrapped. Oudit (1976) showed that freshly harvested roots could be kept in good condition for up to 4 weeks when stored in PE bags. Coating with paraffin wax is commonly applied to roots exported from Costa Rica. Young *et al.* (1971) and Thompson and Arango (1977) found that this treatment kept them in good condition for 1–2 months at room temperature in Bogotá in Colombia.

### Cauliflower

Menjura Camacho and Villamizar (2004) studied ways of reducing wastage of cauliflowers in the central markets of Bogotá and found that the best treatment was packing them in LDPE with 0.17% perforations. Menniti and Casalini (2000) found that PVC film wrapped around each cauliflower reduced weight loss and retarded yellowing during storage for 5 days at 20°C. Curd colour, odour and compactness of the curd remained acceptable for up to 14 days at 6°C in non-perforated PE bags (Rahman *et al.*, 2008). The highest AA content (41 mg/100 g and 23 mg/100 g) and β-carotene content (26 IU/100 g and 14 IU/100 g) were in non-perforated PE bags after 7 days and 14 days of storage, respectively, at 6°C. Kaynaş *et al.* (1994) stored the cultivar 'Iglo' in PE film bags 30 µm thick, PE film bags with one 5 mm diameter hole per kilogram head, 15 µm PVC film bags or not wrapped, at 1°C and

90–95% RH for various times. They found that they could be stored for a maximum of 6 weeks in PVC film bags with a shelf-life of 3 days at 20°C, which was double the storage life of those stored non-wrapped. The permeability of the PVC film used in the studies was given as 200 g of water vapour/m<sup>2</sup>/day and 12,000 ml O<sub>2</sub>/m<sup>2</sup>/day/atm.

### Celery

The equilibrium atmospheres within OPP and LDPE bags containing celery were reached after 10 days and were 8–9 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> and 8 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>, respectively (Gómez and Artes, 2004b). The celery stored in OPP was rated to have an appearance most similar to that at harvest. Escalona *et al.* (2005) described an MA packaging for pallet loads by using a silicone membrane system for shipping celery to distant markets. The design of a system for 450 kg of celery per pallet was studied using an impermeable film with silicone membrane windows at 0.2 and 0.3 dm<sup>2</sup>/kg, to reach an atmosphere of approximately 3–4 kPa O<sub>2</sub> + 7–8 kPa CO<sub>2</sub>. Two kinds of silicone membrane windows were chosen, with permeability ranges of 4500–5250 ml O<sub>2</sub>/dm<sup>2</sup>/day and 10,125–14,000 ml CO<sub>2</sub>/dm<sup>2</sup>/day. The storage conditions for a trial were 21 days at 5°C and 95% RH followed by a shelf-life in air of 2 days at 15°C and 70% RH. As a control, partially unsealed packages were used. MA packaging resulted in slightly better quality and lower decay than those in the control. Rizzo and Muratore (2009) found that the most suitable MA packaging was polyolefin film (coextruded PE and PP) with an antifogging additive and storage at 4 ± 1°C and 90% RH for 35 days. They found tiny accumulations of condensate but it did not reduce shelf-life. Gómez and Artes (2004b) reported that decay developed in non-wrapped celery, but wrapping in either OPP or LDPE inhibited decay, decreased the development of pithiness and retained the sensory quality, reducing both the development of the butt end cut browning and chlorophyll degradation.

### Cherimoya

Melo *et al.* (2002) found that the cultivar ‘Fino of Jete’ stored at 12 ± 1°C and 90–95% RH had a marketable life in air of 2 weeks, while fruits packed with zeolite film could be stored for 4 weeks.

### Cherry

Bertolini (1972) stored the cultivar ‘Durone Neo I’ at 0°C for up to 20 days in air, 20 kPa CO<sub>2</sub> + 17 kPa O<sub>2</sub> or in 0.05 mm PE film bags. Cherries stored in the PE bags retained their freshness best as well as the colour of both the fruit and the stalks. The cultivars ‘V-690618’, ‘V-690616’ and ‘Hedelfingen’ were stored in polyolefin film bags with O<sub>2</sub> permeabilities of 3000, 7000 and 16,500 ml/m<sup>2</sup> (in 24 h at 23°C) for 6 weeks at 2 ± 1°C. The bags with the lowest O<sub>2</sub> permeabilities produced the best results and the cherries had reduced levels of decay. For the cultivar ‘Hedelfingen’, MA packaging reduced decay by 50% and significantly increased firmness (Skog *et al.*, 2003). Massignan *et al.* (2006) sealed the cultivar ‘Ferrovia’ in plastic bags and stored the cherries at 0.5°C in air for 30 days with 2 days of shelf-life at 15°C. They found that the cherries retained their quality throughout the storage and shelf-life period. The equilibrium atmosphere inside the bags was 10 kPa CO<sub>2</sub> + 10 kPa O<sub>2</sub>. Kucukbasmac *et al.* (2008) packed the cultivar ‘0900 Ziraat’ in either Xtend® CH-49 bags or PE bags in three sizes (500 g, 700 g and 1000 g) and stored them at 0°C for up to 21 days plus 3 days at 20°C to simulate shelf-life. The quality and firmness of the cherries from all three Xtend® CH-49 bags were maintained even after an additional 3 days shelf-life and were better than those stored in PE bags. Celikel *et al.* (2003) stored the cultivar Merton Bigarreau at 0°C in polystyrene trays (350 g/tray) over-wrapped with PVC stretch film or P-Plus PP film at different permeability, or left some not wrapped as a control. MA packaging doubled the storage life, compared with the control, to 8 weeks. The

weight loss reached 22.5% for the control, 4.5% with PVC and 0.3–0.5% with PP after 8 weeks. Less permeable films, PP-90 and PE-120, also maintained the fruit quality and flavour better than high-permeable film. Alique *et al.* (2003) found that storage of the cultivar 'Navalinda' at 20°C in 15% micro-perforated films preserved fruit acidity and firmness while slowing the darkening of colour, loss of quality and decay and thus prolonging the shelf-life. However, they recommended that the levels of hypoxia reached at 20°C in micro-perforated films should restrict its use only to the distribution and marketing processes. Padilla-Zakour *et al.* (2004) stored the cultivars 'Hedelfingen' and 'Lapins' in the micro-perforated LDPE bags, called LifeSpan L204, which equilibrated at 4–5 kPa O<sub>2</sub> + 7–8 kPa CO<sub>2</sub>, and LifeSpan 208, which equilibrated at 9–10 kPa O<sub>2</sub> + 8–9 kPa CO<sub>2</sub>. After 4 weeks at 3°C with 90% RH, the fruit in the LDPE bags had green and healthy stems, better colour, better appearance and eating quality when compared with non-wrapped fruit. There was slightly better quality for LifeSpan 204, which had a lower O<sub>2</sub> permeability. Horvitz *et al.* (2004) stored the cultivar 'Sweetheart' at 0°C in LDPE bags for up to 42 days and found no significant changes in colour, firmness, TSS, pedicel dehydration and rotting. In PVC bags there was rapid pedicel deterioration, which affected approximately 50% of the fruit after 7 days of storage.

Sweet cherries were stored with the essential oils thymol, eugenol or menthol separately in trays sealed within PE bags. During 16 days storage at 1°C and 90% RH it was found that these treatments reduced weight loss, delayed colour changes and maintained fruit firmness compared with cherries packed without essential oils. Pedicels remained green in treated cherries while they became brown in those without essential oils. The equilibrium atmosphere inside the bags was 2–3 kPa CO<sub>2</sub> + 11–12 kPa O<sub>2</sub>, which was reached after 9 days with or without the essential oils. The essential oils reduced moulds and yeasts and total aerobic mesophilic colonies (Serrano *et al.*, 2006).

### Cherry, Suriname

Santos *et al.* (2006 a, b) harvested Suriname cherry (*Eugenia uniflora*) at three different maturities based on skin colour (pigment initiation, red-orange and predominant red) and stored them in different temperatures either not packaged or in PVC bags. Storage at 10°C or 14°C in PVC bags maintained TSS, TA, sugars and AA, and also resulted in a lower rate of increase in total carotenoids in fruits in the red-orange maturity stage for 8 days. Santos *et al.* (2006b) showed that fruit in PVC bags at 10°C or 14°C had a lower disease incidence and fruit shrinkage, giving a 4-day increase in postharvest life compared with those stored non-wrapped. Red-orange was the most suitable harvest maturity stage and 10°C the best temperature of those that were tested.

### Chestnut, Japanese

It was reported by Lee *et al.* (1983) that the best storage results for the cultivar 'Okkwang' Japanese chestnut (*Castanea crenata*) were obtained when the cherries were sealed in 0.1 mm PE bags (30 cm × 40 cm) with 9–11 pin holes. The total weight loss was < 20% after 8 months of storage; the CO<sub>2</sub> concentration in the bags increased to 5–7 kPa and the O<sub>2</sub> concentration was reduced to 7–9 kPa. Panagou *et al.* (2006) showed that the atmosphere at equilibrium was 10.5 kPa O<sub>2</sub> + 10.9 kPa CO<sub>2</sub> for micro-perforated PET/PE (12 μm/40 μm) permeable film at 0°C and 8.3 kPa O<sub>2</sub> + 12 kPa CO<sub>2</sub> at 8°C. The total storage period was 110 days and sucrose content increased during the first 40 days in PET/PE micro-perforated and macro-perforated films, but changed only slightly thereafter, though starch and AA decreased throughout the storage period. In some less permeable film packages the O<sub>2</sub> concentration fell below 2 kPa, resulting in the development of off-odours. To create different O<sub>2</sub> and CO<sub>2</sub> compositions Homma *et al.* (2008) stored chestnuts in one, two or five layers of PE bags. The proportion of sound chestnuts decreased for all treatments after 2 months

and starch breakdown was high in one or two layers of PE bag but was decreased when five layers of PE bag were used.

### Chilli

Fresh chillies are often marketed in MA packaging. Fruits of the cultivar 'Nok-Kwang' were sealed in 0.025 mm PE film and stored at 18°C or 8°C. At both temperatures PE film reduced the weight loss below 1% and increased CO<sub>2</sub> concentration to 2–5 kPa (Eum and Lee, 2003). Amjad *et al.* (2009) found that the optimum thickness of PE film for maximum storage life was 15 µm at 7°C for 10 days.

### Citrus

Porat *et al.* (2004) found that packaging citrus fruits in 'bag-in-box' Xtend® films effectively reduced the development of chilling injury as well as other types of rind disorders that are not related to chilling, such as rind breakdown, stem-end rind breakdown and shrivelling and collapse of the button tissue (ageing). In all cases, micro-perforated films (0.002% perforated area) that maintained 2–3 kPa CO<sub>2</sub> + 17–18 kPa O<sub>2</sub> inside the package were much more effective in reducing the development of rind disorders than macro-perforated films (0.06% perforated area), which maintained an equilibrium gas content of 0.2–0.4 kPa CO<sub>2</sub> + 19–20 kPa O<sub>2</sub>. After 5 weeks at 6°C and 5 days of shelf-life in ambient conditions, chilling injury was reduced by 75% and rind disorders by 50% in 'Shamouti' orange and by 60% and 40%, respectively, in 'Minneola' tangerines for those in Xtend® compared with non-packaged fruit. Similarly, micro-perforated and macro-perforated Xtend® packages reduced the development of chilling injury after 6 weeks at 2°C and 5 days of shelf-life in 'Shamouti' oranges and 'Star Ruby' grapefruit. Choi *et al.* (2002) successfully stored the cultivar 'Tsunokaori of Tangor' ((*Citrus sinensis* × *C. unshiu*) × *C. unshiu*) [*C. sinensis* × *C. reticulata*], (*C. reticulata* ×

*C. sinensis* × *C. nobilis*) in LDPE bags with or without calcium oxide and potassium permanganate in storage at 4 ± 1°C with 85 ± 5% RH. Calcium oxide decreased water loss of the fruit during storage but the sugar content did not change in any of the treatments. Storage of *C. unshiu* in China in containers with a D45 M2 1 silicone window of 20–25 cm<sup>2</sup>/kg of fruit gave the optimum concentration of < 3 kPa CO<sub>2</sub> + < 10 kPa O<sub>2</sub> (Hong *et al.*, 1983). Del-Valle *et al.* (2009) reported that the optimum equilibrium MA packaging for mandarin segments at 3°C was 19.8 kPa O<sub>2</sub> + 1.2 kPa CO<sub>2</sub>.

### Cucumber

In Europe, cucumbers are commonly marketed in shrink-wrapped plastic sleeves that provide a modified atmosphere which helps to retain their texture and colour (see Fig. 14.7). 'Beit Alpha'-type cucumbers responded favourably to MA packaging with up to 8–9 kPa CO<sub>2</sub>, showing reduced physiological disorders and decay, but CO<sub>2</sub> levels above 10 kPa were injurious. The combination of optimal atmosphere composition and humidity inhibited the cucumbers from toughening and preserved their tender texture and turgidity (Rodov *et al.*, 2003). Cucumbers stored in 100 kPa O<sub>2</sub> in plastic bags retained their fresh appearance but developed off-flavours; and at 5°C the level of O<sub>2</sub> in the bags had fallen to 50 kPa and CO<sub>2</sub> had risen to 20 kPa after 8 days (Srialong *et al.*, 2005).

### Endive

The quality of endive packed on polystyrene trays in LDPE bags or P-Plus PE coverings was commercially acceptable after 3 weeks storage at 5°C. Those covered with vinyl film (called Borden Vinilo) had the lowest microbial counts, with *Pseudomonas* and enterobacteria being the principal microbial contaminants (Venturini *et al.*, 2000). Charles *et al.* (2005) stored endive at 20°C in LDPE bags which gave a 3 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> equilibrium atmosphere after

25 h storage with an O<sub>2</sub>-absorbing packet, compared with 100 h without the O<sub>2</sub>-absorbing packet. After 312 h of storage in both packages the total aerobic mesophile, yeast and mould population growth was reduced compared with those in macro-perforated OPP bags, which maintained gas composition close to that of air but also limited water loss.

### Feijoa

Feijoas were stored at 6°C, 12°C or 17°C and 85% RH without packaging, or in PP bags with 0, 1, 2 or 3 perforations of 0.225 mm. Fruits stored in MA packaging systems deteriorated more slowly than those stored without packaging at 17°C: from 17 up to 25–31 days for the MA packaging at 12°C and 17°C, and 35–38 days for the MA packaging at 6°C. It was concluded that the most favourable storage was PP bags with one perforation at 6°C, which gave a steady-state atmosphere of 8.2 kPa O<sub>2</sub> + 5.8 kPa CO<sub>2</sub> (Castellanos *et al.*, 2016).

### Fig

Fruits of the cultivar ‘Masudohin’ were sealed in 0.04 mm PE film bags; CO<sub>2</sub> was introduced into half of them and they all were stored for 10 days at 0°C followed by 2 days in ambient conditions. The bags with added CO<sub>2</sub> had an atmosphere of 12.3–14.0 kPa O<sub>2</sub> + 10.0–14.2 kPa CO<sub>2</sub> during the first 4 days, which progressively stabilized at 13.2–14.0 kPa O<sub>2</sub> + 1.1–1.2 kPa CO<sub>2</sub>. Those with added CO<sub>2</sub> had better visual quality, a slower rate of softening and less incidence of decay than those without the CO<sub>2</sub> treatment (Park and Jung, 2000). Fruits of the cultivar ‘Roxo de Valinhos’ were packed in 50 µm PP film bags with or without flushing with 6.5 kPa O<sub>2</sub> + 20 kPa CO<sub>2</sub> and stored at 20 ± 2°C and 85 ± 5% RH for 7 days. Those in PP bags retained their appearance and had low weight loss of 1.7–2.5% compared with about 40% for non-wrapped fruit, but there were high levels of decay in all treatments (Souza and

Ferraz, 2009). Villalobos *et al.* (2018) stored the cultivars ‘Cuello Dama Negro’ and ‘San Antonio’ in 100 µm micro-perforated films with 16, 5 or 3 holes or 9 mm macro-perforated film with 5 holes, all at 0°C for 14 days followed by 2 days at 20°C. ‘San Antonio’ fruit in the 100 µm micro-perforated films with 3 holes had delayed changes in the volatile profile, compared with the other treatments, without negative influence on their flavour.

### Grape

The cultivars ‘Kyoho’ and ‘Campbell Early’ packed in 0.03 mm PE film bags were stored for 60 days at 0°C and 90% RH by Yang *et al.* (2007). They found that packaging in PE film maintained fruit firmness, TSS, TA and skin colour, and reduced respiration rate, decay and weight loss compared with those stored non-wrapped. In contrast, grapes in antifogging and perforated film had increased decay and abscission. Yamashita *et al.* (2000) stored the cultivar ‘Italia’ at 1°C and 85–90% RH followed by 25°C and 80–90% RH in three different film bags, all of which had high gas permeability. Fruit in all three films stored well but Cryovac PD-955 gave the longest storage life of 63 days, compared with 11–21 days for fruit stored without packaging. Artes-Hernández *et al.* (2003) found that an equilibrium atmosphere of about 15 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> gave the best results and for the cultivar ‘Autumn Seedless’ this atmosphere was provided by micro-perforated PP film 35 µm thick.

Several workers have studied the use of MA packaging in combinations with other treatments as an alternative to sulfur dioxide treatment to control *Botrytis cinerea* in grapes. Artes-Hernández *et al.* (2007) used ozone (O<sub>3</sub>) at 0.1 µl/l and found that the sensory quality of the fruit was preserved with MA packaging giving an atmosphere of 13–16 kPa O<sub>2</sub> + 8–11 kPa CO<sub>2</sub> and in CA storage in 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub>. Although O<sub>3</sub> did not completely inhibit fungal development, its application increased anti-oxidant compounds. A pre-storage dip of the

cultivar 'Superior' in either 33% or 50% ethanol then storage in sealed Xtend® or PE film bags maintained grape quality for up to 7 weeks with 3 days shelf-life at 20°C. The treatment was similar to or better than storage with SO<sub>2</sub>-releasing pads (Lichter *et al.*, 2005). Artes-Hernández *et al.* (2003) stored the cultivar 'Napoleon' for 41 days in MA packaging or air followed by 4 days in air, all at 0°C, then 3 days at 15°C. The best results were by placing a soaked filter paper with 15 or 10 µl hexenal combined with sealing in micro-perforated PP of 35 µm thickness. MA packaging improved the visual appearance and reduced weight loss to 0.6% compared with non-packaged grapes, which showed a 3.7% weight loss. Hexenal-treated grapes had the highest score for flavour and the lowest level of fungal disease (6%). Valverde *et al.* (2005) showed that the addition of 0.5 µl of eugenol, thymol or menthol inside MA packages containing the cultivar 'Crimson Seedless' reduced weight loss and colour change, delayed rates of pedicel deterioration, retarded °brix:acid ratio development, maintained firmness and reduced decay. Also the total viable counts for mesophilic aerobics, yeasts and moulds were significantly reduced in the grapes packaged with eugenol, thymol or menthol. The authors concluded that the inclusion of one of these essential oils resulted in an additional 3 weeks of storage compared with MA packaging only. The equilibrium atmosphere in PP bags in storage at 1°C was reached in 21 days with 1.2–1.3 kPa CO<sub>2</sub> and 13–14 kPa O<sub>2</sub>. The inclusion of a sachet impregnated with either 0.5 ml thymol or 0.5 ml menthol inside the bag reduced yeasts, moulds and total aerobic mesophylic colonies and also improved the visual aspect of the rachis. The authors postulated that these essential oils could be an alternative to the use of SO<sub>2</sub> in controlling disease in table grapes (Martinez-Romero *et al.*, 2005).

Ben Arie (1996) described a method of MA packaging for sea-freight transport. This involved whole pallet loads of half a tonne of fruit in cartons being wrapped in PE film. The advantage of this compared with individual cartons being lined with PE film was that any condensation remained on the

outside of the box and pre-cooling was more rapid, since the boxes were pre-cooled before the plastic was applied.

### Kiwifruit

'Hayward' kiwifruit stored in plastic trays and over-wrapped with 16 µm stretch film at both 10°C and 20°C ripened more slowly than non-wrapped fruit but to a similar flavour (Kitsiou and Sfakiotakis, 2003). Storage of 'Hayward' for 6 months at 0°C and 95% RH in sealed 50 cm × 70 cm PE bags with an ethylene absorbent resulted in firmer fruit, higher TA and AA and lower weight loss than when stored non-wrapped. The equilibrium atmosphere in the bags was 6–8 kPa O<sub>2</sub> + 7–9 kPa CO<sub>2</sub> after 6 months of storage (Pekmezci *et al.*, 2004).

### Kohlrabi

Storage in antimist OPP bags 20 µm thick reduced weight loss and development of bacterial soft and black rots and extended the storage life to 60 days at 0°C plus 3 days at 12°C (Escalona *et al.*, 2007). The equilibrium atmosphere inside the bags was 4.5–5.5 kPa O<sub>2</sub> + 11–12 kPa CO<sub>2</sub>.

### Lanzones (langsat)

Pantastico (1975) indicated that the skin of the fruit turns brown during retailing and if they are sealed in PE film bags the browning was aggravated, probably due to CO<sub>2</sub> accumulation.

### Lemon

Lemons stored at 25°C and 50–60% RH for up to 21 days in 100 gauge PVC 20% shrink-wrap retained their firmness and TSS, with reduced loss of weight and rotting compared with those stored non-wrapped or waxed (Neri *et al.*, 2004).

### Lemongrass

Lemongrass leaves were hot-water treated at 55°C for 5 min, then hydrocooled at 3°C for 5 min and placed in PE bags, which were then flushed with various levels of O<sub>2</sub> + CO<sub>2</sub>. They were then sealed and stored at 5°C and 90 ± 5% RH for 3 weeks. They found that *E. coli* was inhibited in 1 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub>, coliforms and faecal coliforms were inhibited in 5 kPa O<sub>2</sub> and *Salmonella* at all gas mixtures for up to 14 days. Fungi were not controlled in any of the atmospheres tested (Samosornsuk *et al.*, 2009).

### Lettuce

Cameron (2003) found that at 5°C browning was retarded when O<sub>2</sub> was below 1 kPa, while fermentation occurred below 0.3–0.5 kPa O<sub>2</sub>; therefore MA packaging that gave an atmosphere between 0.5 and 1 kPa O<sub>2</sub> was recommended. Escalona *et al.* (2006b) concluded that 80 kPa O<sub>2</sub> should be used in MA packaging of fresh-cut butterhead lettuce in combination with 10–20 kPa CO<sub>2</sub> to reduce their respiration rate and avoid fermentation.

### Lime

Storage at 6°C led to chilling injury and at 10°C to ‘ageing’ (Ladaniya, 2004). Packing limes in sealed PE film bags inside cartons in ambient conditions in Khartoum in the Sudan resulted in a weight loss of only 1.3% in 5 days, but all the fruits de-greened more rapidly than those that were packed in cartons without the PE liner, where the weight loss was 13.8% (Thompson *et al.*, 1974c). This de-greening effect could be countered by including potassium permanganate on vermiculite in the bags. The potential of 30 µm HDPE bags with 1 × 40, 2 × 40 or 3 × 40 micro-perforations was studied by Ramin and Khoshbakhat (2008). The greenest and firmest fruits were found in micro-perforated PE bags at a storage temperature of 10°C. At 20°C, fruits kept in micro-perforated bags were consistently greener than fruits that were stored

non-packaged, but decay was high. Decay was highest in fruits with no package and lowest in PE with micro-perforations. Jadhao *et al.* (2007) found that freshness was retained and flavour and acceptability scores were the highest for the cultivar ‘Kagzi’ stored at 8 ± 1°C for up to 90 days in 25 µm PE bags with 0.5% area as punched holes, with minimum ageing and no chilling injury. Those stored without packing were spoiled completely within 20 days. The fruits stored in 100 gauge ventilated PE bags maintained maximum juice content, TA and AA and minimum TSS and spoilage for 30 days.

### Litchi

Ragnoi (1989), working with the cultivar ‘Hong Huai’ from Thailand, showed that fruits packed in sealed 150 gauge PE film bags containing 2 kg of fruit and sulfur dioxide SO<sub>2</sub> pads could be kept in good condition at 2°C for up to 2 weeks, while fruits that were stored without packaging rapidly became discoloured and unmarketable. Anthocyanin levels in the pericarp are important, since they give the fruit an attractive appearance. Somboonkaew and Terry (2009) studied storage in a selection of films and found that anthocyanin concentrations in the cultivar ‘Mauritius’ wrapped in PropaFresh™ film (Table 14.6) were significantly higher than for other plastic films after 9 days of storage (see Fig. 14.4).

In storage at 14°C the cultivar ‘McLean’s Red’ in PP film had 11.3% decay due mainly to *Alternaria alternata* and *Cladosporium* spp. However, fruits that had been dipped

**Table 14.6.** The water and gas permeability of the films used by Somboonkaew and Terry (2009). (More information on the films is available at [www.innoviafilms.com](http://www.innoviafilms.com).)

Packaging films	Water (g/m <sup>2</sup> /24 h)	O <sub>2</sub> (cc/m <sup>2</sup> / 24 h/bar)
PropaFresh™	5	1600
NatureFlex™	360	3
Cellophane™	370	3



for 2 min at 15°C in *Bacillus subtilis* and stored at 14°C in PP film had no decay or pericarp browning and retained their colour and quality. The equilibrium atmosphere inside the PE bags was about 14 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>. Fruits stored in LDPE bags with *B. subtilis* had higher levels of decay and pericarp browning than those in PP bags. Higher yeast populations were observed in LDPE or LDPE with *B. subtilis* during storage at 2°C or 14°C and *Candida*, *Cryptococcus* and *Zygosaccharomyces* spp. were the predominant yeasts. The equilibrium atmosphere inside the LDPE bags was about 3 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> (Sivakumar *et al.*, 2007).

### Longan

Zhang and Quantick (1997) stored the cultivar 'Shixia' in PE film 0.03 mm thick for 7 days at room temperature followed by 35 days at 4°C. Atmospheres of 1, 3, 10 or 21 kPa O<sub>2</sub> were established in the bags and the former two were said to be effective in delaying peel browning and retaining TSS and AA content of the fruit, though taste panels detected a slight off-flavour in fruit stored in 1 kPa O<sub>2</sub>. Seubrach *et al.* (2006) stored fruits of the cultivar 'Daw' in polystyrene boxes covered with 15 µm PVC film or LLDPE film of 10, 15 and 20 µm thickness, all at 4°C and 90–95 % RH. All the MA conditions extended the shelf-life to 20 days compared with 16 days for non-wrapped fruit. Fruit had slightly lower weight loss in the LLDPE films compared with those in the PVC film. Peel colour gradually deteriorated in all packages but less so for those in the PVC film, which also had the best appearance and better overall customer acceptance.

### Loquat

Loquat cultivar 'Algerie' fruits were stored in five types of micro-perforated PP films for up to 6 weeks at 2°C then 4 days non-wrapped at 20°C. The equilibrium atmosphere changed from 1.2 to 8.5 kPa CO<sub>2</sub> and 19.5 to 13 kPa O<sub>2</sub> as film permeability decreased. Peel softening and colour development were decreased,

as were changes in sugars and organic acids, especially after the shelf-life period, compared with the controls. They concluded that the most suitable atmosphere was about 2–4 kPa CO<sub>2</sub> + 16–18 kPa O<sub>2</sub>, achieved in PA-80 and PA-60 films (Ding *et al.*, 2006). Similar results were reported by Amorós *et al.* (2008), who reported that postharvest life at 2°C without wrappings was 2 weeks with a 4-day shelf-life at 20°C, but in the same environment the postharvest life of those in PA-80 or PA-60 films was 6 weeks plus a 4-day shelf-life.

### Mango

Mangoes stored in PE film bags at 21°C had almost twice the storage life of those stored without wraps (Thompson, 1971). Mature fruits of the cultivar 'Keitt' were stored at 20°C in air or in MA jars supplied with humidified CO<sub>2</sub> at 210 ml/min for 2 h before being sealed, resulting in 0.03–0.26 kPa O<sub>2</sub> + 72–82 kPa CO<sub>2</sub> for up to 4 days, or in jars ventilated with 2 kPa O<sub>2</sub> + 50 kPa CO<sub>2</sub> + 48 kPa N<sub>2</sub> at a continuous rate of 210 ml/min for up to 5 days. Both treatments delayed fruit ripening. Fruits showed no signs of internal or external injuries or off-flavour, either immediately after removal from storage or after transfer to air. There were no significant differences between air-stored and CA-stored fruits in all sensory attributes evaluated, but the overall acceptability of fruits stored in MA for over 72 h was lower than that of fruits stored in air (Yahia and Vazquez Moreno, 1993). The cultivar 'Keitt' stored in CA in 0.3 kPa O<sub>2</sub> or MA packaging in three types of LDPE films at 20°C had slower weight losses and remained firmer and with good appearance with a significant delay in ripening compared with controls. However, a few fruits packed in two of the three films developed a fermented flavour after 10 days of storage (Gonzalez-Aguilar *et al.*, 1994).

### Melon

Martinez-Javega *et al.* (1983) reported that sealing fruits individually in PE film bags

0.017 mm thick reduced chilling injury during storage at 7–8°C. Flores *et al.* (2004) also found that MA packaging reduced chilling injury in the cultivar ‘Charentais’, which is subject to chilling injury when stored at temperatures around 2°C. They also found that MA packaging extended the postharvest life of wild-type melons and conferred additional chilling resistance on ethylene-suppressed melons during storage at 2°C and shelf-life at 22°C. Asghari (2009) reported that immersing the cultivar ‘Sensory’ in hot water at 55°C for 3 min or 59°C for 2 min and then sealing them in 30 µm PE bags kept them in good condition at 2.5°C for 33 days. This was longer than either treatment alone.

The effects of packing in ceramic films with thicknesses of 20, 40 or 80 µm and storage at either 3°C or 10°C showed that loss in fresh weight was greatly reduced to below 0.3% after 36 days at 3°C for fruits in ceramic films. CO<sub>2</sub> accumulated up to 15 and 30 kPa at 3°C and 10°C, respectively, in films 80 µm thick. The concentration of acetaldehyde was over 4 times higher in film 80 µm thick compared with the others at both 3°C and 10°C. Fruits in film 40 µm thick retained their firmness, TSS, TA and visual quality better than any other treatment and their storage life at 3°C was 36 days, which was twice as long as in the other treatments. Chilling injury symptoms occurred at 3°C but fruits packed in films 40 µm thick had lower levels (Kang and Park, 2000). Sa *et al.* (2008) stored cantaloupe in X-tend bags for 14 days at 3 ± 2°C and then removed them from the bags and stored them for a further 8 days at 23 ± 2°C and 90 ± 2% RH. The fruits retained their quality in the bags. Pre-treatment with 1-MCP or inclusion of potassium permanganate on vermiculite within the bags gave no additional effects. MA packaging extended the postharvest life of charentais-type melons by delaying ripening, in spite of the high in-package ethylene concentration of 120 µl/l (Rodov *et al.*, 2003).

### Mushroom

Because of the high respiration rates and the very limited positive effects on quality,

Anonymous (2003) suggested that it did not seem beneficial to pack mushrooms in MA packages. However, storage in MA packaging at 10°C and 85% RH for 8 days delayed maturation and reduced weight loss compared with storage without packaging (Lopez-Briones *et al.*, 1993). PVC overwraps on consumer-sized punnets (about 400 g) greatly reduced weight loss, cap and stalk development and discoloration, especially when combined with refrigeration (Nichols, 1971). MA packaging in microporous film delayed mushroom development, especially when combined with storage at 2°C (Burton and Twynning, 1989). Mushrooms were stored at 4°C in containers where atmospheres of 5 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> were maintained and they were subjected to a sequence of temperature fluctuations of 10°C for 12 days. Temperature fluctuations had a major impact on the composition of the atmospheres and on product quality. CO<sub>2</sub> concentrations increased rapidly, reaching a maximum of 16 kPa, while O<sub>2</sub> concentrations decreased to less than 1.5 kPa. The temperature-fluctuating regime resulted in extensive browning, softening, weight loss increase, ethanol increase and infection due to physiological damage and excessive condensation, compared with storage at a constant temperature (Tano *et al.*, 2007). Hu *et al.* (2003) found that there was a positive effect on postharvest life when shiitake were stored in LDPE or PVC film bags. Respiration rates were suppressed by decreasing O<sub>2</sub> and increasing CO<sub>2</sub> concentrations at 5°C, 15°C, 20°C and 30°C, but there was little difference at 0°C. Lin *et al.* (2017) packed button mushrooms in MA packages with 95–100 kPa CO<sub>2</sub>. After 12 h they punctured the film. This CO<sub>2</sub> exposure reduced browning and MDA and increased antioxidant, catalase and POD activities during subsequent storage. Wei *et al.* (2017) reported decreased PPO activities, delayed textural changes and reduced respiration rates and weight loss when pine mushrooms were stored in PE bags. They also tested PVC bags with silicon windows and found that odour changes were delayed and the bags contained lower amounts of ammonia. Both the package types tested resulted in increased catalase activity and AA content.

### Okra

Hardenburg *et al.* (1990) reported that storage in 5–10 kPa CO<sub>2</sub> lengthened shelf-life by about a week. Finger *et al.* (2008) stored the Brazilian cultivar ‘Amarelinho’ at 25°C, 10°C or 5°C wrapped in PVC over a polystyrene tray or left not wrapped. The development of chilling symptoms (surface pitting) occurred at 5°C but was delayed in those covered with PVC film. Those stored at 25°C lost weight rapidly and became wilted, especially those not wrapped, therefore the best treatment was at 10°C covered in PVC. Mota *et al.* (2006) also found that PVC overwraps were efficient in reducing weight loss and in retaining the vitamin C contents and reducing browning during storage.

### Onion

Packing onion bulbs in plastic film is uncommon, because the high humidity inside the bags can cause rotting and root growth (see [Table 14.2](#)).

### Papaya

Wills (1990) reported that MA packaging in plastic film extended the storage life of papaya. Singh and Rao (2005) found that packaging the cultivar ‘Solo’ in LDPE or Pebax-C bags and storing at 7°C or 13°C and 85–90% RH prevented the development of chilling injury symptoms during storage at 7°C or 13°C for 30 days, but fruits stored at 7°C failed to ripen normally and showed blotchy appearance, due to skin scald, and also had internal damage. LDPE and Pebax-C packed fruits showed higher retention of AA, total carotenoids and lycopene after 30 days storage at 13°C and 7 days non-wrapped at 20°C compared with storage non-wrapped throughout. Rohani and Zaipun (2007) reported that the storage life of the cultivar ‘Ekstotika’ could be extended to 4 weeks at 10–12°C when nine to ten fruits were stored in 0.04 mm LDPE bags measuring 72 cm × 66 cm. After 24 h the CO<sub>2</sub> level inside the

bags was about 4–5 kPa and O<sub>2</sub> about 2–3 kPa. After 4 weeks of storage the fruit ripened normally within 3–4 days in ambient temperature when removed from the bags.

### Parsley

Desiccation was the most important cause of postharvest loss. Storage in folded unsealed PE bags at 1°C combined with pre-cooling gave a maximum storage life of about 6 weeks. If desiccation occurred, placing the petioles in water enabled them to recover from a maximum of 10% loss (Almeida and Valente, 2005).

### Passionfruit

Mohammed (1993) found that both yellow and purple passionfruit stored best at 10°C in perforated LDPE film bags 0.025 mm thick.

### Pea

Pariasca *et al.* (2001) stored snow pea pods in polymethyl pentene polymeric films of 25 and 35 µm thickness at 5°C. They found that internal quality was maintained and that appearance, colour, chlorophyll, AA, sugar content and sensory scores were retained better than for those stored without packaging. The equilibrium atmosphere in both thicknesses of bags was around 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>.

### Peach

Malakou and Nanos (2005) treated the cultivar ‘Royal Glory’ in hot water at 46°C containing 200 mM sodium chloride for 25 min, then sealed them in PE bags and stored them at 0°C for up to 2 weeks. This combination resulted in good quality fruit after 1 week and gave an equilibrium atmosphere of > 15 kPa O<sub>2</sub> + < 5 kPa CO<sub>2</sub> inside the bags. Ripening

was slowed in the PE bags compared with those not in bags. However, the gas content inside the bags changed to  $< 3 \text{ kPa O}_2 + > 13 \text{ kPa CO}_2$  within 10 h at room temperature, which was found to damage the fruit. It was therefore recommended that they should be removed from the bags directly after removal from cold storage. De Santana *et al.* (2009) in a comparison of storage of the cultivar 'Douradão' at  $1 \pm 1^\circ\text{C}$  and  $90 \pm 5\%$  RH in LDPE bags of 30, 50, 60 and 75  $\mu\text{m}$  found that 50 and 60  $\mu\text{m}$  were the most suitable and that after 28 days of storage fruit quality was better than those in 30 or 75  $\mu\text{m}$  LDPE. Zoffoli *et al.* (1997) stored the cultivars 'Elegant Lady' and 'O'Henry' at  $0^\circ\text{C}$  in film bags of various permeabilities. They found that the atmospheres inside the packs varied between 10 and 25 kPa  $\text{CO}_2$  and 1.5 and 10 kPa  $\text{O}_2$ . The rates of fruit softening and flesh browning were reduced in all the high- $\text{CO}_2$  packages, but there was some development of mealiness and off-flavours.

### Pear

Geeson *et al.* (1990a) found that when unripe 'Doyenné du Comice' were held in MA packages at  $20^\circ\text{C}$  softening was faster than chlorophyll loss, so that the appearance of the fruit failed to match its internal condition. However, a 4-day extension in shelf-life was obtained by MA packaging of fruits in a part-ripe condition and, although the effects on rate of softening were less than those for unripe fruits, the yellower ground colour was more consistent with the eating qualities of the fruit. For the cultivar 'Conference', Geeson *et al.* (1990b) found that in MA packs the rate of flesh softening was only partially slowed; chlorophyll degradation was completely inhibited but resumed when packs were perforated. The equilibrated atmosphere within the packs after about 3 days was  $5\text{--}9 \text{ kPa CO}_2 + < 5 \text{ kPa O}_2$ . Pears retarded by MA packaging failed to develop the normal sweet, aromatic flavour and succulent, juicy texture of 'eating-ripe' fruit, even when the packs were perforated after 4 days.

### Persimmon

In Korea, fruits are stored for 6 months at  $2\text{--}3^\circ\text{C}$  with minimal rotting or colour change, and storage in the same conditions with ten fruits sealed in PE film bags 60  $\mu\text{m}$  thick was even better (A.K. Thompson, unpublished results). Astringency of persimmons was removed by storing fruits in PE bags that had been evacuated or with total  $\text{N}_2$  or  $\text{CO}_2$  atmospheres, but the fruits stored in the  $\text{CO}_2$  atmosphere were susceptible to flesh browning. Fruits stored under vacuum or an  $\text{N}_2$  atmosphere maintained high quality and firmness for 2 weeks at  $20^\circ\text{C}$  and 3 months at  $1^\circ\text{C}$ . Flesh appearance and taste after 14 weeks of storage at  $1^\circ\text{C}$  was best in the  $\text{N}_2$  atmosphere (Pesis *et al.*, 1986). Liamnimitr *et al.* (2018) designed a bulk MAP for storage of 'Fuyu'. Initially, the lower  $\text{O}_2$  limit of aerobic respiration was determined to be 0.65 kPa, based on the RQ breakpoint. They then used a Michaelis–Menten-type respiration model to predict the  $\text{O}_2$  consumption rate at 1.5 kPa, which was set as the target  $\text{O}_2$  level inside the package in order to avoid fermentation. They compared this bulk MAP with an individual package (LDPE film bag 60  $\mu\text{m}$  thick) and non-packed fruit. After 4 months of storage and subsequent transport of fruit from Japan to Thailand, the  $\text{O}_2$  partial pressure at a steady state in bulk MAP was 1.3–1.6 kPa, which was close to the 1.5 kPa target. Softening and loss of colour of fruits were reduced in the bulk MAP compared with the other two treatments.

### Pineapple

Storage in PE bags led to condensation and mould growth. No condensation was observed in cellophane bags, as they allowed moisture to pass through them, but there was some drying out of the crowns. However, in both types of bag off-flavours were detected due to  $\text{CO}_2$  accumulation and, because of the fruit's spiny nature, bags were considered impractical (Paull and Chen, 2003). Black heart development was less than

10% in storage of the cultivar 'Mauritius' in PE film bags for 2 weeks at 10°C with an equilibrium atmosphere inside the bags of about 10 kPa O<sub>2</sub> + 7 kPa CO<sub>2</sub> (Hassan *et al.*, 1985). They also suggested that pineapple should be kept in PE film bags until consumption in order to avoid the development of black heart.

Minimally processed pineapples are commonly packed in MA and 5 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub> appeared to be the most appropriate atmosphere for storage (Budu *et al.*, 2007). Minimally processed pineapple stored at 5 or 10 kPa O<sub>2</sub> + 10 kPa CO<sub>2</sub> helped to maintain their lightness and visual quality and prolonged the storage life up to 9 days (Tancharoensukjit and Chantanawarangoon, 2008). Pineapples were sliced and dipped in a solution of sodium hypochlorite at 20 mg/l for 30 seconds, then stored at 5 ± 1°C in a flow-through system with different O<sub>2</sub> and CO<sub>2</sub> concentrations. None of the CA treatments appeared to be advantageous, since the slices had little browning and were free of any contamination that would affect the food safety at the end of the storage period (Antoniolli *et al.*, 2007).

## Plantain

The effects of PE film wraps on the postharvest life of plantains may be related to moisture conservation around the fruit as well as the change in the CO<sub>2</sub> and O<sub>2</sub> content. This was shown by Thompson *et al.* (1972b) where fruits stored in moist coir had a longer storage life than fruits that had been stored in dry coir or non-wrapped (Fig. 14.9). There was an added effect when fruits were stored in non-perforated PE bags, which was presumably due to the effects of the changes in the CO<sub>2</sub> and O<sub>2</sub> levels (Table 14.7). Thus the positive effects of storage of pre-climacteric fruits in sealed plastic films may be, in certain cases, a combination of its effects on the CO<sub>2</sub> and O<sub>2</sub> content within the fruit and the maintenance of high moisture content.

## Pomegranate

The cultivar 'Hicaznar' could be kept in cold storage for 2–3 months in 8 µm film, but when packed in 12 µm stretch film the fruits maintained their quality and had no fungal disorders for 3–4 months (Bayram

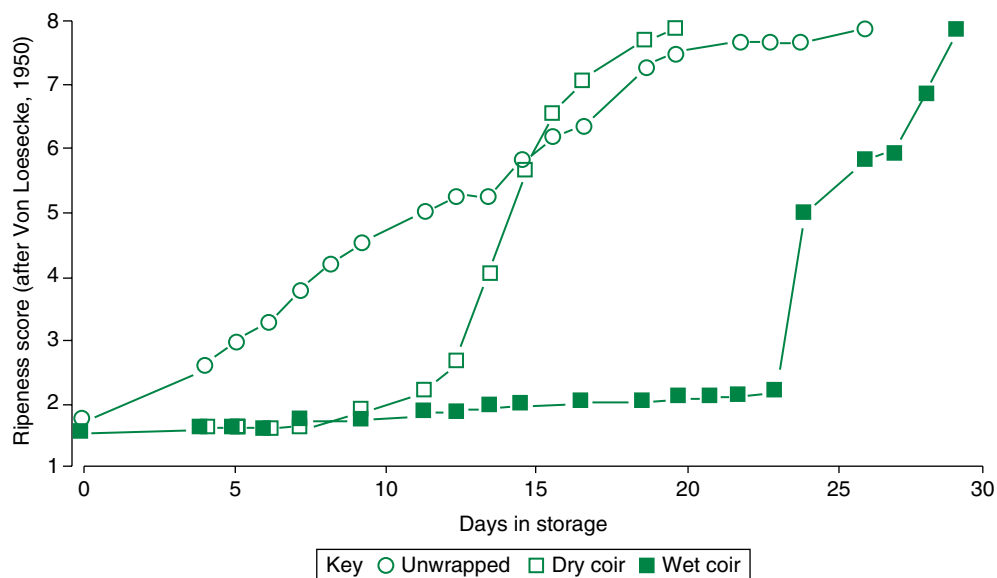


Fig. 14.9. Effects of storage in coir dust on ripening of plantains in Jamaican ambient conditions.

**Table 14.7.** Effects of wrapping and packing material on ripening and weight loss of plantains stored at tropical ambient conditions of 26–34°C and 52–87% RH; adapted from Thompson *et al.* (1972a).

Packing material	Days to ripeness	Weight loss at ripeness
Not wrapped	15.8	17.0%
Paper	18.9	17.9%
Moist coir fibre	27.2	(3.5%) <sup>a</sup>
Perforated PE	26.5	7.2%
PE	36.1	2.6%
LSD ( $p = 0.05$ )	7.28	2.81

<sup>a</sup>The fruit actually gained in weight.

*et al.*, 2009). Porat *et al.* (2009) reported that packaging 4–5 kg in cartons lined with Xtend® film bags could maintain fruit quality for 3–4 months. They found that MA packaging reduced water loss, shrinkage, scald and decay.

### Radish: Daikon

With Japanese radish, O<sub>2</sub> and CO<sub>2</sub> concentrations inside different micro-perforated film bags varied between 0.2 and 15 kPa O<sub>2</sub> and 5 and 16 kPa CO<sub>2</sub> at 15°C after 6 days of storage, depending on the degree of micro-perforation. Films that gave an equilibrium concentration of 9–13 kPa O<sub>2</sub> + 8–11 kPa CO<sub>2</sub> were suitable to maintain most of the qualitative parameters and avoiding the development of off-flavours during 6 days of storage (Saito and Rai, 2005).

### Rambutan

In air the maximum shelf-life for the cultivars ‘R162’, ‘Jit Lee’ and ‘R156’ was given as 15 days at 7.5°C and 11–13 days at 10°C (O’Hare *et al.*, 1994). They also reported that storage in ethylene at 5 µl/l at 7.5°C or 10°C did not significantly affect the rate of colour loss. Fruit have been stored successfully at 10°C in PE film bags by Mendoza *et al.* (1972) but the storage life was given as about 10 days in perforated bags and 12 days

in non-perforated bags. Fruits of the cultivar ‘Rong-Rein’ were stored in PE bags 40 µm thick at 13°C and 90–95% RH, giving an equilibrium atmosphere of 5–8 kPa O<sub>2</sub> + 7–10 kPa CO<sub>2</sub> (Boonyaritthongchai and Kanlayanarat, 2003a). They found that those in MA packaging had reduced pericarp browning, weight loss, respiration rate and ethylene production, giving a storage life of 16 days compared with fruits stored non-wrapped, which was about 12 days. Somboonkaew (2001) stored ‘Rong-Rien’ at 10°C for 22 days sealed in 0.036 mm LDPE bags where the air had been evacuated and then flushed with air, 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>, 10 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> or 15 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>. She found that, irrespective of the initial level, the CO<sub>2</sub> stabilized at about 1 kPa after 3 days then rose to a peak of about 4 kPa after 2–3 weeks. O<sub>2</sub> levels stabilized at around 3 kPa after about 1 week. Little difference was detected in softening except for those in 15 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub>, which were generally softer and less ‘bright’ after storage. On balance the fruit flushed with 5 kPa CO<sub>2</sub> + 5 kPa O<sub>2</sub> appeared best.

### Raspberry

Siro *et al.* (2006) reported that raspberries had a shelf-life of 7 days in MA and 5 days in high O<sub>2</sub> atmospheres at 7°C. *Escherichia coli*, *Salmonella* spp. and *Listeria monocytogenes* were artificially inoculated on to packaged fruits and were able to survive at 7°C. Raspberries showed an enhanced inactivation of *Salmonella* spp. during storage. Generally, MA or high-O<sub>2</sub> atmospheres did not give an increased microbial risk. Jacxsens *et al.* (2003) found that storage of fruit in sealed film packages resulted in a prolonged shelf-life compared with macro-perforated packaging films. They used selective permeable films in order to obtain 3–5 kPa O<sub>2</sub> + 5–10 kPa CO<sub>2</sub> and limited ethylene accumulation. Storage in 95 kPa O<sub>2</sub> + 5 kPa N<sub>2</sub> in a semi-barrier film resulted in the accumulation of too-high CO<sub>2</sub> levels, resulting in fermentation and softening but an inhibition of mould development. Moor

*et al.* (2009) stored the cultivar 'Polka' in Xtend®, LDPE or LDPE flushed with 10 kPa O<sub>2</sub> + 15 kPa CO<sub>2</sub>. Rotting was not suppressed in any of the bags, but after 4 days at 1.6°C those in LDPE bags that had not been flushed were of the best quality. However, after one more day at 6°C, to simulate shelf-life, the fruit in Xtend® bags were best.

### Redcurrant

McKay and Van Eck (2006) described a successful packing system where fruit were placed in fibreboard boxes that were then stacked on pallets and covered with a large plastic bag, which fitted over the pallet. The bags were sealed at the bottom. In storage at -0.9°C to 0.5°C and 93% RH, the O<sub>2</sub> concentration was lowered to 2.5–3.0 kPa by N<sub>2</sub> injection and CO<sub>2</sub> was increased to 20–25 kPa during storage. They found that 93% RH was optimum and recommended pre-cooling to 0°C to reduce the chances of condensation; they also observed that the boxes could absorb some of the moisture, which helped to prevent condensation. They reported that the berries could lose their bright red colour and appear water-soaked if the O<sub>2</sub> level was too low.

### Rhubarb

Unsealed PE film liners in crates reduced rhubarb weight loss (Lutz and Hardenburg, 1968) but there was no mention of other MA effects.

### Roseapple

Worakeeratikul *et al.* (2007) reported storage where the fruits were dipped in whey protein concentrate and then successfully stored in PVC film at 5°C for 72 h.

### Sapodilla

At ambient temperatures in Brazil, fruits stored in PVC film packaging had less weight

loss and softened more slowly than those not wrapped. Decreases in acidity, TSS and total sugar contents were similar in wrapped and non-wrapped fruit (Miranda *et al.*, 2002). In India, Waskar *et al.* (1999) found that 'Kalipatti' could be stored for up to 9 days when packed in 100 gauge PE bags with 1.2% ventilation holes at room temperature of 31.7–36.9°C. They also found that in simple cool stores fruit in PE bags could be stored for 15 days at 20.2–25.6°C and 91–95% RH.

### Soursop

In a comparison between 16°C, 22°C or 28°C the longest storage period (15 days) was at 16°C with fruits packed in HDPE bags without an ethylene absorbent. This packaging also resulted in the longest storage duration at 22°C and 28°C, but the differences between treatments at those temperatures were much less marked (Guerra *et al.*, 1995).

### Spinach

Fresh-cut 'Hybrid 424' was packed in mono-oriented PP or LDPE bags. The type and permeability of film affected off-odour development, but did not influence visual sensory attributes or chlorophyll retention (Piagentini *et al.*, 2002). Díaz-Mula *et al.* (2017) stored baby spinach in MA packages. After 14 days the packages contained 0.3 kPa O<sub>2</sub> and 9.3 kPa CO<sub>2</sub> and off-odours were detected. They found 39 main compounds with olfactory activity, including alcohols associated with lipid peroxidation, sulfur compounds from amino acid degradation and alkanes from lipid autoxidation processes or carotenoid degradation.

### Strawberry

A comparison of fruits in sealed plastic containers using impermeable PP film was made by Stewart (2003). Three internal atmospheres were injected as follows: air;

5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> + 90 kPa N<sub>2</sub>; or 80 kPa O<sub>2</sub> + 20 kPa N<sub>2</sub>. Overall the fruit stored in 80 kPa O<sub>2</sub> + 20 kPa N<sub>2</sub> proved most successful in terms of maintaining firmness, reduction in cell wall breakdown and improvement in appearance and taste.

### Sweetcorn

Othieno *et al.* (1993) showed that the best MA packaging was non-perforated film, or film with the minimum perforation of one 0.4 mm hole per 6.5 cm<sup>2</sup>. They found that cobs packed in PP film with no perforations showed a reduced rate in toughening, weight loss and reduced loss of sweetness, but they tended to develop off-odours. Manleitner *et al.* (2003) found that changes of carbohydrate composition of the cultivar 'Tasty Gold' were influenced by the permeability of the films during 20 days storage at 5°C. Sweetcorn in films with low permeability maintained carbohydrate content during storage, but had low sucrose after subsequent shelf-life of two days at 20°C. Losses of sucrose during the shelf-life phase were lowest in 20 µm LDPE film.

### Taro

MA packaging in PE film bags has been shown to reduce postharvest losses. Both *Xanthosoma* and *Colocasia* stored in Trinidad ambient conditions of 27–32°C showed reduced weight losses (Table 14.8), though the recommended optimum temperature for storage in non-CA or non-MA conditions varied from 7°C to 13°C (Passam, 1982).

### Tomato

At UK ambient temperature with about 90% RH, partially ripe tomatoes, packed in suitably permeable plastic film giving 4–6 kPa CO<sub>2</sub> + 4–6 kPa O<sub>2</sub>, can be expected to have 7 days longer shelf-life than those stored without wrapping (Geeson, 1989). Under these conditions the eating quality was slowed but film packaging that results in higher CO<sub>2</sub> and lower O<sub>2</sub> levels may not only prevent ripening but also result in tainted fruits when they are ripened after removal from the packs. He also showed that film packages with lower water vapour transmission properties can encourage rotting. Batu (1995) showed that there were interactions between MA packaging and temperature in that MA packaging was more effective in delaying ripening at 13°C than at 20°C. MA packaging interacted also with harvest maturity: it was more effective in delaying ripening of fruits harvested at the mature-green stage than for those harvested at a more advanced stage of maturity. Batu and Thompson (1994) stored tomatoes at both the mature-green and the pink stage of maturity either non-wrapped or sealed in PE films of 20 µm, 30 µm or 50 µm thickness at either 13°C or 20°C for 60 days. All non-wrapped pink tomatoes were overripe and soft after 30 days at 13°C and after 10–13 days at 20°C. Green tomatoes sealed in 20 µm and 30 µm film reached their reddest colour after 40 days at 20°C and 30 days at 13°C. Fruits in 50 µm film still had not reached their maximum red colour even after 60 days at both 13°C and 20°C. All green tomatoes sealed in PE film were very firm even after 60 days of storage at 13°C and 20°C. Non-wrapped tomatoes remained

**Table 14.8.** Effects of packaging of the percentage weight loss of two species of taro (modified from Passam, 1982).

Storage time	Colocasia				Xanthosoma			
	Control	Dry coir	Moist coir	PE film	Control	Dry coir	Moist coir	PE film
2 weeks	11	7	0	1	8	7	0	1
4 weeks	27	21	3	5	15	19	4	3
6 weeks	35	30	7	9	28	28	21	6



**Table 14.9.** Effects of packaging material on the quality of yam (*Dioscorea trifida*) after 64 days storage at 20–29°C and 46–62% RH. Fungal score from 0 (= no surface fungal growth) to 5 (= tuber surface entirely covered with fungi). Necrotic tissue estimated by cutting the tuber into two, lengthways, measuring the area of necrosis and expressing it as % of the total cut surface.

Package type	% weight loss	Fungal score	% necrotic tissue
Paper bags	26.3	0.2	5
Sealed 0.03 mm thick PE bags with 0.15 % of the area as holes	15.7	0.2	7
Sealed 0.03 mm thick PE bags	5.4	0.4	4

acceptably firm for about 50 days at 13°C and 20 days at 20°C. Tano *et al.* (2007) stored mature-green tomatoes at 13°C in containers where an atmosphere of some 5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub> was maintained and they were subjected to a sequence of temperature fluctuations of 10°C for 35 days. Temperature fluctuations had a major impact on the composition of the package atmospheres and on product quality. CO<sub>2</sub> concentrations increased rapidly, reaching a maximum of 11 kPa, while O<sub>2</sub> concentrations decreased to less than 1.5 kPa. The temperature-fluctuating regime resulted in extensive browning, softening, weight loss increase, ethanol increase and infection due to physiological damage and excessive condensation, compared with storage at constant temperature. Fagundes *et al.* (2015) showed that MA packaging in BOPP/LDPE bags decreased the ethylene production and respiration rates of cherry tomatoes and reduced weight loss and lycopene biosynthesis at 5°C. In a comparison of air and active MA packaging

(5 kPa O<sub>2</sub> + 5 kPa CO<sub>2</sub>) they found that the latter was preferable and extended their shelf-life to 25 days.

### Wampee

Zhang *et al.* (2006) found that Wampee fruit (*Clausena lansium*) deteriorated quickly after harvest and that 2–4°C was optimum, which delayed the total sugar, acidity and AA losses. MA packaging in 0.025 mm PE film with a packing capacity of 5.4 kg/m had a positive additional extension to their storage life at 2–4°C.

### Yam

Storing yams (*Dioscorea* spp.) in PE bags was shown to reduce weight loss, but had little effect on surface fungal infections and internal browning of tissue (Table 14.9).

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